

Letter

Superconductivity of SrSi_2 and BaGe_2 with the $\alpha\text{-ThSi}_2$ -type structure

J. EVERS and G. OEHLINGER

Institut für Anorganische Chemie der Universität, Meiserstrasse 1, D-8000 München 2 (F.R.G.)

H. R. OTT

Laboratorium für Festkörperphysik ETH, CH-8093 Zürich, Hönggerberg (Switzerland)

(Received June 6, 1979)

Of various divalent metal disilicides MSi_2 and digermanides MGe_2 ($\text{M} \equiv \text{Yb, Ca, Eu, Sr, Ba}$) only EuSi_2 crystallizes in the tetragonal $\alpha\text{-ThSi}_2$ -type structure at ambient pressure [1]. The characteristic feature of this structure is a three-dimensional net built up of three-connected metalloid atoms. The repeat unit of four atoms is planar [2]. However, in this class of materials three additional phases with the $\alpha\text{-ThSi}_2$ -type structure are obtained when the MSi_2 and MGe_2 compounds are heated under pressure and are then quenched to ambient conditions. From X-ray powder investigations this structure was identified for the high pressure-high temperature (HPHT) polymorph of CaSi_2 [3, 4] and more recently, in a systematic study on these compounds, also for that of SrSi_2 [5] and BaGe_2 [6].

Up to now in these MSi_2 and MGe_2 compounds superconductivity has been observed only for the $\alpha\text{-ThSi}_2$ -type phases of CaSi_2 and defect SrSi_{2-x} ($x = 0.5$) [4]. EuSi_2 orders antiferromagnetically at 40 K [7] and is therefore not expected to become superconducting at lower temperatures. The purpose of this work was to search for possible superconductivity in stoichiometric SrSi_2 and BaGe_2 with the $\alpha\text{-ThSi}_2$ -type structure.

High purity samples of SrSi_2 and BaGe_2 were prepared by direct synthesis of electronic grade silicon or germanium respectively with ultrahigh-vacuum-distilled strontium or barium [8] in an inductively heated copper boat. Evaporation losses were determined in preliminary experiments. Ambient pressure phases were transformed by HPHT treatment in a belt-type apparatus [9]. Typical conditions were 40 kbar and 1000 °C for 15 min and subsequent quenching to ambient conditions.

The superconducting transition temperatures T_c were measured inductively using a standard low frequency mutual inductance bridge. We also checked the depression of the critical temperatures with increasing magnetic field.

TABLE 1

Superconducting transition temperatures T_c

<i>Composition</i>	T_c (K)
CaSi ₂	1.37 ± 0.02
CaSi _{1.9}	1.35 ± 0.04
CaSi _{2.2}	1.37 ± 0.02
SrSi ₂	3.1 ± 0.3
BaGe ₂	4.93 ± 0.1

In Table 1 we show our values of T_c for CaSi₂, SrSi₂ and BaGe₂. In the case of CaSi₂ our value is roughly 0.2 K lower than that reported previously [4]. For this reason we also measured T_c for off-stoichiometric samples but found very little influence of the silicon content on the transition temperature (see Table 1). A change of the ratio Ca:Si in the range up to about 10% also left the lattice parameter virtually unaffected; it remained constant within the experimental uncertainty. It may therefore be assumed that the α -ThSi₂-type structure is stable for stoichiometric CaSi₂ only. A reasonable explanation of the somewhat higher value of T_c obtained earlier for CaSi₂ cannot therefore be based on differences in stoichiometry but rather on a possible small content of strontium in the samples used in the former investigations. Commercially available calcium often contains more than 1% Sr impurity. In solid solutions Ca_{1-x}Sr_xSi₂ an increase of T_c above the value for pure CaSi₂ may be expected even for small x since T_c for SrSi₂ is more than twice as high as that of CaSi₂. It is interesting to note that BaGe₂, which in its ambient pressure phase is a semiconductor, has the highest value for T_c , above the values for CaSi₂ and SrSi₂ which in their ambient pressure phases are semimetals.

Unfortunately no clear evidence for a dependence of T_c on the atomic number of the metal atoms can be established for the silicon series. In our pressure and temperature range (45 kbar and 1500 °C) BaSi₂ does not form in the α -ThSi₂-type structure. Two other HPHT modifications are obtained instead, namely the SrSi₂- and EuGe₂-type structures. The same is true for the digermanides since in dimorphic SrGe₂ other structure types than that of α -ThSi₂ are found. However, this particular structure can be prepared in a limited range of solid solutions of BaGe_{2-x}Si_x and Ba_{1-x}Sr_xGe₂. Investigations of these compounds may show the tendency of increasing T_c with increasing radius of the metal atoms, as is indicated by our measurements.

Acknowledgment

We are grateful to Mr. Th. Siegrist and Mr. H. Thomas for their help in part of the experiments. We have also benefited from the financial support

of the Swiss National Science Foundation and the Deutsche Forschungsgemeinschaft.

- 1 J. A. Perri, I. Binder and B. Post, *J. Phys. Chem.*, *63* (1959) 616.
- 2 A. F. Wells, *Structural Inorganic Chemistry*, Clarendon Press, Oxford, 1975.
- 3 M. S. Silverman and J. R. Soulen, *J. Phys. Chem.*, *67* (1963) 1919.
- 4 D. B. McWhan, V. B. Compton, M. S. Silverman and J. R. Soulen, *J. Less-Common Met.*, *12* (1967) 75.
- 5 J. Evers, G. Oehlinger and A. Weiss, *J. Solid State Chem.*, *20* (1977) 173.
- 6 J. Evers, G. Oehlinger and A. Weiss, *Z. Naturforsch., Teil B*, *32* (1977) 1352.
- 7 M. Loewenhaupt, *Z. Phys.*, *267* (1974) 219.
- 8 J. Evers, E. Kaldis, J. Muheim and A. Weiss, *J. Less-Common Met.*, *30* (1973) 83; *31* (1973) 169.
- 9 K.-J. Range and R. Leeb, *Z. Naturforsch., Teil B*, *30* (1975) 889.