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1,3-DIARYL-2-CARBOXYINDOLES AS POTENT NON-PEPTIDE ENDOTHELIN ANTAGONISTS

Amy M. Bunker^{*}, Jeremy J. Edmunds, Kent A. Berryman, Donnelle M. Walker, Michael A. Flynn, Kathy M. Welch, and Annette M. Doherty

Departments of Chemistry and Cardiovascular Therapeutics, Parke-Davis Pharmaceutical Research Division of Warner-Lambert Company, Ann Arbor, Michigan 48105

Abstract: Endothelin-1 is a potent vasoconstrictor that is thought to be involved in many human disease states. We have developed a series of indole non-peptide endothelin antagonists including PD 159110 (31), an ET_A selective antagonist, and PD 159020 (37), a non-selective ET_A/ET_B antagonist. The discovery, synthesis, and structure-activity relationships of this series of compounds are described. Copyright © 1996 Elsevier Science Ltd

Introduction: The family of endothelin peptides consist of endothelin-1 (ET-1), endothelin-2 (ET-2), endothelin-3 (ET-3), and the vasoactive intestinal contractor (VIC). The biological effects of the endothelins has been shown to occur through the interaction of the peptides with specific receptor subtypes. The ETA receptor subtype, which is selective for ET-1, is predominantly found in the vascular smooth muscle. The ETB receptor subtype is non-selective, binding ET-1, ET-2, and ET-3 with similar affinity, and has been found in a variety of tissues including human cultured umbilical vein and human mammary arteries and veins.^{1,2,3,4} The ET_c receptor subtype, which is selective for ET-3, has recently been cloned from dermal melanophores of *Xenopus laevis*, but no mammalian homolog is known.⁵

Endothelin has been implicated in a number of disease states including renal failure, pulmonary hypertension, cerebral ischemia and vasospasm, endotoxic shock, and congestive heart failure.⁶ The ETA receptor is known to mediate a major portion of the vasoconstrictor activity of endothelin in human vessels.

There have been a number of peptide and non-peptide endothelin antagonists reported in the literature. These include potent ETA-selective antagonists such as BQ-123^{7,8} and PD 156707;⁹ non-selective ETA/ETB antagonists such as PD 142893,¹⁰ PD 145065,¹¹ SKF 209670,¹² L-749,329,¹³ as well as, Ro 47-0203 (bosentan);¹⁴ and the ETB-selective antagonist BQ-788.¹⁵ The synthesis of potent, orally active, non-peptide endothelin antagonists with differing selectivity for the ETA and ETB receptors was the objective of this research.

Results and Discussion: Screening of the Parke-Davis compound library afforded several moderately active compounds when tested in rabbit renal artery vascular smooth muscle cells expressing the ETA receptors.^{9,16} We selected the lead structure, compound 1, for synthetic modification to define the structure-activity relationships and to enhance potency. The compound exhibited micro molar binding affinity for the rabbit ETA receptor and was inactive at the rat ETB receptor (rat cerebellum).



Preliminary SAR revealed that an acidic substituent at C-2, a lipophilic substituent at C-3, and a benzyloxy substituent at C-5 were essential for receptor binding activity. However, with the disclosure of a series of indanes^{12,17} as endothelin antagonists, we directed our attention to a series of N1-substituted indoles. While incorporating substituents at C-2, C-3, and C-5 that had been found to be important for activity in the benzothiophene series, we decided to synthesize the two indoles **23** and **24**.⁹ These two compounds both displayed moderate ETA binding affinity and encouraged us to systematically investigate the effect of electron donating substituents on each aromatic ring.

Two series of indoles were prepared. The first series, the N-aryl analogs, are summarized in Table 1. From these compounds it was evident that a methylenedioxy substituent at both R_1 , R_2 and R_5 , R_6 afforded potent compounds in combination with an appropriate substituent at R_7 . In fact, as compounds **22-25** indicate, the optimal substituent at R_7 was the propyloxy group. Increasing electron donation by substituting R_7 and R_8 with methoxy groups afforded compound **26** which demonstrated a binding affinity approaching that of the optimal substitution.

A second series of indoles incorporated a benzyl substituent at N-1 as exemplified in Table 2. While it was apparent that methylenedioxy substituted phenyl rings again afforded active compounds, optimal activity was achieved by modifications of the R7 and R8 substituents. Compounds **27-32** demonstrated that a propyloxy substituent at either R7 or R8 afforded active compounds. Interestingly, dual substitution of R7 and R8 (**36-38**) indicated that ETA selective (**36**) and non-selective (**37**) compounds could be prepared by modification of the R8 substituent.

Synthesis: The initial synthetic route to compounds **23** and **24**, the first two indoles synthesized in this series of compounds, followed a classical route, as depicted in Scheme 1.¹⁸ However, this initial synthetic route required some modification to allow a more flexible approach for the incorporation of a variety of substituents at N-1 and C-3. Furthermore, since the SAR studies indicated that a variety of C-5 and C-6 substituents were important for activity, we chose a variety of benzaldehydes as starting materials for the synthesis of substituent for activity, we chose a variety of benzaldehydes as starting materials for the synthesis of substituents to be introduced at C-3, C-5 and C-6. Unfortunately, the introduction of substituents at N-1 posed a significant problem. SAR studies revealed that electron rich arenes were required at N-1, but the copper catalyzed arylation of nitrogen was not efficient with these substrates. Thus, while bromobenzene efficiently arylates indoles, the use of 1-bromo-3,4-methylenedioxy benzene simply led to decarboxylation in addition to some arylation. The use of 1-iodo-3,4-methylenedioxy benzene improved the yield of product substantially but decarboxylation was still a problem. Fortunately, we discovered that benzyl substituents were well tolerated at N-1 and C-3.^{19,20}

Biological Evaluation: Structure-activity relationships were investigated using IC₅₀ values obtained from receptor binding in Ltk-cells expressing recombinant human receptors (hET_A), and

CHO-K1 cells expressing recombinant human receptors (hETB).^{9,21} Selected compounds were evaluated for antagonist activity by measuring the ability of these compounds to reduce ET-1 stimulated arachidonic acid release (AAR) in cultured rabbit renal vascular smooth muscle cells.^{9,22} In addition, in vitro antagonism of ET-1 stimulated vasoconstriction was carried out in rabbit femoral artery, ETA(pA2), to demonstrate a functional response to antagonism of ETA in this isolated tissue. Inhibition of sarafotoxin-6c stimulated vasoconstriction was carried out in rabbit pulmonary artery, ETB(pA2).²¹

Conclusions: Extensive investigation of electron donating substituents on all three aromatic rings led to the discovery of ET_A selective indoles, such as compound **31**, and compounds that demonstrated affinity to both ET_A and ET_B receptors, such as compound **37**. Selected compounds were evaluated for their ability to inhibit the release of arachidonic acid in rabbit renal artery vascular smooth muscle cells (ET_A) upon stimulation with ET-1. Compound **31**, for example, demonstrated an AAR-A IC₅₀ of 0.48 μ M and effectively inhibited the contraction of ET-1 induced rabbit femoral artery, which are known to express predominantly ET_A receptors, with a pA₂ of 6.9. Compound **37** also demonstrated excellent inhibition of arachidonic acid release, AAR-B, with an IC₅₀ of 0.23 μ M and thereby demonstrating functional ET_B activity. These compounds should prove useful in elucidating the physiological and pathophysiological role of endothelin.

Scheme 1: 1,3-diaryl-2-carboxyindoles



(a) i. 1.0 equiv. KOH, 1.10 equiv. Br2, 21.5 equiv. HOAc, reflux, 95% yield; ii. 3.0 equiv. 3,4-methylenedioxyaniline, 1.10 equiv. KOAc, 0.002 equiv. cupric acetate, 1.10 equiv. Et3N, isopropyl alcohol, 85% yield;
(b) i. 37% formaldehyde, ethanol, reflux, 74% yield; ii. 1.0 equiv. NaCN, 7.1 equiv. DMF, 42°C, 94% yield;
(c) i. 4.0 equiv. 50% NaOH (aq), reflux, 77% yield; ii. 3.0 equiv. TMSCHN2, toluene/methanol 5:1, 93% yield;
(d) 1.4 equiv. Na metal, methanol, 89% yield;
(e) i. 5.0 equiv. (3,4-methylenedioxy)phenyl boronic acid, 1.5 equiv. K2CO3, 0.1 equiv. Pd(PPh3)4, toluene/DMF 5:1, 55% yield; iii. 15.0 equiv. LiOH, THF/MeOH/H2O 3:1:1, 84% yield.

Scheme 2: 1,3-diaryl-2-carboxyindoles



(a) i. 2.5 equiv. N₃CH₂CO₂CH₃, 2.5 equiv. Na metal, MeOH, -19 to 0°C, 49% yield; ii. toluene, reflux, 51% yield;
(b) i. 15 equiv. LiOH, THF/MeOH/H₂O 3:1:1, 50°C, 95% yield; ii. 1.3 equiv. 1-iodo-3,4-methylenedioxy benzene, 1.0 equiv. copper oxide, 2.0 equiv. KOH, DMF, 64% crude yield;
(c) i. 3.0 equiv. TMSCHN₂, toluene/methanol 5:2, 61% yield; ii. 1.05 equiv. pyridinum bromide perbromide, pyridine, 77% yield;
(d) i. 1.5 equiv. LiOH, THF/MeOH/H₂O 3:1:1, 50°C, 71% yield.

Scheme 3: 1-benzyl-3-aryl-2-carboxyindoles



(a) 1.05 equiv. pyridinium bromide perbromide, pyridine, 92% yield;
(b) 1.5 equiv. (3,4-methylene-dioxy)phenyl boronic acid, 0.01 equiv. Pd(PPh3)4, toluene/methanol/sat. aq sodium bicarbonate 1:1:1, 80°C, 82% yield;
(c) 1.3 equiv. NaH, 1.25 equiv. (3,4-methylenedioxy)benzyl chloride, DMF, 73% yield;
(d) 15.0 equiv. LiOH, THF/MeOH/H2O 3:1:1, 53°C, 98% yield.

Structure-Activity Relationships:

Table 1 - 1,3-diaryl-2-carboxyindoles



Ex	n	R1	R2	R3	R5	R6	R7	R8	hET _A Ю 50 (μМ)	hЕТВ IС50 (µМ)
13	0	н	OMe	н	-OCH 2O-		OMe	н	1.1	>25
14	0	н	OMe	Н	-OCH ₂ O		OBn	Н	2.4	21
15	0	OMe	OMe	н	-OCH ₂ O		OMe	Н	1.5	>25
16	0	OMe	н	н	-OCH ₂ O-		OMe	Н	5.5	>25
17	0	-00	H2O	н	н	Н	OMe	н	12	>25
18	0	-OCH ₂ O		н	OMe	Н	н	н	2.3	>25
19	0	-00	:H ₂ O	н	OMe	Н	OMe	Н	3.5	>25
20	0	-00	H ₂ O	Н	н	н	a	Н	10	>25
21	0	-00	H2O	н	н	Н	н	OMe	7.4	>25
22	0	-OCH ₂ O-		н	-OCH ₂ O-		н	Н	4.1	22
23	0	-OCH 20-		н	-OCH 20-		OBn	Н	1.3	14
24	0	-OCH ₂ O-		н	-OCH ₂ O-		OMe	н	0.26	15
25	0	-OCH ₂ O-		Н	-OCH 2O-		OPr	н	0.043	12
26	0	-OCH 20-		Н	-OCH 2O-		OMe	OMe	0.08	7.0

Table 2 - 1-benzyl-3-aryl-2-carboxyindoles

Ex	n	R1	R2	R3	R5	R6	R7	R8	hET _A IC 50 (μM)	hЕТВ Ю50(μМ
27	1	-OCH ₂ O-		Н	-OCH 20-		Н	н	1.5	3.3
28	1	-OCH ₂ O		н	-OCH ₂ O-		OBn	н	0.8	2.5
29	1	-OCH2O-		н	-OCH ₂ O-		OMe	н	0.35	2.4
30	1	-OCH ₂ O		н	-OCH ₂ O-		OPr	н	0.08	0.85
31	1	-OCH 2O-		Н	-OCH 2O-		н	OPr	0.047	5.6
32	1	-OCH 2O-		н	-OCH 20-		OPr	Н	0.13	1.1
33	1	-00	H ₂ O	н	Н	ГΗ	н	н	11	14.5
34	1	-OCH 2O-		Н	OMe	Н	OBn	Н	2.7	4.4
35	1	OMe OMe		н	-OCH 20-		OMe	OMe	0.06	0.85
36	1	-OCH 2O-		н	-OCH ₂ O-		OMe	OMe	0.06	1.4
37	1	-OCH 2O-		H	-OCH 2O-		OMe	OBn	0.03	0.05
38	1	-OCH ₂ O		н	-OCH ₂ O-		-OCH2O-		0.24	0.67

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