

Pd-Catalyzed Regioselective Decarboxylative / C-H #-Alkoxyalkenylation of Heterocycles using #-Carboxyvinylethers

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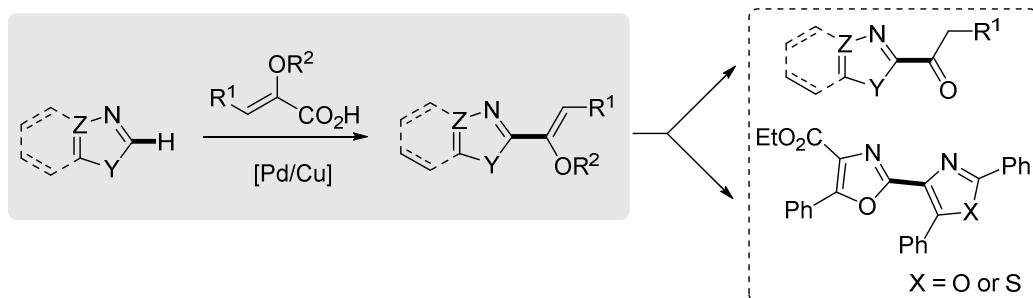
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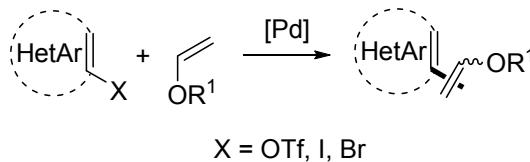
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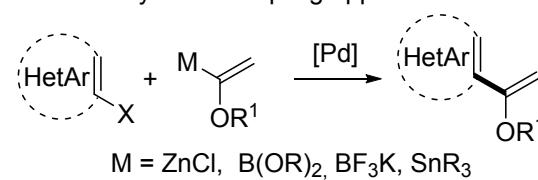
Abstract: A direct introduction of vinyl ethers into C–H bond of heterocycles is reported. For this purpose, decarboxylative direct C–H cross-coupling of 1,3-diazoles with α-carboxyvinylethers as coupling partners was achieved under Pd(0)/Cu(I) cooperative catalysis to produce various α-heteroarylated vinylethers. This methodology was applied to the innovative production of heteroarylated enolizable ketones and naturally-occurring bis-oxa(thia)zoles.

Vinyl ethers are found in many natural products and biologically active molecules. They represent one of the most reactive and synthetically-valuable class of hetero-substituted alkenes for the synthesis of complex organic molecules. Employed as masked-ketones and activated alkenes, vinyl ethers might be involved in number of chemical transformations such as hydrolysis, reduction, cycloaddition reactions, Heck and related cross-coupling reactions to produce poly-functionalized ketones, alcohols, heterocycles and alkenes.^[1-2] In particular, the transition-metal catalyzed heteroarylation of enol ethers has drawn lot of attention. The Heck reaction of non- prefunctionalized vinyl ethers with halides is currently the most step-economical developed strategy.^[3,4] However, this reaction is fraught with difficulties associated with selectivity issues (α/β and Z/E). Therefore, others traditional cross-coupling approaches have been considered using pre-metallated vinyl ethers in order to control the regio- and the stereochemistry (Figure 1).^[2,5] In this context, straightforward, atom-economic, and environmentally benign method for direct heteroarylation of vinyl ethers with a full control of the E/Z stereochemistry and α/β regiochemistry are highly demanded.

■ (Hetero)Arylation of acyclic enol ethers by Heck reaction



■ α -(Hetero)Arylation of acyclic enol ethers by cross coupling approaches



■ **This work:** decarboxylative C-H coupling of heterocycles with α -carboxyvinyl ethers

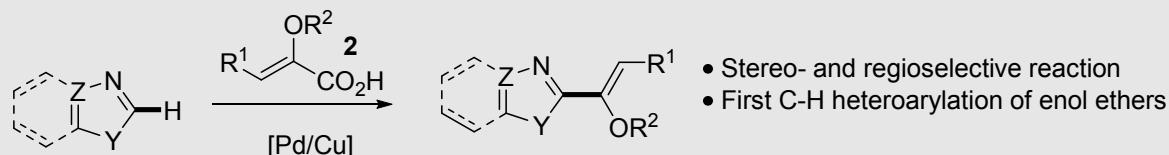
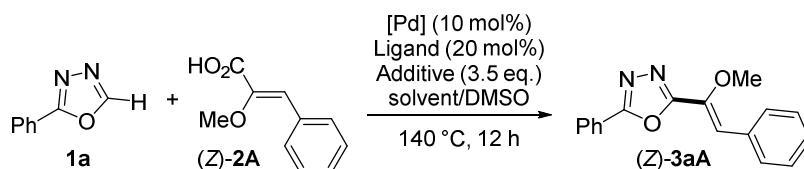


Figure 1. Pd-catalyzed heteroarylation of enol ethers

Faced with the necessity to do ‘*better with less*’, the direct C–H functionalization of heterocycles has emerged as powerful and complementary approach avoiding the generation of metalated heteroarenes, sometimes air-moisture sensitive and unstable.^[6] Since the past decades, explosive achievements have been made in direct C–C bonds formation including arylation, alkenylation, alkylation, carbonylation and acylation. As a complementary approach to dramatically increase the functional diversity, the direct C–H introduction of valuable functions, which offers several options for chemical-modulation and/or reactivity, is of great synthetic interest. While remarkable advances have been made in direct C–H alkenylation of heterocycles,^[7] the transition-metal catalyzed direct C–H α -alkoxyalkenylation of heterocycles with vinyl ethers has not been reported so far.^[8] In line with our ongoing interest in the direct introduction of hetero-substituted alkenes into C–H bond of heterocycles,^[9] and knowing that carboxylic acids are regarded as potential air-stable and easy to handle masked catalytic organometallic building blocks by transition-metal-mediated extrusion of CO₂,^[10-13] we turned our attention to the α -alkoxylated acrylic acids (**2**) as coupling partners for palladium-catalyzed hetero-substituted alkenylation of heterocycles (Figure 1). Herein, Pd(0)-catalyzed and Cu(I)-mediated decarboxylative / C–H α -alkoxyalkenylation of (hetero)arenes using challenging α -alkoxylated acrylic acids (**2**) is reported. This methodology gives access to stereocontrolled *gem*-heteroarylated vinyl ethers, and offers a rational and step-economical route to attractive polysubstituted heteroarylated enol ethers. As an application, the decarboxylative / C–H α -alkoxyalkenylation reaction was applied to afford a novel and convenient entry point into α,β -enolizable α -ketoazole structures^[14] and biologically active C2-C4’ linked azoles.^[15]

Table 1. Decarboxylative coupling: Reaction optimization

Entry	[Pd]	Ligand	[Additive]	Ratio 2A/1a	Solvent/DMSO (8/3)/[Conc.]	Yield [%] ^{a,b}
1	Pd(OAc) ₂	dcpe	CuCO ₃ •Cu(OH) ₂	1.5:1	1,4-Diox./0.2M	n.r.
2	Pd(OAc) ₂	dcpe	CuCO ₃ •Cu(OH) ₂	1.5:1	DMAc/0.2M	23
3	PdBr ₂	dcpe	CuCO ₃ •Cu(OH) ₂	1.5:1	DMAc/0.2M	34
4	Pd(acac) ₂	dcpe	CuCO ₃ •Cu(OH) ₂	1.5:1	DMAc/ 0.2M	41
5	Pd(acac) ₂	dppe	CuCO ₃ •Cu(OH) ₂	1.5:1	DMAc/ 0.2M	62
6	Pd(acac) ₂	dppe	Cu ₂ O	1.5:1	DMAc/ 0.2M	55
7	Pd(acac) ₂	dppe	Cu(OAc) ₂	1.5:1	DMAc/ 0.2M	29
8	Pd(acac) ₂	dppe	Ag ₂ CO ₃	1.5:1	DMAc/ 0.2M	29
9	Pd(acac) ₂	dcpe	CuCO ₃ •Cu(OH) ₂	2/1	DMAc/0.2M	78
10	Pd(acac) ₂	dppe	CuCO ₃ •Cu(OH) ₂	2/1	DMAc/0.1M	96
11	--	--	CuCO ₃ •Cu(OH) ₂	2/1	DMAc/0.1M	n.r.

^aReaction conditions: [Pd] (10 mol%), ligand (20 mol%), **1a** (0.1 mmol), **(Z)-2A** (0.2 mmol), CuCO₃•Cu(OH)₂ (3.5 equiv), solvent/DMSO (8:3), MS 4 Å, 140 °C, 12 h. ^bYield based on isolated product after flash chromatography. 1,4-Diox = 1,4-Dioxane. n.r. = no reaction

We first established reaction conditions to operate the decarboxylative C–H coupling of 1,3,4-oxadiazole **1a** with β -arylated α -methoxyacrylic acid **2A** prepared as pure *(Z)*-isomer through an aldol reaction followed by a dehydroxylation and hydrolysis of the resulting ester^[16] (Table 1). We observed that the heteroarylated enol ether **3Aa** was not produced using strictly the Greaney's conditions.^[11a] However, by switching the 1,4-dioxane/DMSO mixture of solvents for DMAc/DMSO, the reaction occurred leading to the expected enol ether **3Aa** in 23% yield.

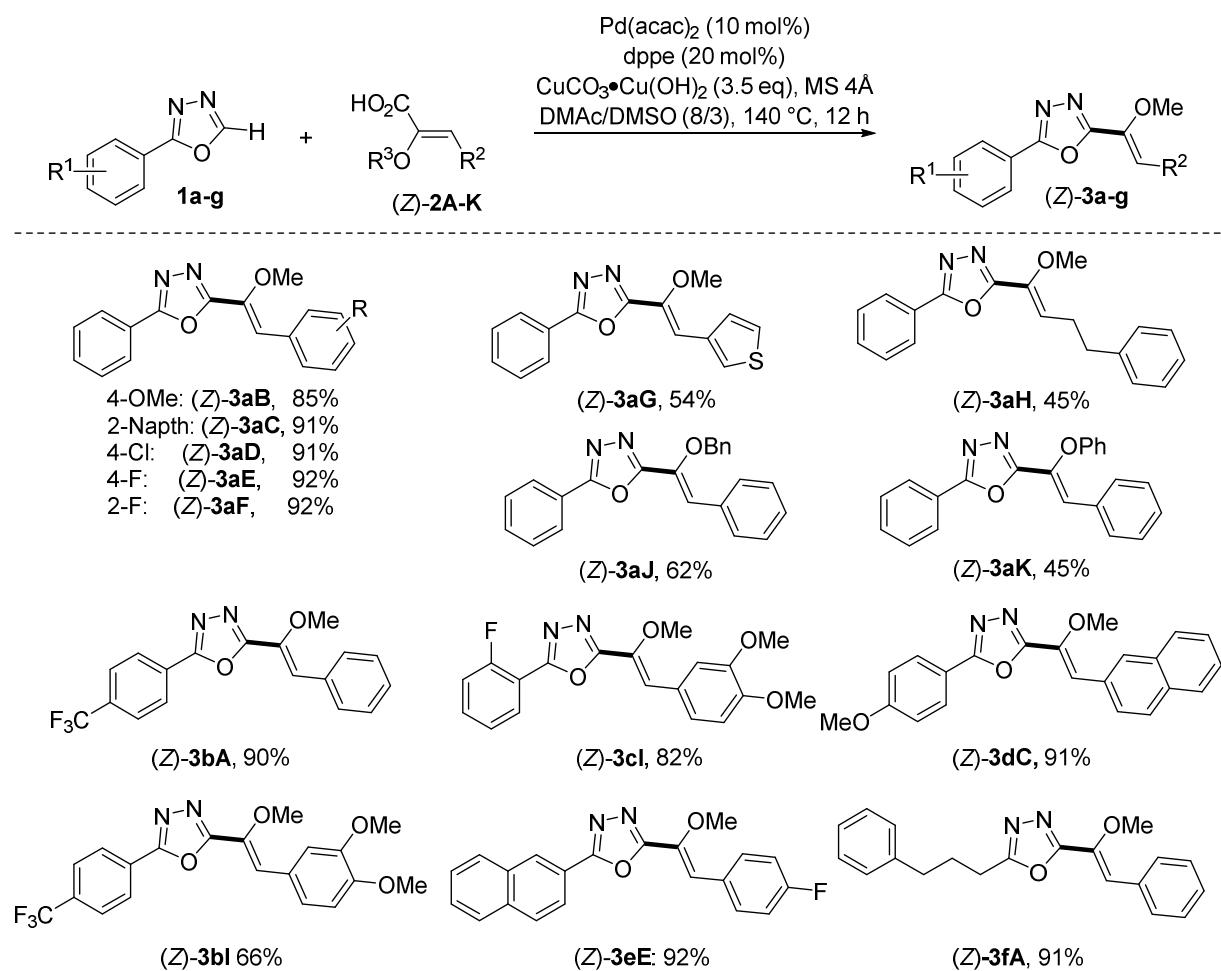
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6 Interestingly, the enol ether **3aA** was only isolated as pure (*Z*)-isomer (Table 1, entries 1 and 2)
7 and the stereochemistry was confirmed by single crystal X-ray structure determination.^[17a,b]

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9 After an additional screening of palladium sources and phosphine ligands,^[17c] enol ether **3aA**
10 was isolated in a fair 62% yield using Pd(acac)₂ catalyst and dppe as ligand. However, the use of
11 other copper sources or silver carbonate as oxidants was detrimental (Table 1, entries 6-8)
12 leading predominantly to protodecarboxylative side product **4** (Figure 2). Full conversion of
13 1,3,4-oxadiazole **1a** as well as quantitative production of (*Z*)-**3aA** were finally reached by
14 reacting 2 equivalents of α -methoxyacrylic acid (*Z*)-**2A** along with reducing of the (*Z*)-**3aA**
15 concentration from 0.2 to 0.1 M (Table 1, entries 9-10). When lowering the amount of [Pd]-
16 charge (from 10 to 5 mol%) and removing the molecular sieve, the efficiency of the Pd(0)-
17 catalyzed and Cu(I)-mediated decarboxylative process is significantly affected.^[17c] Moreover the
18 decarboxylative coupling failed without palladium and ligand^[18] (Table 1, Entry 11).

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34 With the optimized conditions in hands, we turned our attention to the scope and
35 limitations of the Pd-catalyzed and Cu-mediated decarboxylative C–H alkenylation
36 methodology. We first performed the C-2 alkenylation of 1,3,4-oxadiazoles **1a-g** with a broad
37 panel of substituted β -arylated α -methoxyacrylic acids **2A-I**. Overall, the cross-coupling was
38 successfully achieved whatever the electronic effect and the position of substituents on both
39 aromatic units of coupling partners **1** and **2** affording α -methoxyoxadiazolylalkenes (*Z*)-**3** in fair
40 to excellent yields (Scheme 1). The β -heteroarylated α -methoxyacrylic acid **2G** led also to the
41 desired product (*Z*)-**3aG** in 54% isolated yield. We were pleased to find that alkyl groups on both
42 coupling partners **1** and **2** are also tolerated under our optimized experimental conditions. The
43 corresponding 1,2-trisubstituted methoxylated alkenes (*Z*)-**3aH** and (*Z*)-**3fA** were obtained in
44 moderate to good yields. Interestingly, the methodology remains efficient with α -benzyloxy- and
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α -phenoxyacrylic acids (*Z*)-**2J-K** leading to corresponding enol ethers (*Z*)-**3aJ** and (*Z*)-**3aK** in moderate 45–62% isolated yields.

Scheme 1. Scope of decarboxylative / direct C–H α -methoxyalkenylation of 1,3,4-oxadiazoles with various α -alkoxyacrylic acids **2A-K**

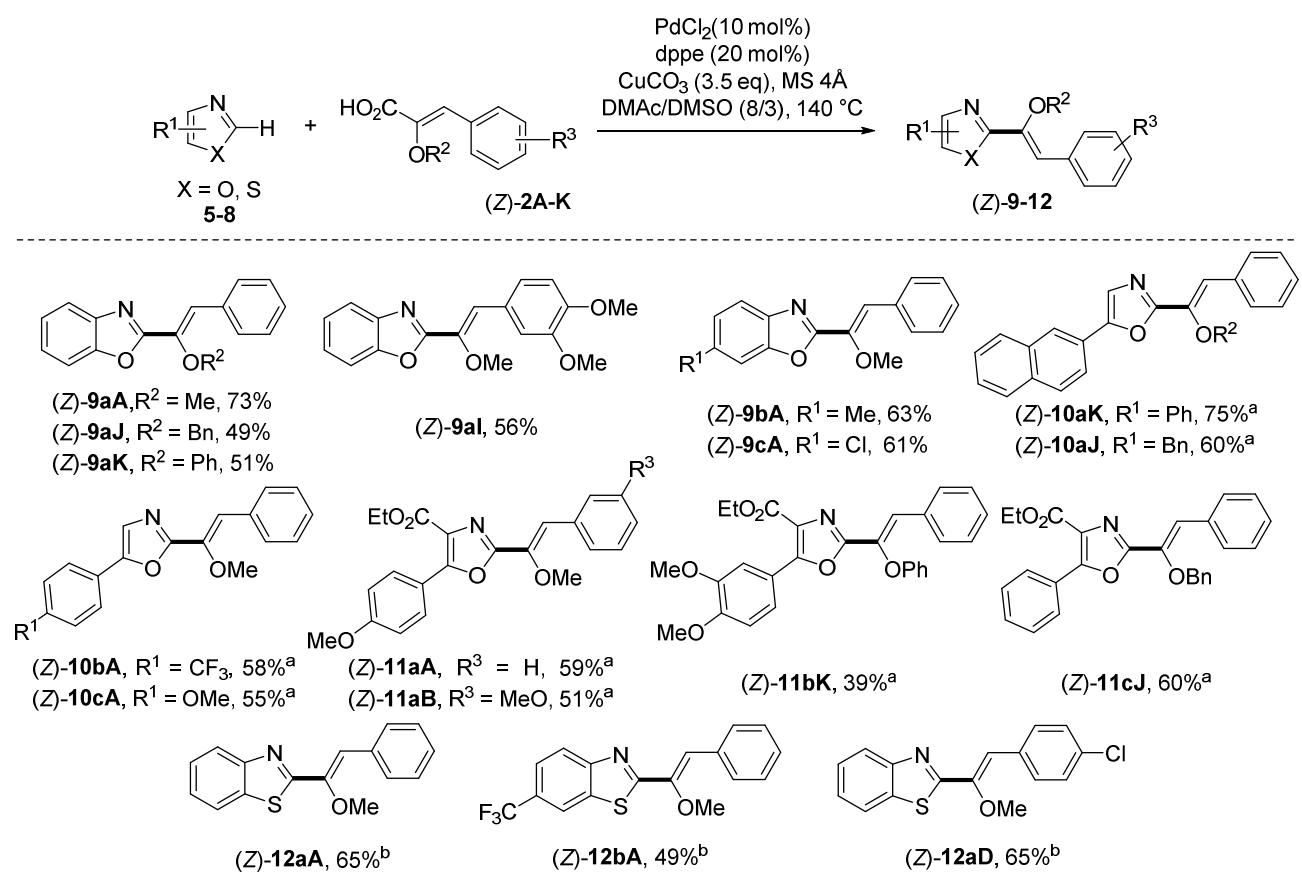


We next sought to investigate the scope of heterocycles (Scheme 2). Interestingly, electronically different heterocycles displayed a good reactivity with α -methoxyacrylic acids (*Z*)-**2A-I**. The α -phenoxy and α -benzyloxy (*Z*)-**2J-K** also proved to be reactive by switching the

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6 palladium source $\text{Pd}(\text{acac})_2$ to PdCl_2 to provide a broad set of heteroarylated enol ethers (*Z*)-**9**-
 7 **11** in fair to excellent yields (Scheme 2).

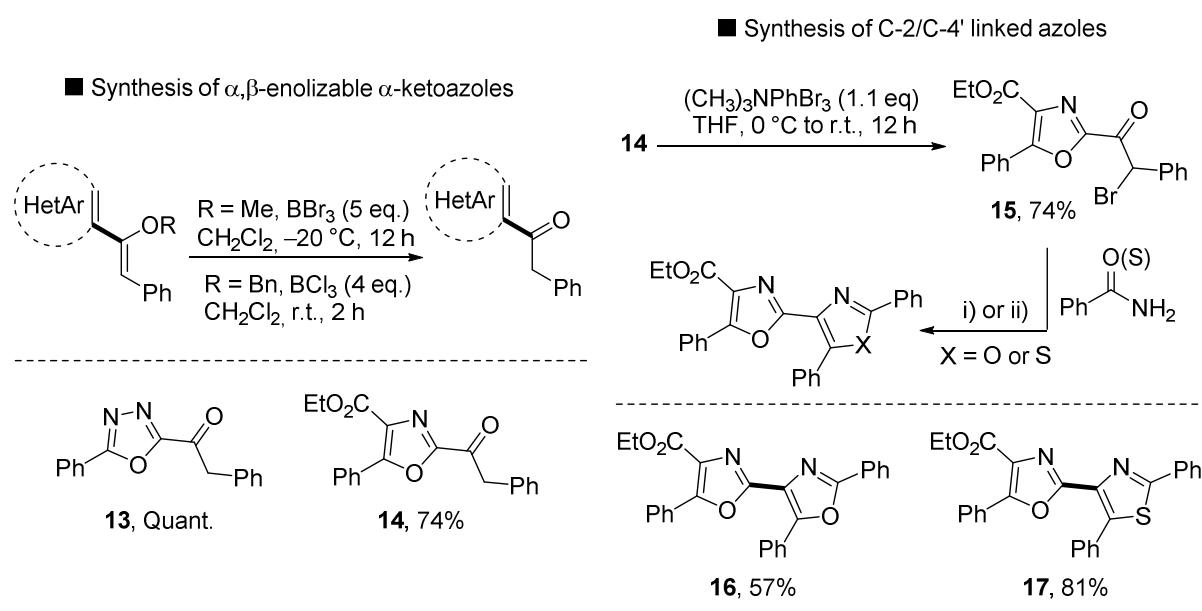
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Scheme 2. Scope of decarboxylative / direct C–H alkenylation of azoles using various α –
 11 alkoxycrylic acids **2A–L** as coupling partners. [a] dcpe instead of dppe. [b] Addition of CuI (10
 12 mol%).



50 The decarboxylative / C–H α -alkoxylalkenylation of variously substituted benzoxazoles **5a–c** as
 51 well as oxazole derivatives, such as 5-arylated oxazoles **6a–c** and 5-arylated ethyl oxazole-4-
 52 carboxylates **7a–d**, were successfully achieved using either dppe or dcpe ligands to produce the
 53 heteroarylated enol ethers **9–11** in 50–73% yields while limiting the main homocoupling side
 54 reaction. α -Phenoxy and benzyloxyalkoxyacrylic acids **2J–K** were also reactive with
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(benzo)oxazole derivatives **5-7** leading to the corresponding heteroarylated enol ethers **9-11J-K** in moderate 35% to good 75% isolated yields. To complete the scope of heterocycles, we found that the cross-coupling of α -methoxyacrylic acid (*Z*)-**2A** with less acidic benzothiazole **8a** under the optimized protocol provided the heteroarylated **12aA** in only 25% isolated yield. However, the reaction could be improved by using CuI as additive. Indeed, this latter is supposed to facilitate the catalytic metalation key-step of less-acidic and more-electron-rich 1,3-diazoles, such as thiazole and imidazole, through the formation of [Cu]-heterocycle pre-complex in order to increase the C2-H acidity.^[19, 20] In that case, the benzothiazolyl enol ether **12aA**, as well as two additional benzothiazolyl enol ethers **12bA** and **12aD**, have been produced in 49% and 65% isolated yields (Scheme 2).

Scheme 3. Synthesis of biologically active structures. Reaction conditions: i) X = O, AgSbF₆ (1 eq), CH₂Cl₂, -20 °C, 12 h; ii) X = S, CaCO₃ (1 eq), EtOH, reflux, 6 h.



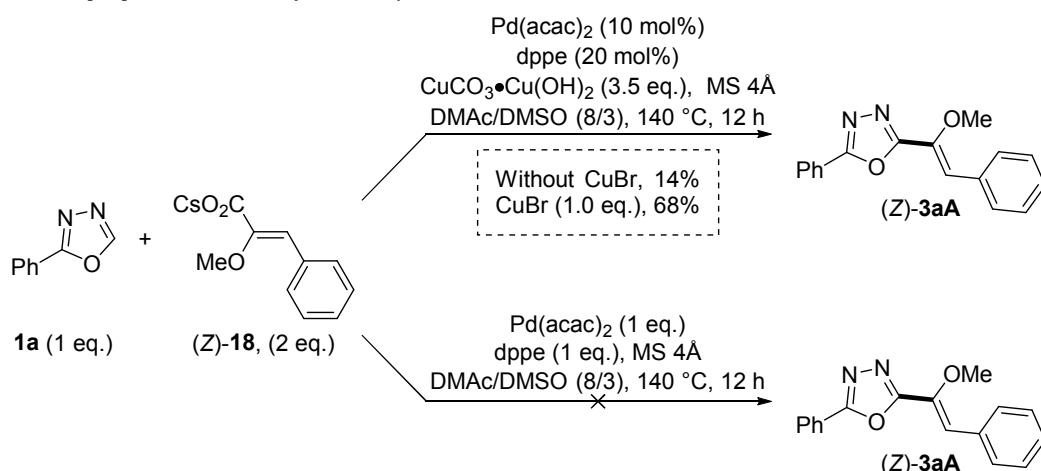
To illustrate the potential usefulness of this novel regio- and stereo-controlled access to heteroarylated enol ethers, two synthetic applications were investigated (Scheme 3). To date and to the best of our knowledge, while the palladium-catalyzed direct C-H arylation of 1,3-diazoles have been already reported,^[21-23] the direct production of valuable 2-(2-oxoalkyl)-1,3-diazoles^[24] through direct C2-H functionalization of 1,3-diazoles remain highly challenging. In this context, the first application of this methodology was directed towards the challenging synthesis of α,β -enolizable α -keto-1,3-diazoles which are immediately available by adding a deprotection step of the enol ether function. Selected as inhibitors of fatty acid amide hydrolase,^[14] the heteroarylated ketones **13** and **14** were produced by treatment of methylated enol ethers (*Z*)-**3aA** and benzylated (*Z*)-**11cJ** with BBr_3 and BCl_3 respectively, according to the nature of the enol ether.^[25, 26] As an additional application, the α,β -enolizable α -ketoazole **14** was used as a building block for the synthesis of naturally-occurring C2-C4' linked bis-oxa(thia)azole.^[15] In particular, the α,β -enolizable α -ketoazole **14** was used both to prepare the C2-C4' linked bis-oxazole **16** applying the Blümlein-Lewis reaction^[27] and the C2-C4' linked oxazole-thiazole **17** via the Hantzsch's condensation.^[28]

To gain insight into the mechanism, we carried out several control experiments (Scheme 4). First, the decarboxylative cross-coupling was executed with cesium (*Z*)- α -methoxycinnamate salt **18** (Scheme 4, eq 1). Interestingly, the addition of CuBr used as electrophile to generate the methoxycinnamate copper salt was required to generate the desired product (*Z*)-**3aA** in good yield. Moreover, under stoichiometric amount of $\text{Pd}(\text{acac})_2$ and dppe as ligand, no cross-coupling product was obtained. Importantly, the Heck coupling reaction of cesium (*Z*)- α -methoxycinnamate salt **18** with phenyl bromide followed by CuBr -mediated protodecarboxylation^[29] led to the exclusive production of the β -diphenylated vinyl ether **19**.

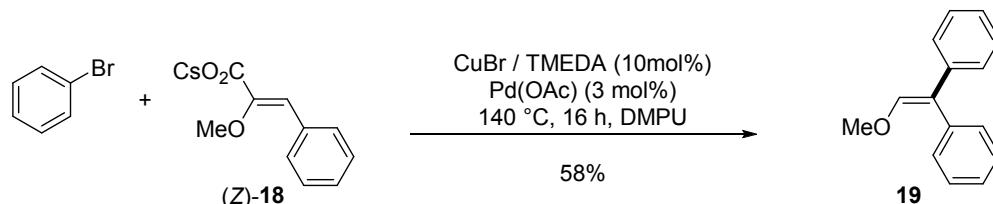
(Scheme 4, eq 2).^[17c] Therefore, the oxidative Heck-type / protodecarboxylative sequence for the synthesis of (Z)-3aA can be discarded. All these data highlighted the importance of CuCO₃•Cu(OH)₂ both in the oxidation step of Pd(0) to Pd(II) and in the crucial generation of copper carboxylate intermediate that may lead to the generation of alkenyl copper intermediate (**II**) through the *ipso*-decarboxylative cupratation process. (Figure 2)

Scheme 4. Additional experiments for mechanism study

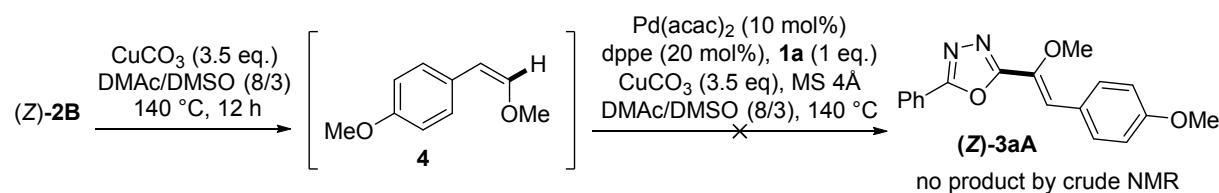
(1) Role of [Cu] in the decarboxylative step



(2) Heck reaction followed by protodecarboxylative process



(3) In situ protodecarboxylation then Fujiwara-Moritani reaction:



This latter may be facilitated by the electron-withdrawing inductive effect of the methoxy group^[30] since we noted that the decarboxylative cross-coupling of 1,3,4-oxadiazole **1a** with cinnamic acid failed under the optimized conditions.^[13, 17c] We then investigated the cross-coupling of the protodecarboxylative enol ether **4** with **1a** under optimal experimental conditions (Scheme 4, eq 3).^[31] The reaction failed precluding a two-steps mechanism involving initially a protodecarboxylation followed by a Fujiwara-Moritani oxidative Heck-type coupling. Finally, the effect of radical scavenger was examined, and 2,2,6,6-tetramethylpiperidine 1-oxyl (TEMPO) did not inhibit the reaction.^[17b] Thus a radical process is unlikely involved.

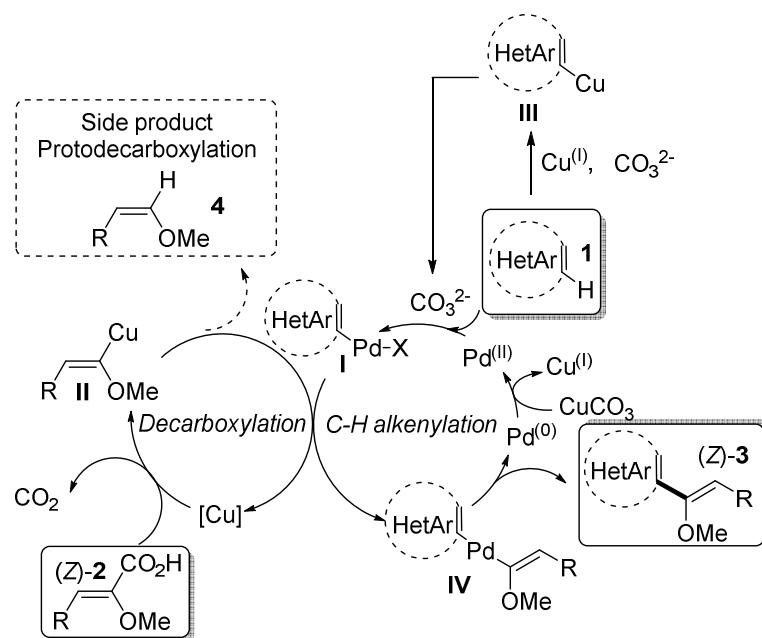


Figure 2. Proposed catalytic pathways.

Based on previous reports^[10-12] and our control experiments, a plausible reaction pathway is delineated in Figure 2. The success of this coupling depends on the interception of the generated alkenyl copper intermediate (**II**) produced by copper-catalyzed extrusion of CO_2 , and the C-2-palladated heterocycle (**I**) through a transmetalation step. The generation of the C-2-palladated key-intermediate (**I**) is highly dependent both on the intrinsic acidity and nucleophilicity of C2-H

site of 1,3-(oxa)diazoles and may be considered according to two pathways, (i) the catalytic palladation of heterocycles with $\text{Pd}(\text{OAc})_2$ through base-assisted concerted^[34] or none-concerted (carbanionic-type) process,^[35] or (ii) the transmetalation of the *in situ* generated heteroaryl copper intermediate (**III**) to $\text{Pd}(\text{OAc})_2$.^[19] Base on our experimental observations, we presume that most acidic oxazole derivatives are appropriate for direct palladation with or without [Cu]-assistance.^[19e] By contrast, the use of CuI as additive for the success of the coupling with benzothiazoles indicates a copper-assisted pre-activation of the heterocycle leading probably to the generation of benzothiazolylcopper intermediate **III**.^[19-20]

In conclusion, we have reported a practical, straightforward, and unprecedented regio- and stereoselective direct introduction of enol ethers onto C–H bonds of heterocycles by decarboxylative/ direct C–H alkenylation of various azoles under palladium catalysis using α -carboxylenol ether as new hetero-substituted alkenylating agent. This procedure offer a rational and step-economical route to attractive regio- and stereocontrolled 1,2-polyheteroarylated enol ethers in moderate to good yield. Moreover, this methodology constitutes a novel and convenient entry point into the synthesis of α,β -enolizable α -ketoazoles and C2-C4' linked bis-azoles.

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Supporting information: Experimental procedures, characterization data, details on the X-ray analyses of compound (Z)-**3aA**, and ^1H and ^{13}C NMR spectra for compounds.

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