

Central European Journal of Chemistry

Synthesis of 1-(pyrrolidin-2-ylmethyl)-1*H*-azoles and their piperidine-derived homologues

Research Article

Sergey Zhersh^{1,2}, Oleksandr V. Karpenko¹, Vasyl Ripenko¹, Andrey A. Tolmachev^{1,2}, Oleksandr O. Grygorenko^{2*}

¹Enamine Ltd., Kyiv 01103, Ukraine

²Department of Chemistry, Taras Shevchenko National University of Kyiv, Kyiv 01601, Ukraine

Received 10 July 2013; Accepted 2 September 2013

Abstract: A convenient preparation of 1-(pyrrolidin-2-yl)-1*H*-pyrazoles, -imidazoles, and -1*H*-1,2,4-triazoles, 1-(piperidin-2-yl)-1*H*-pyrazoles and -1*H*-1,2,4-triazoles, and 1-(piperidin-3-yl)-1*H*-1,2,4-triazoles by alkylation of azoles (*viz.* pyrazoles, imidazoles, and triazoles) with *N*-Cbz-prolinol mesylate or its analogues and subsequent deprotection is reported. The two-step method allows for synthesis of the title compounds in 16–65% yields. The utility of the procedure has been demonstrated by multigram preparation of a 15-member building block mini-library for the lead-oriented synthesis of compound libraries. These building blocks perfectly fit the definition of low-molecular-weight hydrophilic three-dimensional templates, which leave much room for the lead-oriented synthesis of the compound libraries.

Keywords: Azoles • Saturated nitrogen heterocycles • Molecular rigidity • Lead-oriented synthesis • Alkylation © Versita Sp. z o.o.

1. Introduction

Contemporary drug discovery relies heavily on the careful choice of chemotypes that can serve as the starting points for the design and synthesis of compound libraries. The so-called lead-likeness methodology for drug design, which relies on a number of concepts such as privileged structures, three-dimensional molecular shape, conformational restriction, and control of physicochemical properties of the potential lead compounds, have received much attention in the chemical literature recently [1–4]. The need for the "lead-oriented synthesis" [5] has guided the efforts of organic chemists towards novel low-molecular-weight hydrophilic three-dimensional scaffolds.

Among the numerous possible ways to design such structures we have turned our attention to the approach shown schematically in Fig. 1: a highly polar aromatic heterocycle is mounted onto a three-dimensional saturated template bearing an easily modifiable functional group (such as amino) [6–9]. The use of this approach is encountered among natural compounds (e.g. alkaloids

nicotine **1** and anabasine **2**, Fig. 2). Other examples comprise of synthesis of compounds of established biological activity, *e.g.* inhibitors of deacetylases (**3**) [10], tyrosine kinases (**4**) [11], or thrombin-activated fibrinolyse (**5**) [12]. In this work, we have focused on the scaffolds **6** – **11** obtained by linking the azole and pyrrolidine (or piperidine) heterocyclic rings by a single methylene unit (Table 1). Some examples of biologically relevant compounds derived from these templates are given in Fig. 2. Recently, derivatives of **7** have attracted additional attention as enantioselective catalysts [13,14] and chiral ionic liquids [15–19].

2. Experimental procedures

2.1. General

The solvents were purified according to the standard procedures. Compounds **13**, **17**, and **18** were prepared according to the methods reported in the literature [20]. All other starting materials were purchased from Acros, Merck, Fluka, and UkrOrgSyntez. All the chiral starting

^{*} E-mail: gregor@univ.kiev.ua

materials were used as racemates. Analytical TLC was performed using Polychrom SI F254 plates. Column chromatography was performed using Kieselgel Merck 60 (230–400 mesh) as the stationary phase. ¹H and ¹³C NMR spectra were recorded on a Bruker 170 Avance 500 spectrometer (at 499.9 MHz for Protons and 124.9 MHz for Carbon-13). Chemical shifts are reported in ppm downfield from TMS (¹H, ¹³C) as an internal standard. Elemental analyses were performed at the Laboratory of Organic Analysis, Department of Chemistry, Kyiv National Taras Shevchenko University. Mass spectra were recorded on an Agilent 1100 LCMSD SL instrument (chemical ionization (CI)) or Agilent 5890 Series II 5972 GCMS instrument (electron impact ionization (EI)).

2.2. General procedure for the synthesis of 6–11

To a suspension of NaH (0.12 mol) in DMF (200 mL), compound 14, 15, or 16 (0.1 mol) in DMF (100 mL) was added. After the hydrogen evolution was complete, compound 13, 17, or 18 was added, and the resulting mixture was heated at 60°C (14) or 100°C (15 or 16) for 8 h, then cooled and evaporated in vacuo. The residue was quenched with H₂O (1000 mL) and EtOAc (300 mL). The organic phase was separated, and aqueous phase was extracted with EtOAc (2×300 mL). The combined organic extracts were washed with brine (2×400 mL), dried over Na₂SO₄, and evaporated to dryness. The residue was purified by column chromatography (Hexanes - EtOAc (1:1) as an eluent) and then dissolved in MeOH (400 mL). 10% Pd-C (2-3 g) was added, and the mixture was hydrogenated at 1 bar until the reaction was complete (NMR control, usually 1 day). The catalyst was filtered off, the filtrate was evaporated, and the crude product was purified by column chromatography



Figure 1. Scaffolds for lead-oriented synthesis By combining saturated and aromatic nitrogen heterocycles.

 $(CH_2CI_2 - MeOH - Et_3N (9 : 1 : 0.1))$ to give **6** - **11** (Table 1). To obtain the final compounds as dihydrochlorides, **6** - **11** (0.05 mol) were taken up in 10% HCl in dioxane (50 mL) and evaporated to dryness.

1-(Pyrrolidin-2-ylmethyl)-1*H***-pyrazole (6a):** yield 58 g (44%). Yellowish liquid. ¹H NMR (CDCl₃), δ 7.47 (s, 1H), 7.42 (s, 1H), 6.20 (s, 1H), 4.15 (dd, *J* = 13.6 Hz and 4.8 Hz, 1H), 3.99 (dd, *J* = 13.6 Hz and 7.9 Hz, 1H), 3.49–3.56 (m, 1H), 2.91–2.96 (m, 1H), 2.83–2.89 (m, 1H), 2.14 (s, 1H), 1.79–1.90 (m, 1H), 1.64–1.79 (m, 2H), 1.36–1.44 (m, 1H). ¹³C NMR (CDCl₃), δ 139.5, 129.7, 105.3, 58.6, 57.1, 46.4, 29.0, 25.2. Anal. calcld. for C₈H₁₃N₃ C 63.55, H 8.67, N 27.79. Found C 63.94, H 8.30, N 27.96. MS (APCl): 152 (MH⁺).

4-Methyl-1-(pyrrolidin-2-ylmethyl)-1H-pyrazole (**6b**): yield 106 g (65%). Yellowish liquid. ¹H NMR (CDCl₃), δ 7.22 (s, 1H), 7.14 (s, 1H), 4.03 (dd, *J* = 13.6 Hz and 4.4 Hz, 1H), 3.87 (dd, *J* = 13.6 Hz and 7.7 Hz, 1H), 3.42–3.48 (m, 1H), 2.87–2.92 (m, 1H), 2.78–2.84 (m, 1H), 2.28 (s, 1H), 1.99 (s, 3H), 1.75–1.84 (m, 1H), 1.62– 1.73 (m, 2H), 1.30–1.39 (m, 1H). ¹³C NMR (CDCl₃), δ 139.7, 128.5, 115.7, 58.5, 56.8, 46.3, 28.9, 25.1, 8.8. Anal. calcld. for C₉H₁₅N₃ C 65.42, H 9.15, N 25.43. Found C 65.63, H 8.94, N 25.11. MS (APCI): 166 (MH⁺).

3,5-Dimethyl-1-(pyrrolidin-2-ylmethyl)-1*H***pyrazole (6c) dihydrochloride:** yield 29 g (19%). White powder. Mp 178–179 °C. ¹H NMR (DMSO-*d*⁶), δ 10.59 (br s, 1H), 9.93 (br s, 1H), 9.83 (br s, 1H), 6.12 (s, 1H), 4.58 (dd, *J* = 14.7 Hz and 7.5 Hz, 1H), 4.44 (dd, *J* = 14.7 Hz and 5.7 Hz, 1H), 3.86–3.95 (m, 1H), 3.21–3.28 (m, 1H), 3.07–3.14 (m, 1H), 2.35 (s, 3H), 2.20 (s, 3H), 1.91–2.04 (m, 2H), 1.80–1.89 (m, 1H), 1.64–1.72 (m, 1H). ¹³C NMR (DMSO-*d*⁶), δ 146.4, 143.2, 106.7, 58.9, 47.9, 44.9, 28.0, 22.8, 12.6, 11.3. Anal. calcld. for C₁₀H₁₉Cl₂N₃ C 47.63, H 7.59, CI 28.12, N 16.66. Found C 47.50, H 7.74, CI 27.89, N 16.38. MS (APCI): 180 (MH⁺).

1-(Pyrrolidin-2-ylmethyl)-1H-imidazole(7a):yield 82 g (54%). White amorphous solid. ¹H NMR (D_2O) , δ 7.83 (s, 1H), 7.29 (s, 1H), 7.14 (s, 1H), 4.49(dd, J = 14.7 Hz and 4.5 Hz, 1H), 4.41 (dd, J = 14.7 Hzand 8.9 Hz, 1H), 4.00–4.07 (m, 1H), 3.41–3.46 (m,1H), 3.32–3.38 (m, 1H), 2.24–2.30 (m, 1H), 2.03–2.18



Figure 2. Biologically active derivatives of the scaffolds defined in Fig. 1 (the scaffolds are shown in red).

Entry No.	Substrate	Azole	Product	Yield, %
1	OMs N Cbz 13	N N H 14a	N H 6a	44
2		N N H 14b		65
3		N N H H 14c	N N N H 6c	19
4		N N H 15a		54
5		N N H 15b		43
6		N N H 15c	N N H 7c	16
7		N N H 15d	N N H 7d	17
8		N N H 16a	N N H 8a	56
9		N N H H 16b		23
10	Cl Cbz 17	N N H 14a	N N H 9a	22
11		N N H 14c	N H 9b	26

Table	1.	. Synthesis of 1-(pyrrolidinylmethyl)-1H-azoles and their piperidine-derived homologues 6 - 11. Conditions: (i) NaH (1 eq), DMF, 60°C
		or 100°C, 8 h; (ii) H ₂ , 10% Pd-C, MeOH, 1 atm, rt, monitored by NMR.



Continued Table 1. Synthesis of 1-(pyrrolidinyImethyl)-1*H*-azoles and their piperidine-derived homologues **6** – **11**. Conditions: (i) NaH (1 eq), DMF, 60°C or 100°C, 8 h; (ii) H₂, 10% Pd-C, MeOH, 1 atm, rt, monitored by NMR.

(m, 2H), 1.79–1.87 (m, 1H). ^{13}C NMR (D₂O), δ 138.0, 128.5, 120.1, 60.4, 47.3, 45.9, 27.6, 22.8. Anal. calcld. for C₈H₁₃N₃ C 63.55, H 8.67, N 27.79. Found C 63.42, H 8.76, N 27.65. MS (APCI): 152 (MH⁺).

2-Methyl-1-(pyrrolidin-2-ylmethyl)-1*H*-imidazole (7b): yield 71 g (43%). Yellowish liquid. ¹H NMR (DMSO-*d*⁶), δ 7.05 (s, 1H), 6.70 (s, 1H), 3.71–3.80 (m, 2H), 3.28 (quint, *J* = 6.7 Hz, 1H), 3.09 (br s, 1H), 2.79 (t, *J* = 6.7 Hz, 2H), 2.29 (s, 3H), 1.65–1.77 (m, 2H), 1.55– 1.64 (m, 1H), 1.28–1.36 (m, 1H). ¹³C NMR (DMSO-*d*⁶), δ 144.3, 126.4, 120.4, 58.9, 50.8, 46.4, 29.5, 25.6, 13.4. Anal. calcld. for C₉H₁₅N₃ C 65.42, H 9.15, N 25.43. Found C 65.17, H 9.49, N 25.36.

MS (EI): 165 (M⁺), 96, 70.

2-Ethyl-1-(pyrrolidin-2-ylmethyl)-1*H*-imidazole (7c): yield 23.5 g (16%).White amorphous solid. ¹H NMR (D₂O), δ 7.18 (s, 1H), 7.02 (s, 1H), 4.29–4.38 (m, 2H), 3.91 (quint, *J* = 7.4 Hz, 1H), 3.39–3.44 (m, 1H), 3.27–3.33 (m, 1H), 2.77 (q, *J* = 7.3 Hz, 2H), 2.19–2.25 (m, 1H), 2.00–2.17 (m, 2H), 1.74–1.82 (m, 1H), 1.28 (t, *J* = 7.3 Hz, 3H). ¹³C NMR (D₂O), δ 150.6, 125.6, 120.4, 59.8, 46.3, 45.8, 27.7, 22.6, 19.2, 11.2. Anal. calcld. for C₁₀H₁₇N₃ C 67.00, H 9.56, N 23.44. Found C 67.32, H 9.47, N 23.15. MS (APCI): 180 (MH⁺).

2-IsopropyI-1-(pyrrolidin-2-yImethyI)-1*H***imidazole (7d):** yield 14.5 g (17%). Yellowish liquid. ¹H NMR (CDCl₃), δ 6.93 (s, 1H), 6.86 (s, 1H), 3.86 (dd, *J* = 13.8 Hz and 5.4 Hz, 1H), 3.77 (dd, *J* = 13.8 Hz and 7.8 Hz, 1H), 3.34 (quint, *J* = 6.4 Hz, 1H), 2.94–3.06 (m, 2H), 2.87–2.95 (m, 1H), 1.94 (br s, 1H), 1.77–1.92 (m, 2H), 1.69–1.77 (m, 1H), 1.36–1.44 (m, 1H), 1.31 (d, J = 7.0 Hz, 3H). 1.29 (d, J = 7.0 Hz, 3H). ¹³C NMR (CDCl₃), δ 153.1, 127.2, 119.0, 59.1, 50.9, 46.4, 29.3, 26.0, 25.1, 22.1, 22.0. Anal. calcld. for C₁₁H₁₉N₃ C 68.35, H 9.91, N 21.74. Found C 68.07, H 10.02, N 21.70. MS (APCI): 194 (MH⁺).

1-(Pyrrolidin-2-ylmethyl)-1H-1,2,4-triazole (8a): yield 56.6 g (56%). Yellowish liquid. ¹H NMR (CDCl₃), δ 8.15 (s, 1H), 7.90 (s, 1H), 4.17 (dd, *J* = 13.8 Hz and 3.6 Hz, 1H), 4.05 (dd, *J* = 13.8 Hz and 8.0 Hz, 1H), 3.37 (quint, *J* = 6.6 Hz, 1H), 2.92 (td, *J* = 6.6 Hz and 1.0 Hz, 2H), 2.34 (br s, 1H), 1.88–1.94 (m, 1H), 1.67– 1.80 (m, 2H), 1.40–1.47 (m, 1H). ¹³C NMR (CDCl₃), δ 151.7, 143.5, 57.4, 54.7, 46.4, 28.9, 25.3. Anal. calcld. for C₇H₁₂N₄ C 55.24, H 7.95, N 36.81. Found C 54.99, H 8.17, N 36.52. MS (APCI): 153 (MH⁺).

3,5-Dimethyl-1-(pyrrolidin-2-ylmethyl)-1H-1,2,4triazole (8b): yield 35 g (23%). Yellowish liquid. ¹H NMR (CDCl₃), δ 3.89 (dd, J = 13.7 Hz and 4.9 Hz, 1H), 3.79 (dd, J = 13.7 Hz and 7.8 Hz, 1H), 3.49 (quint, J = 6.8 Hz, 1H), 2.81–2.90 (m, 2H), 2.34 (s, 3H), 2.23 (s, 3H), 2.18 (br s, 1H), 1.78–1.85 (m, 1H), 1.61–1.77 (m, 2H), 1.32–1.40 (m, 1H). ¹³C NMR (CDCl₃), δ 159.4, 152.6, 58.0, 53.1, 46.4, 29.1, 25.3, 13.9, 12.1. Anal. calcld. for C₉H₁₆N₄ C 59.97, H 8.95, N 31.08. Found C 59.96, H 8.61, N 31.33. MS (EI): 180 (M⁺), 111, 70.

2-(1*H***-pyrazol-1-ylmethyl)piperidine (9a):** yield 24.4 g (22%). Yellowish liquid. ¹H NMR (CDCl₃), δ 7.51



Scheme 1. 1-(Pyrrolidinylmethyl)-1H-azoles reported in the literature.

(s, 1H), 7.40 (s, 1H), 6.22 (s, 1H), 4.10 (dd, J = 13.6 Hz and 4.1 Hz, 1H), 3.94 (dd, J = 13.6 Hz and 8.7 Hz, 1H), 2.97–3.04 (m, 2H), 2.55 (td, J = 11.5 Hz and 2.5 Hz, 1H), 2.04 (br s, 1H), 1.79 (d, J = 11.5 Hz, 1H), 1.60 (t, J = 14.0 Hz, 2H), 1.27–1.46 (m, 2H), 1.14 (qd, J = 11.8 Hz and 3.4 Hz, 1H). ¹³C NMR (CDCl₃), δ 139.8, 130.1, 105.2, 58.1, 56.5, 46.6, 30.1, 26.0, 24.3. Anal. calcld. for C₉H₁₅N₃ C 65.42, H 9.15, N 25.43. Found C 65.73, H 9.06, N 25.62.

MS (APCI): 166 (MH⁺).

2-[(3,5-Dimethyl-1H-pyrazol-1-yl)methyl] piperidine (9b): yield 12 g (26%). Yellowish liquid. ¹H NMR (CDCl₃), δ 5.73 (s, 1H), 3.78–3.88 (m, 2H), 2.97– 3.06 (m, 2H), 2.56 (td, J = 11.8 Hz and 2.5 Hz, 1H), 2.48 (br s, 1H), 2.20 (s, 3H), 2.18 (s, 3H), 1.72–1.80 (m, 1H), 1.53–1.60 (m, 2H), 1.25–1.47 (m, 2H), 1.10–1.20 (m, 1H). ¹³C NMR (CDCl₃), δ 147.7, 139.5, 104.7, 56.7, 54.1, 46.6, 30.2, 26.0, 24.4, 13.5, 11.2. Anal. calcld. for C₁₁H₁₉N₃ C 68.35, H 9.91, N 21.74. Found C 68.26, H 10.15, N 21.51. MS (APCI): 194 (MH⁺).

2-(1*H***-1,2,4-triazol-1-ylmethyl)piperidine (10a):** yield 24.7 g (22%). Yellowish liquid. ¹H NMR (CDCl₃), δ 8.04 (s, 1H), 7.89 (s, 1H), 4.08 (dd, *J* = 13.6 Hz and 4.2 Hz, 1H), 3.99 (dd, *J* = 13.6 Hz and 8.3 Hz, 1H), 2.93–3.00 (m, 2H), 2.52 (td, *J* = 11.7 Hz and 2.6 Hz, 1H), 1.75–1.79 (m, 2H), 1.52–1.61 (m, 2H), 1.24–1.40 (m, 2H), 1.06–1.16 (m, 1H). ¹³C NMR (CDCl₃), δ 152.3, 143.7, 55.8, 55.5, 46.5, 30.0, 25.9, 24.2. Anal. calcld. for C₈H₁₄N₄ C 57.81, H 8.49, N 33.70. Found C 58.06, H 8.49, N 34.01. MS (APCI): 167 (MH⁺).

2-[(3,5-Dimethyl-1H-1,2,4-triazol-1-yl)methyl] piperidine (10b): yield 19 g (60%). Yellowish liquid. ¹H NMR (CDCl₃), δ 3.81–3.90 (m, 2H), 2.98–3.04 (m, 2H), 2.57 (td, *J* = 11.6 Hz and 1.7 Hz, 1H), 2.38 (s, 3H), 2.29 (s, 3H), 2.03 (br s, 1H), 1.79 (d, *J* = 13.5 Hz, 1H), 1.56– 1.62 (m, 2H), 1.28–1.46 (m, 2H), 1.11–1.21 (m, 1H). ¹³C NMR (CDCl₃), δ 159.5, 152.6, 56.1, 53.7, 46.5, 30.0, 25.8, 24.1, 13.7, 11.9. Anal. calcld. for C₁₀H₁₈N₄ C 61.82, H 9.34, N 28.84. Found C 61.89, H 9.07, N 29.10. MS (APCI): 195 (MH⁺).

3-(1*H***-1,2,4-triazol-1-ylmethyl)piperidine** (11a) **dihydrochloride:** yield 150 g (52%). White powder. Mp 198–200 °C. ¹H NMR (D₂O), δ 9.65 (s, 1H), 8.69 (s, 1H), 4.48–4.52 (m, 2H), 3.46 (t, J = 11.7 Hz, 2H), 2.91–3.00 (m, 2H), 2.52–2.59 (m, 1H), 2.01 (d, J = 14.8 Hz, 1H), 1.89 (d, J = 12.5 Hz, 1H), 1.72–1.81 (m, 1H), 1.41 (qd, J = 12.5 Hz and 3.5 Hz, 1H). ¹³C NMR (D₂O), δ 144.6, 142.1, 53.6, 45.9, 44.1, 33.5, 25.2, 21.3. Anal. calcld. for C₈H₁₆Cl₂N₄ C 40.18, H 6.74, N 23.43. Found C 39.87, H 6.51, N 23.62. MS (APCI): 167 (MH⁺).

3-[(3,5-Dimethyl-1H-1,2,4-triazol-1-yl)methyl] piperidine (11b): yield 87.6 g (45%). Yellowish liquid. ¹H NMR (CDCl₃), δ 3.70–3.82 (m, 2H), 2.86 (t, *J* = 13.6 Hz, 2H), 2.50 (td, *J* = 11.8 Hz and 2.1 Hz), 2.30 (s, 3H), 2.22 (s, 3H), 1.97 (br s, 1H), 1.56–1.71 (m, 3H), 1.31– 1.40 (m, 1H), 1.04–1.13 (m, 1H). ¹³C NMR (CDCl₃), δ 159.2, 152.1, 51.4, 50.2, 46.8, 37.6, 28.8, 25.5, 13.7, 11.9. Anal. calcld. for C₁₀H₁₈N₄ C 61.82, H 9.34, N 28.84. Found C 61.64, H 9.48, N 29.08. MS (APCI): 195 (MH⁺).

3. Results and discussion

Only a few reports on the synthesis of compounds 6 - 11 can be found in the literature: namely, the preparation of compounds 6a [13] and 7a [14,18] has been documented (Scheme 1). Both compounds were obtained by alkylation of the corresponding azole anions with *N*-Boc-prolinol tosylate (12). For the preparation of compounds 6 - 8, we have used an analogous method commencing from *N*-Cbz-prolinol mesylate (13) (Table 1), which is more reactive than *N*-Boc-prolinol tosylate, is shelf-stable and can be easily obtained on a hundred gram scale [20]. Azoles 14 - 16 were used to generate anions for the reaction with 13 (Table 1, Entries 1–9).

It was found that anions of pyrazoles **14a–c** were more reactive towards alkylation with **13** than anions of imidazoles **15a–d** or 1,2,4-triazoles **16a,b**. Whereas **13** and anions of **14a–c** reacted smoothly in DMF already at 60°C, in the case of **15a–d** or **16a,b**, the reaction was successful only at 100°C. It should be noted that in the case of **16**, the alkylation proceeded regioselectively at N-1 atom of the heterocycle. The corresponding Cbz derivatives were not characterized but subjected to catalytic hydrogenation (10% Pd-C, MeOH, 1 atm, rt) to give pure **6** – **8** in 16–65% overall yields.

Table	2.	Physicochemical parameters of N-methyl derivatives of the
		amines 6 – 11.

Parameter	Value range
Molecular weight (MW)	165207
Calculated logarithm of octanol-water partition coefficient (cLogP)	0.201.93
Number of H-bond donors	0
Number of H-bond acceptors	23
Fraction of sp ³ carbon atoms (Fsp ³)	0.670.82
Number of rotatable bonds (RotB)	23
Total polar surface area (TPSA), ${\rm \AA}^2$	21.034.0



Figure 3. Physicochemical parameters of *N*-methyl derivatives of the amines 6 – 11 (black dots within the ellipse), shown together with the lead-likeness criteria as given by Churcher *et al.* [5] (dark grey rectangle). The white arrow shows that there is much room for the design of compound libraries.

As was described previously by our group [20], the mesylate of benzyl 2-(hydroxymethyl)piperidine-1-carboxylate is not stable enough to be isolated. Therefore the corresponding chloride **17** was used in the reaction with the azole anions (Table 1, Entries 10–13). This change did not affect the outcome of the reaction sequence described above: the corresponding products

References

- [1] M.P. Gleeson, J. Med. Chem. 51, 817 (2008)
- [2] T.I. Oprea, A.M. Davis, S.J. Teague, P.D. Leeson, J. Chem. Inf. Comp. Sci. 41, 1308 (2001)
- [3] P.D. Leeson, B. Springthorpe, Nat. Rev. Drug Discovery 6, 881 (2007)
- [4] A. Polinsky, In: C. G. Wermuth (Ed.), Practice of Medicinal Chemistry, 3rd edition (Academic Press/ Elsevier, Amsterdam, 2008) 244
- [5] A. Nadin, C. Hattotuwagama, I. Churcher, Angew.

9a,b and **10a,b** were obtained in 22–60% yields. The method was equally efficient for the preparation of 1-(pyrrolidin-2-yl)-1*H*-1,2,4-triazoles **11a,b** (45–52% yields) starting from mesylate **18** [20] and anions of **16a,b** (Table 1, Entries 14 and 15).

Compounds **6** – **11** were obtained as rather hygroscopic liquids or oils, therefore any aqueous work-up should be avoided during their isolation and/ or purification. In some cases, the compounds were isolated as dihydrochlorides; 10% HCl in dioxane was used for that purpose. Although the synthetic scheme included chromatographic purifications, it was easily scaled up for the preparation of 100 g of the final products in a single run.

Analysis of the physicochemical properties of the simplest (*N*-methyl) derivatives of the amines 6 - 11 (Table 2) showed that these building blocks perfectly fit the definition of low-molecular-weight hydrophilic threedimensional templates [21]. They provide an opportunity for the lead-oriented synthesis of the compound libraries even if the strictest current criteria of lead-likeness are considered [5] (Fig. 3).

4. Conclusions

A convenient approach to multigram synthesis of 1-(pyrrolidin-2-yl)-1*H*-pyrazoles, -imidazoles, and -1*H*-1,2,4-triazoles, 1-(piperidin-2-yl)-1*H*-pyrazoles and -1*H*-1,2,4-triazoles, and 1-(piperidin-3-yl)-1*H*-1,2,4-triazoles was developed. The method allowed for the preparation of the target compounds in 16–65% yields. The building blocks obtained comply with the strictests definitions of templates for lead-oriented synthesis of the compound libraries.

Acknowledgements

The authors thank Prof. Igor V. Komarov for his help with Instant Jchem calculations.

Chem. Int. Ed. 51, 1114 (2012)

- [6] N.V. Shevchuk, K. Liubchak, K.G. Nazarenko, A.A. Yurchenko, D.M. Volochnyuk, O.O. Grygorenko, A.A. Tolmachev, Synthesis 44, 2041 (2012)
- [7] A.V. Borisov, V.V. Voloshchuk, M.A. Nechayev, O.O. Grygorenko, Synthesis 45, 2413 (2013)
- [8] A.G. Yaremenko, D.M. Volochnyuk, V.V. Shelyakin, O.O. Grygorenko, Tetrahedron 69, 6799 (2013)
- [9] R. Kombarov, A. Altieri, D. Genis, M. Kirpichenok,

V. Kochubey, N. Rakitina, Z. Titarenko, Mol. Diversity 14, 193 (2010)

- [10] C.A. Brooks, C.H.-T. Chen, Y.S. Cho, L. Jiang, G. Liu, M. Shultz, PCT International Patent WO 2012/025164, 2012.03.01
- [11] J.J. Cui, L.A. Funk, L. Jia, P.-P. Kung, J.J. Meng, M.D. Nambu, M.A. Pairish, H. Shen, M.B. Tran-Dube, PCT International Patent WO 2006/021881, 2006.03.02
- [12] C. Kallus, H. Heitsch, A. Lindenschmidt, S. Grueneberg, H. Szillat, PCT International Patent WO 2005/105781, 2005.11.10
- [13] S. Chandrasekhar, T.P. Kumar, K. Haribabu, C.R. Reddy, C.R. Kumar, Tetrahedron: Asymmetry 22, 697 (2011)
- [14] D. Dang, P. Wu, C. He, Z. Xie, C. Duan, J. Am. Chem. Soc. 132, 14321 (2010)

- [15] L. Zhang, J.-P. Cheng, S. Luo, X. Mi, S. Liu, Y. Qiao, H. Xu, Org. Biomol. Chem. 6, 567 (2008)
- Y. Abe, T. Hirakawa, S. Nakajima, N. Okano,
 S. Hayase, M. Kawatsura, T. Itoh, Y. Hirose, Adv. Synth. Cat. 350, 1954 (2008)
- [17] S. Luo, X. Mi, L. Zhang, S. Liu, H. Xu, J.-P. Cheng, Tetrahedron 63, 1923 (2007)
- [18] S. Luo, X. Mi, L. Zhang, S. Liu, H. Xu, J.-P. Cheng, Angew. Chem. Int. Ed. 45, 3093 (2006)
- [19] D. Xu, S. Luo, H. Yue, L. Wang, Y. Liu, Z. Xu, Synlett 2569 (2006)
- [20] S. Zhersh, V.V. Buryanov, O.V. Karpenko, O.O. Grygorenko, A.A. Tolmachev, Synthesis 3669 (2011)
- [21] Instant JChem 6.0.2 (ChemAxon, Budapest, Hungary, 2013) (www.chemaxon.com)