

## Preparation of Isoxazoline *N*-Oxide Derivatives from a Nitro-sugar and *S*-Ylides

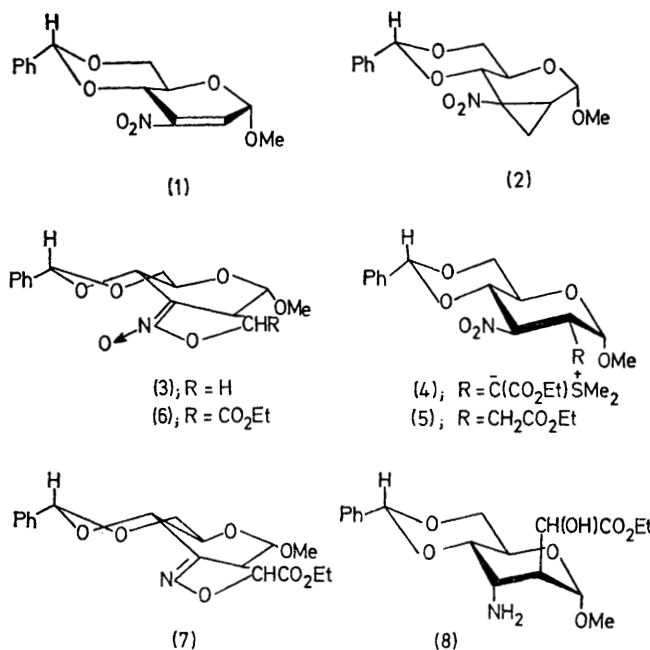
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**Summary** Axial attack predominates over equatorial attack in the addition of dimethylsulphoxonium methylide and ethyl dimethylsulphuranylideneacetate to the nitro-

sugar (**1**); axial attack selectively gave the 1,4-addition products (**3**) and (**6**), whereas equatorial attack afforded the 1,2-addition products (**2**) and (**4**), respectively.

VERY few examples are known of reactions between nitro-olefins and *S*-ylides,<sup>1</sup> where the product is a cyclopropane<sup>2,3</sup> or stabilized *S*-ylide derivatives.<sup>3</sup> We report preparations of the isoxazoline *N*-oxides (3) and (6) from the nitro-sugar (1)<sup>4</sup> and *S*-ylides which are the first examples of 1,4-addition between a nitro-olefin and *S*-ylides.



Treatment of (1) with dimethylsulphoxonium methylide in Me<sub>2</sub>SO at 0° C gave a mixture of the cyclopropane (2) (16.5%), m.p. 131–132 °C, [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 35.0° (c 1, CHCl<sub>3</sub>),  $\nu_{\max}$  (KBr) 1550 (NO<sub>2</sub>) cm<sup>-1</sup>, and the isoxazoline *N*-oxide (3) (56.2%), m.p. 178.5–179.5 °C; [ $\alpha$ ]<sub>D</sub><sup>20</sup> – 139.2° (c 1, CHCl<sub>3</sub>),  $\nu_{\max}$  (KBr) 1640 (C=N) cm<sup>-1</sup>, which were separated by column chromatography on silica gel. Elemental analyses of both compounds corresponded to C<sub>15</sub>H<sub>17</sub>NO<sub>6</sub>. The cyclopropane structure of (2) was determined from its i.r. (the presence of NO<sub>2</sub> absorption band) and n.m.r. (the presence of cyclopropane ring protons at  $\delta$  2.61, 2.18, and 1.98)

<sup>†</sup> Satisfactory elemental analysis has not yet been obtained for (4).

<sup>1</sup> B. M. Trost and L. S. Melvin, Jr. in 'Sulfur Ylides, Emerging Synthetic Intermediates,' Academic Press, New York, 1975.

<sup>2</sup> J. Asunkis and H. Shechter, *J. Org. Chem.*, 1968, **33**, 1164.

<sup>3</sup> T. Sakakibara, T. Takamoto, R. Sudoh, and T. Nakagawa, *Chem. Letters*, 1972, 1219.

<sup>4</sup> H. H. Baer and F. Kienzle, *Canad. J. Chem.*, 1967, **45**, 983.

<sup>5</sup> B. Radatus and B. Fraser-Reid, *Canad. J. Chem.*, 1972, **50**, 2909.

<sup>6</sup> P. M. Collins, D. Gardiner, S. Kumar, and W. G. Overend, *J.C.S. Perkin I*, 1972, 2611.

spectra. The *allo*-configuration of (2) was assigned on the basis of the value of  $J_{1,2}$  (6.3 Hz)<sup>5</sup> and that of long range coupling (1.0 Hz) between the *exo*-proton of the cyclopropane ring and H-4. The structure of (3) was deduced from a comparison of its i.r. and n.m.r. spectra with those of (6) described below.

Reaction of (1) with ethyl (dimethylsulphuranylidene)-acetate in tetrahydrofuran (THF) at 0 °C gave a mixture of the stabilized *S*-ylide (4)<sup>†</sup> (ca. 22%, as syrup) and the isoxazoline *N*-oxide (6) (54.9%), m.p. 151.5–152.0 °C (decomp.), [ $\alpha$ ]<sub>D</sub><sup>20</sup> – 121° (c 1, CHCl<sub>3</sub>), which were separated by preparative t.l.c. The former, having the characteristic absorption band of the ester carbonyl group of the ylide species at 1630 cm<sup>-1</sup> in its i.r. spectrum and two methyl signals (SMe<sub>2</sub>) in its n.m.r. spectrum, was characterized as the reduced compound (5), m.p. 91.5–92.0 °C, [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 72° (c 1, CHCl<sub>3</sub>),  $\nu_{\max}$  (KBr) 1720 (CO<sub>2</sub>Et) and 1550 (NO<sub>2</sub>) cm<sup>-1</sup>,  $J_{1,2}$  3.1 and  $J_{2,3} = J_{3,4} = 11.1$  Hz, obtained by catalytic hydrogenation of (4) over 10% palladium on charcoal.

The structure of (6) was determined on the basis of its elemental analysis (molecular formula C<sub>18</sub>H<sub>21</sub>NO<sub>8</sub>) and i.r. [KBr, no  $\nu$  (NO<sub>2</sub>),  $\nu$  (C=N) 1645 and  $\nu$  (CO<sub>2</sub>Et) 1720 cm<sup>-1</sup>] and n.m.r. (CO<sub>2</sub>Et, and  $J_{1,2}$  6.0 Hz) spectroscopy. This was supported by the fact that treatment of (6) with Ph<sub>3</sub>P in boiling benzene for 15 h afforded the isoxazoline (7) (91%), m.p. 159–160 °C, [ $\alpha$ ]<sub>D</sub><sup>20</sup> – 120° (c 0.5, CHCl<sub>3</sub>). The *arabino*-configuration of (6) was assigned chemically. Reduction of (6) with Raney nickel afforded a mixture of at least two products as identified by n.m.r. spectroscopy, from which crystals of the *allo*-pyranoside (8) were isolated in 35% yield, m.p. 167–168 °C, [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 64.2° (c 1, CHCl<sub>3</sub>),  $\nu_{\max}$  (KBr) 3380 (OH), 3310 (NH), and 1718 and 1700 (CO<sub>2</sub>Et) cm<sup>-1</sup>,  $J_{1,2}$  0,  $J_{2,3}$  1.3, and  $J_{3,4}$  3.8 Hz.

No other product was formed in either reaction, as shown by n.m.r. spectroscopy, suggesting that the 1,4-addition occurred selectively when the *S*-ylides approached C-2 axially, whereas 1,2-addition took place when the *S*-ylides approached equatorially. It is noteworthy that similar nucleophilic addition of dimethylsulphoxonium methylide to the 3-aryldio-derivatives of (1) gave only the 1,4-adducts regardless of the direction of approach of the nucleophile.<sup>6</sup>

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