PHOTOCYCLIZATION OF N-CHLOROACETYLEENZYLAMINES TO ISOQUINOLINE DERIVATIVES

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The Pomeranz-Fritsch method¹ is useful for the preparation of 7- and 5,7-substituted isoquinolines which are difficult to be synthesized by other isoquinoline syntheses. However, the reaction usually proceeds only in a strong acid, which sometimes causes side reactions.

Various novel heterocycles have been synthesized photochemically from N-chloroacetylphene-thylamines.² Among these photoreactions, the most typical one is the formation of 1,2,4,5-tetrahydro-3H-3-benzazepin-2-ones and it is expected to be extended for the synthesis of iso-quinoline derivatives.

When N-chloroacetylbenzylamines (I) in aqueous ethanol or acetonitrile were irradiated, the corresponding N-acetyl or N-hydroxyacetyl derivatives were isolated without any detectable formation of cyclization products. However, m-hydroxy derivatives (II) gave easily 3-oxo-1,2,-3,4-tetrahydroisoquinolines (III) in fair yields.

A 50% aqueous acetonitrile solution of II (ca. 10 mM) was irradiated with a 100 W high pressure mercury lamp under nitrogen for 1-3 hr. Tetrahydroisoquinolines III, whose structures were easily confirmed by mass and nmr spectra, were isolated by recrystallization and by chromatography, if necessary. The results are shown in Table.

Table. Photocyclization of II to III.

II ,	R ¹	R ²	R ³	III	R ¹	R ²	Yield (%)	шр°С
IIa	OH	Н	Н	IIIa IIIa'	OH H	H OH	55 7	236-238 227-228
Пþ	OH	Œ	н	IIIb	OMe ^a	OMe ^a	45	191-194
IIc	OH	H	Me	IIIc IIIc'	CH H	OH H	7 4 13	228-230 218-220
IId	ОH	Н	сн ₂ с ₆ н ₅	IIId'	н Эн	CH H	69 16	200-201 170-171
IIe	OH	ŒН	СH ₂ C ₆ H ₅	IIIe	OH	OH	64	220-223
IIf	OH	Н	CH ₂ Ar ^{1,b}	IIIf	OH	Н	49	174-175
I I g			CH ₂ Ar ¹ ,b	IIIg IIIg'	OH H	H OH	36 7	229-231 169-171
IIh	OH	OH	CH ₂ Ar ² ,c	IIIh	OH	OH	59	260-263

a; treated the reaction mixture with diazomethane.

The common initial step of these photocyclizations has been proposed to be the ejection of an electron from the singlet excited state of the aromatic nucleus. The clear difference in reactivity between I and II can be interpreted by the ease of the electron ejection from phenols and/or the reactivity of the resultant phenoxy radicals. Further synthetic application of this reaction is in progress.

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b; $Ar^1 = 4$ -methoxyphenyl. c; $Ar^2 = 3,4,5$ -trimethoxyphenyl.