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Formal Total Synthesis of (±)-Occidol

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Formal Total Synthesis of (\pm) -Occidol

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Abstract: The conversion of tetralone 1 to methylketone 7, a valuable intermediate of occidol 8, has been accomplished in four steps (reduction, mesylation, cyanation, and Grignard reaction with methylmagnesium bromide).

Keywords: Cyanation, Grignard reaction, occidol, tetralones

Occidol **8**, a principal constituent of essential oil of Thuja species, ^[1] belongs to the eudesmane class of sesquiterpenes. Several syntheses of occidol **8** have been accomplished by a variety of methods. ^[2] In some of these approaches, substituted α -tetralones have been smoothly converted to occidol **8**. To date, no attempt has been made to synthesize occidol from substituted β -tetralones. The recently reported ^[3] 5,8-dimethyl β -tetralone **1** prompted us to study its conversion to occidol **8**.

Tetralone 1 was reduced with sodium borohydride to an oily alcohol 2, which was converted to its mesylate derivative 3 by adopting standard procedures. The conversion to the nitrile derivative was accomplished by heating $^{[4]}$ with sodium cyanide and hexamethyl-phosphoramide. The nitrile 4 was formed along with products 5 and 6.

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The separation of **4** from the reaction mixture by silica-gel chromatography was successful; however, compounds **5** and **6** could not be separated because of their close Rf values. The nitrile **4**, on heating with methylmagnesium bromide in ether (3M) at 50°C for 72 h, produced methylketone **7** in 80%. The IR and ¹H NMR spectral data compared to literature values^[2] confirmed the assigned structure. The one-step conversion of **7** to occidol **8** has already been reported,^[2] so this synthesis constitutes the formal total synthesis of **8**.

EXPERIMENTAL

Unless otherwise stated, IR spectra were taken on Nicolet FT instrument. ¹H and ¹³C NMR spectra were recorded on Brucker AM 300-MHz instrument in CDCl₃. Mass spectra were run on gas chromatograph Hewlett Packard 5890 Quadrupolar 5972 Series S. Melting points were recorded on an

electrothermal melting-point apparatus. The solution is diluted with water, extracted with ether, washed with brine, dried (MgSO₄), and evaporated under reduced pressure. Column chromatography was performed on silica gel (Merck grade 60, 70–230 mesh). The spectral and analytical data of all new compounds have been reported in the Experimental section. Microanalyses were carried out at the Chemistry Department, IVIC, Caracas.

5,8-Dimethyl-1,2,3,4-tetrahydro-2-naphthol 2

Sodium borohydride (980 mg, 26 mmol) in ethanol (70 mL) was added to a solution of **1** (2.26 g, 13 mmol) in ethanol (70 mL). The mixture was stirred for 2 h at room temperature, decomposed with acetic acid, diluted with water, and extracted with diethyl ether. The organic layer was washed, dried, and purified on silica gel (hexane–diethyl ether 4:6) to obtain **2** (2.17 g, 95%); m/z 176 (M⁺) and 158 (M⁺-H₂O); ν_{max} 3375 cm⁻¹ (OH); ¹H: δ 1.85–1.78 (m, 2H, CH₂), 2.21 (s, 3H, C5-Me), 2.26 (s, 3H, C8-Me), 3.1–2.5 (m, 4H, C1-2H, C4-2H), 4.13–4.10 (m, 1H, CHOH), 6.79 (bs, 1H, C-6) and 6.84 (bs, 1H, C-7); ¹³C: δ 136.7 (C-10), 135 (C-9), 130.5 (C-8), 129.6 (C-5), 128.3 (C-6), 126.8 (C-7), 67.8 (C-2), 35.6 (C-1), 31.4 (C-3), 27.5 (C-4), 19.4 (C-12) and 19.2 (C-11). (Found: C, 82.15; H, 9.37. C₁₂H₁₆O requires C, 81.77%; H, 9.15%.)

5,8-Dimethyl-1,2,3,4-tetrahydro-2-cyanonaphthalene 4

Mesylchloride (1.2 mL, 14.5 mmol) was added to a solution of **2** (2.17 g, 12.3 mmol) in pyridine (12 mL). The resulting brown solution was stirred at room temperature for 24 h, diluted with water, and extracted with ether. The organic extract was washed with a 5% solution of copper sulfate, washed with brine, dried, and evaporated to give **3** (2.41 gr, 77%); m/z 254 (M⁺), 175 (M⁺-MeSO₂) and 158 (M⁺-MeSO₂OH); 1 H: δ 2.15–2.22 (m, 2H, CH₂), 2.26 (s, 3H), 2.33 (s, 3H) (C-5, C-8), 2.81–3.03 (m, 4H) (C-1 and C-4), 3.09 (s, 3H, MeSO₃), 5.16–5.28 (m, 1H, CHOSO₂Me), 6.86 (bs, 1H), and 6.92 (bs, 1H) (C-6 and C-7); 13 C: δ 136.5 (C-10), 135.5 (C-9), 130.2 (C-8), 129.7 (C-5), 128.7 (C-6), 126.9 (C-7), 78.5 (C-2), 38.7 (C-13), 33.5 (C-3), 28.9 (C-1), 26.4 (C-4), 19.4 (C-12) and 19.3 (C-11).

The mesylate 3 (2.0 g, 7.9 mmol), powdered sodium cyanide (1.16 g, 23.7 mmol), and hexamethylphosphoramide (28 mL) was heated at 120°C for 72 h under argon. The reaction mixture was diluted with water, extracted with ether, dried and evaporated. The resulting material was purified on silica gel by eluting with hexane to afford a mixture of 5 and 6 (641 mg, 38%); m/z 158 (M^+) and 156 (M^+).

Further elution (hexane-diethyl ether 8:2) yielded **4** (521 mg, 36%), mp $107-110^{\circ}$ C; m/z 185 (M⁺), 170 (M⁺-Me), 158 (M⁺-HCN) and 143

(M⁺-HCN-Me); $\nu_{\rm max}$ 2234 cm⁻¹ (CN); $^{1}{\rm H}$: δ 1.99–2.07 (m, 2H, CH₂), 2.21 (s, 3H), 2.28 (s, 3H) (C-5 and C-8), 2.77–3.06 (m, 5H) (C-1, C-4 and CHCN), 6.80 (bs, 1H), 6.87 (bs, 1H); (aromatic protons) $^{13}{\rm C}$: δ 136.3 (C-10), 135.8 (C-9), 134.3 (C-8), 128.7 (C-6), 127.9 (C-5), 127.3 (C-7), 122.3 (C-13), 29.6 (C-1), 27.6 (C-3), 25.9 (C-2), 20.7 (C-12), 19.3 (C-11). (Found: C, 84.68; H, 8.38. $C_{13}{\rm H}_{15}{\rm N}$ requires C, 84.32%; H, 8.10%.)

5,8-Dimethyl-1,2,3,4-tetrahydro-2-naphthyl-ethan-l-one 7

A solution of methylmagnesium bromide in dry ether (1.8 mL, 3M) was added dropwise to a solution of **4** (502 mg, 2.7 mmol) in tetrahydrofuran (11 mL). The reaction mixture was heated at 50°C for 72 h, and then hydrochloric acid (10%) was cautiously added and extracted with ether. The organic extract was washed, dried, evaporated, and chromatographed on silica gel (hexane–diethyl ether 8 : 2) to obtain **7** (441 mg, 80%); m/z 202 (M⁺); 187 (M⁺-Me), 144 (M⁺-Me-COMe); ν_{max} 1708 cm⁻¹ (CO); ¹H: 1.81–1.71 (m, 2H, CH₂), 2.23 (s, 3H, MeCO) 2.28 (s, 6H; Me), 2.91–2.73 (m, 5H) (C-1, C-2, and C-4), 6.81 (bs, 1H, Ar-H), 6.86 (bs, 1H, Ar-H); ¹³C: δ 211.3 (C-13), 136.5 (C-10), 135.5 (C-9), 134.9 (C-8), 134.1 (C-5), 128.4 (C-6), 127.1 (C-7), 48.4 (C-2), 29.4 (C-3), 28.1 (C-14), 27.8 (C-1), 25.2 (C-4), 19.4 (C-12) and 19.2 (C-11). (Found: C, 83.38; H, 9.15. C₁₄H₁₈O requires C, 83.12%; H, 8.97%.)

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