50. Palladium-Catalyzed Ene-Type Cyclizations of Terpenoid 1,6-Enynes

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Palladium-catalyzed ene-type cyclizations of 1,6-enynes 2 yield methylidene-cyclopentane derivatives 3. Removal of protecting groups from 3 furnishes the diols 4 and 5, whose configurations were established by a combination of X-ray and ¹H-NMR techniques. The predominant formation of isomers 4 is discussed in the context of two alternative hypothetical mechanisms previously proposed for this type of cyclization.

1. Introduction. – The discovery of significant ulcer healing properties in modified acyclic terpenes has led to substantial efforts in their synthesis as well as in their pharmacological characterization [1] [2]. Our interest in this area has focussed on substituted farnesyl derivatives bearing a polar group at each end of a C₁₀ chain [3]. To gain insight into their possible mechanisms of action, we investigated the chemical reactivity of some of the more active compounds of this series. This has already resulted in the discovery of an acetylenic acetal isomerization [4], a synthesis of halogenated furans [5], and an ynol-to-enone rearrangement [6].

Within this context, we decided to evaluate the propensity of a selected set (2a-h) of these terpenoids towards intramolecular cyclizations. This series shares an 1,6-enyne unit embedded into other functionalities. 1,6-Enynes are known to undergo Alder-ene reactions involving the acetylene function as the enophile to yield cyclopentanes containing an exocyclic C=C bond (for reviews, cf. [7] [8]). Resembling truncated prostaglandin derivatives, these molecules might advantageously combine the cytoprotective properties of the open-chain terpenoids with the antisecretory (gastric acid secretion inhibitory) activity [9] claimed for a large number of synthetic prostanoids [10]. Unfortunately, the desired pharmacological effect was not observed. We do, however, wish to document the cyclopentane synthesis, since it represents a useful extension of a known reaction to hitherto unemployed polyfunctional substrates. Given their relative ease of preparation, this type of cyclopentane may also serve as attractive intermediate in the synthesis of more elaborate prostanoids.

2. Results. – The Pd-catalyzed intramolecular ene-type cyclization of enynes 2 yields methylidene-cyclopentane derivatives 3 (*Scheme*). The starting enynes 2a-d were prepared by low-temperature addition of the corresponding (substituted) acetylides to the known [11] aldehyde 1, followed by standard O-silylation of the intermediate propargylic alcohol. Compound 2a was further elaborated into 2e-h by lithiation and subsequent quenching with the appropriate acid derivative. In a similar manner, aldehyde 1 was converted into 7a, b by MnO₂ oxidation following acetylide or methyl-acetylide addition and into α,β -unsaturated ester 8 by a 3-nitropropionate addition/ β -elimination/silylation sequence. Compounds 2 and 8 represent mixtures of racemic diastereoisomers that are not discriminated by ¹H-NMR spectroscopy.

Intramolecular ene reactions of 1,6-enynes are known and have, in the past, been accomplished either thermally [7] or under transition-metal catalysis. In a series of papers, Trost [8] has scrutinized the chemistry of enynes in the presence of Pd compounds and recommended various catalyst systems for different modes of intramolecular cyclizations. In our studies, we employed catalytic amounts (5%) of Pd(OAc)₂ in the presence of Ph₃P (6%). Toluene was used as the solvent and slight heating (50–70°) was usually applied, since the cyclizations are somewhat too slow at room temperature (e.g. 30% conversion for $2c \rightarrow 3c$ after 48 h) to reach completion within practical times. In general, no systematic variation of the catalyst system was carried out. Selected examples do, however, indicate the superiority of Pd(OAc)₂ over Pd(dba)₂, PdCl₂(CH₃CN)₂, and

Table 1. Ene Reaction of Enynes 2 in the Presence of Pd(OAc)₂ (5%) and Ph₃P (6%)

Substrate	Conditions		Product	Yield [%]
	Temp. [°]	Time [h]		
2a	25	15	3a	60
2c	70	19	3c	66
2d	75	7	3d	69
2e	60	4	- 3e	68
2f	50	24	3f	55
2g	50	15	3g	81
2h	60	12	3h	36

Pd(Ph₃P)₄ for this particular purpose. Ph₃P gave the best results among the ligands employed, surpassing Bu₃P, 1,2-bis(diphenylphosphino)ethane, and tris[4-(dimethylamino)phenyl]phosphine, that were ineffective at room temperature.

The yields given in *Table 1* refer to isolated products from g-scale reactions. Actual conversion yields are surmised to be consistently higher. However, they could not be quantified by GC due to the thermal lability of the THP (tetrahydropyran-2-yl) protecting group. Some loss is certainly due to on-column decomposition during silica-gel chromatography.

The propargylic OH function clearly needs protection. Absence of the silyl group in 2a caused the intended ene reaction to produce, at room temperature, mainly untractible polar materials containing only small amounts (ca. 3%) of the desired carbocycle. The conversion $2a \rightarrow 3a$ was accompanied by a side reaction generating dimer 9 as an unspecified mixture of diastereoisomers. This type of cross coupling has previously been exploited in a synthetically useful manner [12]. It presumably arises via a sequence involving oxidative addition of Pd^{2+} into the C-H bond of the terminal acetylene, followed by PdH_2 insertion and reductive elimination [13]. Formation of 9 could not be suppressed by changes in the catalyst system. It was, however, kept below 20% by working in dilute (0.005M) solution at room temperature. Replacement of the silyl by an acetyl protecting group in 2a avoided the formation of 9, but the cyclization was less clean and produced a lower overall yield (40%) of the corresponding acetylated cyclopentane derivative.

Curiously, **2b** failed to cyclize under the examined set of conditions. The $Pd(OAc)_2/Ph_3P$ catalyst left the starting material unchanged even after 24 h at 60°. Prolonged heating to 80° produced a complex mixture containing only small amounts of the desired **3b**, that could not be obtained in pure form. The presence of Et_3N in the reaction mixture was ineffective at room temperature and destructive at $> 60^\circ$. Nor did the employment of $PdCl_2(CH_3CN)_2$ or $Pd(dba)_2$ produce any isolable cyclopentane **3b**.

Extension of this cyclization to substrates other than 1,6-enynes met with little success. Neither the 1,6-dienyl alcohol 8 nor the enynones 7a,b reacted in the desired manner. Nearly quantitative amounts of starting material were usually recovered from

reactions involving 8. Similar failures have been reported for *Lewis*-acid-mediated ene cyclizations of 1,6-dienes [14]. On the other hand, complex mixtures were obtained from 7a, while 7b cyclized to furan 10 in 40% yield. Although undesired in the present context, this side reaction has previously been reported as a useful synthetic method to convert monofunctional aroylacetylenes into 2-arylfurans [15].

The cyclopentane derivatives 3 were obtained as viscous oils of limited stability. Their formation involves the generation of an additional stereocenter at the five-membered ring and of two new C=C bonds, so that a number of possible diastereoisomers may be anticipated. However, with the exception of 3a and 3f, the cyclopentanes appeared to be chromatographically and spectroscopically homogenous, even though they are necessarily mixtures of diastereoisomers due to the stereogenic nature of the THP protecting group. Compound 3a was recognized as a ca. 2:1 mixture of isomers by the presence of a second set of signals in the ¹H-NMR spectrum. Amide 3f, on the other hand, was partially separable by column chromatography into two isomeric compounds, isolated in a 2:1 ratio. To avoid complications arising from the THP-group stereocenter, configurational assignments were not carried out at this stage, but rather inferred from the characterization of the corresponding deprotected alcohols 4 and 5.

Substrate	Conditions	Conditions			Yield [%]
	Solvent	Temp. [°]	Time [h]		
3a	EtOH	50	15	4a/5a	80
3c	EtOH	40	15	4e	61
3d	MeOH	25	12	4i	90
3e	EtOH	40	8	4 e	31
3f ¹)	MeOH	25	2	4f	87
$3f^2$)	MeOH	25	2	5f	71
3g	EtOH	25	48	6j	50
3h	MeOH	40	8	6k	42

Table 2. Removal of the Protecting Groups by 2n HCl

Removal of the protecting groups was effected by dilute HCl in EtOH or MeOH (Table 2). Careful examination (GC) of the crude reaction mixtures revealed the presence of a single compound following deprotection of 3c, 3d, and 3e. Two isomers 4a/5a, present in a ratio of ca. 1:2, were isolated after the deprotection of 3a. Amides 4f and 5f were obtained as pure isomers by individual deprotection of their already separated precursors 3f. Somewhat surprisingly, the free alcohols corresponding to 3g and 3h could not be obtained, nor could their transient presence be detected by reaction monitoring using TLC. Spontaneous ring closure with elimination of H_2O produced furans 6j and 6k.

¹H-NMR Spectroscopy clearly established **4a/5a** and **4f/5f** each as pair of geometrical isomers across the five-membered ring rather than across one of the newly formed C=C bonds. The latter were generated with complete stereocontrol in all cases. Coupling constants near 16 Hz established (*E*)-configuration for the disubstituted chain C=C bond. Intensive nuclear *Overhauser* enhancements (15–20%) between the quaternary

 $^{(2\}alpha, 5\alpha)$ -Isomer.

²) $(2\alpha,5\beta)$ -Isomer.

Me group and the exocyclic methylidene H-atom confirm exclusive (Z)-configuration in all substituted cases. Unfortunately, transannular through-space contacts between the substituents at the stereogenic centers seem to be weak and varying within the series, resulting in consistently small, if any, nuclear *Overhauser* effects (1D experiments). Precedents in the literature for configurational assignments are not available, since this particular five-membered-ring substitution pattern has previously been observed only as part of an annelated or bridged system.

After a number of failures, we succeeded to crystallize the bis(3,5-dinitrobenzoate) 11 of diol 4c. Crystals suitable for X-ray analysis were grown by slow evaporation of solvent from a MeCN solution.

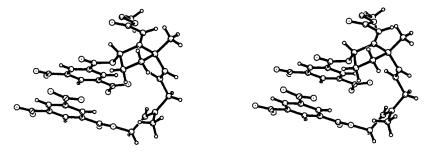


Fig. 1. Stereoscopic representation of compound 11

In the crystal, the C₆ diene chain folds back upon the five-membered ring so as to allow a partial stacking of the two electron-poor aromatic rings (Fig. 1). This obviously overrides an unfavorable endo ring-chain 1,4-interaction resulting from a near synplanar alignment of the disubstituted chain C=C bond with one of the ring C-C bonds. Inspection of the crystal packing reveals that the molecules point towards each other with their aromatic rings. This results in the formation of layered stacks that are in tight van-der-Waals contacts with each other. No solvent is incorporated in the crystal lattice. The pucker of the cyclopentane ring represents an essentially strainless conformation. The quaternary Me group occupies a pseudo-equatorial position and is disposed trans to the aroylated ring OH substituent. This finding prompted a careful ¹H-NMR investigation of the parent ester 4c. Complete assignments were carried out by H,H-COSY techniques in combination with 2D nuclear Overhauser (ROESY) spectroscopy (Fig. 2).

Again, no direct contacts between the stereogenic substituents are observed. However, one of the four ring CH₂ protons does share contacts to both the allylic ring H-atom

Fig. 2. Selected NOE's for diol 4c

and the ring Me group. This clearly requires the three substituents to point into the same direction and constitutes an independent confirmation of the relative configuration in question.

With this information in hand, the configurational characterization of the whole series became feasible. Amide 4f was recognized as the cis- $(2\alpha,5\alpha)$ -isomer based on a similar relay of strong NOE's. This effect is not observed in the trans- $(2\alpha, 5\beta)$ -isomer 5f. Instead, a weak (2%) NOE appeared between the ring Me and OH groups. The chemical shifts (in CDCl₃) of 4f and 5f are almost identical with two exceptions. The quaternary Me group resonates 0.09 ppm downfield in 5f and the resonances of the four ring CH₂ protons, nicely separated in 4f, are partially overlapping and remain unassigned. By analogous argumentation 4a, 4e, and 4i were assigned the relative cis-configuration. Additional evidence was gathered from weak but reproducible NOE's between the ring allylic H-atom and Me substituent (2%) of 4a and between the former and the H-atoms of the disubstituted C=C bond ($\sim 4\%$) of 5a. Because of the furan (6i, k) formation mentioned above, the NMR analysis was extended to include ketones 3g and 3h. A chemical-shift comparison of 3c-h was undertaken with reference to the now assigned pair of isomers $(2\alpha, 5\alpha)$ - and $(2\alpha, 5\beta)$ -3f. For the purpose of comparison, these compounds may be considered homogenous, since the THP-group stereocenter does not produce ¹H-NMR discernible epimers. Similar to the deprotected series, a characteristic downfield shift ($\Delta\delta \approx 0.15$ ppm) of the quaternary Me group together with an upfield shift $(\Delta\delta \approx 0.1 \text{ ppm})$ of its geminal olefinic H-atom is observed for the trans- $(2\alpha, 5\beta)$ -3f isomer. This is in contrast to all remaining compounds under comparison that have virtually identical shifts for these positions and are, therefore, assigned to be cis-isomers with respect to the ring Me and OH substituents.

All compounds shown represent racemic mixtures. Attempts to induce optical activity by the use of enantiomerically pure diphosphine ligands have been unavailing.

3. Discussion. – Based on the assignments above, we conclude that, except for 3a, the predominantly or exclusively formed cyclopentane retains a cis-1,3-relationship between the ring OH substituent and the C_6 diene chain. This selectivity can be reconciled with the 'hydropalladation mechanism' suggested by Trost [8] for this type of ene reaction. This mechanism postulates an olefin insertion into a previously formed Pd—C σ -bond as the actual ring-forming step. The stereoelectronic constraints for such a process require an (ideally) parallel alignment of the Pd—C with the reacting C=C bond. Either face of this C=C bond is accessible for such an arrangement (Fig. 3). Approach from the Si side (A) enforces considerable proximity between the silyloxy and the COR substituents. This interaction is completely absent in the alternative conformation (B). Instead, a synperi-

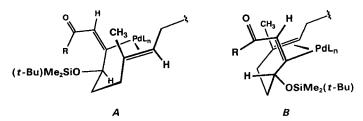


Fig. 3. Hypothetical reaction intermediates

clinal relationship between the silylated O-atom and the Pd-atom (including phosphine ligands) occurs. It appears to be sterically less demanding assuming olefin complexation by Pd within a square-planar coordination geometry. In the context of this model, the observed stereochemistry of the ene reaction is rationalized as the preferred formation of the sterically less encumbered intermediate (B). The lack of selectivity in the formation of 3a is readily attributed to the absence of the decisive acetylenic substituent. However, no explanation is provided for the unselective cyclization of 2f.

In both A and B, the O substituent occupies a pseudoequatorial position. An equivalent treatment of a set of analogous intermediates with a different chain pucker, resulting in a pseudoaxially disposed O-atom, leads to the same conclusion. Rationalization of the observed stereochemistry is not obvious within Trost's alternative 'cyclopalladation mechanism' [8], that entails C-C bond formation from a π -Pd-enyne complex to yield a fused five-membered palladocycle containing σ -bonded Pd⁴⁺. Considerable uncertainties as to the exact geometry of the enyne complexes virtually preclude an evaluation of their relevant conformations. Nor does inspection of the palladocycles reveal any distinguishing features. Even though both the hydropalladation and the cyclopalladation pathway represent hypothetical mechanisms, they did provide useful guidance for an explanation of our experimental results.

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Experimental Part

We gratefully acknowledge the skilful technical assistance of Christine Abrecht

- 1. General. Commercially available reagents were used without further purification. THF was distilled under Ar from Na with benzophenone ketyl as indicator. Toluene was distilled from CaH₂. Reactions were monitored by TLC on 10×2.5 cm precoated silica 60F-254 plates (E. Merck & Co.). Chromatograms were developed in appropriate hexane/AcOEt mixtures and spots visualized with 4-methoxybenzaldehyde (4% in EtOH/H₂SO₄) stain at 150°. Column chromatography (silica 60, 0.040-0.063 mm, E. Merck & Co.) refers to the procedure of Still et al. [16]. IR: Nicolet 170-SX FT-IR spectrometer; samples measured as thin films unless otherwise indicated; bands given in cm⁻¹. ¹H-NMR: Bruker AC-250 or Bruker AM-400; samples measured in CDCl₃ (unless otherwise indicated) with TMS as internal standard; chemical shifts (δ) given in ppm, coupling constants (J) in Hz. EI-MS: MS9, updated with a ZAB console, VG, Aldrincham, UK; ionization energy 70 eV, ion source at 250°, m/z in % of the base peak. Abbreviations used for frequently observed mass fragments: DHP = dihydropyran, THP = tetrahydropyran-2-yl, THPOH = tetrahydropyran-2-ol.
- 2. Cyclization Substrates. 2.1. (RS,E,E)-6,10-Dimethyl-12-{[(RS)-tetrahydro-2H-pyran-2-yl]oxy}dodeca-6,10-dien-1-yn-3-ol. A 1.6M soln. of BuLi in hexane (87.2 ml, 140 mmol) was added slowly to a soln. of acetylene (3.9 g, 150 mmol) in THF (70 ml) at -75° . After 15 min, a soln. of (E,E)-4,8-dimethyl-10-{[(RS)-tetrahydro-2H-pyran-2-yl]oxy}deca-4,8-dienal (1.13.3 g, 47.5 mmol) in THF (60 ml) was added dropwise. The mixture was left at -75° for 1 h, warmed to -30° , and kept at that temp. for another h. After purging with N₂, it was poured on ice-cold brine (ca. 1 l) and extracted three times with Et₂O. The combined extracts were washed with brine, dried (Na₂SO₄), filtered, and concentrated. FC on silica (500 g; hexane/AcOEt 4:1): 13.1 g (90%) of a colorless oil. IR: 3423m, 3306m, 2942s, 2854s, 2112w, 1666w, 1442m, 1384m, 1262m, 1200m, 1116s, 1074s, 1022s, 904m, 868m, 811m. 1 H-NMR (250 MHz): 5.34, 5.15 (2 br. t, $J \approx 7$, 2 C=CH); 4.64 ($\sim t$, OCHO); 4.33 (m, CHOH); 4.27, 4.03 (2 br. t)

- J = 12, 7, C=CHC H_2 O); 3.90, 3.52 (2m, pyran CH₂O); 2.45 (d, J = 2, C=CH); 2.27–2.02 (m, 3 CH₂C=C); 1.90–1.46 (m, 4 CH₂); 1.66, 1.61 (2 s, 2 CH₃). MS: 222 (1, [M DHP]⁺), 204 (1, [M THPOH]⁺), 188 (1), 186 (1), 171 (2), 161 (1), 145 (2), 137 (4), 119 (8), 109 (12), 91 (10), 85 (100).
- 2.2. (RS,E,E)-3- $\{f(\text{tert-}Butyl)\ dimethylsilyl]$ oxy $\}$ -6,10-dimethyl-12- $\{f(RS)$ -tetrahydro-2H-pyran-2-yl- $oxy\}$ dodeca-6,10-dim-l-yne (2a). A soln. of the preceding alcohol (15.3 g, 50 mmol) in CH_2Cl_2 (150 ml) was treated, at r.t., sequentially with (t-Bu)Me $_2$ SiCl (9 g, 60 mmol) and DBU (9.12 g, 60 mmol). After 2 h, most of the solvent was removed and the residue diluted with hexane and washed exhaustively with H_2O . The hexane layer was dried (MgSO₄), filtered, and concentrated to a slightly yellow oil. FC on silica (500 g, hexane/AcOEt 95:5): 18.3 g (87%) of 2a as a colorless oil. IR: 3310m, 2932s, 2856s, 2105s, 1668s, 1477m, 1385sm, 1358m, 1254s, 1202m, 1095s, 1023s, 897s, 788s. 1 H-NMR (250 MHz): 5.40, 5.17 (2 br. t, t) t) t) t) t0 t0. t0 t0 t0. t0 t0 t0 t1, t0 t1, t1, t2, t3, t3, t3, t3, t4, t5, t5, t6 t6, t7, t7, t7, t8, t7, t8, t8, t8, t8, t9, t9,
- 2.3. (RS,E,E)-7,11-Dimethyl-13- $\{f(RS)$ -tetrahydro-2H-pyran-2-yl $\}$ oxy $\}$ trideca-7,11-dien-2-yn-4-ol. Prepared from propyne (18 g, 450 mmol) and 1 (28 g, 100 mmol) in the presence of BuLi (1.6M in hexane; 112.5 ml) as described in 2.1. Purification by silica-gel chromatography (1.2 kg; hexane/AcOEt 4:1) furnished 20.2 g (63%) of a colorless oil. IR: 3426m, 2942s, 2871s, 2235w, 1672w, 1442m, 1384m, 1265m, 1200m, 1148s, 1119s, 1023s, 905m, 867m, 812m. 1 H-NMR (250 MHz): 5.55, 5,35 (2 br. t, $J \approx 7$, 2 C=CH); 4.71 (m, CHOH); 4.64 ($\sim t$, OCHO); 4.24, 4.01 (2 br. dd, J = 12, 6.8, C=CHC H_2 O); 3.89, 3.52 (2m, pyran CH $_2$ O); 2.25-2.09 (m, 3 C=CCH $_2$); 1.98-1.49 (m, 4 CH $_2$); 1.87 (d, J = 2, CH $_3$ C=C); 1.74, 1.68 (2s, 2 CH $_3$). MS: 190 (0.5), 175 (2), 159 (2), 157 (2), 145 (3), 133 (2), 122 (8), 109 (6), 95 (12), 91 (10), 85 (100), 67 (42).
- 2.4. (RS,E,E)-10{[(tert-Butyl)dimethylsilyl]oxy}-3,7-dimethyl-1-{[(RS)-tetrahydro-2H-pyran-2-yl]oxy}-trideca-2,6-dien-11-yne (**2b**). By silylation of the preceding alcohol (2 g, 6.24 mmol) with (t-Bu)Me₂SiCl (1.12 g, 7.43 mmol) in the presence of DBU (1.13 g; 7.43 mmol) as described for **2a**. Chromatography on silica (100 g; hexane/AcOEt 95:5): 2.17 g (80%) of **2b** as a colorless oil. IR: 2931s, 2855s, 2230w, 1664w, 1465m, 1444m, 1360m, 1253s, 1200m, 1081s, 1024s, 837s, 777s. ¹H-NMR (250 MHz): 5.36, 5.13 (2 br. t, $J \approx 7$, 2 C=CH); 4.63 ($\sim t$, OCHO); 4.28 (m, CHOSi); 4.26, 4.04 (2 br. dd, J = 12, 7.5, C=CHCH₂O); 3.91, 3.52 (2m, pyran CH₂O); 2.19–2.0 (m, 3 C=CCH₂); 1.82 (d, J = 2, CH₃=C); 1.68, 1.61 (2s, 2 CH₃); 1.80–1.48 (m, 4 CH₂); 0.90 (s, t-Bu); 0.12, 0.09 (2s, 2 MeSi). MS: 349 (1, [M THP]⁺), 275 (2, [M THPOH and t-Bu]⁺), 265 (2), 207 (3), 183 (6), 159 (18), 133 (18), 105 (22), 97 (20), 85 (100), 75 (36).
- 2.5. Ethyl (RS,E,E)-4-Hydroxy-7,11-dimethyl-13-{[(RS)-tetrahydro-2H-pyran-2-yl]oxy}trideca-7,11-dien-2-ynoate. A soln. of ethyl propiolate (9.8 g, 100 mmol) in THF (200 ml) was cooled to -90° . BuLi (1.6M in hexane; 62.5 ml) was added dropwise, followed, after 30 min, by a soln. of 1 (28 g, 100 mmol) in THF (50 ml). After 1 h at -70° , the mixture was poured (without prior warming) on ice-cold brine and extracted three times with hexane. The combined extracts were washed with H_2O , dried (Na₂SO₄), filtered, and concentrated. Silica-gel chromatography (1 kg; hexane/AcOEt 4:1): 21.2 g (56%) of the title compound as a colorless oil. IR: 3408s, 2940s, 2855s, 2235m, 1712s, 1443m, 1385m, 1366m, 1248s, 1200m, 1117m, 1075s, 1022s, 904m, 866m, 812m, 751m. 1 H-NMR (250 MHz): 5.34, 5.16 (2 br. t, 2 C=CH); 4.65 (\sim t, OCHO); 4.44 (t, J = 6.5, CHOH); 4.33–4.14, 4.07–3.98 (2m, CO₂CH₂C-3, CCHCH₂O); 3.90, 3.53 (2m, pyran CH₂O); 2.30–2.01 (m, 3 C=CCH₂); 1.95–1.50 (m, 4 CH₂); 1.65, 1.60 (2s, 2 CH₃); 1.32 (t, J = 7, CO₂CH₂CH₃). MS: 293 (1, [M THP] $^+$), 277 (1, [M THPO] $^+$), 185 (4), 181 (6), 166 (4), 163 (5), 135 (6), 117 (5), 93 (8), 85 (100).
- 2.7. $8-\{[(\text{tert-Butyl})dimethylsilyl]oxy}\}octyl$ (RS,E,E)-4-Hydroxy-7,11-dimethyl-13- $\{[(\text{RS})\text{-tetrahydro-}2H\text{-pyran-}2\text{-yl}]oxy}\}$ trideca-7,11-dien-2-ynoate. A soln. of $8-\{[(\text{tert-butyl})dimethylsilyl]oxy}\}$ octyl propiolate (5 g, 16 mmol) in THF (70 ml) was treated, at -100° , with BuLi (1.6m in hexane, 10 ml), followed after 45 min by soln. of 1 (4 g, 14.3 mmol) in THF (10 ml). After 1 h at -100° , the soln. was warmed to -10° within another, then poured on ice-cold NH₄Cl soln. and extracted twice with hexane. The combined extracts were washed with H₂O, followed

by brine, dried (MgSO₄), filtered, and concentrated to a yellow oil (9.55 g). Chromatography on silica (500 g; hexane/AcOEt 95:5 \rightarrow 9:1) yielded 7.1 g (84%) of the title compound as a slightly yellow oil. IR: 3400*m*, 2931*s*, 2856*s*, 2235*w*, 1714*s*, 1465*w*, 1388*w*, 1249*s*, 1097*m*, 1023*m*, 836*m*. ¹H-NMR (250 MHz): 5.29, 5.10 (2 br. *t*, 2 C=CH); 4.59 (*m*, OCHO); 4.38 (*m*, CHOH); 4.28–4.16, 4.04–3.91 (2 *m*, C=CHC H_2 O); 4.11 (*t*, J = 6.5, CO₂CH₂); 3.90–3.78, 3.55–3.42 (2 *m*, pyran CH₂O); 3.55 (*t*, J = 6.5, CH₂OSi); 2.98 (*d*, J = 5, OH); 2.25–2.0 (*m*, 3 C=CCH₂); 1.64, 1.59 (2 *s*, 2 CH₃); 1.90–1.42 (*m*, 6 CH₂); 1.30 (*m*, 4 CH₂); 0.88 (*s*, *t*-Bu); 0.07 (*s*, 2 MeSi). MS: 491 (5, [*M* - THPO]⁺), 313 (6), 278 (4), 261 (42), 85 (100).

- 2.8. $8-\{f(\text{tert-}Butyl)dimethylsilyl]oxy\} octyl (RS,E,E)-4-\{f(\text{tert-}Butyl)dimethylsilyl]oxy}-7.11-dimethyl-13-\{f(RS)-tetrahydro-2H-pyran-2-yl]oxy\} trideca-7.11-dim-2-ynoate (2d). Silylation of the preceding hydroxy ester (7 g, 11.8 mmol) with <math>(t\text{-Bu})\text{Me}_2\text{SiCl}$ (2.3 g; 15.3 mmol) in the presence of DBU (2.33 g; 15.3 mmol) was carried out as described for 2a. Chromatography on silica (200 g; hexane/AcOEt 9:1): 7.3 g (87%) of 2d as a slightly yellow oil. IR: 2931s, 2857s, 2230w, 1717s, 1673w, 1468m, 1386m, 1360m, 1251s, 1098s, 1024s, 837s, 778s. $^1\text{H-NMR}$ (250 MHz): 5.38, 5.15 (2 br. t, $J \approx 7$, 2 C=CH); 4.63 ($\sim t$, OCHO); 4.43 (t, J = 6.8, CHOSi); 4.34, 4.03 (2 br. dd, J = 12, 6.8, C=CHC H_2 O); 4.16 (t, J = 6.7, CO $_2$ CH $_2$); 3.91, 3.53 (2m, pyran CH $_2$ O); 3.60 (t, J = 6.7, CH $_2$ OSi); 2.19–1.99 (m, 3 C=CCH $_2$); 1.89–1.45 (m, 6 CH $_2$); 1.68, 1.60 (2s, 2 CH $_3$); 1.32 (m, 4 CH $_2$); 0.91, 0.90 (2s, 2 t-Bu); 0.15, 0.11, 0.05 (3s, 4 MeSi). MS: 621 (0.5, [M THP] $^+$), 605 (0.5, [M THPO] $^+$), 547 (3, [M THPOH and t-Bu] $^+$), 479 (2), 415 (2), 310 (6), 287 (12), 213 (14), 185 (16), 159 (28), 85 (100). Anal. calc. for C $_4$ 0H $_7$ 4O $_6$ Si $_2$ (707.20): C 67.94, H 10.55; found: C 67.74, H 10.85.
- 2.9. Diethyl {(RS,E,E)-3-{[(tert-Butyl)dimethylsilyl]oxy}-6,10-dimethyl-12-{[(RS)-tetrahydro-2H-pyran-2-yl]oxy}dodeca-6,10-dien-1-ynyl} Phosphonate (2e). BuLi (1.6m in hexane; 11 ml) was added dropwise to a soln. of 2a (5 g, 11.9 mmol) in THF (40 ml) at -78° . Diethylchlorophosphate (2.42 g, 14 mmol) was introduced after 90 min and the resulting mixture kept at -78° for another h. After warming to -20° , it was poured on sat. NH₄Cl soln. and extracted twice with Et₂O. The combined extracts were washed with H₂O, dried (Na₂SO₄), and filtered. Solvent removal left a yellow oil, that was chromatographed on silica (400 g; hexane/AcOEt 8:2 \rightarrow 7:3): 4.66 g (70%) of 2e as a viscous oil. IR: 2932s, 2860m, 2195m, 1661w, 1470m, 1442m, 1263m, 1098m, 1026s, 976m, 836m, 776m. ¹H-NMR (250 MHz): 5.37, 5.14 (2 br. t, $J \approx 7$, 2 C=CH); 4.66 (\sim t, OCHO); 4.41 (dt, J = 3.4, 6.8, CHOSi); 4.31–3.96 (m, 3 CH₂O); 3.92, 3.51 (2m, pyran CH₂O); 2.20–2.0 (m, 3 C=CCH₂); 1.91–1.46 (m, 4 CH₂); 1.69, 1.60 (2s, 2 CH₃); 1.37 (t, J = 7.1; 2 POCH₂CH₃); 0.90 (s, t-Bu); 0.15, 0.11 (2s, 2 MeSi). MS: 499 (12, $[M (t-Bu)]^+)$, 471 (14, $[M THP]^+)$, 455 (8, $[M THPO]^+)$, 397 (24, $[499 THPOH]^+)$, 369 (34, $[397 C_2H_4]^+$), 323 (24, $[396 EtOH]^+)$, 306 (72), 230 (76), 185 (90), 149 (42), 117 (74), 85 (100). Anal. calc. for $C_{29}H_{53}O_6$ PSi (556.80): C 62.56, H 9.59; found: C 61.98, H 9.83.
- 2.10. (RS,E,E)-4-{ $f(\text{tert-}Butyl) dimethylsilyl] oxy}-N,N,7,11-tetramethyl-13-{<math>f(\text{RS})$ -tetrahydro-2H-pyran-2-ylJoxy}trideca-7,11-dien-2-ynamide (2f). Compound 2a (7.14 g, 17 mmol) was dissolved in THF (60 ml) and deprotonated at -78° with BuLi (1.6M in hexane; 14.8 ml). The resulting soln. was briefly warmed to -50° , recooled to -78° , and treated dropwise with dimethyl chlorocarbamate (2.56 g, 23.8 mmol). The mixture was allowed to reach 0° within 1 h, poured on ice-cold NH₄Cl soln. and extracted twice with hexane. The combined extracts were washed with H₂O, dried (MgSO₄), and concentrated to a yellow oil. It was chromatographed on silica (500 g; hexane/AcOEt 8:2 \rightarrow 7:3): 7.9 g (94%) of 2f as a pale yellow oil. IR: 2931s, 2856m, 2230w, 1644s, 1397m, 1257m, 1185m, 1094s, 1023s, 839s, 779m. ¹H-NMR (250 MHz): 5.37, 5.14 (2 br. t, $J \approx 7$, 2 C=CH); 4.63 (\sim t, OCHO); 4.47 (t, J = 6.3, CHOSi); 4.25, 4.02 (2 br. dd, J = 12, 6.8, C=CHCH₂O); 3.90, 3.51 (2m, pyran CH₂O); 3.20, 2.97 (2s, 2 NCH₃); 2.19–1.98 (m, 3 C=CCH₂); 1.87–1.45 (m, 4 CH₂); 1.68, 1.60 (2s, 2 CH₃); 0.90 (s, t-Bu); 0.13, 0.11 (2s, 2 MeSi). MS: 491 (2, M^+), 467 (2, $[M \text{CH}_3]^+$), 434 (6, $[M (t \text{Bu})]^+$), 406 (6, $[M \text{THP}]^+$), 389 (4, $[M \text{THPOH}]^+$), 322 (13), 258 (21), 240 (16), 198 (62), 190 (26), 185 (26), 165 (44), 159 (17), 124 (27), 85 (78), 72 (100).
- 2.11. (RS, E, E)-9-{[(tert-Butyl)dimethylsilyl]oxy}-12,16-dimethyl-18-{[(RS)-tetrahydro-2H-pyran-2-yl]-oxy}octadeca-12,16-dien-7-yn-6-one (**2g**). A soln. of **2a** (4.2 g, 10 mmol) in THF (30 ml) was treated dropwise with BuLi soln. (1.6м in hexane; 6.9 ml) at -78° . The mixture was transferred after 2 h via cannula into a soln. of caproic anhydride (2.35 g, 11 mmol) in Et₂O kept at -20° . The resulting soln. was allowed to reach r.t. within 1 h, poured into ice-cold NaHCO₃ soln., and extracted with hexane. The extracts were washed with H₂O, dried (Na₂SO₄), and concentrated to a yellowish oil (5.4 g). Chromatography on silica (500 g; hexane/AcOEt 97:3): 2.43 g (47%) **2g** as a colorless oil. IR: 2931s, 2857m, 2210w, 1679s, 1466m, 1360m, 1255m, 1097s, 1029s, 837s, 781m. ¹H-NMR (250 MHz): 5.37, 5.15 (2 br. t, $J \approx 7$, 2 C=CH); 4.62 ($\sim t$, OCHO); 4.46 (t, J = 6, CHOSi); 4.26, 4.02 (2 br. dd, J = 12, 7, C=CHCH₂O); 3.91, 3.51 (2m, pyran CH₂O); 2.54 (t, J = 7.3, COCH₂); 2.18–1.98 (m, 3 C=CCH₂); 1.88–1.46, 1.37–1.21 (2m, 7 CH₂); 1.68, 1.60 (2s, 2 CH₃); 0.91 (s, t-Bu); 0.90 (m, CH₃CH₂); 0.15, 0.11 (2s, 2 MeSi). MS: 359 (2.5, [M THPOH and $t \text{Bul}^+$), 285 (5), 267 (4), 225 (10), 185 (6), 159 (8), 119 (6), 105 (7), 99 (16), 85 (100).

- 2.12. (RS, 2E, 6E, 2IZ)-10- $\{f$ (tert-Butyl) dimethylsilyl $\}$ oxy $\}$ -3,7-dimethyl-1- $\{f$ (RS)-tetrahdydro-2H-pyran-2-yl $\}$ oxy $\}$ triaconta-2,6,2I-trien-1I-yn-13-one (2h). Prepared from 2h (5.4 g, 12.8 mmol), BuLi (1.6M in hexane; 9.6 ml), and oleic anhydride (8.4 g, 15.4 mmol) as described in 2.1I. The crude dark yellow oil was chromatographed on silica (500 g; hexane/AcOEt 97:3): 6.2 g (70%) 2h as a pale yellow oil. 1R: 2927s, 2855s, 2215m, 1679s, 1464m, 1359w, 1254m, 1200w, 1093s, 1023s, 839s, 779m. 1 H-NMR (250 MHz): 5.34 (m, 3 C=CH); 5.14 (5.14
- 2.13. (E,E)-6,10-Dimethyl-12- $\{[(RS)$ -tetrahydro-2H-pyran-2-yl]oxy $\}$ dodeca-6,10-dien-1-yn-3-one (7a). A soln. of the alcohol described in 2.1 (13.1 g, 42.7 mmol) in CH₂Cl₂ (50 ml) was added to a rapidly stirred suspension of MnO₂ (130 g) in CH₂Cl₂ (400 ml) at 0°. Stirring was continued for 4 h while warming to r.t. Solids were removed by filtration over compressed Na₂SO₄ and the concentrated filtrate was chromatographed on silica (500 g; hexane/AcOEt 4:1): 7.7 g (59%) of 7a as a colorless oil. IR: 3250m, 2941s, 2871s, 2090s, 1683s, 1443m, 1385m, 1353m, 1200m, 1115s, 1077m, 1022s, 905m, 869m, 813m. ¹H-NMR (250 MHz): 5.35, 5.15 (2 br. t, $J \approx 7$, 2 C=CH); 4.63 ($\sim t$, OCHO); 4.24, 4.02 (2 br. dd, J = 12, 7, C=CHCH₂O); 3.90, 3.51 (2m, pyran CH₂O); 3.23 (s, C=CH); 2.69 (t, J = 7.5, COCH₂); 2.35 (t, J = 7.5, COCH₂CH₂); 2.19–2.00 (t, C=CCH₂CH₂C=C); 1.95–1.45 (t, 3 pyran CH₂); 1.67, 1.61 (2s, 2 CH₃). MS: 135 (10, [C₉H₁₁O]⁺), 85 (100, [THP]⁺). Anal. calc. for C₁₉H₂₈O₃ (304.43): C 74.96, H 9.27; found: C 74.54, H 9.46.
- 2.14. (E,E)-7,11-Dimethyl-13- $\{f(RS)$ -tetrahydro-2H-pyran-2-yl]oxy $\}$ trideca-7,11-dien-2-yn-4-one (**7b**). The alcohol described in 2.3 (1.6 g, 5 mmol) was oxidized with MnO₂ (16 g) as in the previous example. Silica-gel chromatography (100 g; hexane/AcOEt 95:5) afforded 0.96 g (60%) of **7b** as a colorless oil. IR: 2941s, 2869s, 2220s, 1674s, 1441m, 1387m, 1353m, 1283w, 1252m, 1163m, 1118m, 1076m, 1023s, 905m, 874m, 816m. ¹H-NMR (250 MHz): 5.37, 5.13 (2 br. t, $J \approx 7$, 2 C=CH); 4.63 ($\sim t$, OCHO); 4.24, 4.02 (2 br. dd, J = 12, T, C=CHC H_2 O); 3.91, 3.50 (2m, pyran CH₂O); 2.62 (t, t = 8.3, COCH₂); 2.32 (br. t, COCH₂CH₂C); 2.16–1.98 (t, C=CCH₂CH₂C=C); 2.03 (t, CH₃C=C); 1.92–1.46 (t, 3 pyran CH₂); 1.67, 1.60 (2t, 2 CH₃). MS: 233 (3, t = THPt = 1, 134 (12), 123 (20), 107 (62), 85 (100), 67 (80).
- Ethyl (RS,E,E,E)-4-{[(tert-Butyl)dimethylsilyl]oxy}-7,11-dimethyl-13-{[(RS)-tetrahydro-2Hpyran-2-yl]oxy \text{trideca-2,7,11-trienoate (8). A soln. of 1 (41 g, 146 mmol) and ethyl 3-nitropropionate (21.5 g; 146 mmol) in dry DMF was treated dropwise with DBU (22.5 g, 148 mmol). The resulting soln. was stirred overnight at r.t., then poured on ice-cold 2N HCl (~ 1.5 l), and extracted three times with a 1:1 mixture of hexane and Et₂O. The extracts were washed with H₂O followed by brine, dried (Na₂SO₄), and concentrated. Chromatography on silica (1.5 kg; hexane/AcOEt 4:1 containing 0.1% TEA): 27.3 g (50%) of a yellowish oil. A portion of this material (5.93 g, 15.6 mmol) was dissolved in pyridine together with (t-Bu)Me₂SiCl (2.7 g, 17.9 mmol). DBU (4.6 g, 30 mmol) had to be added to initiate the reaction. After 3 h at r.t., the mixture was poured on ice and extracted twice with hexane. The extracts were washed with H₂O, dried (Na₂SO₄), concentrated, and purified on silica (300 g; hexane/AcOEt 100:0→95:5): 7.41 g (96%) of 8 as a slightly yellow oil. IR: 2939s, 2958s, 1722s, 1659m, 1452m, 1366m, 1258s, 1196s, 1162s, 1118s, 1077m, 1025s, 977m, 837s, 776m. ¹H-NMR (250 MHz): 6.92 (dd, J = 15.5, 4.8, CH=CCO₂); $5.96 (dd, J = 15.5, 1.7, C = CHCO_2); 5.36, 5.11 (2 \text{ br. } t, 2 C = CHCH_2); 4.63 (\sim t, OCHO); 4.35 - 4.15 (m, CHOSi, CHOSi,$ CO_2CH_2 ; 4.20, 4.03 (2 br. dd, J = 12, 7, C=CHC H_2O); 3.90, 3.52 (2m, pyran CH_2O); 2.17–1.95 (m, 3 C=CHC H_2O), 1.90-1.45 (m, 4 CH₂); 1.68, 1.58 (2s, 2 CH₃); 1.30 (t, J = 7.3, CO₂CH₂CH₃); 0.91 (s, t-Bu); 0.05, 0.03 (2s, 2 MeSi). MS: 409 (1, [M - THP]⁺), 335 (3), 243 (9), 187 (8), 159 (10), 147 (16), 119 (32), 105 (8), 93 (14), 85 (100). Anal. calc. for C₂₈H₅₀O₅Si (494.79): C 67.97, H 10.19; found: C 68.00, H 10.30.
- 3. Ene-Reactions. 3.1. General Procedure. Solid Pd(OAc)₂ (5 mol-%) together with Ph₃P (6 mol-%) were placed in a dried flask and put under Ar by three cycles of evacuation. Dry toluene (5 ml) was injected to dissolve the catalyst components. A soln. of the substrate 2, dissolved in toluene, was introduced via cannula and the resulting mixture heated with stirring as indicated in Table 1. A positive pressure of Ar was maintained during that time, and samples for reaction monitoring were withdrawn with an Ar flushed syringe. In all cases, the product was found to move slightly faster on TLC in hexane/AcOEt mixtures. The completion of the reaction was usually accompanied by the precipitation of Pd black. After cooling, it was filtered off on Celite. The filtrate was concentrated and the residual crude material purified by FC on silica as indicated below. On-column decomposition clearly is a problem, as small-scale chromatography gave consistently better recoveries. All compounds 3 were obtained as moderately stable, viscous, colorless-to-pale-yellow oils that were freed of residual solvents by evacuation to 0.001 mbar for 48 h.
- 3.2. (RS)-2-{{(E,E)-6-{3-{{(tert-Butyl)dimethylsilyl]oxy}-1-methyl-2-methylidenecyclopentyl}-3-methyl-hexa-2,5-dienyl}oxy}tetrahydro-2H-pyran (3a). From 2a (1.05 g, 2.5 mmol). Chromatography on silica (150 g,

hexane/AcOEt 98:2 \rightarrow 95:5) separated 0.19 g of dimer 9 from 0.63 g of the less polar 3a. IR: 3070w, 2955s, 2857s, 1668w, 1467m, 1359m, 1254s, 1200m, 1118s, 1024s, 973m, 879m, 836s, 776s. ¹H-NMR (250 MHz, major diastereoisomer): 5.46 (d, J = 15.3, CH=CHCH₂); 5.34 (dt, J = 15.3, 6.5, CH=CHCH₂); \sim 5.4 (partially obscured t, C=CHCH₂O); 5.10, 4.86 (2m, C=CH₂); 4.63 (\sim t, OCHO); 4.45 (m, CHOSi); 4.24, 4.02 (2 br. dd, J = 12, 7, C=CHCH₂O); 3.90, 3.53 (2m, pyran CH₂O); 2.70 (br. d, J = 6.3, C=CCH₂C=C); 1.98–1.87 (m, 5 CH₂); 1.64 (s, CH₃); 1.19 (s, CH₃); 0.92 (s, t-Bu); 0.09, 0.07 (2s, 2 MeSi). MS: 405 (2, [M - CH₃]⁺), 261 (15, [M - THPOH and t-Bu]⁺), 193 (10), 187 (14), 159 (60), 131 (14), 93 (14), 85 (100), 75 (60).

- 3.3. Ethyl $\{(Z)\text{-}5\text{-}\{(\text{tert-butyl})\text{dimethylsilyl}]\text{ox}\}\text{-}2\alpha\text{-}\{(E,E)\text{-}4\text{-methyl-}6\text{-}\{(RS)\text{-}tetrahydro\text{-}2\text{H-}pyran-2\text{-}yl]\text{ox}\}\text{hexa-}1\text{-}4\text{-}dienyl}\}\text{-}2\text{-}methylcyclopentylidene}\}\text{acetate}$ (3c). From 2c (1.05 g, 2.1 mmol) to yield, after chromatography (200 g; hexane/AcOEt 97:3), 0.7 g of 3c as a colorless oil. IR: 2955s, 2855s, 1720s, 1661w, 1466m, 1371m, 1334w, 1299w, 1253m, 1199s, 1146m, 1117m, 1028m, 840s, 776m. $^{1}\text{H-}\text{NMR}$ (400 MHz): 5.58 (s, C=CHCO₂); 5.53 (d, J=15.5, CH=CHCH₂); 5.44 (m, CHOSi); 5.41 (dt, J=15.5, 7, CH=CHCH₂); ~5.38 (partially obscured t, C=CHCH₂O); 4.62 (\sim t, OCHO); 4.23, 4.01 (2m, C=CHCH₂O); 4.14 (2dq, CO₂CH₂); 3.89, 3.52 (2m, pyran CH₂O); 2.71 (br. d, J=6.8, C=CCH₂C=C); 2.06–1.97, 1.88–1.42 (2m, 5 CH₂); 1.64 (s, CH₃); 1.27 (t, J=7, CO₂CH₂CH₃); 1.13 (s, CH₃); 0.82 (s, t-Bu); 0.11, 0.04 (2s, 2 MeSi). MS: 447 (1, [M-EtO]⁺), 435 (2, [M-t-Bu]⁺), 407 (1, [M-THP]⁺), 391 (2, [M-THPO]⁺), 333 (100, [M-THPOH and t-Bu]⁺), 259 (38, [333 CO and EtOH]⁺), 231 (12), 198 (46), 185 (24), 171 (30), 159 (65), 85 (74), 75 (96). Anal. calc. for C₂₈H₄₈O₅Si (492.77): C 68.25, H 9.82; found: C 68.58, H 10.35.
- 3.4. $8-\{[(\text{tert-}Butyl)dimethylsilyl]oxy\} \text{octyl}$ $\{(Z)-5\alpha-\{[(\text{tert-}Butyl)dimethylsilyl]oxy}\}-2-\text{methyl-}2\alpha-\{(E,E)-4-\text{methyl-}6-\{[(RS)-\text{tetra}hydro-2H-pyran-2-yl]oxy}\} \text{hexa-}1,4-\text{dienyl}\} \text{cyclopentylidene}\} \text{acetate}$ (3d). From 2d (7.1 g, 10 mmol). The crude product was chromatographed on silica (250 g; hexane/AcOEt 98:2): 4.88 g (69%) of 3d as a colorless oil. IR: 2931, 2858s, 1721s, 1660w, 1467m, 1387m, 1334m, 1298w, 1253s, 1198s, 1146s, 1099s, 1063s, 1024s, 836s, 776s. 1 H-NMR (250 MHz): 5.59 (d, J = 1.1, C=CHCO₂); 5.55 (d, J = 15.5, CH=CHCH₂); 5.42 (dt, J = 15.5, 6.5, CH=CHCH₂); ~5.45 (partially obscured m, C=CHCH₂O, CHOSi); 4.65 (\sim t, OCHO); 4.26, 4.01 (2 br. dd, J = 12, 7, C=CHCH₂O); 4.08 (t, J = 6.7, CO₂CH₂); 3.91, 3.54 (t, pyran CH₂O); 3.60 (t, t = 6.5, CH₂OSi); 2.72 (br. t, t = 7, C=CCH₂C=C); 2.10-1.97, 1.92-1.42, 1.42-1.23 (3t, 11 CH₂); 1.65, 1.13 (2t, 2 CH₃); 0.90, 0.83 (2t, 2 t -Bu); 0.12 (t, MeSi); 0.05 (t, 3 MeSi). MS: 691 (0.5, [t CH₃]⁺), 649 (2, [t (t -Bu)]⁺), 605 (2, [t THPO]⁺), 547 (68, [649 THPOH]⁺), 473 (6), 415 (10), 345 (8), 305 (12), 231 (60), 213 (44), 185 (24), 171 (48), 159 (78), 85 (80), 75 (100). Anal. calc. for C₄₀H₇₄O₆Si₂(707.20): C 67.94, H 10.55; found: C 68.20, H 10.52.
- 3.5. Diethyl $\{\{(Z)-5\alpha-\{[(\text{tert-}Butyl)dimethylsilyl]oxy\}-2\alpha-\{(E,E)-4-methyl-6-\{[(RS)-tetrahydro-2H-pyran-2-yl]oxy\}\}$ hexa-1,4-dienyl $\}$ -2-methylcyclopentylidene $\}$ methyl $\}$ Phosphonate (3e). From 2e (5.9 g, 10.6 mmol). Purification on silica (500 g; hexane/AcOEt 9:1 \rightarrow 8:2) furnished 4 g of 3e as a sligthly yellow oil. IR: 2955s, 2855s, 1640w, 1452w, 1388w, 1352w, 1250s, 1119m, 1055s, 1028s, 965s, 838m, 779m. 1 H-NMR (250 MHz): 5.56 (d, J = 16, CH=CHCH₂); 5.44 (partially obscured dt, CH=CHCH₂); 5.45–5.30 (m, CHOSi, C=CHCH₂O, C=CHP); 4.63 (m, CHCHO); 4.26 and m 4.01 (2 br. dd, d) = 12, 7, d=CHCH₂O); 4.13–3.95 (d), 2 POCH₂); 3.91, 3.51 (d), pyran d=0, 2.72 (br. d, d) = 7, d=CCH₂C=C); 2.14–2.0, 1.91–1.44 (d), 5 CH₂); 1.63, 1.14 (d)s, 2 CH₃); 1.30, 1.27 (d)t, d=7, 2 POCH₂CH₃); 0.87 (d)s, d=1, d1, d
- 3.6. $\{(Z)-5-\{f(\text{tert-}Butyl)dimethylsilyl\}oxy\}-2\alpha-\{(E,E)-4-methyl-6-\{f(RS)-tetrahydro-2H-pyran-2-yl\}-oxy\}hexa-1,4-dienyl\}-2-methylcyclopentylidene\}-N, N-dimethylacetamide (3f). From 2f (7.6 g, 15.5 mmol). Chromatography of the crude material on silica (500 g; hexane/AcOEt 8:2) furnished 0.35 g of the less polar <math>(2\alpha,5\beta)$ -isomer and 2.0 g of the more polar $(2\alpha,5\alpha)$ -isomer together with 1.8 g of mixed fractions. The latter were rechromatographed (300 g silica, same eluent) to yield another 0.42 g of $(2\alpha,5\beta)$ -isomer and 0.46 g $(2\alpha,5\alpha)$ -isomer together with 0.86 g of mixed fractions.
- $(2\alpha,5\beta)$ -Isomer: IR: 2931s, 2855s, 1639s, 1467m, 1392m, 1253m, 1130s, 1056s, 1029s, 836s, 776m. ¹H-NMR (250 MHz): 5.74 (d, J=1.5, C=CHCON); 5.49-5.32 (m, 3 C=CH); 5.11 (m, CHOSi); 4.61 ($\sim t$, OCHO); 4.25, 4.01 (2 br. dd, J=12, 7, C=CHC H_2 O); 3.88, 3.50 (2m, pyran CH $_2$ O); 2.93, 2.91 (2s, 2 NCH $_3$); 2.69 (br. d, C=CCH $_2$ C=C); 1.94-1.45 (m, 5 CH $_2$); 1.62, 1.27 (2s, 2 CH $_3$); 0.83 (s, t-Bu); 0.05, 0.01 (2s, 2 MeSi). MS: 491 (0.5 M^+), 476 (3, $[M-CH_3]^+$), 434 (100, [M-t-Bu] $^+$), 406 (4, $[M-THP]^+$), 389 (4, $[M-THPOH]^+$), 350 (8, [434 DHP] $^+$), 332 (4, [434 THPOH] $^+$), 258 (11), 198 (96). Anal. calc. for $C_{28}H_{49}NO_4Si$: (491.79): C 68.38, H 10.04, N 2.85; found: C 68.54, H 10.35, N 2.84.

 $(2\alpha,5\alpha)$ -Isomer: IR: 2932s, 2855s, 1639s, 1465m, 1395m, 1253m, 1118s, 1023s, 837s, 777m. ¹H-NMR (250 MHz): 5.77 (d, J = 1.8, C=CHCON); 5.53 (d, J = 15.5, CH=CHCH₂); 5.42 (partially obscured dt, J = 15.5, 5.3, CH=CHCH₂); 5.37 (br. t, C=CHCH₂O); 5.13 (m, CHOSi); 4.61 ($\sim t$, OCHO); 4.24, 3.99 (2 br. dd, J = 12, 7, C=CCH₂O); 3.88, 3.50 (2m, pyran CH₂O); 2.93, 2.91 (2s, 2 NCH₃); 2.71 (br. d, J = 6.3, C=CCH₂C=C); 1.93–1.37

- $(m, 5 \text{ CH}_2); 1.63, 1.13 (2s, 2 \text{ CH}_3); 0.82 (s, t-\text{Bu}); 0.04, 0.01 (2s, 2 \text{ MeSi}). \text{ MS: same as for } (2\alpha, 5\beta)-\text{isomer. Anal. calc. for } C_{28}H_{49}NO_4Si (491.79): C 68.38, H 10.04, N 2.85; found: C 68.39, H 10.40, N 2.85.$
- 3.8. (Z)-1- $\{(Z)$ -5 α - $\{(\text{tert-}Butyl)$ dimethylsilyl]oxy $\}$ -2 α - $\{(E,E)$ -4-methyl-6- $\{(RS)$ -tetrahydro-2H-pyran-2-yl]oxy $\}$ hexa-1,4-dienyl $\}$ -2-methylcyclopentylidene $\}$ nonadec-10-en-2-one (3h). From 2h (6.1 g, 8.9 mmol). Chromatography on silica (400 g; hexane/AcOEt 98:2) yielded 2.2 g of 3h as a pale yellow syrup. IR: 2926s, 2854s, 1694m, 1633m, 1462m, 1365w, 1252m, 1116s, 1056s, 1025s, 836s, 777m. 1 H-NMR (250 MHz): 5.92 (d, J = 1.4, C=CHCO); 5.58–5.27 (m, 5 olefin. H, CHOSi); 4.61 (\sim t, OCHO); 4.26, 4.02 (2 br. dd, CH=CHCd2O); 3.90, 3.53 (2m, pyran CH₂O); 2.73 (br. d, J = 7, C=CCH₂C=C); 2.42 (d, J = 7, COCd2CH₂C, 2.09–1.17 (m, 18 CH₂C); 1.65, 1.12 (2g, 2 CH₃C, 3, 88 (m, CH₃CH₂C, 0.82 (g, g-Bu); 0.11, 0.04 (2g, 2 MeSi). MS: 684 (0.5, g-1, 669 (10, g-CH₃C+3, 627 (3, g-1, 649) (4, 59) (4, g-1, 649) (5, 59) (4, g-1, 659 (42), 85 (92), 72 (100).
- 4. Removal of the Protecting Groups. 4.1. General Procedure. The protected cyclopentane 3 was dissolved in alcohol ($\sim 1 \text{ mmol/}10 \text{ ml}$). 2 N HCl (0.15 equiv.) was added dropwise and the reaction mixture treated as indicated in Table 2. After cooling, it was poured into ice-cold brine containing 10% NaHCO₃ soln. and extracted with Et₂O. The combined extracts were washed with H₂O, followed by brine, and dried (Na₂SO₄). The crude material was purified by FC on silica using the eluents indicated below. Appropriate fractions were concentrated, filtered, and finally pumped dry at 0.001 mbar for 48 h. The pure products 4, 5, or 6 were obtained as viscous oils.
- 4.2. $3\alpha-[(E,E)-6-Hydroxy-4-methylhexa-1,4-dienyl]-3-methyl-2-methylidenecyclopentan-1\alpha-ol (4a)$ and $3\alpha-[(E,E)-6-Hydroxy-4-methylhexa-1,4-dienyl]-3-methyl-2-methylidenecyclopentan-1<math>\beta$ -ol (5a). From 3a (0.62 g, 1.47 mmol). Chromatography of the crude product (100 g silica; hexane/AcOEt 3:1) furnished 104 mg of 5a and 13 mg of the slightly more polar 4a together with 145 mg of mixed fractions.

Data of **5a**: IR: 3334s, 3076w, 3021w, 2959s, 2868s, 1661w, 1651w, 1443m, 1382w, 1326w, 1074m, 1004s, 973s, 905m. 1 H-NMR (400 MHz, (D₆)DMSO): 5.45 (d, J = 15.5, CH=CHCH₂); 5.27 (dt, J = 15.5, 7, CH=CHCH₂); 5.25 (partially obscured t, C=CHCH₂O); 5.08, 4.80 (2m, C=CH₂); 4.91 (d, J = 5.6, CHOH); 4.47 (t, J = 5.2, CH₂OH); 4.27 (m, CHOH); 3.92 (\sim t, $J \approx$ 6, CH₂O); 2.62 (br. d, J = 7.2, C=CCH₂C=C); 1.89–1.80, 1.60–1.40 (2m, 2 ring CH₂); 1.53, 1.14 (2s, 2 CH₃). MS: 207 (14, $[M - \text{CH}_3]^+$), 204 (3, $[M - \text{H}_2\text{O}]^+$), 189 (10, $[207 - \text{H}_2\text{O}]^+$), 171 (8), 145 (15), 134 (22), 119 (80), 105 (53), 95 (48), 93 (100), 91 (76).

Data of **4a**: IR: 3328s, 3075w, 3024w, 2959s, 2928s, 2866m, 1670w, 1655w, 1448m, 1382w, 1327w, 1077m, 999s, 974s, 899m. ¹H-NMR (400 MHz, (D₆)DMSO): 5.48 (d, J = 15.6, CH=CHCH₂); 5.35–5.25 (m, CH=CHCH₂, C=CHCH₂O); 5.10, 4.79 (2m, C=CH₂); 4.90 (d, J = 5.3, CHOH); 4.48 (t, J = 5.2, CH₂OH); 4.32 (m, CHOH); 3.93 ($\sim t$, $J \approx 5.4$, CH₂O); 2.64 (br. d, J = 6.8, C=CCH₂C=C); 1.87–1.78, 1.75–1.67 (2m, ring CH₂); 1.54, 1.10 (2s, 2 CH₃); 1.48–1.33 (m, 2 ring CH₂). MS: same as for **5a**.

- 4.3. Ethyl {(Z)-5 α -Hydroxy-2 α -[(E,E)-6-hydroxy-4-methylhexa-1,4-dienyl]-2-methylcyclopentylidene}} acetate (4c). From 3c (0.63 g, 1.28 mmol). Chromatography (60 g silica; hexane/AcOEt 3:1): 0.23 g, of 4c as a colorless oil. IR: 3427m, 2963m, 2869m, 1691s, 1645m, 1446w, 1371m, 1344m, 1301m, 1209s, 1154w, 1088w, 1027m, 973m. ¹H-NMR (400 MHz): 5.74 (d, J = 1.8, C=CHCO₂); 5.46 (d, J = 15.6, CH=CHCH₂); 5.42 (partially obscured t, C=CHCH₂O); 5.39 (dt, J = 15.6, 6.4, CH=CHCH₂); 4.94 (m, CHOH); 4.84 (d, J = 2.3, CHOH); 4.21 (d, J = 7.1, CO₂CH₂); 4.15 (d, J = 6.9, CH₂OH); 2.71 (d, J = 6.3, C=CCH₂C=C); 2.10 (ddt, J = 12.5, 4.6, 7.2, H_{β} -C(4)); 1.90 (ddd, J = 11.8, 6.6, 4.6, H_{α} -C(3)); 1.81 (ddt, J = 12.5, 9.5, 6.6, H_{α} -C(4)); 1.64 (dd, J = 11.8, 9.5, 7.2, H_{β} -C(3)); 1.32 (t, J = 7.1, CO₂CH₂CH₃); 1.18 (s, CH₃). MS: 276 (4, [M H_2 O]⁺), 230 (34, [276 EtOH]⁺), 179 (42), 134 (54), 119 (70), 105 (56), 95 (100). Anal. calc. for C₁₇H₂₆O₄ (294.39): C 69.36, H 8.90; found: C 69.11, H 9.07.
- 4.4. Diethyl $\{\{(Z)-5\alpha-Hydroxy-2\alpha-\{(E,E)-6-hydroxy-4-methylhexa-1,4-dienyl\}-2-methylcyclopentylidene\}-methyl\}$ Phosphonate (4e). From 3e (0.41 g, 0.74 mmol). Chromatography on silica (20 g; hexane/AcOEt 1:1): 82 mg of 4e as a pale yellow oil. IR: 3392s, 2935s, 2869m, 1671w, 1629m, 1445m, 1392m, 1292w, 1214s, 1164m, 1025s,

- 972*s*, 827*m*. ¹H-NMR (250 MHz): 5.65 (br. *s*, CHO*H*); 5.51–5.33 (*m*, 4 olef. H); 4.87 (*m*, CHOH); 4.15 (*d*, J = 7.5, CH₂OH); 4.08 (*m*, 2 POCH₂); 2.71 (br. *d*, J = 5.8, C=CCH₂C=C); 2.16–2.02, 1.95–1.66, 1.55–1.44 (3*m*, 2 ring CH₂); 1.64 (*s*, CH₃); 1.34, 1.33 (2*t*, J = 7.3, 2 POCH₂CH₃); 1.17 (*s*, CH₃). MS: 341 (5, $[M OH]^+$), 340 (2, $[M H_2O]^+$), 325 (4, $[340 CH_3]^+$), 311 (8, $[340 C_2H_3]^+$), 297 (3), 295 (3), 283 (6), 273 (5), 261 (76, $[325 H_2O]^+$) and EtOH]⁺), 243 (24), 230 (25), 145 (25), 131 (48), 117 (48), 105 (64), 91 (100).
- 4.5. {(*Z*)-5α-Hydroxy-2α-[(E,E)-6-hydroxy-4-methylhexa-1,4-dienyl]-2-methylcyclopentylidene}-N,N-dimethylacetamide (4f). From $(2\alpha,5\alpha)$ -3f (1.78 g, 3.62 mmol). Chromatography (100 g of silica; hexane/AcOEt 1:1→1:4): 0.93 g of 4f as a pale yellow oil. IR: 3404s, 2942s, 2868m, 1651m, 1601s, 1495w, 1449w, 1397m, 1309w, 1260w, 1172w, 1144w, 1085w, 1057w, 1023m, 974m. ¹H-NMR (400 MHz, (D₆)DMSO): 6,06 (*d*, *J* = 1.7, C=CHCON); 5.78 (*d*, *J* = 16, CHOH); 5.50 (*d*, *J* = 15.6, CH=CHCH₂); 5.36 (*dt*, *J* = 15.6, 6.8, CH=CHCH₂); 5.29 (br. *t*, *J* ≈ 7, C=CHCH₂O); 4.67 (br. *t*, CHOH); 4.48 (*t*, *J* = 5.2, CH₂OH); 3.93 (~ *t*, *J* ≈ 5.6, CH₂OH); 3.03, 2.89 (2s, 2 NCH₃); 2.66 (br. *d*, *J* = 6.8, C=CCH₂C=C); 1.95–1.86, 1.84–1.77, 1.63–1.54, 1.46–1.38 (4m, 2 ring CH₂); 1.55, 1.15 (2s, 2 CH₃). MS: 293 (1, *M* +), 276 (4, [*M* − OH]+), 275 (4, [*M* − H₂O]+), 262 (1), 260 (1), 258 (2), 246 (3), 232 (2), 230 (3), 208 (8), 196 (48), 178 (22), 155 (24), 141 (32), 119 (22), 105 (22), 93 (22), 91 (38), 72 (100). Anal. calc. for C₁₇H₂₇NO₃ (293.41): C 69.59, H 9.28, N 4.77; found: C 69.57, H 9.21, N 4.63.
- 4.6. $\{(Z)-5\beta-Hydroxy-2\alpha-f(E,E)-6-hydroxy-4-methylhexa-1,4-dienyl\}-2-methylcyclopentylidene\}$ N, N-dimethylacetamide (5f). From $(2\alpha,5\beta)$ -3f (0.67 g, 1.36 mmol). Chromatography on silica (20 g; hexane/AcOEt 1:1 \rightarrow 1:9): 284 mg of 5f. IR: 3401s, 2932s, 2873s, 1651m, 1600s, 1496m, 1448w, 1397m, 1304w, 1259w, 1176w, 1145w, 1014m, 979m, 866w. ¹H-NMR (400 MHz, (D₆)DMSO): 6.07 (d, J=1.6, C=CHCON); 5.88 (d, J=1.6, CHOH); 5.45 (d, J=15.6, CH=CHCH₂); 5.33 (dt, J=15.6, 6.8, CH=CHCH₂); 5.27 (br. $t, J\approx7$, C=CHCH₂O); 4.64 (m, CHOH); 4.47 (t, J=5.6, CH₂OH); 3.92 ($\sim t, J=5.6$, CH₂OH); 3.02, 2.89 (2 s, 2 NCH₃); 2.65 (br. d, J=6.8, C=CCH₂C=C); 1.92–1.83, 1.71–1.58 (2 m, 2 ring CH₂); 1.55, 1.21 (2 s, 2 CH₃). MS: same as for 4f.
- 4.7. 8-Hydroxyoctyl { (Z)-5 α -Hydroxy-2 α -(E,E)-6-hydroxy-4-methylhexa-1,4-dienyl]-2-methylcyclopentylidene} acetate (4i). From 3d (0.2 g, 0.28 mmol). Chromatography on silica (20 g; hexane/AcOEt 6:4 \rightarrow 1:1): 100 mg of 4i as a colorless oil. IR: 3400s, 2931s, 2859m, 1690s, 1645m, 1461m, 1389m, 1346m, 1300m, 1208s, 1155m, 1062m, 1010m. H-NMR (250 MHz): 5.74 (d, J = 1.7, C=CHCO); 5.50–5.32 (m, 3 olef. H); 4.96 (m, CHOH); 4.88 (d, d = 2, CHOH); 4.14 (d, CO₂CH₂, C=CHCH₂O); 3.64 (d, d = 1.7, C=CHCO); 2.18–2.03, 1.96–1.29 (d, 8 CH₂); 1.65, 1.19 (d, d) d = 2, CH₃). MS: 376 (2, d) d = 1, 358 (2, d) d = 2, CHOH₁(O)⁺), 248 (4, d) d = HO(CH₂)8OH]⁺), 230 (26, d) d = 1, 151 (56), 147 (48), 134 (64), 119 (56), 105 (50), 95 (88), 93 (62), 91 (48), 81 (58).
- 4.8. (RS, E, E)-6-[5,6-Dihydro-4-methyl-2-pentyl-4 H-cyclopenta[b]furan-4-yl]-3-methylhexa-2,5-dien-1-ol (6j). From 3g (2.1 g, 4.05 mmol). Chromatography on silica (150 g; hexane/AcOEt 9:1 \rightarrow 8:2): 0.62 g of 6j. IR: 3346m, 2930s, 2859s, 1724w, 1671w, 1633w, 1556w, 1452m, 1378w, 1315w, 1232m, 1150w, 1089w, 1000m, 973s, 931w, 797w. ¹H-NMR (250 MHz): 5.81 (s, furan H); 5.61 (d, J = 15.5, CH=CHCH₂); 5.40 (br. t, C=CHCH₂O); 5.28 (dt, J = 15.5, 7, CH=CHCH₂); 4.15 (d, J = 7, CH₂OH); 2.68 (br. d, $J \approx 7$, C=CCH₂C=C); 2.66, 2.57 (2t, J = 7.7, 2 C=C(O)CH₂); 2.34, 2.20 (2dt, J = 11, 5.5, ring CH₂); ~ 1.6 (partially obscured m, CH₂); 1.42–1.27 (m, 2 CH₂); 1.64, 1.26 (2s, 2 CH₃); 0.90 (m, CH₃CH₂). MS: 302 (45, M^+), 287 (100, $[M CH_3]^+$), 217 (86, $[M C_5H_9O]^+$), 159 (22), 151 (30), 145 (20), 131 (18), 119 (22), 105 (35), 91 (38).
- 4.9. (RS, E, E)-6- {2-[(Z)-Heptadec-8-enyl]-5,6-dihydro-4-methyl-4H-cyclopenta[b]furan-4-yl}-3-methyl-hexa-2,5-dien-1-ol (**6k**). From **3h** (0.55 g, 0.8 mmol). Chromatography on silica (70 g; hexane/AcOEt 95:5): 156 mg of **6k** as a pale yellow oil. IR: 3321m, 3004w, 2925s, 2854s, 1672w, 1634w, 1556m, 1480m, 1374w, 1343w, 1311w, 1292m, 1000m, 972m, 797w. 1 H-NMR (250 MHz): 5.80 (s, furan H); 5.62 (d, J=15.5, trans CH=CHCH₂); 5.42 (br. t, C=CHCH₂O); 5.34 (m, cis CH=CH); 5.30 (dt, J=15.5, dt, dt (br. dt) dt) dt (br. dt) dt) dt (br. dt) dt) dt (br. dt) dt) dt0 (br. dt) dt0 (br. dt0) dt0 (br. dt0)
- 5. Compound 11. 5.1. (E,E)-6- $\{3\alpha$ -[(3,5-Dinitrobenzoyl)oxy]-2-[(Z)-(ethoxycarbonyl)methylidene]- 1β - $methylcyclopentyl\}$ -3-methylhexa-2,5-dienyl 3,5-Dinitrobenzoate (11). Diol 4c (200 mg, 0.678 mmol) was dissolved in MeCN (10 ml). Et₃N (0.368 ml, 2.72 mmol) followed by 3,5-dinitrobenzoyl chloride (627 mg, 2.72 mmol) was introduced at 0°, and the resulting soln. was stirred overnight at r.t. Then, it was diluted with ice-cold brine and extracted twice with AcOEt. The combined extracts were washed with H₂O and brine, dried (Na₂SO₄), filtered, and concentrated to a brown oil. Chromatography on silica (90 g; hexane/AcOEt 9:1 \rightarrow 4:1) produced a yellow oil that spontaneously crystallized. Yield: 374 mg (81%). M.p. 110.6–112°. IR (KBr): 3082m, 2935m, 1730m, 1626m, 1547m, 1461m, 1345m, 1275m, 1202m, 1165m, 1081m, 1030m, 973m, 926m, 725m. ¹H-NMR (250 MHz): 9.22, 9.19 (2m, m) = 2.2, 2 arom. H); 9.15, 9.09 (2m, m) = 2.2, 4 arom. H); 6.48 (m, CO₂CH); 5.92 (m) = 1.6, C=CHCO₂); 5.70–5.47

 $(m, 3 \text{ olef. H}); 4.96 (d, J = 7, C=CHCH_2O); 4.03 (m, CO_2CH_2CH_3); 2.91 (br. d, J = 7, C=CCH_2C=C); 2.45-2.30, 2.10-1.80, 1.72-1.58 (3 m, 2 ring CH_2); 1.81, 1.30 (2 s, 2 CH_3); 1.17 (t, J = 7, CO_2CH_2CH_3). MS: 470 (8, <math>[M - (\text{dinitrobenzoic acid})]^{++}), 425 (2), 403 (6), 258 (18, <math>[M - 2 \text{ (dinitrobenzoic acid})]^{++}), 195 (100), 185 (58), 149 (64), 119 (36), 105 (42), 93 (68), 91 (56), 75 (82).$

5.2. X-Ray Analysis of 11. Crystal Data: Crystals were grown by slow evaporation of MeCN at r.t. Space group triclinic P_1^- ; cell dimensions: a = 10.400(2), b = 11.009(3), c = 15.338(3) Å; $\alpha = 100.69(2)^\circ$, $\beta = 108.43(2)^\circ$, $\gamma = 98.95(2)^\circ$, D = 1.42 Mg/m³, Z = 2; μ (MoK α) = 0.11 mm⁻¹. Data Collection: Data were collected on a Nicolet R3m four-circle diffractometer fitted with a LTI cooling apparatus. Temp. 190 K; wavelength: 0.71069 Å; scan mode $\Theta/2\Theta$; scan speed 3.3°/min minimum speed; strong reflections measured up to 10°/min; scan width 1.8°; 2Θ range 0–50°; peak: background ratio 5:1, total data measured: 5996 excluding standards; total observed: 3875; rejection criterion $I > 2.5\sigma(I)$; number of parameters: 441; weights $w = 1/\sigma(F) + 0.001|F|^2$.

Structure Determination and Refinement: The structure was determined by direct methods using the Nicolet SHELXTL PLUS (MicroVAX II) system. The Me group at the ester function is disordered in two positions, the calculated occupation factors are 0.69 and 0.31, respectively. Refinement proceeded to convergence at R = 0.049 with anisotropic refinement of all non-H-atoms. Coordinates and thermal parameters have been deposited with the Crystallographic Data Centre, Cambridge, University Chemical Lab., Cambridge CB2 1EW, UK.

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