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Revised Structure and Absolute Configuration of the Sesquiterpene (+)-Bazzanene

By Akihiko Matsuo and Shûichi Hayashi

(Department of Chemistry, Faculty of Science, Hiroshima University, Hiroshima 730, Japan)

Summary A previous assignment of the structure of (+)-bazzanene is revised based on chemical correlation with (+)-trichodiene (II); the absolute configuration has been determined to be as shown in structure (I), a diastereoisomer of (+)-trichodiene.

In a previous investigation on terpenoids of the liverworts (Hepaticae), we isolated a novel bicyclic sesquiterpene hydrocarbon, bazzanene, $C_{15}H_{24}$; $[\alpha]_D$ +48.0°, from the leafy liverwort, Bazzania pompeana, and presented a gross structure for it based on chemical and spectral evidence.1 Since the proposed structure was inconsistent in some respects with a recent examination of the off-resonance ¹³C n.m.r. spectrum, the structure of the hydrocarbon was reinvestigated in detail, resulting in a revision to structure (I) corresponding to a diastereoisomer of (+)-trichodiene (II), $C_{15}H_{24}$; $[\alpha]_D + 21^\circ$, which has been isolated recently from the fungus Trichosecium roseum by Nozoe et al.2 We report evidence for the revised structure and absolute configuration.

Although the off-resonance ¹³C n.m.r. spectrum† of bazzanene [4 s (δ 153·4 and 132·4: olefinic; δ 50·2 and 36·7: aliphatic), 1 d (δ 120·4 olefinic), 7 t (δ 106·2: 1 olefinic; δ 38.9, 37.0, 32.3, 28.0, 27.8, and 23.4: 6 CH₂), and 3 Me q $(\delta 23.6, 23.4, \text{ and } 17.6)$] was consistent with the structure of trichodiene, the ¹H n.m.r. spectra of the two hydrocarbons were slightly different. Accordingly, (+)-bazzanene was oxidized with m-chloroperbenzoic acid in CH₂Cl₂ to give as the major product the monoepoxide (III), $\mathrm{C_{15}H_{24}O}$ (M+ 220); [lpha]_D +36·6°, $^{\dagger}_{+}$ containing an intact exo double bond and as a minor product the diepoxide (IV), $C_{15}H_{24}O_2$ (M+ 236). The ¹H n.m.r. spectrum of the diepoxide (IV) was rather different from that of the diepoxide derived from trichodiene, but the spectra of the monoepoxide (III) resembled the spectra of trichodiene monoepoxide; its ¹³C n.m.r. spectrum[†] [4 s (δ 159·1, 56·9, 49.8, and 36.0), 1 d (\delta 58.9), 7 t (\delta 106.8, 38.9, 37.0, 31.8, 26.6, 24.5, and 23.4), and $3 \neq (\delta 23.7, 23.1, and <math>19.7$)] was also reasonably explained by structure (III). As it was thought from the above results that (+)-bazzanene was a diastereoisomer of (+)-trichodiene, a chemical correlation was carried out as follows. Oxidation of the monoepoxide (III) with OsO4 in pyridine for 30 h3 gave the epoxyglycol (V), $C_{15}H_{26}O_3$ (M+ 254); $[\alpha]_D$ -22.0°, glycol fission of which with NaIO₄ afforded the nor-epoxy-ketone (VI), $C_{14}H_{22}O_2$ (M⁺ 222); [α]_D +88·2°, containing a cyclopentanone unit. Reduction of (VI) with LiAlH4 in ether afforded two epimeric diols, (VIIa), $C_{14}H_{26}O_2$ (M+ 226); m.p. $95 \cdot 5 - 96$ °C; $[\alpha]_{\rm D} = 28 \cdot 8$ °, and "(VIIb), $C_{14}H_{26}O_{2}$ (M+ 226); m.p. 98 - 99 °C; $[\alpha]_{\rm D} + 5 \cdot 1$ °, in a 2:1 ratio. The major diol (VIIa) was oxidized with CrO3 in pyridine into the hydroxy-ketone (VIII), $C_{14}H_{24}O_2$ (M^+ 224); $[\alpha]_D$ +95.7°. The i.r., ¹H n.m.r., and mass spectra of (VIII) thus obtained coincided in all respects with those of the hydroxy-ketone prepared from (+)-trichodiene via the corresponding nor-epoxy-ketone.2

Further, the o.r.d. and c.d. spectra of the epoxy-ketone (VI) $\{[\phi]_{330} + 5230 \text{ (max)}, [\phi]_{312} 0, [\phi]_{288} - 4610 \text{ (min)}; [\theta]_{312} + 1380 \text{ (max)} \text{ (in dioxan)} \}$ and the hydroxy-ketone (VIII) $\{[\phi]_{330} + 6190 \text{ (max)}, [\phi]_{311} 0, [\phi]_{288} - 4620 \text{ (min)}; [\theta]_{313} + 1310 \text{ (max)} \text{ (in dioxan)} \}$ showed a positive Cotton effect.⁴ These facts indicated that (+)-bazzanene and (+)-trichodiene had the same absolute configuration of the methyl-cyclopentane unit. Thus, the structure of (+)-bazzanene is the same as that of (+)-trichodiene except for the relative position of the endo double bond, i.e. the configuration of the tertiary methyl group in the cyclobexane ring

Treatment of (I) with OsO₄ in pyridine for 1.5 h^3 caused attack of the reagent from the less hindered site against the axial tertiary methyl group to the glycol (IX), $C_{15}H_{26}O_2$ (M^+ 238); m.p. 87.5-88 °C; $[\alpha]_D$ +73.0°, which had an equatorial secondary OH group and an axial tertiary OH group as well as an unchanged *exo* methylene group; the glycol (IX) gave the hydroxy-cyclohexanone (X), $C_{15}H_{24}O_2$ (M^+ 236); $[\alpha]_D$ +119°, upon Sarett oxidation. In order to determine the absolute configuration of the cyclohexane unit the glycol (IX) was converted into the

HO...

(IX)
$$R = H$$

(XI) $R = COC_6H_{\lambda}Br-\rho$

(III)

(III)

(IV)

(IV)

(VII)

(VII)

(VIII)

(VIII)

(VIII)

(VIII)

(VIIII)

(VIIII)

Reagents: i, m-ClC₆H₄CO₃H in CH₂Cl₂; ii, OsO₄ in pyridine; iii, NaIO₄; iv, LiAlH₄ in Et₂O; v, CrO₃ in pyridine.

p-bromobenzoate (XI), $C_{22}H_{20}BrO_3$ (M⁺ 420 and 422); $[\alpha]_D + 27 \cdot 7^\circ$, the molecular rotation of which $([M]_D + 113^\circ)$ showed a decrease compared with that $([M]_D + 174^\circ)$ of the original glycol. In the glycol, accordingly, the carbon atom bearing the secondary equatorial OH group had the (R) configuration, as shown in structure (IX), according to the benzoate rule.⁵ In addition, the α-hydroxy-ketone (X) showed a positive Cotton effect in the o.r.d. and c.d. spectra $\{[\phi]_{326} + 5180 \text{ (max)}, [\phi]_{303} 0, [\phi]_{287} - 1280 \text{ (min)}; [\theta]_{308} + 835 \text{ (max) (in dioxan)}\}$ owing to the contribution of the axial methyl group on the β-carbon of the carbonyl group.⁶

On the basis of the above evidence, the structure and absolute configuration of (+)-bazzanene was shown to be as in the stereostructure (I) which corresponds to a diastereoisomer of (+)-trichodiene (II). It is thought that (+)-bazzanene is an important precursor in the biogenesis of (+)- α -pompene (= isogymnomitrene) and (-)- β -pompene⁷ (= gymnomitrene)⁸ which has been isolated from the same liverwort.

† The ¹³C n.m.r. spectra were determined in CDCl₃ solutions; values of δ are given in p.p.m.

‡ All new compounds (III)—(VII) and (IX)—(XI) gave spectral data in good accord with the assigned structures. Optical rotations were measured in CHCl₃ solutions and i.r. and ¹H n.m.r. spectra in CCl₄ solutions.

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