CHEMISTRY LETTERS, pp. 371-372, 1988.

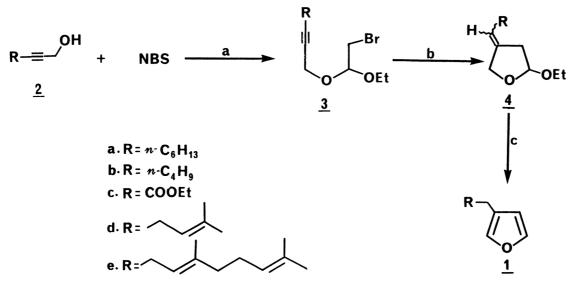
A Radical Cyclisation Route to 3-Alkyl Furans. Synthesis of Perillene and Dendrolasin

Adusumilli SRIKRISHNA<sup>\*</sup> and Gajendran SUNDERBABU Department of Organic Chemistry, Indian Institute of Science, Bangalore - 560 012, India

A general three step synthesis of 3-alkyl furans, including Perillene and Dendrolasin, from alk-2-ynyl alcohols is described <u>via</u> radical cyclisation of 2-bromo-1-ethoxyethyl alk-2-ynyl ethers to 2-ethoxy-4-alkylidene tetrahydrofurans.

Currently radical cyclisation is widely accepted as a powerful tool in organic synthesis.<sup>1)</sup> Its use in the synthesis of variety of butyrolactones is well documented.<sup>2,3)</sup> 3-Alkyl furan (<u>1</u>) is an important functional moiety frequently encountered in variety of terpenoids, however, there are not many general useful synthetic methodologies to this important functional moiety.<sup>4)</sup> We now wish to describe a general strategy to the synthesis of 3-alkyl furans (<u>1</u>) from alk-2-ynyl alcohols using radical cyclisation as the key step. This, incidentally establishes that a mixture of propargyl alcohol and ethyl vinyl ether can serve as an isoprene equivalent.

The methodology is depicted in Scheme 1; radical cyclisation of bromoacetal 3, obtained by bromination of ethyl vinyl ether in the presence of alk-2-ynyl alcohol 2, generates the 2-ethoxy-4-alkylidene tetrahydrofuran (4), which on acid catalysed aromatisation leads to 3-alkyl furan (1). The requisite alk-2-ynols (2) were obtained either by condensing formaldehyde with 1-lithioalkynes(for 2a,b) or by alkylating lithium salt of prop-2-ynol THP ether with alkyl halide followed



Scheme 1. a. CH<sub>2</sub>Cl<sub>2</sub>, -40 <sup>O</sup>C, CH<sub>2</sub>=CH-OEt, 1.5-2 h; b. n-Bu<sub>3</sub>SnCl (0.15 equiv.)/ NaCNBH<sub>3</sub> (1.5 equiv.)/cat.AIBN/t-BuOH; c. PTSA/benzene/RT.

Entry	% yieid of		
	<u>3</u>	<u>4</u> b)	<u>1</u>
a	98	97	89
b	98	76	67
С	96	55	65
d	80	75	67
е	80	72	75

Table 1. Synthesis of 3-alkyl furans via radical cyclisation<sup>a)</sup>

a) All the compounds were purified by silicagel column chromatography

and yields refer to the isolated and chromatographically pure products.

b) Mixture of stereoisomers ( $\approx$  1:1).

by pyridinium p-toluenesulfonate catalysed hydrolysis (for 2c-e). The key radical precursors  $3^{\#}$  were obtained in over 80% yield (Table 1) by a slow addition of ethyl vinyl ether (l.2 equiv., l.5-2 h) to a cold (-40  $^{\circ}$ C) solution of alcohol 2 (l equiv.) and NBS (l.2 equiv.) in methylene chloride. The radical cyclisation (3 - 4) was best achieved by in situ generated catalytic tri-n-butyltinhydride (n-Bu<sub>3</sub>SnCl/NaCNBH<sub>3</sub>) in refluxing t-BuOH (1 to 3 h) in the presence of a catalytic amount of azobisisobutyronitrile (AIBN). The cyclised product (4)<sup>#</sup> was obtained as a mixture of stereoisomers (lpha 1:1) as evidenced by NMR and the mixture was used as such in the next aromatisation step. Finally, the cyclised products 4 were transformed to 3-alkyl furans (1)<sup>#</sup> by treatment with a catalytic amount of p-toluenesulfonic acid in benzene (5-8 h) at room temperature. The yields of bromination, cyclisation and aromatisation are summarised in Table 1.

The generality of this methodology is exemplified by the synthesis of two naturally occuring terpenoids Perillene  $(1d)^{4,5}$  and Dendrolasin  $(1e)^{4,6}$ starting from readily available dimethylallyl bromide and geranyl bromide (Table 1, entries d and e).<sup>7)</sup> Currently, this work is being extended to establish the flexibility of this methodology to multiply substituted furans available in nature.

<sup>#</sup> Spectral data for <u>3a</u>: IR (neat), 2300, 2240, 1120, and 1050 cm<sup>-1</sup>; <sup>1</sup>H NMR (60 MHz,  $CCl_A$ ),  $\delta$  4.75(1H,t,J=6 Hz), 4.15(2H,t,J=2 Hz), 3.60(2H,m), 3.3(2H,d,J=6 Hz), 2.15 (2H,m), 1.2-1.8(8H,m), 1.2(3H,t,J=7 Hz), 0.9(3H,t,J=6 Hz); <u>4a</u> IR (neat), 1190, 1130, 1100, 1050, 1030, 1000, 930 cm<sup>-1</sup>; <sup>1</sup>H NMR (60 MHz,  $CCl_4$ ), $\delta$ 5.1-5.4(1H,m,olefinic), 5.1(1H, m,H-2), 4.15-4.35(2H,m,H-5), 3.2-3.9(2H,m,-OCH<sub>2</sub>CH<sub>3</sub>), 2.25-2.6(2H,m,H-3), 1.7-2.2(2H,m,allylic), 0.8-1.6(14H,m); la IR (neat), 1500, 1460, 1030, 780, 730 cm<sup>-1</sup>; <sup>1</sup>H NMR (60 MHz, CCl<sub>4</sub>), 7.25(1H,t,J=1.5 Hz), $\delta$  7.15(1H,br s), 6.2(lH,br s), 2.4(2H,br t,J=7 Hz,), 1.25(10H,m) 0.85(3H,t). Similarly, all other compounds gave satisfactory spectral data. Perillene (ld) and Dendrolasin (le) exhibited spectral data identical to those reported in literature.

## References

- Selectivity and synthetic application of radical cyclisation reactions, Tetra-1) hedron Symposia in Print, ed byB. Giese, Tetrahedron, <u>41</u>, 3887-4302 (1985); A. Srikrishna, Current Science, <u>56</u>, 392 (1987). Y.Ueno, O. Moriya, K. Chino, M. Watanabe, and M. Okawara, J. Chem. Soc., Perkin Trans. 1, <u>1986</u>, 1351. A. Srikrishna, J. Chem. Soc., Chem. Commun., <u>1987</u>, 587. T. Mandai, M. Takeshita, K. Mori, M. Kawada, and J. Otera, Chem. Lett., 1983. 1909 and references cited therein
- 2)
- 3)
- 4) 1983, 1909 and references cited therein.
- 5) R. Bernardi, C. Cardani, D. Ghiringhelli, A. Selva, A. Baggini, and M. Pavan, Tetrahedron Lett., <u>1967</u>, 3893.
- 6)
- A. Quilico, F. Piozzi, and M. Pavan, Tetrahedron, 1, 177 (1957);
  Y. Hirose, M. Abu, and Y. Sekiya, Nippon Kagaku Zasshi, 82, 725 (1961). For a similar methodology using haloacetal route for 3-alkyl furans see G. Stork, and R. Mook, J. Am. Chem. Soc., <u>105</u>, 3720 (1983). 7)
- (Received August 17, 1987)