PRESSURE INDUCED PHASE TRANSITION IN MnTe₂

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Energy dispersive high-pressure powder X-ray experiments have been performed for MnTe₂ up to a pressure of 20 GPa. MnTe₂ undergoes a discontinuous transformation from the cubic pyrite type structure to the orthorhombic marcasite type structure at 7.0 ± 0.5 GPa upon increasing pressure. The transformation is accompanied by a large reduction in the specific volume ($\Delta V/V = 0.18$) which probably reflects different magnetic properties of the two modifications of MnTe₂.

The TX_2 (T = Cr, Mn, Fe, Co, Ni; X = P, As, Sb, S, Se, Te) compounds often crystallize in one of several related structure types [1,2]. Among these, the pyrites (FeS₂-p), marcasites (FeS₂-m) and arsenopyrites (FeAsS-ap) have the largest number of representatives. As a function of temperature and composition (e.g. in ternary solid solution phases) conversions $m \rightarrow p$ are well established [3,4] and continuous second-order transitions are found between m- and p-type phases [2]. The MnX_2 (X = S, Se, Te) compounds [1,5,6] crystallize in the p type structure over their entire temperature range of stability. However, recently a structural transformation (probably associated with a change in spin state) was found in MnS₂-p, when hydrostatic pressure was applied [7,8]. The high pressure modification (not stable at ambient conditions) is of the m-type structure and is characterized relative to the p-type structure by taking a much lower unit cell volume per formula unit. Such relatively large variations in unit cell volume between different modifications most likely represent differences in physical (magnetic) properties [9]. A similar situation to that of MnS₂ may possibly exist for the isostructural phases MnSe₂ and MnTe₂. In this work we report on a high pressure energy dispersive X-ray diffraction study on $MnTe_2$.

Powder samples of MnTe₂ were synthesized from

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the elements (Mn, crushed flakes, 99.99%, Johnson, Matthey & Co.; Te, powder, 99.999%, Koch Light Laboratories) using the sealed silica capsule technique. After the first heat treatment at 650° C for 5 d the samples were cooled to room temperature and crushed before being subjected to two further similar annealing cycles (500° C, 10 d). The product obtained by this procedure contained only minor traces of free Te.

Powder X-ray diffraction data were collected at ambient conditions with a Guinier camera (Cu K α_1 radiation, Si as internal standard). The cubic lattice constant for MnTe₂-p (space group Pa3) a =695.32(12) pm agrees with already reported values [1,10]. High-pressure powder X-ray diffraction data were obtained by the energy dispersive technique, using a tungsten tube. Pressures up to 20 GPa were generated in a gasketed diamond anvil cell [11], and the pressure was determined using the ruby fluorescence technique [12]. The spectra were partly masked by the presence of Te characteristic radiation peaks and reflections resulting from the inconel gasket material. The positions of the Bragg reflections of MnTe₂ were obtained through a curve fitting procedure. Calculated powder X-ray intensities with the LAZY-PULVERIX programme [13] were used as guide in the indexing of the X-ray diagrams.



Fig. 1. Variation of the specific volume of $MnTe_2$ for pressures up to 18 GPa. Open and filled symbols refer to increasing and decreasing pressures, respectively. The full lines represent a fit of the Birch equation of state to the data points.

The p-type modification of MnTe₂ was found to exist at pressures up to 7.0 ± 0.5 GPa. The variation in specific volume with pressure is shown in fig. 1. At pressures p > 7.0 GPa the reflections characteristic for the p-type structure disappeared and a new set of reflections appeared. A meaningful interpretation of the pattern consisting of only 4-5 reflections requires input of some structural informations. As the high pressure modification of MnS₂ is unambiguously of the m type [7,8], we assume this to hold also for MnTe₂. The intensity of the observed reflections were in good agreement with those calculated for MnTe₂-m, using positional parameters for FeTe₂ [14] (space group Pnnm). The variation of the specific volume of MnTe₂-m with pressure is shown in fig. 1. The $p \rightarrow m$ transformation is accompanied by a volume decrease of about 18%. This volume collapse is somewhat larger than that found for MnS_2 (15%) [7]. However, the pressure required for triggering the transformation is lowered from 14 GPa for MnS₂ to 7 GPa for MnTe₂.

A fit of the Birch equation of state [15] to the data points yield the parameters $B_0 = 14.9$ GPa, $B'_0 = 16.3$ GPa, and $B_0 = 9.2$ GPa, $B'_0 = 4.4$ GPa for the p- and mtype modifications, respectively. Attempts to fit the data points to a zeroth-order Birch equation of state $(B'_0 = 4, \text{ fixed value}; B_0 \sim 40 \text{ GPa})$ were not satisfactory, indicating that the unusual large value of B'_0 in the pyrite type structure may be real. Due to the large scattering of the data points which results from overlapping characteristic lines, mentioned before, the values of B_0 and B'_0 may have errors of about 20% to 30%. Ultrasonic measurements should be made to improve the accuracy of B_0 resulting in an improvement of the error of B'_0 . The B_0 and B'_0 values in the marcasite type high pressure structure have been determined for completeness. However, they cannot be compared with the values in the pyrite type structure, as the marcasite structure is only hypothetical at low pressures.

Some TX₂ compounds (e.g. FeS₂ [4], OsTe₂ [4]) are known to crystallize with both the m- and the ptype structures, and transformations $m \rightarrow p$ are established as a function of temperature. For FeS₂ and OsTe₂ the pyrite modifications take the smallest specific volume ($\Delta V = -2.9 \times 10^6$ pm³ for OsTe₂). This shows, that only small differences in the specific volume are connected with the different packings of the X₂ dumbbells in the p-, m-, and ap-type structures. Therefore, the huge ΔV of the p \rightarrow m transformations in MnS₂ and MnTe₂ can probably not be attributed to the structural rearrangement (reorientation of the X₂²⁻ pair).

A main distinction between the MnX₂-p and other TX_2 compounds is the magnetic state of the metal atom. The MnX₂ compounds exhibit paramagnetic and long range antiferromagnetic properties associated with ~5 unpaired electrons [1,16,17] (d⁵ high spin state) contrary to the diamagnetic properties found in e.g. FeX_2 [1]. This difference in the magnetic properties is probably responsible for the observed reduction of the specific volume on going from MnX_2 to the corresponding FeX₂ compounds. Therefore we are tempted to conclude that the volume collapse accompanying the $p \rightarrow m$ transformation in MnS₂ and MnTe₂ reflects mainly a change of the magnetic state. However, for verification, measurements of spin dependent properties at high pressures are required. The Mössbauer data for 57 Fe substitution in MnX₂ [18,19] demonstrate a high to low spin transition for the iron atoms upon increasing pressure (i.e. going from the pto the m-type structure) and thus lend support to the assumption of coupled magnetic and structural transformations in MnX₂. A similar high pressure X-ray diffraction study of MnSe₂ is in progress.

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Volume 112A, number 8

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