

## Published on Web 07/12/2007

## *O*-Aryloxycarbonyl Hydroxamates: New $\beta$ -Lactamase Inhibitors That Cross-Link the Active Site

Pauline N. Wyrembak,<sup>†</sup> Kerim Babaoglu,<sup>‡</sup> Ryan B. Pelto,<sup>†</sup> Brian K. Shoichet,<sup>‡</sup> and R. F. Pratt<sup>\*,†</sup>

Department of Chemistry, Wesleyan University, Middletown, Connecticut 06459, and Department of Pharmaceutical Chemistry, University of California, San Francisco, 1700 4th Street, San Francisco, California 94158-2330

Received April 4, 2007; E-mail: rpratt@wesleyan.edu

The effective lifetime of  $\beta$ -lactams as antibiotics can be extended by concurrent treatment of patients with  $\beta$ -lactamase inhibitors.<sup>1,2</sup> The  $\beta$ -lactamase inhibitors in commercial production at present, however, are of limited spectrum and are largely specific to the class A enzymes.<sup>1</sup> New classes of inhibitor with broader specificity covering classes B, C, and D  $\beta$ -lactamases would be welcome since the prevalence of these enzymes, and thus  $\beta$ -lactam resistance among bacteria, continues to grow. We describe in this communication a new type of  $\beta$ -lactamase inhibitor with an unusual mechanism of action involving the covalent cross-linking of active side residues.

The depsipeptides of general structure 1 are  $\beta$ -lactamase substrates.<sup>3,4</sup> Although the aza analogues **2** display little or no substrate activity,<sup>5</sup> we were encouraged to try the oxa analogues **3** because of the inhibitory properties of vanadate/ hydroxamic acid complexes.<sup>6</sup> Compounds 4 and 5 were therefore obtained from careful reaction of the appropriate hydroxamic acid and chloroformate (the latter carboxyl-protected in the case of 5) in the presence of imidazole (Supporting Information). An alkoxy side chain was chosen since the alkyl or aryl analogues were unstable to the Lössen rearrangement.<sup>7</sup> An NMR spectrum of <sup>15</sup>N-4 in DMSO-d<sub>6</sub> showed a <sup>15</sup>N resonance at 161.1 ppm coupled (J = 93 Hz) to a proton at 11.75 ppm. This clearly identifies the product as the O-acyl rather than N-acyl hydroxamic acid. Compounds 4 and 5 hydrolyzed in aqueous buffer (20 mM MOPS, pH 7.5), yielding benzyl Nhydroxycarbamate, the phenol, and, presumably, bicarbonate; pseudo-first-order rate constants ( $k_0$ ) of 2.5  $\times$  10<sup>-4</sup> and 2.78  $\times$  $10^{-4}$  s<sup>-1</sup>, respectively, were obtained.



Compound **4** inhibited, essentially irreversibly, the class C  $\beta$ -lactamase of *Enterobacter cloacae* P99 in a time-dependent fashion, as evident from Figure 1. At low inhibitor/enzyme concentration ratios, the final activity of the enzyme was not zero, which suggested that some turnover accompanied the inhibition reaction (background hydrolysis of **4** was not sufficient to explain the final activity). A greater excess of inhibitor did completely inactivate the enzyme (Figure 1). A plot of residual activity versus concentration of **4** (Figure 2) suggested that about two turnovers accompanied inhibition. These data were fitted to Scheme 1, where EI is likely to be a hydrolyzable acyl enzyme<sup>4</sup> which can also partition to a dead end complex EI'. These fits, shown as solid

9548 J. AM. CHEM. SOC. 2007, 129, 9548-9549



**Figure 1.** Activity of the P99  $\beta$ -lactamase (0.25  $\mu$ M) as a function of time in the presence of **4** (0  $\mu$ M,  $\blacklozenge$ ; 0.5  $\mu$ M,  $\blacksquare$ ; 2.5  $\mu$ M,  $\blacklozenge$ ).



*Figure 2.* Activity of the P99  $\beta$ -lactamase (0.25  $\mu$ M) after complete reaction with **4** ( $\bullet$ ) and **5** ( $\bigcirc$ ) at various concentrations (0–1.0  $\mu$ M).

Scheme 1



lines in the figures, yielded  $k_1$  and  $k_2/k_3$  values of  $6.1 \pm 0.2 \times 10^3$  s<sup>-1</sup> M<sup>-1</sup> and 2.0  $\pm$  0.1, respectively.

Compound **5** was also an inhibitor of the P99 enzyme (Figure 2). Experiments analogous to those described above yielded values of  $k_1$  and  $k_2/k_3$  of  $5.4 \pm 0.3 \times 10^3 \text{ s}^{-1} \text{ M}^{-1}$  and  $1.00 \pm 0.05$ , respectively. It is interesting that **5**, bearing the *m*-carboxy sub-

 <sup>&</sup>lt;sup>†</sup> Wesleyan University.
<sup>‡</sup> University of California, San Francisco.



**Figure 3.** Top: crystal structure of the active site of the AmpC  $\beta$ -lactamase after inhibition by **4**, showing the carbamate cross-link between Ser64 and Lys315. The electron density is contoured at the  $3\sigma$  level. Bottom: the same view of the wild-type enzyme,<sup>10</sup> showing a clear gap between Ser64 and Lys315.

stituent, is not a better inhibitor than 4. This result is contrary to what would be expected from comparable substitution in the depsipeptides 1. A *m*-carboxy group in 1 is thought to interact specifically with the P99 active site.<sup>8</sup> The results suggest that 1 and 5 may not bind to the active site in the same way. It should be noted, however, that inactivation of the enzyme by 5 was competitively inhibited by *p*-nitrobenzene boronic acid, which is itself a competitive inhibitor of the P99 enzyme.<sup>9</sup>

An electrospray mass spectrum of the inhibited enzyme was obtained. Enzyme (10  $\mu$ M) and 4 (5 mM) were incubated together in MOPS buffer (above) for 5 min, after which time the enzyme was inactive. The protein was then precipitated with trichloroacetic acid, washed, and dried, and an ES+ mass spectrum obtained. The spectrum showed an increase in protein mass of 29, in good agreement with the mechanism of inactivation described below.

A 1.8 Å resolution crystal structure of the inhibited AmpC class C  $\beta$ -lactamase was also obtained, as described in the Supporting Information. The only observable difference from the structure of the native enzyme<sup>10</sup> was at the active site. In monomer A of the structure, the  $O_{\gamma}$  oxygen of Ser64 is flipped some 180° (Ser64C<sub> $\alpha$ </sub>C<sub> $\beta$ </sub>O<sub> $\gamma$ </sub>C) and forms part of an unprecedented carbamate bridge to N<sub> $\xi$ </sub> of Lys315 (Figure 3). Tyr150 has moved aside slightly to accommodate insertion of a carbonyl, but O<sub> $\xi$ </sub> remains within hydrogen-bonding distance of the inserted carbonyl oxygen (Figure



S1, Supporting Information). Lys73 remains hydrogen bonded to Tyr150  $O_{\zeta}$ . Coordinates of the structure have been deposited in the RCSB protein data bank as entry 2P9V.

The mechanism of inhibition of the P99  $\beta$ -lactamase by **4** and **5** can thus, from the data available at present, be represented by the sequence shown in Scheme 2. This represents a novel cross-linking of the active site and a previously unobserved specific modification of one of the two conserved lysine residues of the  $\beta$ -lactamase active site. Inhibition of class A  $\beta$ -lactamases by clavulanic acid and penicillin sulfones has been shown to involve cross-linking of the active site serine to the conserved Ser130.<sup>11,12</sup>

We have also observed that **4** and **5**, and other derivatives of these compounds, inhibit the class A TEM  $\beta$ -lactamase. We plan further experiments to determine the scope of these compounds against  $\beta$ -lactam-recognizing enzymes.

Acknowledgment. This research was supported by the National Institutes of Health through Grant AI 17986 to R.F.P. and GM 63815 to B.K.S. K.B. is supported by a Ruth L. Kirschstein National Research Service Award fellowship (GM 076883).

**Supporting Information Available:** Synthetic procedures for compounds **4** and **5** and the kinetics methods. Details of the crystallographic procedures and statistics are also provided. This material is available free of charge via the Internet at http://pubs.acs.org.

## References

- (1) Georgopapadakou, N. Exp. Opin. Invest. Drugs 2004, 13, 1307.
- (2) Buynak, J. D. Biochem. Pharmacol. 2006, 31, 930.
- (3) Pratt, R. F.; Govardhan, C. P. Proc. Natl. Acad. Sci. U.S.A. 1984, 81, 1302.
- (4) Govardhan, C. P.; Pratt, R. F. Biochemistry 1987, 26, 3385.
- (5) Cabaret, D.; Garcia Gonzalez, M.; Wakselman, M.; Adediran, S. A.; Pratt, R. F. *Eur. J Org. Chem.* 2001, 141.
- (6) Bell, J. H.; Pratt, R. F. Biochemistry 2002, 41, 4329.
- (7) Renfrow, W. B., Jr.; Hauser, C. R. J. Am. Chem. Soc. 1937, 59, 2308.
- (8) Ahn, Y.-M.; Pratt, R. F. Bioorg. Med. Chem. 2004, 12, 1539.
- (9) Nagarajan, R.; Pratt, R. F. Biochemistry 2004, 43, 9664
- (10) Usher, K. C.; Blaszczak, L. C.; Weston, G. S.; Shoichet, B. K.; Remington, S. J. Biochemistry 1998, 37, 16082.
- (11) Brown, R. P. A.; Aplin, R. T.; Schofield, C. J. Biochemistry 1996, 35, 12421.
- (12) Kuzin, A. P.; Nukaga, M.; Nukaga, Y.; Hujer, A.; Bonomo, R. A.; Knox, J. R. Biochemistry 2001, 40, 1861.

JA072370U