

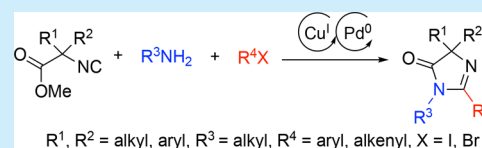
Cooperative Pd/Cu Catalysis: Multicomponent Synthesis of Tetrasubstituted Imidazolones from Methyl α -Isocyanoacetates, Primary Amines, and Aryl(vinyl) Iodides

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S Supporting Information

ABSTRACT: Three-component reaction of methyl α,α -disubstituted α -isocyanoacetates, primary amines, and aryl(vinyl) halides in the presence of Pd(OAc)₂ (0.05 equiv) and Cu₂O (1.0 equiv) provided 2,3,5,5-tetrasubstituted imidazolones via the formation of three chemical bonds. A copper-mediated migratory insertion of the isocyano group into the N–H bond of the amine followed by lactamization and Pd-catalyzed cross-coupling of the in situ generated amidinyl copper species with aryl(vinyl) halides accounted for the reaction outcome.



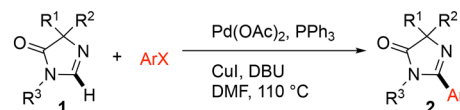
Imidazolone, a nonaromatic heterocycle, is found in bioactive natural products¹ and is a key structural element responsible for the luminescent properties of green fluorescent proteins (GFP).² It is a core structure of a number of pharmaceuticals displaying potent inhibitory activities against fatty acid synthases³ and angiotensin II receptor antagonists.⁴ For example, irbesartan, a marketed drug for the treatment of hypertension, contains this heterocycle.⁵ Different synthetic methods toward the substituted imidazolones⁶ have been developed including: (a) the transformation of the Erlenmeyer azlactones;⁷ (b) the cyclization of the amino acid derived formamidines,⁸ α -amidoamides,⁹ and α -isocyanoacetamides;¹⁰ and (c) the functionalization of the simple imidazolones.¹¹ Recently, Hoarau, Bischoff, and co-workers reported a Pd-catalyzed direct C–H functionalization of imidazolones **1** for the synthesis of the C-2 functionalized counterpart **2** (Scheme 1a).¹² Pirali et al. developed a synthesis of the C-2 arylated imidazolones by reaction of the α -isocyanoacetamides with benzynes (Scheme 1b).¹³

Metal-catalyzed domino processes initiated by isocyanide insertion reaction has recently been developed into a powerful tool for the rapid construction of heterocycles.^{14–17} In connection with our research program aimed at exploiting the diverse reactivities of the isocyano group,¹⁸ we reported a silver nitrate-catalyzed synthesis of the 3,5,5-trisubstituted imidazolones **1** from methyl α -isocyanoacetates **3** and primary amines **4**.¹⁹ As a continuation of this research topic, we describe herein a three-component synthesis of 2,3,5,5-tetrasubstituted imidazolones **2** by a Pd/Cu-catalyzed reaction of methyl α,α -disubstituted α -isocyanoacetates **3** with primary amines **4** and aryl(vinyl) halides **5** (Scheme 1c).

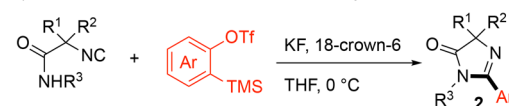
Methyl α,α -dibenzyl α -isocyanoacetates (**3a**, $R^1 = R^2 = \text{Bn}$), benzylamine (**4a**), and 4-iodotoluene (**5a**) were chosen as test substrates to evaluate the feasibility of our approach. Combining AgNO₃, the optimum metal salt for catalyzing the reaction between **3a** and **4a**, with Pd catalyst afforded only an intractable

Scheme 1. Synthesis of Tetrasubstituted Imidazolones

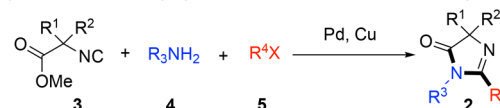
a) Hoarau, Bischoff and co-workers: C–H arylation of imidazolones



b) Pirali and co-workers: α -addition of α -isocyanoacetamides to benzynes

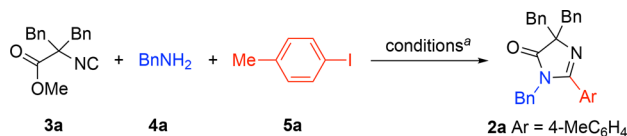


c) This work: one-step synthesis of imidazolones from α -isocyanoacetates



reaction mixture (entry 1, Table 1). Screening other cocatalysts (entries 2–11) indicated that Cu₂O formed the most productive couple with Pd leading to the desired three-component adduct **2a** in 72% isolated yield (entry 7). The impact of base on the reaction outcome was also examined. Among the two best copper copromoters, addition of a base increased the product yield in the case of CuI (entries 4 vs 5), but exerted a deleterious effect on the reaction mediated by Cu₂O (entries 7 vs 12–14). The reaction temperature of 130 °C seemed optimum as the reaction carried out at other temperatures led to the product with diminished yields (entries 7 vs 15, 16). Performing the reaction in DMSO (entry 17) or with reduced loading of Cu₂O (entries 18, 19) resulted in low yields of **2a**. However, reaction proceeded smoothly in the absence of phosphine ligand to provide **2a**, albeit with reduced yield (entries 21).²⁰ Finally, performing the reaction with Pd or Cu alone afforded **2a** in much reduced

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Table 1. Synthesis of 2,3,5,5-Tetrasubstituted Imidazolones **2a**: Survey of Reaction Conditions^a

entry	Pd/ligand	Lewis acid	base	yield (%) ^b
1	Pd(OAc) ₂ /PPh ₃	AgNO ₃	DBU	degradation
2	Pd(OAc) ₂ /PPh ₃	AgOTf	DBU	33
3	Pd(OAc) ₂ /PPh ₃	Yb(OTf) ₃	DBU	23
4	Pd(OAc) ₂ /PPh ₃	CuI	DBU	68
5	Pd(OAc) ₂ /PPh ₃	CuI		57
6	Pd(OAc) ₂ /PPh ₃	CuCl		55
7	Pd(OAc) ₂ /PPh ₃	Cu ₂ O		75(72) ^h
8	Pd(OAc) ₂ /PPh ₃	CuOTf		58
9	Pd(OAc) ₂ /PPh ₃	CuOAc		21
10	Pd(OAc) ₂ /PPh ₃	[(MeCN) ₄ Cu]BF ₄		50
11	Pd(OAc) ₂ /PPh ₃	Cu(OTf) ₂		6
12	Pd(OAc) ₂ /PPh ₃	Cu ₂ O	DBU	28
13	Pd(OAc) ₂ /PPh ₃	Cu ₂ O	Et ₃ N	57
14	Pd(OAc) ₂ /PPh ₃	Cu ₂ O	Cs ₂ CO ₃	27
15 ^c	Pd(OAc) ₂ /PPh ₃	Cu ₂ O		66
16 ^d	Pd(OAc) ₂ /PPh ₃	Cu ₂ O		66
17 ^e	Pd(OAc) ₂ /PPh ₃	Cu ₂ O		41
18 ^f	Pd(OAc) ₂ /PPh ₃	Cu ₂ O		34
19 ^g	Pd(OAc) ₂ /PPh ₃	Cu ₂ O		54
20	Pd(dba) ₂	Cu ₂ O		66
21	Pd(OAc) ₂	Cu ₂ O		62
22		Cu ₂ O		43 ⁱ
23	Pd(OAc) ₂ /PPh ₃			21

^aStandard conditions: **3a** (0.1 mmol), **4a** (0.15 mmol), **5a** (0.2 mmol), Lewis acid (1.0 equiv), Pd catalyst (5 mol %), ligand (10 mol %), base (1.0 equiv), DMF (1.0 mL), 130 °C. ^bNMR yield using CH₂Br₂ as an internal standard. ^c110 °C. ^d150 °C. ^eReaction was performed in DMSO. ^f10 mol % of Cu₂O was used. ^g25 mol % of Cu₂O was used. ^hYield of isolated product in parentheses. ⁱYield referred to C2-unsubstituted imidazolone resulting from the Cu-promoted amination/lactamization sequence (cf. ref 19).

yield indicating a strong synergistic effect of these two metals in catalyzing/mediating this transformation (entries 22, 23). Overall, the optimum conditions found consisted of performing the reaction in DMF at 130 °C in the presence of Pd(OAc)₂ (5 mol %), PPh₃ (10 mol %), and Cu₂O (1.0 equiv). Under these conditions, imidazolone **2a** was isolated in 72% yield. The reaction was chemoselective since direct *N*-arylation of primary amine **4a** by aryl iodide was not observed under these conditions.

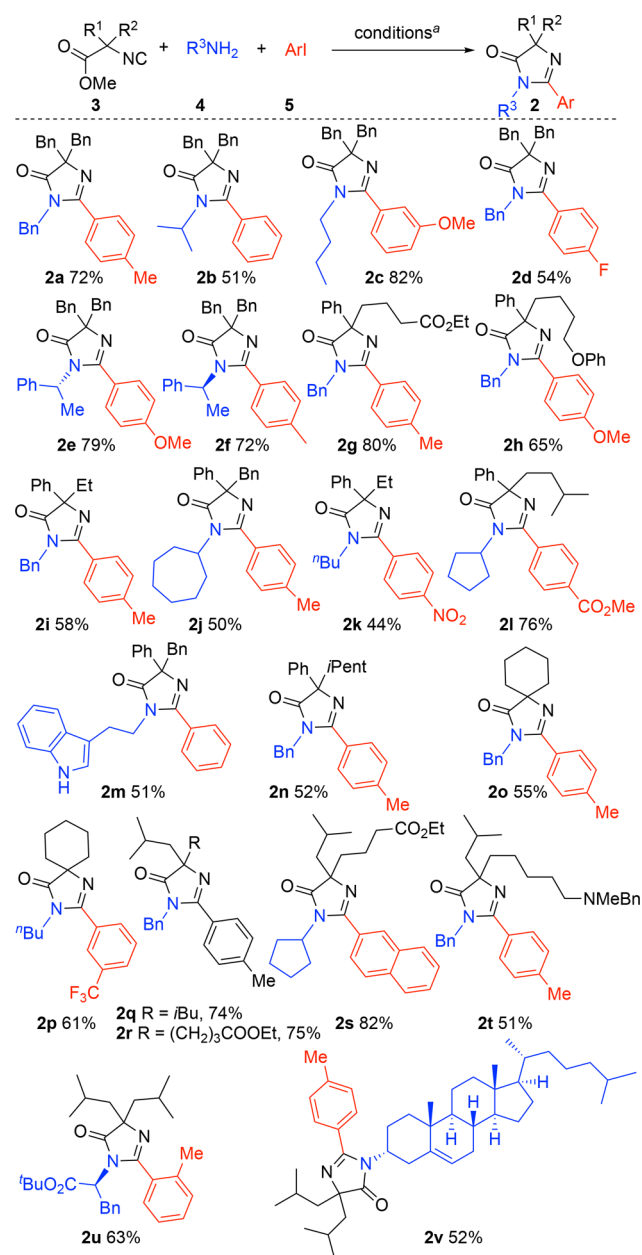
This novel three-component reaction displayed a broad application scope (Scheme 2). Aryl iodides bearing electron-donating (Me, OMe) and electron-withdrawing groups (F, CF₃, NO₂) at different positions were appropriate substrates for this reaction. Various alkyl amines, linear, branched, and functionalized, such as tryptamine (**2m**), participated well in this reaction. However, aniline failed to participate in this reaction. Reactions involving chiral amines (**2e**, **2f**) and α -amino ester (**2u**) afforded the three-component adducts in good yields without racemization. The α,α -disubstituted α -isocyanoacetates bearing different substituents reacted with amines and aryl iodide to afford the adducts, including spiroimidazolones (**2o**, **2p**), without event. The protocol can also be applied to complex natural products. Thus, reaction of 3 α -amino-5-cholestene^{21,22} with 4-iodotoluene (**5a**) and methyl α,α -diisobutyl isocyanoacetate afforded **2v** in 52% yield. The α -nonsubstituted and α -monosubstituted α -isocyanoacetates²³ and methyl 2-isocyanobenzoate failed to undergo the three-component coupling reaction. Finally,

performing the reaction of **3a**, **4a**, and **5a** at 1.0 mmol scale afforded the three-component adduct **2a** in 90% isolated yield.

Vinyl bromides, including the 1,2-disubstituted, 1,1-disubstituted, and tetrasubstituted ones, reacted smoothly with **3** and **4** to provide the 2-vinyl substituted imidazolones (**2w–2y**) in good yields (Scheme 3a). Based on this novel three-component reaction, more complex domino process can be designed for the one-pot synthesis of the polyheterocyclic compounds. For example, reaction of *ortho*-bromobenzylamine (**6**) with methyl isocyanacetates **3** furnished the tricyclic compounds **7a** and **7b**, respectively (Scheme 3b).

To gain insight on the reaction mechanism, the following control experiments were carried out (Scheme 4a). The reaction of the presynthesized 3,5,5-trisubstituted imidazolone **1a** with iodide **5a** under our optimized conditions led only to the recovery of **1a**.²⁴ In accordance with Hoarau and Bischoff's observation on the importance of the base on the C2–H arylation of imidazolones, the same reaction performed in the presence of benzylamine (1.5 equiv) afforded indeed **2a**, albeit in only 26% yield. The low yield of **2a** indicated nevertheless that the formation of transient imidazolone **1a** followed by C–H arylation, although feasible, was not the main reaction pathway under our reaction conditions. Taking into consideration the above results, a possible reaction sequence leading to imidazolone **2** is depicted in Scheme 4b. Coordination of the isocyano and the amino groups to the copper affording complex **A**. Migratory insertion from **A** would afford **B**. Intramolecular

Scheme 2. Three-Component Synthesis of 2-Aryl Substituted Imidazolones 2

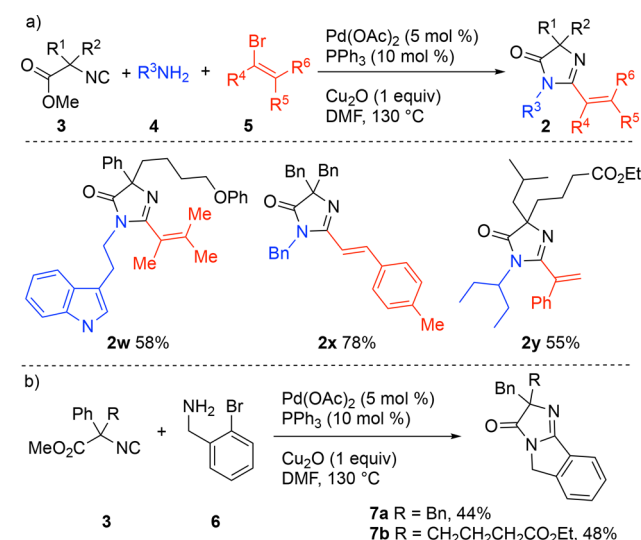


^aConditions: 3 (0.1 mmol), 4 (0.15 mmol), 5 (0.2 mmol), Cu₂O (1.0 equiv), Pd(OAc)₂ (5 mol %), PPh₃ (10 mol %), DMF (1.0 mL), 130 °C.

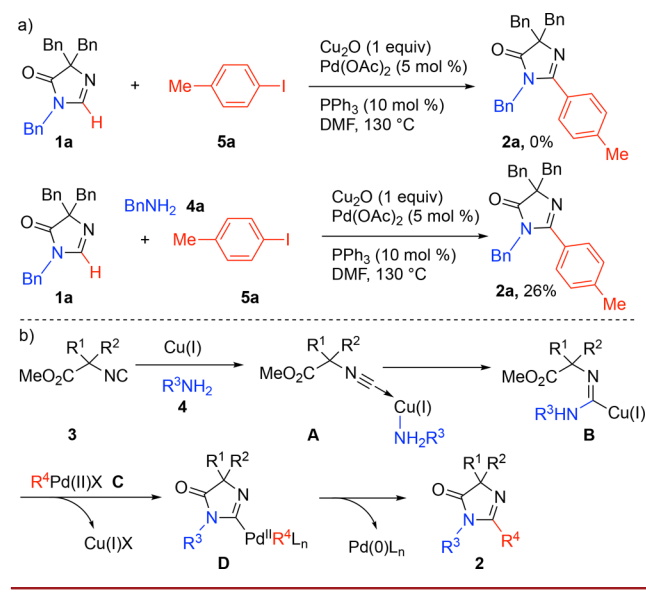
amidation followed by transmetalation of the organocopper species with R⁴Pd(II)X C (R⁴ = aryl, vinyl), generated in situ by oxidative addition of 5 to Pd(0), would produce complex D.²⁵ Reductive elimination would then furnish imidazolone 2 with concurrent regeneration of the palladium(0) catalyst.

In summary, we have developed a palladium-catalyzed, Cu₂O-mediated three-component reaction of methyl α,α -disubstituted α -isocyanoacetates 3 with primary amines 4 and aryl(vinyl) iodides 5. Three chemical bonds were created in this process leading to diversely substituted 2,3,5,5-tetrasubstituted imidazolones 2 in good yields. This multicomponent reaction, applicable to a wide range of substrates, constitutes novel access to this medicinally important nonaromatic heterocycle.

Scheme 3. Three-Component Synthesis of 2-Vinyl Substituted Imidazolones and Tricyclic Compounds



Scheme 4. Possible Reaction Pathway and Control Experiments



■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.7b03479.

Experimental procedures, product characterization data, and ¹H and ¹³C NMR spectra for new compounds (PDF)

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Notes

The authors declare no competing financial interest.

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