

LETTERS TO THE EDITOR

Synthesis of 4-(Hydroxyphenyl)-1,2,4-triazoles

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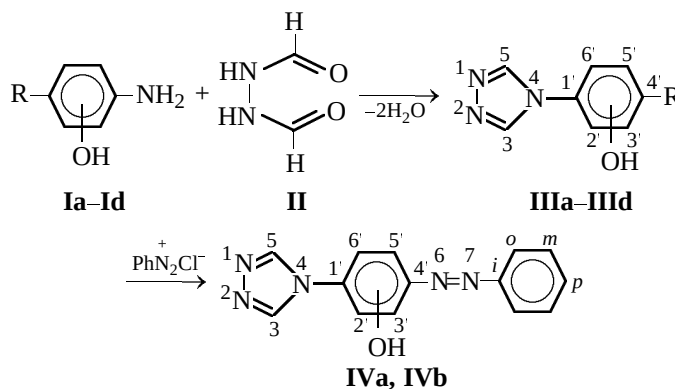
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Derivatives of 1,2,4-triazole are potential ligands for the synthesis of magnetoactive coordination compounds possessing optoelectronic, electrochemical, electrophysical, and thermochemical properties [1–7]. Certain 1,2,4-triazoles are applied as drugs [8–10], herbicides, pesticides, and growth regulators [11–13],

while 4-substituted 1,2,4-triazoles exhibit radioprotective activity [14, 15].

In the present work we developed a procedure for preparing previously unknown 4-substituted 1,2,4-triazoles, specifically 4-(*o*-, *m*-, *p*-hydroxyphenyl)-1,2,4-triazoles, and studied their azo coupling with phenyldiazonium chloride.



2-OH (Ia, IIIa, IVa), 3-OH (Ib, IIIb, IVb), 4-OH (Ic, IIIc); R = H (Ia-Ic, IIIa-IIIc); 2-OH; R = NO₂ (Id, IIId).

4-Substituted 1,2,4-triazoles were prepared by fusing diformylhydrazine (II) with amino phenols I at 114–170°C. Triazoles III are stable crystalline substances with high melting points and soluble in ethanol, DMSO, and aqueous ethanol (1:1).

Treatment of solutions of compounds IIIa and IIIb in methanol (aniline, NaNO₂, conc. HCl) gave azo derivatives IVa and IVb, respectively.

4-(*o*-Hydroxyphenyl)-1,2,4-triazole (IIIa). A mix-

ture of 8.8 g of diformylhydrazine (II), 10.9 g of *o*-aminophenol (Ia), and a little hydroquinone was heated at 145–150°C for 0.5 h under argon and then cooled to 70°C, and 20 ml of ethanol. The mixture was stirred for 5 min, cooled to 20°C, diluted with 100 ml of absolute ether, and left to stand overnight in a refrigerator. The precipitate that formed was filtered off and recrystallized from aqueous ethanol (1:1). Yield 6.65 g (46%), grayish yellow crystals, mp 214–216°C. IR spectrum, ν , cm⁻¹: 3400 (OH), 3100

(CH), 1600 (C=N), 1520, 1475, 1440 (C=C_{arom}), 1285 (C-N). ¹H NMR spectrum, ν , ppm: 7.32–7.81m (CH_{arom}), 9.20 s (2H, C^{3,5}H), 10.92 br.s (1H, OH). ¹³C NMR spectrum, δ_C , ppm: 150.53 (C²), 143.17 (C^{3,5}), 129.79 (C⁵), 125.63 (C⁴), 121.71 (C¹), 119.70 (C³), 119.95 (C⁶). ¹⁵N NMR spectrum, δ_N , ppm: –203.5 (N⁴), –62.0 (N^{1,2}). Found, %: C 59.62; H 3.95; N 26.12. C₈H₇N₃O. Calculated, %: C 59.63; H 4.38; N 26.07.

4-(*m*-Hydroxyphenyl)-1,2,4-triazole (IIIb) was prepared similarly to IIIa from 4.4 g of compound II and 5.5 g of *m*-aminophenol at 114–115°C, reaction temperature 1 h. Yield 4.37 g (54%), grayish yellow crystals, mp 230–231°C (from ethanol). IR spectrum, ν , cm^{–1}: 3410 (OH), 3100 (CH), 1605 (C=N), 1520, 1490, 1440 (C=C_{arom}), 1220 (C–N). ¹H NMR spectrum, δ , ppm: 6.85–7.33 m (CH_{arom}), 9.05 s (2H, C^{3,5}H), 10.01 s (1H, OH). ¹³C NMR spectrum, δ_C , ppm: 158.87 (C³), 141.71 (C^{3,5}), 135.19 (C¹), 131.25 (C⁵), 115.42 (C⁶), 112.02 (C⁴), 108.60 (C²). ¹⁵N NMR spectrum, δ_N , ppm: –193.8 (N⁴), –59.7 (N^{1,2}). Found, %: C 59.58; H 4.30; N 25.92. C₈H₇N₃O. Calculated, %: C 59.63; H 4.38; N 26.07.

4-(*p*-Hydroxyphenyl)-1,2,4-triazole (IIIc) was prepared similarly to IIIc from 4.4 g of compound II and 5.5 g *p*-aminophenol at 165–170°C, reaction time 0.5 h. Yield 4.82 g (64%), grayish yellow crystals, mp 285–286°C (from ethanol). IR spectrum, ν , cm^{–1}: 3400 (OH), 3090 (CH), 1590 (C=N), 1520, 1480 (C=C_{arom}), 1230 (C–N). ¹H NMR spectrum, δ , ppm: 6.90–17.45 m (C_{arom}), 8.92 s (2H, C^{3,5}), 9.85 s (1H, OH). ¹³C NMR spectrum, δ_C , ppm: 157.43 (C⁴), 141.82 (C^{3,5}), 125.85 (C¹), 123.33 (C², 6'), 116.30 (C^{3,5}). ¹⁵N NMR spectrum, δ_N , ppm: –194.4 (N⁴), –59.7 (N^{1,2}). Found, %: C 59.48; H 4.25; N 26.02. C₈H₇N₃O. Calculated, %: C 59.63; H 4.38; N 26.07.

4-(2-Hydroxy-4-nitrophenyl)-1,2,4-triazole (IIId) was prepared similarly to IIIa from 4.4 g of compound II and 7.7 g of 2-amino-5-nitrophenol at 175–180°C, reaction time 1 h. Yield 6.68 g (68%), black crystals, mp 360–365°C. IR spectrum, ν , cm^{–1}: 3410 (OH), 3105 (CH), 1590 (C=N), 1520, 1495, 1425 (C=C_{arom}), 1360, 865 (NO₂), 1210 (C–N). ¹H NMR spectrum, δ , ppm: 7.72–7.83 m (3H, C₆H₃), 8.95 s (2H, C^{3,5}H), 14.33 br.s (1H, OH). ¹³C NMR spectrum, δ_C , ppm: 151.36 (C²), 147.93 (C⁴), 143.41 (C^{3,5}), 127.69 (C¹), 126.55 (C⁶), 115.29 (C⁵), 111.92 (C³). ¹⁵N NMR spectrum, δ_N , ppm: –200.4 (N⁴), –59.9 (N^{1,2}), –7.2 (NO₂). Found, %: C 46.85; H 2.71; N 26.95. C₈H₆N₄O₃. Calculated, %: C 46.60; H 2.91; N 27.18.

4-(2-Hydroxy-4-phenylazophenyl)-1,2,4-triazole (IVa). To a cold (0°C) solution of 1.61 g of 4-(*o*-hydroxyphenyl)-1,2,4-triazole (IIIa) in 20 ml of methanol, a cold (0°C) solution of phenyldiazonium chloride (2 g of aniline and 1.65 g of NaNO₂ in 6 ml of conc. HCl and 50 ml of water) was added. After 5-min stirring, a saturated solution of sodium acetate was added until turbidity, and the mixture was left to stand overnight. The precipitate that formed was filtered off, washed on the filter with cold water, and dried in a vacuum. Yield 1.56 g (60%), bright yellow crystals, mp 235–237°C (from DMF). IR spectrum, ν , cm^{–1}: 3440 (OH), 3100 (CH), 1610, 1595 (N=N, C=N), 1530, 1500, 1425 (C=C_{arom}), 1205 (C–N). ¹H NMR spectrum, δ , ppm: 7.27–8.01 m (3H, C₆H₃), 7.54–7.85 m (5H, C₆H₅), 8.93 (2H, C^{3,5}H). ¹³C NMR spectrum, δ_C , ppm: 153.81 (C⁶), 151.81 (C_i), 144.88 (C³), 143.00 (C^{3,5}), 131.05 (C_p), 129.40 (C_m), 124.38 (C⁴), 122.27 (C¹, C_o), 120.21 (C²), 117.22 (C⁵). ¹⁵N NMR spectrum, δ_N , ppm: –200.8 (N⁴), –125.5 (N⁶), –121.5 (N⁷), –57.7 (N^{1,2}). Found, %: C 63.25; H 4.29; N 26.18. C₁₄H₁₁N₅O. Calculated, %: C 63.38; H 4.17; N 26.40.

4-(3-Hydroxy-4-phenylazophenyl)-1,2,4-triazole (IVb) was prepared similarly to IVa from 1.73 g of 4-(*m*-hydroxyphenyl)-1,2,4-triazole (IIb), 2.0 g of aniline, 1.65 g of NaNO₂, 1.65 g of sodium acetate at 0°C. Yield 1.51 g (58%), bright yellow crystals, mp 274–276°C. IR spectrum, ν , cm^{–1}: 3430 (OH), 3105 (CH), 1610, 1585 (N=N, C=N), 1535, 1510, 1410 (C=C_{arom}), 1205 (C–N). ¹H NMR spectrum, δ , ppm: 7.00–7.66 m (3H, C₆H₃), 7.48–7.85 m (5H, C₆H₅), 8.85 s (C^{3,5}H), 10.96 br.s (1H, OH). ¹³C NMR spectrum, δ_C , ppm: 162.26 (C³), 152.38 (C_i), 144.30 (C^{3,5}), 137.49 (C⁴), 135.17 (C¹), 131.32 (C_p), 129.55 (C_m), 122.75 (C_o), 118.27 (C⁵), 117.04 (C⁶), 112.12 (C²). ¹⁵N NMR spectrum, δ_N , ppm: –199.3 (N⁴), –120.4 (N⁷), –116.0 (N⁶), –58.1 (N^{1,2}). Found, %: C 63.38; H 4.22; N 26.12. C₁₄H₁₁N₅O. Calculated, %: C 63.38; H 4.17; N 26.40.

The IR spectra were obtained on a Specord IR-75 instrument in KBr. The ¹H and ¹³C NMR spectra were recorded at 20°C on a Bruker-400 spectrometer (400.13 and 100.62 MHz, respectively) in DMSO-*d*₆, internal reference HMDS.

REFERENCES

1. Haashoot, I.G., *Coord. Chem. Rev.*, 2000, vol. 131, p. 200.
2. Lavrenova, L.G. and Larionov, S.V., *Koord. Khim.*, 1998, vol. 24, no. 6, p. 403.

3. Koningsbruggen, P.I., Haasnoot, I.G., Kooijman, H., Reedijk, I., and Spek, A.L., *Inorg. Chem.*, 1997, vol. 36, no. 11, p. 2487.
4. Lavrenova, L.G., Shakirova, O.G., Shvedenkov, Yu.G., Ikorskii, V.N., Varnek, V.A., Sheludyakova, L.A., and Larionov, S.V., *Koord. Khim.*, 1999, vol. 25, nos. 3–4, p. 208.
5. Shakirova, O.G., Virovets, A.V., Naumov, D.Y., Shvedenkov, Y.G., Elokhina, V.N., and Lavrenova, L.G., *Inorg. Chem. Commun.*, 2002, no. 5, p. 690.
6. Shakirova, O.G., Lavrenova, L.G., Shvedenkov, Yu.G., Berezevskii, G.A., Naumov, D.Yu., Sheludyakova, L.A., Dolgushin, G.V., and Larionov, S.V., *Koord. Khim.*, 2004, vol. 30, no. 7, p. 507.
7. Virovets, A.V., Lider, E.V., Elokhina, V.N., Bushuev, M.B., and Lavrenova, L.G., *Zh. Strukt. Khim.*, 2005, vol. 46, no. 2, p. 366.
8. Mashevskaya, I.V., Anikina, L.V., Vikharev, Yu.B., Safin, V.A., Kol'tsova, S.V., and Maslivets, A.N., *Khim.-Farm. Zh.*, 2001, vol. 35, no. 8, p. 11.
9. Pavlov, P.A., Basov, N.Yu., and Pavlov, P.P., *Khim.-Farm. Zh.*, 1998, vol. 32, no. 7, p. 33.
10. Zor'kina, A.V., Kostin, Ya.V., Inchina, V.I., Serenov, L.N., and Smirnov, L.D., *Khim.-Farm. Zh.*, 1998, vol. 32, no. 5, p. 16.
11. Shtents, E., Mendel', G., and Shuster, F., *Mikrobiologiya*, 1980, vol. 42, no. 2, p. 239.
12. Pankaja, K., *Pectic. Sci.*, 1974, vol. 5, no. 3, p. 255.
13. Selezneva, E.S., Belousova, Z.P., Ten'gaev, E.I., Kadatskaya, A.V., Zvereva, I.A., and Poletaeva, E.N., *Khim.-Farm. Zh.*, 2003, vol. 37, no. 8, p. 14.
14. Kulyabko, P.N., Dubenko, R.G., and Pel'kis, P.S., *Radiobiologiya*, 1969, no. 9, p. 633.
15. Yakushinskii, V.G., *Usp. Khim.*, 1975, vol. 44, no. 3, p. 531.