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C(3)-Carboxy-cephem

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Summary C(3)-carboxy-cephem and its subsequent conversion into C(3)-ketones, amides, and various N-cephem acylamines is described.

ALTHOUGH the C(3)-formyl group of the cephem molecule was reported in 1966 by Woodward¹ et al. in the total synthesis of cephalosporin C, its conversion into the C(3)carboxylic acid has not been reported.²

We now describe the synthesis of C(3)-carboxy-cephem and its subsequent conversion into other C(3)-derivatives, for example, ketones, amides, and various N-cephem acylamines.

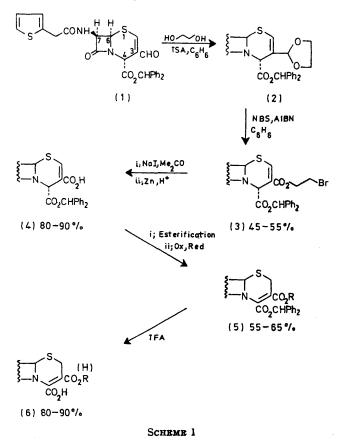
Treatment of the ethylene acetal (2) [m.p. 142–143°] with 1·1 equiv. of N-bromosuccinimide (NBS) and a trace of azobisisobutyronitrile (AIBN) in refluxing benzene, followed by chromatography on silica gel, gave 45–55% of the β -bromoethyl ester (3) [m.p. 129–130°] plus 10–20% of (1) (Scheme 1). Conversion of the β -bromointo the β -iodo-compound (>95%) followed by zinc-acetic acid cleavage gave the C(3)-carboxy- Δ^2 -cephem (4).

Esterification of (4) with diazoalkanes $(CH_2N_2, Ph_2CN_2, p-N_2CH \cdot C_6H_4 \cdot NO_2)$ or with the sodium carboxylate† of (4) and an alkyl iodide [*e.g.*, n-propyl (75%) or isopropyl (70%) iodide] in HMPA³ followed by shifting the double bond (ox-red of S-1) gave (5; R=Me) [ν_{max} (CHCl₃) 1795—1800 cm⁻¹] m.p. 201° decomp., *p*-nitrobenzoate (PNB) m.p. 165—166°. Cleavage (PCl₅) of the C(7)-side chain provided the nucleus of (5), which was then acylated to provide the various C(7)-derivatives. Cleavage of the benzhydryl ester with trifluoroacetic acid (TFA) then gave (6), the C(3), C(4)-diacid (pK_a = 4.0,8.0) resulting from the cleavage of the dibenzhydryl ester.

Biological tests show that, in general, the electron withdrawing effect of the C(3)-esters enhances both the gram negative and the gram positive activity, however, the

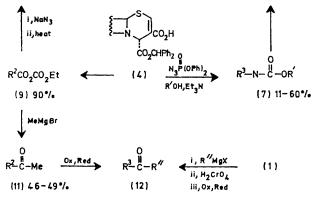
† The sodium carboxylate of (4) is soluble in EtOAc.

compounds show significantly reduced activity against penicillin resistant Staph. aureus.



The synthesis of the C(3)-carboxylic acid allows the preparation of many previously unreported C(3)-deri-





SCHEME 2

 $\mathbf{R}^{\mathbf{s}} = \Delta^{\mathbf{s}} - \mathbf{C}(\mathbf{3})$ -substituted benzhydryl 7-(2-thienyl)acetamidocephem-4-carboxylate $R^3 = \Delta^3$ -C(3)-substituted benzhydryl 7-(2-thienyl)acetamidocephem-4-carboxylate

vatives of the cephem molecule. Various Δ^{s} -N-cephem-Oalkyl carbamates (7) [R' = Me (11%), Et (34%), PNB(60%); for example, have been obtained using the modi-

[‡] The C(3)-PNB carbamate-C(4)-acid has m.p. 149-150°.

§ Identical i.r. and n.m.r. to the product obtained by W. Spitzer et al. from an independent synthesis, personal communication.

¶ Optimum yields are obtained using 3 equiv. of Grignard reagent. Apparently two equiv. are tied up with the amide and the C(4)-proton.

†† One of the methyl diastereomeric carbinols is crystalline (m.p. 133-134°).

11 Secondary acetates are readily prepared from the diastereoisomeric carbinols.

1 R. B. Woodward, K. Heusler, J. Gosteli, P. Naegeli, W. Oppolzer, R. Ramage, S. Ranganathan, and H. Vorbrüggen, J. Amer. Chem. Soc., 1966, 88, 852.

¹ For **0**⁺-C(3)-formyl cephalosporins see J. W. Chamberlin and J. B. Campbell, J. Medicin. Chem., 1967, **10**, 966.
⁹ J. E. Shaw, D. C. Kunerth, and J. J. Sherry, Tetrahedron Letters, 1973, 689.
⁴ T. Shioiri, K. Ninomiya, and S. Yamada, J. Amer. Chem. Soc., 1972, **94**, 6203.

Catalytic reduction of the Δ^2 -acylazide [ν_{max} (CHCl₃) 2143 cm⁻¹] gave the Δ^2 -primary-amide (67%), while secondary and tertiary amides are available from the acid chloride or mixed anhydride.

C(3)-Ketones were prepared from the Grignard (MeMgBr) reaction on the mixed anhydride (9) to give 46-49% of the Δ^2 -C(3)-acetyl (11) [m.p. 139-140°] plus 40-50% of (4). It is, however, more convenient to prepare such C(3)ketone derivatives via the Grignard¶ reaction on the Δ^{\bullet} -C(3)-formyl derivative (1) [Yields R'' = Me (45-65%), Et, or Ph (30-35%)] to give diastereoisomeric carbinols (separated in the case of $R'' = Me, \dagger \dagger Et. \ddagger \ddagger$). Oxidation to the corresponding ketones followed by double-bond isomerization then gave (12) $[v_{max} (CHCl_3) 1795-1800 \text{ cm}^{-1}]$ $R'' = Me (m.p. 201^{\circ} decomp.)$, Et (m.p. 220° decomp.), Ph (m.p. 209° decomp.). The corresponding C(4)-acids were then obtained by TFA cleavage.

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