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THE REGIOSELECTIVE EFFECT OF THE ALLYLIC SILICON OR OXYGEN ATOM IN 1,3-DIPOLAR CYCLOADDITION OF NITRONES

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1,3-Dipolar cycloaddition of nitrones on several allyl silanes, allyl ethers, and allyl alcohol was investigated. The silicon atom in dipolarophiles controlled the regioselectivity in favor for 5-(silylmethyl)isoxazolidines, while the oxygen atom for 4-(alkoxymethyl)isoxazolidines.

The 1,3-dipolar cycloaddition of nitrones has been widely used for the construction of various natural products.^{1,2)} The reaction is generally highly regio- and/or stereoselective and the N-O bond of the isoxazolidine ring formed can be effectively cleaved under mild conditions. The regioselectivity is of interest in both synthetic and theoretical aspects and is well documented in the reactions with mono- or di-substituted dipolarophiles which possess polar substituent(s) attached on the C-C double bond.³⁻⁷⁾

On the contrary, 1,2-dialkyl substituted olefins had been rarely used as dipolarophiles until 1978, when McAlduff, Caramella, and Houk reported the cycloaddition of benzonitrile oxide to 3-substituted cyclopentenes and found that the regioselectivity was reversed depending on the electron donating or withdrawing nature of the substituents.⁸⁾ On the other hand, we found previously that, in the nitrone cycloaddition, the allylic oxygen atom seemed to affect the regioselectivity in favor of formation of 4-(alkoxymethyl)isoxazolidine as shown below.⁹⁾

(3:1)

These results prompted us to investigate the effect of allylic heteroatoms on the

regioselectivity of 1,3-dipolar cycloaddition of nitrones. In the present communication, we wish to report preliminarily the marked effect of the allylic silicon and oxygen atoms.

The reactions were conducted with a mixture of nitrone and excess dipolarophile at room temperature or at elevated temperature in a sealed tube. The products were separated by column chromatography or recrystallization and characterized by 1 H NMR spectra.¹⁰⁾ The results are shown in Table 1.

At first, we followed the reaction of a terminal alkene, 1-hexene,¹¹⁾ and confirmed regioselective production of 5-butylisoxazolidine <u>la</u> in 70% yield(run 1). None of regioisomers were detected but a very minute amount of stereoisomer <u>la</u>' was isolated by preparative TLC.¹²⁾ In the case of (E)-2-octene (run 2), two regioisomers <u>2a</u> and <u>2b</u> were obtained almost in equal amount, as expected from the similar electronic nature of both alkyl substituents.

The effect of the allylic silicon atom was investigated in runs 3 - 6. The reactions with allyltrimethylsilane showed the same regioselectivity as the run 1 and 5-(trimethylsilylmethyl) isoxazolidines <u>3a</u> and <u>4a</u> were isolated in moderate yield, respectively (runs 3 and 4). Regioselective effect of the silicon atom became clear when 1-trimethylsilyl-(Z)-2-butene was used as dipolarophile (run 5). The allylic silyl group remarkably controls the selectivity to afford 5-(trimehtyl-silylmethyl) isoxazolidine <u>5a</u>. This tendency is kept in the reaction with 3-trimethylsilylcyclohexene, though the yield was low (run 6).

On the other hand, the allylic oxygen atom was found to affect in opposite direction (runs 7 - 9). The reaction with 3-ethoxypropene proceeded in similar yield as that of the run 1, but gave clearly two regioisomers 7a and 7b (run 7). The formation of 7b may be due to the presence of the allylic oxygen atom. Reversed regioselective effect of the oxygen atom compared with the silicon atom was further confirmed by the reaction with 1-methoxy-(E)-2-butene (run 8): 4-(methoxymethyl)isoxazolidine 8b was mainly obtained. The allylic hydroxyl group also showed regioselectivity in favor of producing 4-(hydroxymethyl)isoxazolidines (9b + 9c). The slight decrease in regioselectivity in the run 9 compared with the run 8 can be partially attributed to the hydrogen bonding of the hydroxyl group to the O4N of the nitrone so as to locate both components to give 9a.

The present results show that the allylic silicon and oxygen atoms affect the regioselectivity of 1,3-dipolar cycloaddition of nitrones in opposite direction. Further works on fundamental aspects are now in progress.

Run Dipolar phile	ro- Nitr R _l CH=N		Conditions	Yield/%	Products ^{a)}	Regioisomeric ratio (<u>a/b</u>)
C4H9	PhCO	Ph	r.t. (43h) benzene	70 C ₄ H _c	$\int_{1}^{\text{COPh}} \sqrt{1} \frac{1}{2} (3)^{1} \sqrt{2} \sqrt{2}$	100/0
C5H11	PhCO	Ph	r.t. (70h) benzene	64 ^C 5 ^H 1	N-Ph 5 11	DPh 47/53 −Ph
Me ₃ Si	-сн ₂ сн	2 ^{CH} 2 ⁻	ll0°C (12h) benzene	63 Me ₃ Si		100/0
Me ₃ Si	Ph	Me	165°C (12h)	70 Me ₃ Sia	Ph N-Me <u>4a</u> (71:29)	100/0
Me ₃ Si	Ph	Me	190°C (24h) benzene	31 Me ₃ Si	\longrightarrow Me ₃ Si \longrightarrow	²h ∙Me 98/2
$\bigcup_{Me_3 Si}$	Ph	Me	190°C (24h)	15 Me ₃ S	N-Me i <u>6a</u> (52:48)	100/0
Eto	PhCO	Ph	r.t. (41h) benzene	71 EtO		Ph 91/9
MeO	PhCO	Ph	r.t. (40h) benzene	90 MeQ		Ph 6/94
HO	PhCO	Ph	r.t. (36h) benzene	94 HO	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \underline{9a}\\\end{array}\end{array} \end{array} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\$	Ph $28/72^{b}$
					9c (75	-Ph

Table	1.	Cycloaddition of Nitrones with Some Allylsilanes,
		Allyl Ethers, or Related Dipolarophiles

a) Stereoisomeric ratio is shown in parentheses. b) $\frac{9a}{(9b + 9c)}$.

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