A Synthesis of Talaromycin B

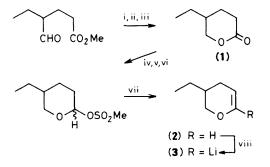
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The nucleophilic cleavage of the oxirane (4) by the organocuprate derived from 6-lithio-3-ethyl-3,4-dihydro-2*H*-pyran (3) was the key step in a synthesis of racemic talaromycin B (7).

The fungus *Talaromyces stipitatus* infects wood-shavingsbased chicken litter. Two isomeric toxic metabolites have been isolated from cultures of *T. stipitatus* which appear to block outward potassium fluxes and thus lead to muscle dysfunction.¹ Talaromycins A (8) and B (7) were identified as the toxic metabolites by an elegant application of twodimensional n.m.r. cross relaxation spectroscopy.² We report a brief synthesis³ of talaromycin B which confirms the relative stereochemistry.

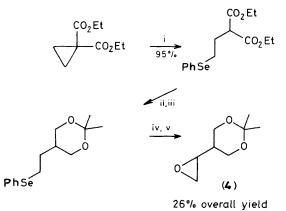
The key dihydropyran (2) [59% overall from (1); b.p. 140—141 °C at 760 mmHg; v_{max} (film) 3 070m, 1 650s, and 1 078s cm⁻¹; $\delta_{\rm H}$ (90 MHz, CDCl₃) 6.3 (1H, m), 4.65 (1H, m), 4.0 (1H, dd with further fine splitting, J 10, J' 2 Hz), 3.5 (1H, dd with further splitting, J 10, J' 8 Hz), 1.15—2.2 (5H, m), and 0.95 (3H, distorted t, J 7 Hz); M^+ 112.0888, C₇H₁₂O



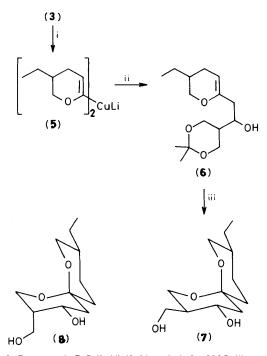
Scheme 1. Reagents: i, H₂, PtO₂-SnCl₂-60% aq. EtOH; ii, NaOH, room temperature; iii, conc. HCl; iv, Buⁱ₂AlH-toluene, -78°C; v, aq. AcOH; vi, MeSO₂Cl-pyridine; vii, 100-110°C, pyridine, 1¹/₂ h; viii, BuⁱLi-tetrahydrofuran.

requires *M* 112.088 81] and the oxirane (4) [b.p. 110–115 °C (bath) at 18 mmHg; v_{max} (film) 1 268, 1 250, 1 200, and 830 cm⁻¹ (all s); $\delta_{\rm H}$ (90 MHz, CDCl₃) 3.7–4.2 (4H, m), 3.03 (1H, ddd, *J* 7, *J*' 4.5, *J*" 4.0 Hz), 2.72 (1H, dd, *J* 4.5, *J*' 4.0 Hz), 2.54 (1H, dd, *J* 4.5, *J*' 3.0 Hz), 1.4 (6H, s), and 1.4–1.7 (1H, m); *M*⁺ 158.0941, C₈H₁₄O₃ requires *M* 158.094 288] were prepared by standard procedures as shown in Schemes 1 and 2, respectively.

The fulcrum of our synthetic plan involved the use of a 6-lithio-3,4-dihydro-2*H*-pyran as a masked bifunctional carbonyl anion equivalent.⁴ Central to the use of such acyl



Scheme 2. Reagents: i, PhSeNa–EtOH, room temperature, 18 h; ii, LiAlH₄–Et₂O, reflux, 4 h; iii, 2-methoxypropene, H^+ –CH₂Cl₂; iv, H_2O_2 (excess), pyridine (2 equiv.)–CH₂Cl₂; v, *m*-ClC₆H₄CO₃H–CH₂Cl₂.



Scheme 3. Reagents: i, CuI; ii, (4) (0.66 equiv.), 0-20 °C; iii, aq. HCl.

anion equivalents for the synthesis of the 4-hydroxy-1,7dioxaspiro[5.5]undecane moiety of talaromycin B is the recent discovery that relatively stable organocuprates⁵ prepared from 6-lithio-3,4-dihydro-2*H*-pyrans cleanly cleave monosubstituted oxiranes.⁶ Thus metallation of the dihydropyran (2) with 1.3 equivalents of Bu^tLi in tetrahydrofuran⁷ gave (3) which was converted into the organocuprate (5) with CuI (Scheme 3). Reaction of (5) with 0.66 equivalents of the oxirane (4) at 0–20°C gave the unstable intermediate (6) which without purification was treated with aqueous HCl. As expected, hydrolysis of the 1,3-dioxane moiety of (6) was accompanied by ring closure to the more thermodynamically stable¹ talaromycin B (7). Chromatographic purification on silica gel G eluting with 5:2 (v/v) benzene–dioxane gave (7) in 23% overall yield from (2) and (4). Recrystallisation from ethyl acetate–hexane gave pure (7) [m.p. (sealed tube) 135–136.5°C; v_{max} (KBr) 3 350, 1 380, 1 187, 1 085, 1 075, 1 060, 1 040, 1 035, 895, and 870 cm⁻¹ (all s); $\delta_{\rm C}$ (22.6 MHz, CDCl₃) 96.8, 65.9, 64.8, 61.4, 60.85, 45.9, 44.1, 36.6, 35.1, 25.1, 24.7, and 11.1 p.p.m. *M*⁺ 230.151 36, C₁₂H₂₂O₄ requires *M* 230.151 80]. The 400 MHz ¹H n.m.r. spectrum was identical in every detail with published spectra of natural talaromycin B.^{1,2}

By this route gram quantities of (7) can be prepared from cheap, readily available starting materials. Further applications of lithiated dihydropyrans to the synthesis of natural 4-hydroxy-1,7-dioxaspiro[5.5]undecanes are under investigation.

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