
Reaction of Organozinc Reagents Formed in Reaction of Dibromomalonic Acid Dialkyl Esters and Zinc, with 2-Arylmethylenemalonic Acid Dinitriles and 3-Aryl-2-cyanopropenoic Acid Methyl Esters

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Abstract—Organozine compounds, obtained from dibromomalonic acid dialkyl esters and zinc, react with 2-arylmethylmalonic acid dinitriles and 3-aryl-2-cyanopropenoic acid methyl esters forming 3-aryl-2,2-dicyanocyclopropane-1,1-dicarboxylic acid dialkyl esters and 3-aryl-2-cyanocyclopropane-1,1,2-tricarboxylic acid trimethyl esters. 1 H and 13 C NMR, $^{3}J_{CH}$ constants are considered. **DOI:** 10.1134/S1070363206120152

Earlier the organozinc reagents obtained from dialkyl esters of dibromomalonic acid have been applied for cyclopropanation of arylidenmalonic acid alkyl esters [1].

In the present work we have studied an interaction of organozinc reagents **IIa** and **IIb**, formed of di-

bromomalonic acid dialkyl esters **Ia** and **Ib** and zinc, with electrophilic substrates **IIIa–IIIg**, in order to obtain cyclopropane compounds containing in the ring two geminal alkoxycarbonyl or cyano groups and simultaneously another cyano- or ester group. The reaction is found to proceed according to the following scheme:

In the ether-tetrahydrofuran-HMPTA medium, organozinc compounds **IIa** and **IIb** react regiospecifically with electrophilic substrates **IIIa-IIIg**, attacking C³ carbon atom and giving intermediate com-

pounds **IVa–IVk**. Under the reaction conditions, intermediates **IVa–IVk** spontaneously undergo cyclization forming corresponding products of cyclopropanation **Va–Vk**, dialkyl 3-aryl-2,2-dicyanocyclopropane-

1,1-dicarboxylates and trimethyl 3-aryl-2-cyanocyclo-propane-1,1,2-triicarboxylates.

The structure of compounds Va-Vk was proven by analytical data and IR and ¹H NMR spectroscopy. The IR spectra of compounds Va-Vk contain characteristic absorption bands of the ester carbonyl groups $(1720-1725 \text{ cm}^{-1})$ and nitrile groups (2240-1720)2245 cm⁻¹). In ¹H NMR spectra of compounds Va-Vh we observed characteristic signals at 3.84-4.91 ppm, belonging to methine proton (CH), as well as proton signals of methyl- and ethyl-substituents of two ester groups. In ¹H NMR spectra of compounds Vi–Vk we observed characteristic signals belonging to methine proton (CH) at 3.80–3.87 ppm, and there were also proton signals of methyl substituents of three ester groups. The same set of proton signals indicates the formation of these substances as the same diastereoisomer.

The structure of compounds Vc and Vk was confirmed also by ¹³C NMR spectra recorded both with and without decoupling from protons. The ¹H and ¹³C signals assignment was made on the basis of twodimensional HSQC, HMBC and NOESY experiments. Signals of two nitrile and two carbonyl carbons in structure Vc were identified according to spin-spin coupling constants ${}^{3}J_{CH}$ with cyclopropane ring proton H³. The constants of the carbons located in *cis*-position to a proton ($\delta_{\rm CO}$ 163.26 ppm, $^3J_{\rm CH}$ 6.3 Hz; $\delta_{\rm CN}$ 111.45 ppm, $^3J_{\rm CH}$ 6.1 Hz) are larger than for *trans*-carbons ($\delta_{\rm CO}$ 161.23 ppm, $^3J_{\rm CH}$ 4.1 Hz; $\delta_{\rm CN}$ 109.36 ppm, ${}^3J_{\rm CH}$ 4.6 Hz). Signals of methoxy groups protons are assigned according to the cross peaks with corresponding carbonyls in HMBC spectrum: protons at 3.98 ppm give cross peak with carbon at 163.26 ppm, and protons at 3.80 ppm with carbon at 161.23 ppm. Carbons of methoxy groups are assigned according to the cross peaks with corresponding protons in HSQC spectrum (due to direct constants). These conclusions are confirmed by the 2D NOESY data for compound Vc: cis-OMe protons give cross peak with proton H³, and trans-OMe protons with ortho-protons of the aromatic substituent. In compound Vk the value of vicinal coupling constant between H^3 proton and carbon of nitrile group ${}^3J_{CH}$ 4.4 Hz allows to assume that in this structure interacting atoms occupy trans-position relative to the cyclopropane plane.

EXPERIMENTAL

IR spectra were recorded on a Specord IR-75 spectrometer for individual compounds dispersed in mineral oil. ¹H NMR spectra were measured from solutions of compounds **Va**, **Vd–Vj** in CDCl₃ on

Tesla BS-567A spectrometer at 100 MHz, and for compound **Vb** from solution in CDCl₃ on a Mercury–Plus-300 spectrometer at 300 MHz using HMDS as internal reference. 1D- and 2D-NMR spectra of compounds **Vc** and **Vk** from solutions in CDCl₃ were taken on a Bruker DRX-400 instrument (400 MHz for ¹H and 100 MHz for ¹³C) relative to TMS as internal reference.

Dialkyl 3-aryl-2,2-dicyanocyclopropane-1,1-dicarboxylates (VIa–VIh). A dibromomalonic acid dialkyl ester 0.024 mol was added to 2 g of zinc crushed in a fine shavings in 7 ml of ether and 10 ml of tetrahydrofuran. Thr mixture was heated up to the beginning of reaction, then reaction continued spontaneously. After the end of reaction the mixture was refluxed for 5 min, then cooled and poured off from zinc into a flask containing 0.02 mol of 2-arylmethylenemalonic dinitrile and 1,5 ml of HMPTA, and refluxed for 60 min. After cooling, the mixture was hydrolyzed with 5% acetic acid and reaction product was extracted with benzene, the solvents were evaporated and the residue was recrystallized from methanol.

Dimethyl 3-phenyl-2,2-dicyanocyclopropane-1,1-dicarboxylate (Va). Yield 76%, mp 125–126°C. IR spectrum (mineral oil), ν, cm $^{-1}$: 1720, 2240. 1 H NMR spectrum, δ, ppm: 3.72 s (3H, COOCH₃), 3.91 s (1H, CH), 3.95 s (3H, COOCH₃), 7.11–7.44 m (5H, C₆H₅). Found, %: C 63.29; H 4.20; N 9.74. C₁₅H₁₂N₂O₄. Calculated, %: C63.38; H 4.25; N 9.85.

Dimethyl 3-(4-bromophenyl)-2,2-dicyanocyclopropane-1,1-dicarboxylate (Vb). Yield 70%, mp $100-102^{\circ}$ C. IR spectrum (mineral oil), ν, cm⁻¹: 1725, 2245. ¹H NMR spectrum, δ, ppm: 3.79 s (3H, COOCH₃), 3.90 s (1H, CH), 3.97 s (3H, COOCH₃), 7.23–7.56 m (4H, BrC₆H₄). Found, %: C 49.54; H 2.98; N 7.63. C₁₅H₁₁BrN₂O₄. Calculated, %: C 49.61; H 3.05; N 7.71.

Dimethyl 3-(4-chlorophenyl)-2,2-dicyanocyclo-propane-1,1-dicarboxylate (**Vc**). Yield 77%, mp 103–104°C. IR spectrum (mineral oil), ν, cm⁻¹: 1720, 2245. ¹H NMR spectrum: (400 MHz, δ, ppm, *J*, Hz): 3.80 s (3H, OMe-*trans*); 3.93 t (1H, H³, 4*J* 0.8); 3.98 s (3H, OMe-*cis*); 7.31 dd (2H, H_o, *J* 8.8, 0.8); 8.74 d (2H, H_m, *J* 8.8). ¹³C NMR spectrum (100 MHz, δ, ppm, *J*, Hz): 16.46 d ($^{\text{C}2}$, *J* 2.8); 39.51 d.t ($^{\text{C}3}$, *J* 170.4, 4.6); 46.32 d ($^{\text{C}1}$, *J* 3.4); 54.11 q (OMe-*trans*, *J* 149.3); 54.99 q (OMe-*cis*, *J* 149.4); 109.36 d (CN-*trans*, *J* 4.6); 111.45 d (CN-*cis*, *J* 6.1); 125.58 t.d ($^{\text{C}}$, *J* 8.0, 2.3); 129.48 d.d ($^{\text{C}}$ m, *J* 168.1, 5.1); 130.08 d.d.d ($^{\text{C}0}$ n, *J* 162.3, 6.8, 2.7); 136.08 t.t ($^{\text{C}0}$ n, *J* 163.26 d.q (CO-*cis*, *J* 6.3, 4.0). Found, %: 56.43; H 3.42; N 8.72.

 $C_{15}H_{11}ClN_2O_4$. Calculated, %: C 56.53; H 3.48; N 8.79.

Dimethyl 3-(4-methoxyphenyl)-2,2-dicyanocyclopropane-1,1-dicarboxylate (Vd). Yield 67%, mp 101–102°C. IR spectrum (mineral oil), v, cm⁻¹: 1725, 2245. ¹H NMR spectrum, δ, ppm: 3.75 s (6H, CH₃O, COOCH₃), 3.91 s (1H, CH), 3.95 s (3H, COOCH₃), 6.78–7.22 m (4H, MeOC₆H₄). Found, %: C 61.07; H 4.43; N 8.84. C₁₆H₁₄N₂O₅.Calculated, %: C 61.14; H 4.49; N 8.91.

Diethyl 3-phenyl-2,2-dicyanocyclopropane-1,1-dicarboxylate (**Ve**). Yield 69%, mp 71–72°C. IR spectrum (mineral oil), ν, cm⁻¹: 1720, 2240. ¹H NMR spectrum, δ, ppm (J, Hz): 1.16 t, 1.36 t (6H, COOC H_3 CH $_2$, J 7.4), 3.91 s (1H, CH), 4.14 q, 4.40 q (4H, COOCH $_3$ CH $_2$, J 7.4), 7.16–7.29 m (5H, C $_6$ H $_5$). Found, %: C 65.30; H 5.11; N 8.89. C $_{17}$ H $_{16}$ N $_2$ O $_4$. Calculated, %: C65.38; H 5.16; N 8.97.

Diethyl 3-(4-bromophenyl)-2,2-dicyanocyclo-propane-1,1-dicarboxylate (Vf). Yield 70%, mp 86–87°C. IR spectrum (mineral oil), ν, cm⁻¹: 1725, 2240. ¹H NMR spectrum, δ, ppm (J, Hz): 1.20 t, 1.42 t (6H, COOC H_3 CH $_2$, J 7.4), 3.84 s (1H, CH), 4.16 q, 4.40 q (4H, COOC H_3 C H_2 , J7.4), 7.12–7.49 m (4H, BrC $_6$ H $_4$). Found, %: C 52.12; H 3.77; N 7.09. C $_{17}$ H $_{15}$ BrN $_2$ O $_4$. Calculated, %: C 52.19; H 3.86; N 7.16.

Diethyl 3-(4-chlorophenyl)-2,2-dicyanocyclo-propane-1,1-dicarboxylate (Vg). Yield 69%, mp 85–87°C. IR spectrum (mineral oil), ν, cm⁻¹: 1725, 2240. ¹H NMR spectrum, δ, ppm (J, Hz): 1.19 t, 1.43 t (6H, COOC H_3 CH $_2$, J 7.4), 3.86 s (1H, CH), 4.16 q, 4.40 q (2H, COOCH $_3$ CH $_2$, J 7.4), 7.17–7.26 m (4H, ClC $_6$ H $_4$). Found, %: C 58.79; H 4.30; N 8.02. C $_{17}$ H $_{15}$ ClN $_2$ O $_4$. Calculated, %: C 58.88; H 4.36; N 8.08.

Diethyl 3-(4-methoxyphenyl)-2,2-dicyanocyclopropane-1,1-dicarboxylate (Vh). Yield 67%, mp 75–77°C. IR spectrum (mineral oil), ν, cm⁻¹: 1725, 2240. ¹H NMR spectrum, δ, ppm, (J, Hz): 1.20 t, 1.42 t (6H, COOC H_3 CH $_2$, J 7.4), 3.75 s (3H, CH $_3$ O), 3.85 s (1H, CH), 4.16 q, 4.39 q (4H, COOC H_3 C H_2 , J 7.4), 6.76–7.23 m (4H, MeOC $_6$ H $_4$). Found, %: C 63.04; H 5.24; N 8.11. C $_{18}$ H $_{18}$ N $_2$ O $_5$. Calculated, %: C 63.15; H 5.30; N 8.18.

3-Aryl-2-cyanocyclopropane-1,1,2-tricarboxylic acid trialkyl esters (Vi–Vk). The procedure is similar to a technique used for preparation of compounds

VIa–**VIh** except that instead of 2-arylmethylenemalonic acid dinitrile methyl ester of a 2-aryl-3-cyanopropenoic acid has been used.

Trimethyl 3-(4-bromophenyl)-2-cyanocyclopropane-1,1,2-tricarboxylate (Vi). Yield 68%, mp 92–93°C. IR spectrum (mineral oil), v, cm⁻¹: 1730, 2235. ¹H NMR spectrum, δ, ppm: 3.66 s (3H, COOCH₃), 3.77 s (3H, COOCH₃), 3.83 s (1H, CH), 3.87 s (3H, COOCH₃), 7.15 d (2H, 4-BrC₆H₄), 7.45 d (2H, 4-BrC₆H₄). Found, %: C 48.47; H 3.50; N 3.49. C₁₆H₁₄BrNO₆. Calculated, %: C 48.50; H 3.56; N 3.54.

Trimethyl 3-(3-bromophenyl)-2-cyanocyclopropane-1,1,2-tricarboxylate (Vj). Yield 65%, mp 104–106°C. IR spectrum (mineral oil), v, cm⁻¹: 1730, 2245. ¹H NMR spectrum, δ, ppm: 3.68 s (3H, COOCH₃), 3.75 s (3H, COOCH₃), 3.82 s (1H, CH), 3.87 s (3H, COOCH₃), 7.08–7.45 m (4H, 3-BrC₆H₄). Found, %: C 48.40; H 3.48; N 3.47. $C_{16}H_{14}BrNO_6$. Calculated, %: C 48.50; H 3.56; N 3.54.

Trimethyl 3-(2,4-dichlorophenyl)-2-cyanocyclo- propane-1,1,2-tricarboxylate (Vk). Yield 66%, mp 116–117°C. IR spectrum (mineral oil), ν, cm⁻¹: 1735, 2245. ¹H NMR spectrum.: (400 MHz), δ, ppm (*J*, Hz): 3.78 s (3H, COOCH₃); 3.81 d (1H, CH, *J* 1.0); 3.83 s (3H, COOCH₃); 3.94 s (3H, COOCH₃); 7.29 dd (1H, H⁵, *J* 8.4, 2.0); 7.34 d.d (1H, H⁶, *J* 8.4, 1.0); 7.46 d (1H, H³, *J* 2.0). ¹³C NMR (100 MHz, δ, ppm, *J*, Hz): 31.32 d (C², *J* 2.5); 37.90 d.d (C³, *J* 169.7, 4.6); 47.34 d (C¹, *J* 3.0); 53.72 q (MeO, *J* 148.4); 53.85 q (MeO, *J* 149.0); 54.81 q (MeO, *J* 149.2); 112.03 d (CN, *J* 4.4); 126.28 d.d.d (C¹, *J* 7.9, (C⁴, *J* 10.4, 3.9, 2.5); 162.81 d.q (CO, *J* 5.0, 4.1); 163.87 d.q (CO, *J* 5.7, 4.0); 164.55 d.q (CO, *J* 5.2, 4.0). Found, %: C 49.73; H 3.3.35; N 3.59. C₁₆H₁₃Cl₂NO₆. Calculated, %: C 49.76; H 3.3.39; N 3.63.

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