

## Note

Synthesis of alkyl 1-thio- $\beta$ -D-galactopyranosides

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Among the different methods<sup>1–4</sup> for the synthesis of alkyl 1-thio- $\beta$ -D-glycopyranosides, the method of Černý and Pacák<sup>1,2</sup> has several advantages. It avoids the use of thiols, which have unpleasant smells, and starts with the easily accessible *O*-acetyl-1-thio- $\beta$ -D-glycopyranoses<sup>1</sup>. In a previous communication<sup>5</sup>, we described the synthesis of alkyl 1-thio- $\beta$ -D-xylopyranosides by this method. In this work, the synthesis of alkyl 1-thio- $\beta$ -D-galactopyranosides by the same method is reported.

## EXPERIMENTAL

A solution of 2,3,4,6-tetra-*O*-acetyl-1-thio- $\beta$ -D-galactopyranose<sup>6</sup> (30 mmoles) and alkyl iodide (30 mmoles) in acetone (30 ml) was mixed with a solution of potassium carbonate (30 mmoles) in water (15 ml), and gently refluxed for 90 min. The organic layer was separated, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated in *vacuo*. Because none of the galactose acetates crystallized, and t.l.c. of the reaction mixture indicated partial deacetylation of the acetates, the crude syrup was deacetylated with sodium methoxide<sup>7</sup>. The alkyl 1-thio- $\beta$ -D-galactosides were then crystallized from the appropriate solvent (Table I).

TABLE I

ALKYL 1-THIO- $\beta$ -D-GALACTOPYRANOSIDES

Aglycon group	Yield (%)	Crystallization solvent	M.p. (degrees)	[ $\alpha$ ] <sub>D</sub> <sup>24</sup> <sub>5890</sub> (degrees)	Found (%)		Formula	Calc. (%)	
					C	H		C	H
Pentyl	75	water–acetone	111–112	–39.5	49.5	8.4	C <sub>11</sub> H <sub>22</sub> O <sub>5</sub> S	49.6	8.3
3-Pentyl	53	butanone	140–142	–40.2	49.4	8.5	C <sub>11</sub> H <sub>22</sub> O <sub>5</sub> S	49.6	8.3
Heptyl	68	water	95–96	–36.0	52.9	8.7	C <sub>13</sub> H <sub>26</sub> O <sub>5</sub> S	53.0	8.9
Octyl	63	methanol	106–107	–35.0	54.4	9.3	C <sub>14</sub> H <sub>28</sub> O <sub>5</sub> S	54.5	9.2
3-Phenyl-propyl	51	water–ethanol	97–98	–38.4	57.3	7.1	C <sub>15</sub> H <sub>22</sub> O <sub>5</sub> S	57.3	7.1

The melting points were determined with a Mettler FP2 instrument and are uncorrected. The optical rotations were measured on 0.5% solutions in methanol with a Perkin-Elmer Model 141 photoelectric polarimeter. T.l.c. tests were performed on Silica Gel G (Merck) with acetic acid–water–ethyl acetate (1:1:3). Detection was effected with 5% sulphuric acid in ethanol (10 min at 120°).

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