

Synthesis and bioactivity of C-29 brassinosteroid analogues with different functional groups at C-6

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Abstract

In this paper, we report the synthesis and bioactivity of four synthetic analogues of 28-homobrassinosteroids, in order to evaluate the influence in bioactivity when the C-6 keto group is replaced by different functional groups. The synthetic analogues are 6-deoxo-28-homocastasterone [(22*R*,23*R*)-stigmasta-2 α ,3 α ,22,23-tetraol], 6 α -hydroxy-28-homocastasterone [(22*R*,23*R*)-stigmasta-2 α ,3 α ,6 α ,22,23-pentaol], 6 β -hydroxy-28-homocastasterone [(22*R*,23*R*)-stigmasta-2 α ,3 α ,6 β ,22,23-pentaol], and [(22*R*,23*R*)-6 α -fluorostigmasta-2 α ,3 α ,22,23-tetraol].

Results indicate that replacement of the 6-keto moiety by an β or α hydroxyl group led to a decrease in activity, whereas the 6-deoxo analogue showed a very low activity, confirming the importance of an electronegative moiety at C-6 to observe hormonal potency. The 6 α -fluorinated analogue elicited a low activity, similar to that of the 6-deoxo analogue.

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1. Introduction

Brassinosteroids (BRs) are recognized as a class of plant steroidal hormones that control many developmental and physiological processes including regulation of gene expression, cell division, differentiation, and homeostasis (Kripach et al., 1999; Sakurai, 1999a).

Since the discovery of brassinolide (BL) (**1a**) (Fig. 1) (Grove et al., 1979) as the growth-promoting factor responsible for the activity of *Brassica* pollen extracts, more than 40 related compounds have been isolated from a wide variety of plant species (Bajguz and Tretny, 2003; Hayat and Ahmad, 2003). The exogenous applica-

tion of BRs has shown a wide spectrum of responses, such as stem elongation, inhibition of root growth, leaf epinasty and xylem differentiation, brought about by changes in enzyme activity and gene expression (Cutler et al., 1991; Arteca, 1995; Mussig et al., 2002).

Brassinolide (**1a**) and castasterone (**2a**) are C₂₈ brassinosteroids bearing an *S*-methyl group at C-24 in the side chain of its 5 α -ergostane structure.

Though C₂₈ brassinosteroids are the most ubiquitous in nature, other BRs with a different steroidal side-chain are known. Among them, 28-homobrassinolide (HBL) (**1b**) and 28-homocastasterone (HCS) (**2b**) are the most active C₂₉ BRs with a 5 α -stigmastane structure (Sakurai et al., 1999b) (Fig. 1); both compounds have been widely used in field trials.

Previous reports (Sakurai, 1999a; Sung et al., 2000) indicated that in the C₂₈ series bioactivities decrease within the following order: C-6 lactone, C-6 ketone,

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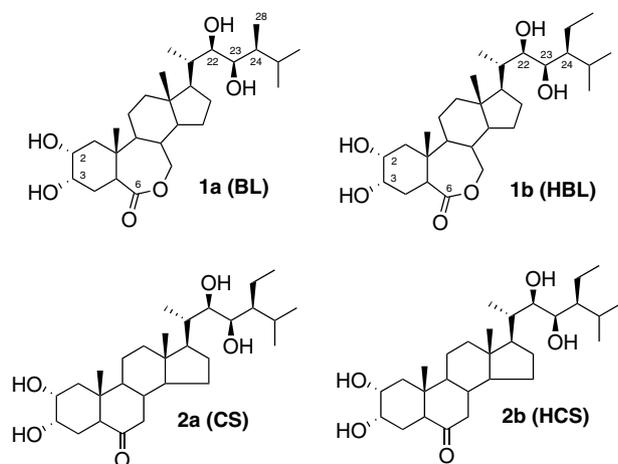


Fig. 1. Some C₂₈ and C₂₉ natural brassinosteroids.

C-6 α or β hydroxy, C-6 deoxy (compounds **1a**, **2a**, **4a**, **5a** and **3a**, respectively, Figs. 1 and 2).

In this paper, we report the synthesis and bioactivity of four new synthetic analogues of the C₂₉ series, in order to evaluate the influence on bioactivity when the C-6 lactone or keto groups are replaced by different functionalities.

New synthetic analogues are 6-deoxy-28-homocasterone [(22*R*,23*R*)-stigmasta-2 α ,3 α ,22,23-tetraol] (**3b**), 6 α -hydroxy-28-homocasterone [(22*R*,23*R*)-stigmasta-2 α ,3 α ,6 α ,22,23-pentaol] (**4b**), 6 β -hydroxy-28-homocasterone [(22*R*,23*R*)-stigmasta-2 α ,3 α ,6 β ,22,23-pentaol]

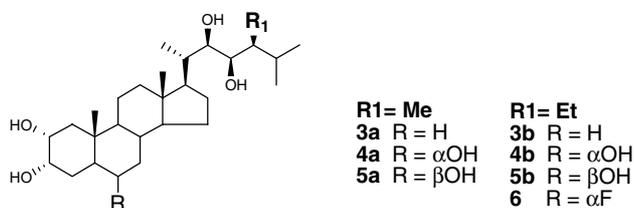


Fig. 2. Structures of brassinosteroids analogues.

(**5b**), and [(22*R*,23*R*)-6 α -fluorostigmasta-2 α ,3 α ,22,23-tetraol] (**6**) (Fig. 2).

2. Results and discussion

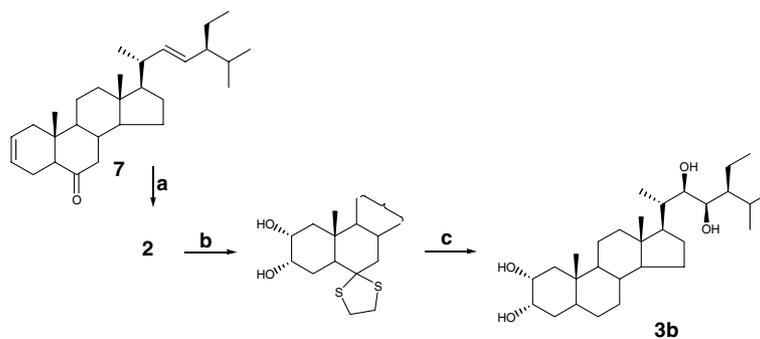
Synthesis of compound **3b** is shown in Scheme 1. Dienone **7** was obtained in four steps from stigmaterol following the procedure described by McMorris et al. (1994). 28-Homocasterone (**2b**) was obtained from dienone **7** by osmium-catalyzed asymmetric dihydroxylation (CAD) (Sharpless et al., 1992) on double bonds at C-2 and C-22, using K₃Fe(CN)₆ as cooxidant and hydroquinidine-1,4-phthalazinediyl diether [(DHQD)₂-Phal] as chiral ligand. Subsequent treatment of compound **2b** with ethanedithiol and boron trifluoride etherate gave a dithiane intermediate, which was reduced by tri-*n*-butyltin hydride (Anastasia et al., 1985) to afford the desired compound **3b** in 68% yield.

Syntheses of compounds **4b** and **5b** are shown in Scheme 2. Reduction of **7** with sodium in ethanol, afforded the thermodynamically more stable equatorial 6 α -alcohol (**8**), along with a 12% of the 6 β -epimeric compound (**9**). Stereochemistry at C-6 of both isomers was deduced from examination of the multiplicity pattern of the H-6 in the ¹H NMR spectra.

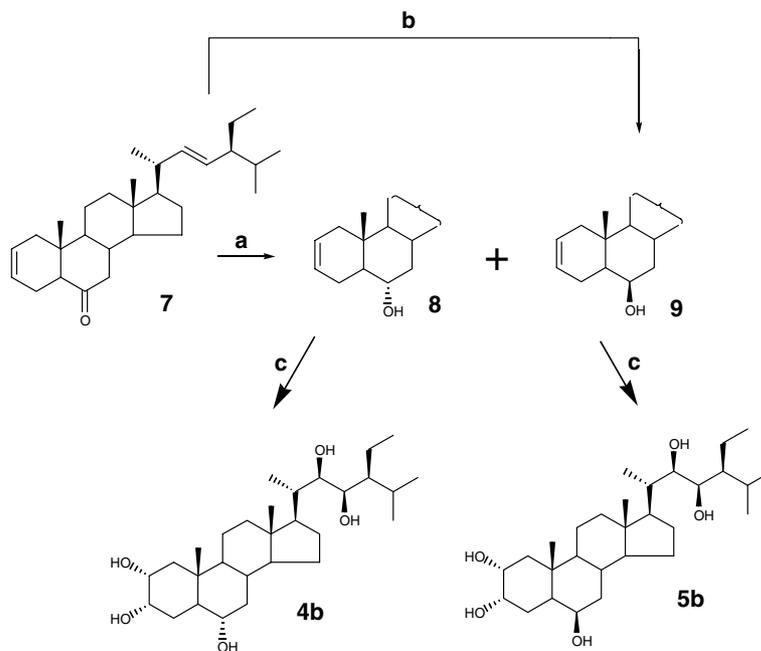
CAD of double bonds of compounds **8** and **9** gave compounds **4b** and **5b**, respectively, along with their corresponding 22*S*,23*S* diastereomers in a minor proportion (ratio 3:1), and with more than 30% of the starting material that could be recycled. Yields, after purification, were 32% and 30%, respectively.

In order to improve the yield of compound **9**, the dienone **7** was reduced using lithium tri-*t*-butoxyaluminium hydride in THF/methanol. The use of this bulky reducing reagent ensured hydride attack from the less hindered equatorial direction yielding only compound **9**.

Several efforts for introducing fluorine groups in the structure of brassinosteroids have been made (Jin et al., 1993; Back et al., 1999; Jiang et al., 1993; Ramírez et al., 2000; Galagovsky et al., 2001); however, few stud-



Scheme 1. Synthesis of compound **3b** Reagents and conditions: (a) K₂OsO₄, K₃Fe(CN)₆, (DHQD)₂Phal, K₂CO₃, *t*-BuOH, H₂O, CH₃ SO₃ NH₂, r.t., 9 days; (b) HSCH₂CH₂ SH, BF₃·Et₂O, CH₂Cl₂, r.t., 30 min; (c) *n*-Bu₃SnH, AIBN, benzene, reflux, 2 h.



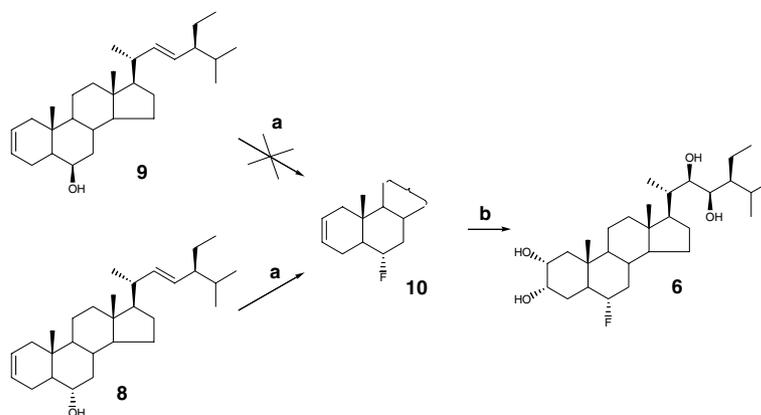
Scheme 2. Synthesis of compounds **4b** and **5b**. Reagents and conditions: (a) Na, EtOH, reflux, 3 h; (b) $(t\text{-BuO})_3\text{AlH}$, THF, MeOH, r.t., 1 day; (c) K_2OsO_4 , $\text{K}_3\text{Fe}(\text{CN})_6$, $(\text{DHQD})_2\text{Phal}$, K_2CO_3 , $t\text{-BuOH}$, H_2O , $\text{CH}_3\text{SO}_3\text{NH}_2$, r.t., 9 days.

ies on the effect on bioactivity have been reported. Synthesis of fluorinated analogue **6** is shown in Scheme 3. Stereoselective introduction of fluorine is usually achieved by bimolecular nucleophilic substitution ($\text{S}_{\text{N}}2$) of an hydroxyl group with diethylaminosulfur trifluoride (DAST) (Rozen et al., 1979). Treatment of **9** with DAST at -78°C gave no reaction, whereas further attempts at higher temperatures led to decomposition of the substrate. This behaviour might be explained by the fact that the axial 19-methyl could block the $\text{S}_{\text{N}}2$ reaction on the axial 6β -hydroxyl group. A similar case was reported by Kumar et al. (1995) during the transformation of a hindered 17β -hydroxy steroid into a 17α -fluoro compound by DAST treatment. The desired

compound was finally obtained starting from the corresponding 17α -hydroxy substrate, suggesting an $\text{S}_{\text{N}}1$ mechanism via a carbocationic intermediate (Shellhamer et al., 1996).

In fact, the treatment of the 6α -hydroxy steroid **8** with DAST yielded the expected compound (22*E*)- 6α -fluoro- 5α -stigmasta-2,22-diene (**10**).

The configuration at C-6 was established from analysis of the ^1H NMR spectra, as shown in Fig. 3. Fig. 3(a) shows the multiplet corresponding to the H-6 in the hydroxylated precursor **8**, consistent with two axial-axial couplings of 10.8 Hz ($J_{6\beta\text{H}-5\alpha\text{H}}$ and $J_{6\beta\text{H}-7\alpha\text{H}}$), and an axial-equatorial coupling of 4.6 Hz ($J_{6\beta\text{H}-7\beta\text{H}}$). Fig. 3(b) shows that, besides the large germinal $J_{\text{H-F}}$



Scheme 3. Synthesis of fluorinated analogue **6**. Reagents and conditions: (a) DAST, CH_2Cl_2 , -78°C , 20 min. (b) K_2OsO_4 , $\text{K}_3\text{Fe}(\text{CN})_6$, $(\text{DHQD})_2\text{Phal}$, K_2CO_3 , $t\text{-BuOH}$, H_2O , $\text{CH}_3\text{SO}_3\text{NH}_2$, r.t., 9 days.

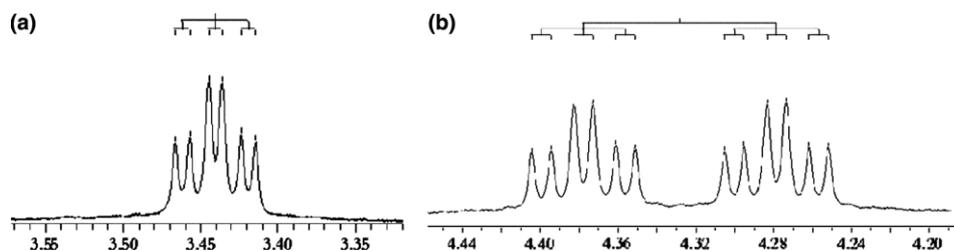


Fig. 3. (a) 6 β -H signal for compound **8**. (b) 6 β -H signal for compound **10** (CDCl₃, 500 MHz). The same $J_{\text{H-H}}$ pattern is observed, confirming that fluorination proceeded with retention of the configuration.

coupling, the same pattern is observed for H-6 in compound **10**, confirming that fluorination occurred with retention of configuration.

Finally, CAD on **10** gave, after purification, the desired compound **6** in 35% yield. All of the chemical structures were confirmed by HRMS and NMR spectroscopic analyses (DEPT-135, COSY-45 and HETCOR experiments).

Bioactivities were evaluated at four different doses for each compound using a modified RLIT (Takeno and Pharis, 1982) on local *Chuy* rice cultivar, using 28-homocastasterona (**2b**) as a control compound, and are shown in Fig. 4.

Results indicate that replacement of the 6-keto moiety by a β hydroxy group (**5b**) led to a similar bioactivity at the highest dose used, but to a decrease in lower doses. On the other hand, the replacement by an α hydroxy group (**4b**) led to a decrease in the activity at all doses. These results are in agreement with preliminar data reported by Brosa (1999) using the *Bahia* rice cultivar.

The 6-deoxo analogue (**3b**) showed very low activity, confirming the importance of an electronegative moiety at C-6 for hormonal potency (Sakurai, 1999a).

The 6 α -fluorinated analogue (**6**) elicited a low activity, similar to that of the 6-deoxo analogue (**3b**). Many previous studies showed different effects on bioactivity due to the replacement of an hydrogen or an oxygenated functional group by fluorine. Two examples may be mentioned. On the one hand, Back et al. (1999) found that the 25-fluoro analogues of brassinolide and castas-

terone showed almost no activity while the corresponding 25-hydroxy analogues had potent biological activity. On the other hand, 5 α -fluoro-28-homocastasterone and 5 α -fluoro-28-homotyphasterol showed increased bioactivity compared with their related 5-hydroxy analogues (Ramírez et al., 2000; Schaefer et al., 2002).

The low bioactivity elicited by compound **6** indicates that, in this case, the presence of a 6-fluorine atom is not recognized in the biological system as an electronegative group like a 6-keto or a 6-hydroxy one, notwithstanding the described isosteric and electronic similarities between an hydroxy and a fluoro group (Welch, 1987).

3. Experimental

3.1. General experimental procedures

Melting points (m.p.) were determined on a Fisher Johns apparatus and are uncorrected. ¹H-NMR and ¹³C NMR spectra were recorded on a Bruker AM-500 at 500 MHz and at 125.1 MHz, respectively; chemical shifts (δ) are given in ppm downfield from TMS as internal standard. Coupling constant (J) values are in Hz. ¹⁹F-NMR spectra were recorded on a Brüker AM-500 at 470.4 MHz, and chemical shifts (δ) are given in ppm upfield from CFCl₃ as the internal standard. High-resolution mass spectra (EI and FAB) were obtained for all new compounds on a ZAB SEQ instrument (VG-Micromass). FAB were obtained using a DS (disulfide) matrix. Low resolution mass spectra were recorded on a Shimadzu QP-5050 at 70 eV. Unless otherwise indicated, all solvents and reagents used were of commercial grade. Reactions were monitored by TLC on precoated plates with silica gel F₂₅₄ 0.2 mm (Merck). CC was carried out on silica gel 60, 0.04–0.063 mm (Merck).

3.2. Rice lamina inclination test

Rice seedlings (*Oryza sativa*, cv *Chuy*) were washed with EtOH (1 min) and water and were then left in water at 30 °C for two days (with a 16 h photoperiod). Germinated seeds were cultivated on agar under the same

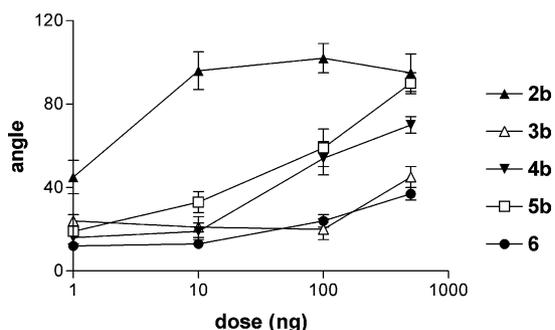


Fig. 4. Bioactivity in the rice lamina inclination test. Values are angles (degrees) between the lamina and the sheath, representing the means of 20 replicates \pm standard error. Average angle of control: 13 ± 2 .

growing conditions for four days. Intact seedlings (4–5 cm long) were inoculated with 0.5 μ l of the test compound solution (in EtOH) just under the 2nd leaf joint. After 48 h in the dark (at 30 °C), the magnitude of the induced angle between the leaf and the sheath was measured.

3.3. Syntheses

3.3.1. (22*R*,23*R*)-stigmasta-2 α ,3 α ,22,23-tetraol (**3b**)

Treatment of 28-homocasterone (**2b**) (15 mg) (McMorris et al., 1994) with ethanedithiol (50 μ l) and boron trifluoride etherate (50 μ l) for 30 min afforded the corresponding dithiane as a glass, which was dissolved, without purification, in dry benzene (5 ml), and treated with tri-*n*-butyltin hydride (0.1 ml) in the presence of 2,2'-azobis(isobutyronitrile) (3 mg). After work-up, the obtained product was purified by silica gel TLC (CH₂Cl₂/CH₃CN 9:1) to afford **3b** (10 mg), m.p.: 223–224 °C. ¹H NMR (CDCl₃:CD₃OD, 95:5): 0.70 (H-18, 3H, *s*), 0.81 (H-19, 3H, *s*), 0.89 (H-21, 3H, *d*, *J* = 6.2 Hz), 0.92–0.99 (H-26, H-27 and H-29, 9H, *m*), 3.59 (H-22, 1H, *dd*, *J* = 8.4 Hz, 1.3 Hz), 3.69 (H-23, 1H, *dd*, *J* = 8.4 Hz, 1.3 Hz), 3.72 (H-2 β , 1H, *m*), 4.05 (H-3 β , 1H, *dd*, *J* = 6.0 Hz, 3.0 Hz). ¹³C NMR (CDCl₃:CD₃OD, 95:5): 11.6 (*q*, C-18), 11.7 (*q*, C-21), 13.0 (*q*, C-19), 13.5 (*q*, C-29), 19.0 (*t*, C-28), 19.3 (*q*, C-26), 20.8 (*t*, C-11), 20.8 (*q*, C-27), 23.8 (*t*, C-15), 27.7 (*t*, C-6), 28.5 (*t*, C-16), 29.0 (*d*, C-25), 32.7 (*t*, C-7), 34.5 (*t*, C-4), 36.0 (*d*, C-8), 36.5 (*d*, C-20), 37.6 (*s*, C-10), 38.5 (*s*, C-5), 39.8 (*t*, C-1), 40.9 (*t*, C-12), 42.2 (*s*, C-13), 46.4 (*d*, C-24), 52.3 (*d*, C-17), 53.6 (*t*, C-9), 55.5 (*d*, C-14), 68.1 (*d*, C-3), 68.7 (*d*, C-2), 72.2 (*d*, C-23), 74.3 (*d*, C-22). MS (FAB) *m/z* (rel. int.): 487 [(M + Na)⁺, 10], 447 (8), 429 (12), 231 (30), 136 (100). HRMS (FAB): Calculated for C₂₉H₅₂O₄Na (M + Na)⁺: 487.3763, found: 487.3770.

3.3.2. (22*E*)-stigmasta-2,22-dien-6 α -ol (**8**)

Sodium metal (53 mg, 2.3 mmol) was cut into small pieces and added to a refluxing solution of (22*E*)-stigmasta-2,22-dien-6-one **7** (100 mg, 0.25 mmol) in anhydrous EtOH (20 ml) (McMorris et al., 1994). After cooling, the mixture was diluted with H₂O and extracted with CH₂Cl₂. The combined organic layers were washed, dried and evaporated in vacuo. Chromatographic purification of the crude product (hexane:EtOAc, 95:5) gave **9** (12 mg), m.p.: 107–109 °C. ¹H NMR (CDCl₃): 0.72 (H-18, 3H, *s*), 0.80 (H-27, 3H, *d*, *J* = 6.5 Hz), 0.81 (H-29, 3H, *t*, *J* = 7.4 Hz), 0.85 (H-26, 3H, *d*, *J* = 6.5 Hz), 0.97 (H-19, 3H, *s*), 1.02 (H-21, 3H, *d*, *J* = 6.7 Hz), 2.30 (H-4 β , 1H, *m*), 3.85 (H-6 α , 1H, *m*), 5.02 (H-23, 1H, *dd*, *J* = 15.1 Hz, 8.5 Hz), 5.16 (H-22, 1H, *dd*, *J* = 15.1 Hz, 8.6 Hz), 5.56–5.71 (H-2 and H-3, 2H, *m*).

¹³C NMR (CDCl₃): 12.2 (*q*, C-18), 12.2 (*q*, C-29), 15.2 (*q*, C-19), 19.0 (*q*, C-26), 20.8 (*t*, C-11), 21.0 (*q*,

C-21), 21.2 (*q*, C-27), 24.3 (*t*, C-15), 25.4 (*t*, C-28), 26.4 (*t*, C-4), 28.9 (*t*, C-16), 31.9 (*d*, C-25), 34.2 (*d*, C-8), 34.6 (*s*, C-10), 39.7 (*t*, C-12), 39.9 (*t*, C-1), 40.5 (*d*, C-20), 41.7 (*t*, C-7), 42.5 (*s*, C-13), 44.5 (*d*, C-5), 51.2 (*d*, C-24), 54.2 (*t*, C-9), 56.1 (*d*, C-14), 56.3 (*d*, C-17), 70.5 (*d*, C-6), 125.6 (*d*, C-3), 125.7 (*d*, C-2), 129.3 (*d*, C-23), 138.3 (*d*, C-22). MS (EI) *m/z* (rel. int.): 412 (M⁺, 2), 351 (1), 271 (3), 253 (3), 199 (2), 55 (100).

HRMS (EI): Calculated for C₂₉H₄₈O: 412.3705, found: 412.3712.

Further elution with the same solvent gave **8** (70 mg), m.p.: 119–120 °C. ¹H NMR (CDCl₃): 0.68 (H-18, 3H, *s*), 0.77 (H-19, 3H, *s*), 0.80 (H-27, 3H, *d*, *J* = 6.5 Hz), 0.81 (H-29, 3H, *t*, *J* = 7.4 Hz), 0.85 (H-26, 3H, *d*, *J* = 6.4 Hz), 1.02 (H-21, 3H, *d*, *J* = 6.7 Hz), 2.40 (H-4 β , 1H, *m*), 3.44 (H-6 β , 1H, *ddd*, *J* = 10.7 Hz, 10.7 Hz, 4.6 Hz), 5.02 (H-23, 1H, *dd*, *J* = 15, 1 Hz, 8.5 Hz), 5.16 (H-22, 1H, *dd*, *J* = 15.1 Hz, 8.6 Hz), 5.56–5.71 (H-2 and H-3, 2H, *m*).

¹³C NMR (CDCl₃): 12.1 (*q*, C-29), 12.2 (*q*, C-18), 13.0 (*q*, C-19), 18.9 (*q*, C-26), 20.8 (*t*, C-11), 21.0 (*q*, C-21), 21.0 (*q*, C-27), 24.2 (*t*, C-15), 25.4 (*t*, C-28), 25.8 (*t*, C-4), 28.8 (*t*, C-16), 31.9 (*d*, C-25), 34.2 (*d*, C-8), 35.4 (*s*, C-10), 39.7 (*t*, C-12), 40.0 (*t*, C-1), 40.4 (*d*, C-20), 41.4 (*t*, C-7), 42.4 (*s*, C-13), 48.7 (*d*, C-5), 51.2 (*d*, C-24), 53.6 (*t*, C-9), 56.0 (*d*, C-17), 56.2 (*d*, C-14), 71.6 (*d*, C-6), 125.2 (*d*, C-3), 125.5 (*d*, C-2), 129.4 (*d*, C-23), 138.2 (*d*, C-22). MS (EI) *m/z* (rel. int.): 412 (M⁺, 2), 351 (1), 314 (1), 271 (4), 229 (3), 55 (100).

HRMS (EI): Calculated for C₂₉H₄₈O: 412.3705, found: 412.3699.

3.3.3. (22*E*)-stigmasta-2,22-dien-6 β -ol (**9**)

Compound **7** (150 mg, 0.36 mmol) was dissolved in 15 ml of a 9:1 mixture of THF and MeOH under an inert atmosphere. To this stirred solution, lithium tri-*t*-butoxy-aluminium hydride (280 mg, 1.1 mmol) was added. After 24 h, stirring the solution was poured into 5% HCl and the whole was extracted with CH₂Cl₂. The crude product was purified by Si gel CC (hexane:EtOAc, 95:5) to afford **9** (140 mg, 92%).

3.3.4. (22*R*,23*R*)-stigmasta-2 α ,3 α ,6 α ,22,23-pentaol (**4b**)

A mixture of **8** (60 mg, 0.14 mmol), *t*-BuOH:water (1:1, 10 ml), (DHQD)₂-Phal (60 mg, 0.08 mmol), methansulfonamide (45 mg, 0.26 mmol), potassium ferricyanide (150 mg, 0.46 mmol), potassium carbonate (62 mg, 0.45 mmol), and potassium osmate dihydrate (7 mg, 0.02 mmol) was stirred at room temperature for 9 days. An excess of NaHSO₃ was added until no evolution of bubbles was observed. Layers were separated and the aqueous phase was thoroughly extracted with CH₂Cl₂:MeOH (95:5).

Combined organic layers were sequentially washed with 0.25 M H₂SO₄ and 2% NaOH. Purification by column chromatography (CH₂Cl₂/acetonitrile gradient)

separated **4b** from its 22*S*, 23*S* diastereomer, with 32% yield, m.p.: 227–225 °C. ¹H NMR (CDCl₃:CD₃OD, 95:5): 0.67 (H-18, 3H, *s*), 0.82 (H-19, 3H, *s*), 0.89 (H-21, 3H, *d*, *J* = 6.2 Hz), 0.92–0.99 (H-26, H-27 and H-29, 9H, *m*), 3.29 (H-6β, *ddd*, *J* = 10.8 Hz, 10.7 Hz, 4.7 Hz), 3.56 (H-22, 1H, *dd*, *J* = 8.4 Hz, 1.3 Hz), 3.67 (H-23, 1H, *dd*, *J* = 8.4 Hz, 1.3 Hz), 3.71 (H-2β, 1H, *m*), 3.98 (H-3β, 1H, *dd*, *J* = 6.0 Hz, 3.0 Hz). ¹³C NMR (CDCl₃:CD₃OD, 95:5): 11.7 (*q*, C-21), 11.8 (*q*, C-18), 13.1 (*q*, C-19), 13.4 (*q*, C-29), 18.7 (*t*, C-28), 19.2 (*q*, C-26), 20.7 (*t*, C-11), 20.9 (*q*, C-27), 23.8 (*t*, C-15), 27.5 (*t*, C-4), 28.4 (*t*, C-16), 28.9 (*d*, C-25), 34.1 (*d*, C-8), 36.9 (*d*, C-20), 37.9 (*s*, C-10), 39.7 (*t*, C-1), 40.7 (*t*, C-12), 41.3 (*t*, C-7), 42.2 (*s*, C-13), 46.4 (*d*, C-24), 48.9 (*d*, C-5), 52.4 (*d*, C-17), 53.6 (*t*, C-9), 55.6 (*d*, C-14), 68.4 (*d*, C-3), 68.5 (*d*, C-2), 69.1 (*d*, C-6), 72.3 (*d*, C-23), 74.2 (*d*, C-22). MS (FAB) *m/z* (rel. int.): 503 [(M + Na)⁺, 25], 305 (10), 231 (22), 154 (31), 55 (100). HRMS (FAB): Calculated for C₂₉H₅₂O₅Na (M + Na)⁺: 503.3712, found: 503.3720.

3.3.5. (22*R*,23*R*)-stigmasta-2α,3α,6β,22,23-pentaol (**5b**)

Compound **9** (75 mg, 0.18 mmol) was tetrahydroxylated as described above for compound **8**. Chromatographic purification (CH₂Cl₂:CH₃CN gradient) afforded **5b** in 30% yield, m.p.: 234–235 °C. ¹H NMR (CDCl₃:CD₃OD, 95:5): 0.70 (H-18, 3H, *s*), 0.89 (H-21, 3H, *d*, *J* = 6.2 Hz), 0.92–0.99 (H-26, H-27 and H-29, 9H, *m*), 1.02 (H-19, 3H, *s*), 3.57 (H-22, 1H, *dd*, *J* = 8.4 Hz, 1.3 Hz), 3.68 (H-23, 1H, *dd*, *J* = 8.4 Hz, 1.3 Hz), 3.74 (H-2β, 1H, *m*), 3.79 (H-6α, 1H, *m*), 4.01 (H-3β, 1H, *dd*, *J* = 6.0 Hz, 3.0 Hz). ¹³C NMR (CDCl₃:CD₃OD, 95:5): 11.7 (*q*, C-18), 11.7 (*q*, C-21), 13.5 (*q*, C-29), 15.6 (*q*, C-19), 18.7 (*t*, C-28), 19.3 (*q*, C-26), 20.7 (*t*, C-11), 21.1 (*q*, C-27), 23.9 (*t*, C-15), 28.1 (*t*, C-4), 28.5 (*t*, C-16), 28.9 (*d*, C-25), 34.1 (*d*, C-8), 37.0 (*d*, C-20), 37.1 (*s*, C-10), 39.7 (*t*, C-1), 40.7 (*t*, C-12), 41.9 (*t*, C-7), 42.3 (*s*, C-13), 46.4 (*d*, C-24), 50.5 (*d*, C-5), 52.7 (*d*, C-17), 54.2 (*t*, C-9), 55.5 (*d*, C-14), 68.5 (*d*, C-3), 69.1 (*d*, C-2), 70.9 (*d*, C-6), 72.3 (*d*, C-23), 74.3 (*d*, C-22). MS (FAB) *m/z* (rel. int.): 503 [(M + Na)⁺, 62], 317 (9), 231 (31), 154 (92), 55 (100). HRMS (FAB): Calculated for C₂₉H₅₂O₅Na (M + Na)⁺: 503.3712, found: 503.3730.

3.3.6. (22*E*)-6α-fluorostigmasta-2,22-diene (**10**)

A solution of **8** (78 mg, 0.19 mmol) in CH₂Cl₂ (10 ml) was slowly added to a solution of DAST (0.1 ml) in CH₂Cl₂ (3 ml) cooled to –78 °C. After 20 min the mixture was warmed to room temperature and poured into water, washed with NaHCO₃ dried and evaporated. Purification of the crude product by Si gel cc graphy (hexane:EtOAc, 98:2) gave **10** (63 mg), m.p.: 95–97 °C. ¹H NMR (CDCl₃): 0.68 (H-18, 3H, *s*), 0.77 (H-19, 3H, *s*), 0.80 (H-27, 3H, *d*, *J* = 6.5 Hz), 0.81 (H-29, 3H, *t*, *J* = 7.4 Hz), 0.85 (H-26, 3H, *d*, *J* = 6.5 Hz), 1.02 (H-

21, 3H, *d*, *J* = 6.7 Hz), 2.38 (H-4β, 1H, *m*), 4.33 (H-6β, 1H, *double m*, *J*_{H,F} = 49.7 Hz), 5.02 (H-23, 1H, *dd*, *J* = 15.1 Hz, 8.5 Hz), 5.16 (H-22, 1H, *dd*, *J* = 15.1 Hz, 8.6 Hz), 5.56–5.71 (H-2 and H-3, 2H, *m*). ¹³C NMR (CDCl₃): 12.1 (*q*, C-29), 12.2 (*q*, C-18), 13.0 (*q*, C-19), 19.0 (*q*, C-26), 20.8 (*t*, C-11), 21.0 (*q*, C-21), 21.0 (*q*, C-27), 24.2 (*t*, C-15), 25.3 (*t*, C-28), 25.4 (*dt*, C-4, *J*_{CF} = 3.8 Hz), 28.7 (*t*, C-16), 31.9 (*d*, C-25), 33.9 (*dd*, C-8, *J*_{CF} = 11.4 Hz), 35.7 (*d*, C-10, *J*_{CF} = 9.4 Hz), 38.0 (*dt*, C-7, *J*_{CF} = 16.3 Hz), 39.6 (*t*, C-12), 39.8 (*t*, C-1), 40.4 (*d*, C-20), 42.5 (*s*, C-13), 46.6 (*dd*, C-5, *J*_{CF} = 15.1 Hz), 51.2 (*d*, C-24), 53.4 (*t*, C-9), 56.0 (*d*, C-17), 56.1 (*d*, C-14), 94.3 (*dd*, C-6, *J*_{CF} = 169.1 Hz), 124.9 (*d*, C-3), 125.3 (*d*, C-2), 129.4 (*d*, C-23), 138.1 (*d*, C-22). MS (EI) *m/z* (rel. int.): 414 (M⁺, 1), 371 (1), 302 (3), 273 (5), 260 (3), 55 (100). HRMS (EI): Calculated for C₂₉H₄₇F: 414.3662, found: 414.3656

3.3.7. (22*R*,23*R*)-6-fluorostigmasta-2,3,22,23-tetraol (**6**)

Compound **10** (60 mg, 0.14 mmol) was tetrahydroxylated as described for compound **8**. Chromatographic purification (CH₂Cl₂:acetonitrile gradient) afforded **6** (24 mg, 35% yield), m.p.: 221–222 °C. ¹H NMR (CDCl₃:CD₃OD, 95:5): 0.67 (H-18, 3H, *s*), 0.82 (H-19, 3H, *s*), 0.89 (H-21, 3H, *d*, *J* = 6.3 Hz), 0.92–0.99 (H-26, H-27 and H-29, 9H, *m*), 3.56 (H-22, 1H, *d*, *J* = 8.9 Hz), 3.68 (H-23, 1H, *d*, *J* = 8.9 Hz), 3.71 (H-2β, 1H, *m*), 3.99 (H-3β, 1H, *d*, *J* = 3.0 Hz), 4.27 (H-6β, *double m*, *J*_{H,F} = 50.0 Hz). ¹³C NMR (CDCl₃:CD₃OD, 95:5): 11.7 (*q*, C-18), 11.7 (*q*, C-21), 13.2 (*q*, C-19), 13.4 (*q*, C-29), 18.8 (*t*, C-28), 19.3 (*q*, C-26), 20.7 (*t*, C-11), 21.0 (*q*, C-27), 23.9 (*t*, C-15), 27.5 (*t*, C-16), 28.3 (*dt*, C-4, *J*_{CF} = 3.6 Hz), 28.9 (*d*, C-25), 33.9 (*dd*, C-8, *J*_{CF} = 10.5 Hz), 36.9 (*d*, C-20), 37.9 (*dt*, C-7, *J*_{CF} = 17.7 Hz), 38.3 (*d*, C-10, *J*_{CF} = 8.8 Hz), 39.6 (*t*, C-1), 40.6 (*t*, C-12), 42.4 (*s*, C-13), 43.3 (*dd*, C-5, *J*_{CF} = 14.6 Hz), 46.4 (*d*, C-24), 52.5 (*d*, C-17), 53.3 (*t*, C-9), 55.9 (*d*, C-14), 68.1 (*d*, C-3), 68.3 (*d*, C-2), 72.3 (*d*, C-23), 74.3 (*d*, C-22), 92.4 (*dd*, C-6, *J*_{CF} = 168.6 Hz). MS (FAB) *m/z* (rel. int.): 505 [(M + Na)⁺, 12], 465 (10), 447 (9), 307 (15), 231 (27), 154 (100), 137 (88). HRMS (FAB): Calculated for C₂₉H₅₁O₄FNa (M + Na)⁺: 505.3669, found: 505.3649.

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References

- Anastasia, M., Allevi, P., Ciuffreda, P., Fiecchi, A., Scala, A., 1985. Synthesis of (2*R*,3*S*,22*R*,23*R*)- and (2*R*,3*S*,22*S*,23*S*)-2,3,22,23-tetrahydroxy-B-homo-7 α -oxa-5 α -ergosta-7-ones, two new brassinolide analogues. *J. Org. Chem.* 50, 321–325.
- Arteca, R.N., 1995. Brassinosteroids. In: Davies, P. (Ed.), *Plant Hormones. Physiology, Biochemistry and Molecular Biology*, second ed. Kluwer Academic Publishers, The Netherlands, pp. 206–213.
- Bajguz, A., Tretyn, A., 2003. The chemical characteristic and distribution of brassinosteroids in plants. *Phytochemistry* 62, 1027–1046.
- Back, T., Janzen, L., Nakajima, S., Pharis, R., 1999. Synthesis and biological activity of 25-methoxy-, 25-fluoro-, and 25-azabassinolide and 25-fluorocasterone: surprising effects of heteroatom substituents at C-25. *J. Org. Chem.* 64, 5494–5497.
- Brosa, C., 1999. Structure–activity relationships. In: Sakurai, A., Yokota, T., Clouse, S. (Eds.), *Brassinosteroids: Steroidal Plant Hormones*. Springer-Verlag, Tokyo, pp. 191–222.
- Cutler, H., Yokota, T., Adam, G. (Eds.), 1991. *Brassinosteroids: Chemistry, Bioactivity and Applications*. ACS Symp. Series, 474. American Chemical Society, Washington, DC.
- Galagovsky, L.R., Gros, E.G., Ramírez, J.A., 2001. Synthesis and bioactivity of natural and C-3 fluorinated biosynthetic precursors of 28-homobrassinolide. *Phytochemistry* 58, 973–980.
- Grove, M., Spencer, G., Rohwedder, W., Mandava, N., Worley, J., Warthen Jr., J., Steffens, G., Flippen-Anderson, J., Cook, J., 1979. Brassinolide, a plant growth-promoting steroid isolated from *Brassica napus* pollen. *Nature* 281, 216–218.
- Hayat, S., Ahmad, A. (Eds.), 2003. *Brassinosteroids: Bioactivity and Crop Productivity*. Kluwer Academic Publishers, The Netherlands.
- Jiang, B., Liu, Y., Zhou, W., 1993. Stereocontrolled synthesis of the 22*E*,24 β (*S*)-trifluoromethyl steroidal side chain and its applications to the synthesis of fluorinated analogues of naturally occurring sterols. *J. Org. Chem.* 65, 6231–6236.
- Jin, F., Xu, Y., Huang, W., 1993. 2,2-Difluoro enol silyl ethers: convenient preparation and application to the synthesis of novel fluorinated brassinosteroids. *J. Chem. Soc. Perkin Trans. I*, 795–799.
- Kripach, V., Zhabinskii, V., de Groot, A. (Eds.), 1999. *Brassinosteroids: A New Class of Plant Hormones*. Academic Press, San Diego.
- Kumar, V., Rodger, C., Bell, M., 1995. Reaction of steroidal 17-acetylenic alcohols with diethylaminosulfur trifluoride and the structural assignments of 17-fluorosteroids. *J. Org. Chem.* 60, 4591–4594.
- Mc Morris, T., Patil, P., Chavez, R., Baker, M., Clouse, S., 1994. Synthesis and biological activity of 28-homobrassinolide and analogues. *Phytochemistry* 36, 585–589.
- Mussig, C., Fischer, S., Altmann, T., 2002. Brassinosteroid-regulated gene expression. *Plant Physiology* 129, 1241–1251.
- Ramírez, J., Gros, E., Galagovsky, L., 2000. Effect on bioactivity due to C-5 heteroatom substituents on synthetic 28-homobrassinosteroid analogs. *Tetrahedron* 56, 6171–6181.
- Rozen, S., Faust, Y., Ben-Yakov, H., 1979. A new method for fluorination of steroids. *Tetrahedron Lett.* 20, 1823–1826.
- Sakurai, A., 1999a. Biosynthesis. In: Sakurai, A., Yokota, T., Clouse, S. (Eds.), *Brassinosteroids: Steroidal Plant Hormones*. Springer-Verlag, Tokyo, pp. 91–111.
- Sakurai, A., Yokota, T., Clouse, S.D. (Eds.), 1999b. *Brassinosteroids: Steroidal Plant Hormones*. Springer-Verlag, Tokyo.
- Schaefer, S., Medeiro, S.A., Ramírez, J.A., Galagovsky, L.R., Pereira-Netto, A.B., 2002. Brassinosteroid-driven enhancement of in vitro multiplication rate for the marubakaido apple rootstock (*Malus prunifolia* (Willd.) Borkh). *Plant Cell Reports* 20, 1093–1097.
- Sharpless, K.B., Amberg, W., Bennani, Y.L., 1992. The osmium-catalyzed asymmetric dihydroxylation: a new ligand class and a process improvement. *J. Org. Chem.* 57, 2768–2771.
- Shellhamer, D., Briggs, A., Miller, B., Prince, J., Scott, D., Heasley, V., 1996. Reaction of aminosulfur trifluorides with alcohols: inversion vs. retention. *J. Chem. Soc. Perkin Trans. II* 5, 973–977.
- Sung, C.Y., Janzen, L., Pharis, R.P., Back, T.G., 2000. Synthesis and bioactivity of 6 α - and 6 β -hydroxy analogues of castasterone. *Phytochemistry* 55, 121–126.
- Takeo, K., Pharis, R.P., 1982. Brassinosteroid-induced bending of the leaf lamina of dwarf rice seedlings: an auxin-mediated phenomenon. *Plant Cell Physiol.* 23, 1275–1281.
- Welch, J.T., 1987. Advances in the preparation of biologically active organofluorine compounds. *Tetrahedron* 43, 3123–3197.