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SYNTHESIS OF NEW 8-METHOXY-4-METHYL-3-(N-[2 - AMINO-(1',3',4')THIA/OXA-DIAZOL-5'-YL]-SUBSTITUTED METHYL)-AMINO THIOCOUMARINS

B. Prasanna ^a & G. V. P. Chandramouli ^a

^a Department of Chemistry, National Institute of Technology, Warangal, A.P., India

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SYNTHESIS OF NEW 8-METHOXY-4-METHYL-3-(N-[2'-AMINO-(1',3',4')THIA/OXA-DIAZOL-5'-YL]-SUBSTITUTED METHYL)-AMINO THIOCOUMARINS

B. Prasanna and G. V. P. Chandramouli Department of Chemistry, National Institute of Technology, Warangal, A.P., India

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8-Methoxy-4-methyl-3-(N-[2'-amino-(1', 3', 4')thia/oxa-diazol-5'-yl] substituted methyl)-amino thiocoumarins **6**(**a**-**f**) and **7**(**a**-**f**), were synthesized by using the unreported 8-methoxy-4-methyl-3-[N-(2'-oxo-2'-methoxy-1'-substituted ethan-1'-yl) amino thiocoumarins as key intermediates.

Keywords: Aminoacids; amino thia/oxa-diazoles; 8-methoxy-4-methyl-thiocoumarin

INTRODUCTION

Many derivatives of coumarins, such as EM 800, are known to be Chromone prodrugs,¹ which are useful as orally active nonsteroidal antiestrogens² and antitumor agents.³ Tazarotene, a substituted benzothiopyran, is used for clinical evaluation of psoriasis.^{4,5} Tertatol is another substituted benzothiopyran that is widely used against hypertension.^{6,7}

Inspired by these observations, and in continuation of our program of synthesizing sulfur heterocycles,⁸ we designed the synthesis of a series of a new 8-methoxy-4-methyl-3-(N-[2'-amino-(1',3',4')thia/oxa-diazol-5'-yl]-substituted methyl)-amino thiocoumarins **6**(**a**-**f**) and **7**(**a**-**f**) starting from 6-methoxy-2-acetyl thiophenol.

Address correspondence to G. V. P. Chandramouli, Department of Chemistry; National Institute of Technology, Warangal 506004, A.P., India. E-mail: gvpc@nitw.ernet.in

RESULTS AND DISCUSSION

6-Methoxy-2-acetyl thiophenol **1** was prepared by acetylating 6methoxy thiophenol with acetyl chloride in anhydrous $AlCl_3$ by using dry DCM as solvent. The acetyl compound **1** was later reacted with chloroacetyl chloride by refluxing in dry acetone containing anhydrous potassium carbonate in order to convert it into the corresponding acetyl thio compounds **2**. Upon further treatment with substituted amino acid esters **3**(**a**-**e**) in presence of dry acetone containing anhydrous potassium carbonate, componds **2** produced a series of 2-acetyl-6-methoxy-1-[S-(1'oxo-3'-aza-4-substituted-5'-oxo-5'-methoxy-pent-1'-yl]-thiophenols **4**(**a**-**e**). These esters, when treated with NaOMe in dry methanol, in turn furnished the desired 8-methoxy-4-methyl-3-[N-(2'-oxo-2'-methoxy-1'-substituted ethan-1'-yl]-amino thiocoumarins **5**(**a**-**e**).

The esters $5(\mathbf{a}-\mathbf{e})$, upon reaction with thiosemicarbazide and semicarbazide, produced a series of required thia/oxa diazole derivatives $6(\mathbf{a}-\mathbf{e})$ and $7(\mathbf{a}-\mathbf{e})$, respectively, as shown in Scheme 1.

By using methyl proline ester **3f** in the place of the aminoacid esters, and by carrying out same sequence of reactions b, c, and d, we got another set products, 4f, 5f, 6f, and 7f, respectively, which are not shown in the above scheme.

The structures of the compounds synthesized were established on the basis of analytical and spectral data (Tables I–III). In the IR spectrum of compound **4b**, vibration bands at 3180 cm⁻¹ and 1710 cm⁻¹ correspond to (CH₃CO) group. The ¹H NMR of the same compound contains signals at 2.3 δ (d, 3H, CH₃), 4.46 δ (s, 2H, CH₂), 4.18 δ (q, 1H, CH), 3.74 δ (s, 3H, OMe), 4.28 δ (d, 3H, –CO–OCH3), 6.88–7.42 δ (m, 3H, Ar–H). In case of compound **5b**, the signal for –CH₂ is absent. The absence of this signal indicated in the participation of –CH₂ in the cyclization to give rise to the thiocoumarin heterocycles. The mass spectrum of **5b** contains molecular ion peak of m/z 306. The ¹H NMR spectra of the compounds **6(a–f)** and **7(a–f)** show the absence of amino acid methyl ester (–COOMe) group.

EXPERIMENTAL

All melting points were uncorrected. The elementary analysis was carried out by CARRLO ERBA STUMENTOZOINE, Itali model, 1108, and IR spectra (cm⁻¹) were recorded on a PERKINE Elmer-282 instrument. The ¹H NMR spectra were recorded on a Varian 300 MHz spectrometer using tetra methyl silane as an internal standard. Chemical shift values



are expressed upon δ scale. Mass spectra were scanned on a Jeol-Jms-300 spectrometer at 70 eV. The purity of the compound was monitored by TLC performed on a silica gel plates (Merck) using ethylecetate and petroleum ether.

2-Acetyl-6-methoxy-S-chloroacetyl Thiophenol 2

To a suspension of 2-acetyl-6-methoxy thiophenol 1 (1.82 g, 0.01 mol) and anhydrous potassium carbonate (2.7 g, 0.02 mol) in dry acetone (20 ml), chloro acetylchloride (1.13 g, 0.01 mol) was added. The resulting mixture was refluxed with stirring for 6 h. The progress of the reaction was monitored by TLC. The reaction mixture was cooled, filtered,

Comp.	Mol. formula	m.p. ($^{\circ}C$)	С	Н	Ν	Ο	\mathbf{S}
4a	$\rm C_{14}H_{17}NO_5S$	123-125	54.01/	5.50/	4.50/	25.69/	10.30/
			53.94	5.45	4.46	26.54	10.32
4b	$C_{15}H_{19}NO_5S$	131 - 133	55.37/	5.89/	4.30/	24.30/	9.85/
			55.35	5.84	4.26	24.23	9.80
4c	$C_{21}H_{23}NO_5S$	142 - 144	62.83/	5.77/	3.49/	19.93/	7.99/
			62.78	5.72	3.45	19.90	8.01
4d	$\mathrm{C}_{23}\mathrm{H}_{24}\mathrm{N}_{2}\mathrm{O}_{5}\mathrm{S}$	155 - 157	62.71/	5.49/	6.36/	18.16/	7.28/
			62.68	5.42	6.34	18.12	7.26
4e	$C_{21}H_{23}NO_6S$	163 - 165	60.42/	5.55/	3.36/	22.99/	7.68/
			60.39	5.52	3.34	22.94	7.62
4f	$C_{17}H_{21}NO_5S$	114 - 116	56.96/	5.68/	4.15/	23.71/	9.50/
			56.92	5.64	4.10	23.67	9.52
5a	$C_{14}H_{15}NO_4S$	162 - 164	57.32/	5.15/	4.77/	21.82/	10.93/
			57.30	5.11	4.72	21.78	10.90
5b	$C_{15}H_{17}NO_4S$	174 - 176	58.62/	5.57/	4.56/	20.82/	10.43/
			58.58	5.52	4.50	20.79	10.40
5c	$C_{21}H_{21}NO_4S$	183 - 185	65.78/	5.52/	3.65/	16.69/	8.36/
	<u>.</u>		65.72	5.49	3.63	16.65	8.32
5d	$C_{22}H_{22}N_2O_4S$	202 - 204	65.39/	5.25/	6.63/	15.16/	7.59/
	- 20 22 2 4		65.34	5.23	6.60	15.12	7.52
5e	C91H91NO5S	158 - 160	63.14/	5.30/	3.51/	20.03/	8.03/
	-21-213.5		63.10	5.26	3.48	19.92	8.08
5f	C17H10NO4S	152 - 154	61.24/	5.74/	4.20/	19.19/	9.62/
01	-17-19-19-4-		61.20	5.71	4.18	19.14	9.58
6a	C14H14N4O2S2	192–194	50.28/	4.22/	16.75/	9.57/	19.18/
	- 1414- 14 - 22		50.24	4.20	16.72	9.51	19.15
6b	C15H16N4O2S2	204 - 206	51.71/	4.63/	16.08/	9.18/	18.40/
	- 1510- 14 - 2.02		51.67	4.60	16.02	9.15	18.42
6c	Co1Ho0N4OoSo	218 - 220	59.41/	4.75/	13.20/	7.54/	15.11/
	-21-20-14-21-2		59.38	4.72	13.16	7.50	15.08
6d	CooHo1N=OoSo	242-244	59.59/	4.57/	15.11/	6.90/	13.83/
	-23213-22		59.55	4.52	15.08	6.87	13.78
6e	Cat HaaN4OaSa	185-187	57 25/	4 58/	12.72/	10.90/	14 56/
00	02111201140302	100 101	57 22	4 55	12.68	10.86	14 54
6f	C17H10N4O0S0	166-168	54 53/	4 84/	14 96/	8 54/	17 12/
01	01711181140202	100 100	54.50	4.78	14.92	8.50	17.08
7a	C14H14N4O2S	172 - 174	52.82/	4 43/	17.60/	15.08/	10.07/
·u	0141114114030	112 111	52.80	4 38	17.57	15.02	10.04
7h	C15H16N4O2S	189-191	54 20/	4 85/	16.86/	14 44/	9.65/
•••	01011014030	100 101	54 17	4.82	16.82	14 41	9.61
7c	Cat HaoN4OaS	242-246	61 75/	4 64/	13.72/	13.72/	7 85/
	0211120114030	212 210	61 71	4 60	13.68	13.69	7.82
7d	CapHatNrOaS	235 - 237	61 73/	4 73/	15.65/	10.00	7 16/
	C79117110 C30	200 201	61 70	4 69	15.61	10.70	7 11
7e	Ca1Ha0N4O4Sa	209-211	59 42/	4.75/	13 20/	15.08/	7 55/
	21-20-14-24-02		59 40	4.73	13 17	15.02	7.51
7f	C17H10N4O2S	194-196	56 97/	5.06/	15 63/	13 39/	8 95/
••	01/11/01/4030	101 100	56.92	5.02	15.60	13 34	8 91

TABLE I Physical Properties and Analytical Data (Calc./Found%) of the Compounds 4(a–f), 5(a–f), 6(a–f), and 7(a–f)

Comp.	$\mathrm{IR}(\mathrm{cm}^{-1})$	$m\!/\!z~(M^{+1})$	$^{1}\mathrm{H}\ \mathrm{NMR}\ (\delta\ \mathrm{ppm})$
5a	1680 (—S—C=O), 1710 (CO),3240(NH)	292	2.22 (s, 3H, CH ₃), 3.62 (s, 3H, OCH ₃), 3.90 (s, 3H, CO–OCH ₃), 4.20 (s, 2H, CH ₂), 6.96–7.40 (m, 3H, Ar–H), 12.16 (s, 1H, NH)
5b	1680 (-S-C=O), 1705 (-CO), 3254 (-NH)	306	1.34 (d, 3H, CH ₃), 2.59 (s, 3H, CH ₃), 4.30 (q, 1H, CH), 6.88–7.42 (m, 3H, Ar–H), 12.38 (s, 1H, NH)
5c	1690 (-S-C=O), 1710 (-CO), 3242 (NH)	382	2.32 (s, 3H, CH ₃), 3.74 (s, 3H, OCH ₃), 4.12 (s, 3H, CO–OCH ₃), 4.14 (t, 1H, CH), 4.32 (d, 2H, CH ₂), 6.90–7.46 (m, 8H, Ar–H)
5d	1692 (-S-C= O), 1704 (CO), 3320 (NH)	421	$\begin{array}{l} 2.12 \ ({\rm s}, 3{\rm H}, {\rm CH}_3), 3.3 \ ({\rm d}, 2{\rm H}, {\rm CH}_2), \\ 3.64 \ ({\rm s}, 3{\rm H}, {\rm OCH}_3), 4.26 \ ({\rm s}, 3{\rm H}, \\ -{\rm CO-OCH}_3), 4.36 \ ({\rm t}, 1{\rm H}, {\rm CH}), \\ 5.20 \ ({\rm s}, 1{\rm H}, {\rm Indole-H}), 6.92{-}7.52, \\ ({\rm m}, 7{\rm H}, {\rm Ar-H}), 12.12 \ ({\rm s}, 1{\rm H}, {\rm NH}) \end{array}$
5e	1694 (-S-C=O), 1705 (CO), 3140 (NH), 3520 (-OH)	408	$\begin{array}{c} 2.34 \ ({\rm s}, 3{\rm H}, {\rm CH}_3), 3.64 \ ({\rm s}, 3{\rm H}, \\ {\rm OCH}_3), 4.00 \ ({\rm s}, 1{\rm H}, {\rm CH}), 4.26 \ ({\rm s}, \\ 3{\rm H}, {\rm CO-OCH}_3), 3.32 \ ({\rm d}, 2{\rm H}, \\ {\rm CH}_2), 6.94{-}7.42 \ ({\rm m}, 7{\rm H}, {\rm Ar-H}), \\ 10.42 \ ({\rm s}, 1{\rm H}, {\rm OH}), 12.14 \ ({\rm s}, 1{\rm H}, \\ {\rm NH}) \end{array}$
5f	1690 (-S-C=O), 1704 (-CO)	332	$\begin{array}{l} 2.10{-}2.18\ (m,6H,(CH_2)_3),2.36\ (s,\\ 3H,CH_3),3.64\ (s,3H,OCH_3),\\ 4.12\ (t,1H,CH),4.30\ (s,3H,\\ CO{-}OCH_3),7.12\ (m,3H,Ar{-}H),\\ 12.14\ (s,1H,NH) \end{array}$

TABLE II IR and ¹H NMR Spectral Data of the Compounds 5(a-f)

the inorganic residue was washed with acetone, and the solvent was removed on a rotavapor. By the addition of water to the residue, the title compound was obtained as an oil, which was purified by column chromatography (EtOAc/petroleum ether 1:9). Yield, 70-74%.

2-Acetyl-6-methoxy-1-[S-(1'-oxo-3'-aza-4'-substituted-5'oxo-5'-methoxy-pent -1'-yl]-thiophenols 4(a-e) and 4f

A mixture of 3(a-f) (0.01 mol), 2-Acetyl-6-methoxy-S-chloroacetyl thiophenol 2 (0.01 mol), and anhydrous K₂CO₃ (2.7 g; 0.02 mol) in 25 ml of dry acetone was refluxed with stirring for 8 h. The progress of the reaction was monitored by TLC. The reaction mixture was cooled, filtered, the inorganic residue was washed with acetone, and the solvent was removed on a rotavapor. The addition of water to the residue gave the title compounds as precipitates, which were collected by filtration. They were washed with water and crystallized from ethanol. Yield, 60–72%.

Comp.	${ m IR}~{ m cm}^{-1}$	$(m\!/\!z) \: M^{+1}$	1 H NMR(δ ppm)
6a	1680 (C=O),	333	2.32 (s, 3H, CH ₃), 2.48 (s, 2H, CH ₂),
	3242 (NH),		$3.82 (s, 3H, OCH_3), 7.30-7.36 (m, 3H, Ar-H)$
	$3415(NH_2)$		9.8 (s, 2H, NH ₂), 12.2 (s, 1H, NH)
6b	1684 (C=O),	347	2.14 (d, 3H, CH ₃), 2.36 (s, 3H, CH ₃),
	3140 (NH),		2.42 (q, 1H, CH), 3.84 (s, 3H, OCH ₃), 7.32–7.38
	$3412(NH_2)$		(m, 3H, Ar–H), 9.6 (s, 2H, NH ₂), 12.8 (s, 1H, NH)
6c	1682 (C=O),	425	$2.30 (s, 3H, CH_3), 2.54 (t, 1H, CH), 3.14 (d, 2H, CH_2),$
	3142 (NH),		3.84 (s, 3H, OCH ₃), 7.34–7.36
	$3416\left(NH_2 ight)$		$(m, 8H, Ar-H), 9.6 (s, 2H, NH_2), 12.1 (s, 1H, NH)$
6d	1690 (C=O),	462	2.32 (s, 3H, CH ₃), 2.58 (t, 1H, CH), (d, 2H, CH ₂),
	3145		3.42 3.86 (s, 3H, OCH ₃), 4.68 (s, 1H, CH), 5.12
	(indole-NH),		(s, 1H, indole–H), 6.94–7.42 (m, 7H, Ar–H),
	3244 (NH),		9.4 (s, 2H, NH ₂), 11.4 (s, 1H, NH)
	$3418(NH_2)$		
6e	1692 (C = O),	441	$2.36 (s, 3H, CH_3), 2.48 (t, 1H, CH), 3.40 (d, 2H, CH_2),$
	3140 (NH),		3.86 (s, 3H, OCH ₃), 6.92–7.40 (m, 7H, Ar–H),
	$3412 (NH_2),$		9.4 (s, 2H, NH ₂), 10.8 (s, 1H, OH), 12.6 (s, 1H, NH)
	3560 (–OH)		
6f	1690 (C = O),	375	2.10 (t, 1H, CH), 2.22–2.26 (complex multiplet
	$3412(NH_2)$		pattern, 6H, (CH ₂) ₃), 2.32 (s, 3H, CH ₃), 3.86 (s, 3H,
			OCH ₃), 6.92–7.24 (m, 3H, Ar–H), 2.42 (s, 2H, NH ₂)
7a	1682 (C = O),	303	2.30 (s, 3H, CH ₃), 2.54 (s, 2H, CH ₂), 3.80
	3180 (NH),		(s, 3H, OCH ₃), 7.34–7.92 (m, 3H, Ar–H),
	$3412(NH_2)$		$9.92 (s, 2H, NH_2), 12.42 (s, 1H, NH)$
7b	1688 (C = O),	333	$2.10 (d, 3H, CH_3), 2.22 (s, 3H, CH_3), 2.48 (q,$
	3142 (NH),		$2H, CH_2), 8.88 (s, 3H, OCH_3), 7.52-7.58 (m, 3H,$
	$3414 \left(NH_2 \right)$		$Ar-H$), 9.6 (s, 2H, NH_2), 12.24 (s, 1H, NH)
7c	1686 (C = O),	409	$2.33 \ (s, 3H, CH_3), 2.52 \ (t, 1H, CH), 3.12 \ (d, 2H, CH_2),$
	3240 (NH),		3.86 (s, 3H, OCH ₃), 6.92–7.38 (m, 8H, Ar–H),
	$3412(NH_2)$		$9.72 (s, 2H, NH_2), 12.51 (s, 1H, NH)$
7d	1692 (C = O),	446	$2.32(s,3H,CH_3),2.56(t,1H,CH),3.46(d,2H,CH_2),$
	3182 (NH),		3.86 (s, 3H, OCH ₃), 5.62 (s, 1H, indole-H),
	$3415 \left(NH_2 \right)$		$6.92-7.30 (m, 7H, Ar-H), 9.4 (s, 2H, NH_2),$
			12.3 (s, 1H, NH)
7e	1688 (C=O),	423	$2.46(t,1H,CH),2.48(s,3H,CH_3),3.46(d,2H,CH_2),$
	3240 (NH),		3.88 (s, 3H, OCH ₃), 6.94–7.42 (m, 7H, Ar–H),
	$3412 (NH_2),$		$9.7 (s, 2H, NH_2), 10.6 (s, 1H, OH), 12.3 (s, 1H, NH)$
	$3540 \left(-OH\right)$		
7f	1690 (C = O),	357	2.10 (t, 1H, CH), 2.26–2.32 (complex multiplet
	$3415(NH_2)$		pattern, 6H, $(CH_2)_3$), 2.34 (s, 3H, CH_3), 3.84
			(s, 3H, OCH ₃), 6.92–7.32 (m, 3H, Ar–H),
			$9.8 (s, 2H, NH_2)$

TABLE III $\,$ IR and 1H NMR Spectral Data of the Compounds $6(a{-}f)$ and $7(a{-}f)$

8-Methoxy-4-methyl-3-[N-(2'-oxo-2'-methoxy-1'substituted Ethan-1'-yl)]-amino Thiocoumarins 5(a–e) and 5f

To a stirred solution of 4(a-f) (0.001 mol) in dry methanol (15 ml) was added dropwise a solution of sodium methoxide (0.01 mol) in methanol (5 ml). The reaction mixture was refluxed for 3 h followed by removal of methanol under reduced pressure. The resultant precipitate was obtained by addition of water. The residue was filtered, washed with water, dried, and crystallized from ethanol. Yield, 58–64%.

8-Methoxy-4-methyl-3-(N-[2'-amino-(1',3',4')thia/oxadiazol-5'-yl]-substituted methyl)-amino Thiocoumarins 6(a–f) and 7(a–f)

A mixture of 5(a-f) (0.001 mol) and thiosemicarbazide or semicarbazide (0.001 mol) in dry ethanol (20 ml) was stirred at room temperature for 5 h. The reaction mixture was poured into ice-cold water. The precipitated solids were collected and recrystallized from chloroform. Yield, 56–60%.

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