

STRUCTURE OF SOLANOCAPSINE

Synthesis of 3 β -acetoxy-22,26-acetylepimino-16 α ,23-epoxy-5 α ,22OH,25 β H-cholestan-23 β -ol

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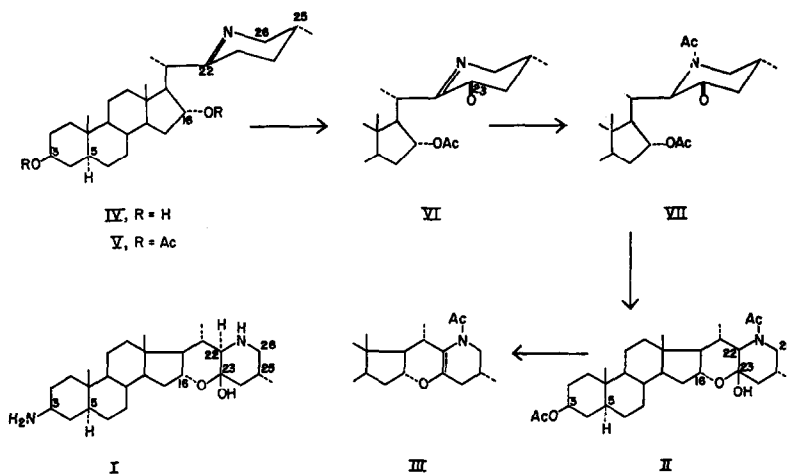
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In a recent publication Ripperger and Schreiber² reported the synthesis of 3 β -amino-22,26-epimino-16 β ,23-epoxy-5 α ,22OH,25 β H-cholestan-23 β -ol (16 β -isomer of solanocapsine) and further submitted evidence (NMR, molecular rotation differences) in support of the presence of a 16 α -epoxy linkage in place of the 16 β in the molecular structure of solanocapsine (I). We have now been able to confirm the correctness of their observation by synthesizing the key solanocapsine derivatives II and III.

Starting from solafioridine³ [IV, (25R)-22,26-epimino-5 α -cholest-22(N)-ene-3 β ,16 α -diol] a minor constituent of Solanum congestiflorum now available synthetically⁴, the synthesis consists in the acetylation of IV-HCl to the 3,16-diacetyl derivative (V), C₃₁H₄₉NO₄, m.p. 173-175°, [α]_D - 56.6°, IR (CCl₄) 1740, 1734, 1240 (OAc), 1658 cm⁻¹ (C=N-), oxidation of V to the 23-oxo compound, VI, C₃₁H₄₇NO₅, m.p. 196-198.5°, [α]_D - 27.5°, IR (CCl₄) 1740, 1736, 1247 (OAc), 1708 (C=O), 1631 cm⁻¹ (C=N-) and reduction of VI followed by acetylation to the 22,26-acetylepimino compound, VII, C₃₃H₅₁NO₆, m.p. 194.5-196.5° [α]_D - 37.7°, IR (CCl₄) 1741, 1738, 1243 (OAc), 1724 (sh, C=O), 1661 cm⁻¹ (NAc), NMR (δ , CDCl₃) 0.82 (2CH₃), 2.03 (2CH₃CO-O-), 2.15 (CH₃CON), MS 557 (Weak M⁺, C₃₃H₅₁NO₆), 497, 454, 437, 394, 343, 283, 182, 155. Hemiketalization of VII with base and reacetylation afforded the 3-deamino-3 β -acetoxy-N'-acetylsolanocapsine (II) (3 β -acetoxy-22,26-acetylepimino-16 α ,23-epoxy-5 α ,22OH,25 β H-cholestan-23 β -ol), m.p. 202-205° (lit. m.p. 174-176°⁵), IR (CHCl₃) 3570 (OH), 1726 (OAc), 1630 cm⁻¹ (N-Ac), MS 515 (M⁺, C₃₁H₄₉O₅N), 497 (M⁺-H₂O). Compound II agreed in properties (m.p., mixture m.p., IR, TLC) with an authentic sample of II prepared from solanocapsine^{5,6}. For confirmation II was further converted into the unsaturated product (III) by refluxing in glacial acetic acid. Its properties, m.p. 262-266°, [α]_D 10.4°, (lit. m.p. 245°, [α]_D¹⁹ 12.4°²) were also in agreement (m.p., mixture m.p., IR, TLC) with an authentic specimen^{2,5}. Barring an unlikely isomerization (C-20

and/or C-22), the synthesis of II and III from solafioridine thus proves the correctness of the structure ascribed to solanocapsine (I).

For the acetylation of IV to V in the above synthesis, an acetylating mixture consisting of acetic anhydride, acetic acid and zinc chloride was utilized. It avoided the formation of the undesirable Δ^{22} -Nac compound³. Manganese dioxide³ in



chloroform proved to be the reagent of choice for the oxidation of V to VI (77% yield). The reduction of the C=N bond in VI was effected with zinc and acetic acid. While the carbonyl moiety remained intact during the reduction, some aromatization of the side chain occurred and the product after acetylation was contaminated with slight amounts of 3 β ,16 α -diacetoxy-20-[2-(5-methylpyridyl)]-5 α -pregnane^{3,7}. An alcoholic KOH base containing water was used in the hemiketalization of VII.

Finally an identical sequence of reactions was also applied to 5,6-dihydropseudo-solasodine B diacetate⁸, the 16 β -isomer of V, for conversion into the 16 β -isomers of II and III. The properties of the compounds obtained thereby were essentially in agreement with those as reported by Ripperger and Schreiber² in their conversion procedure.

REFERENCES

1. Visiting Scientist, National Institutes of Health
2. H. Ripperger and K. Schreiber, Ann. Chem., 723, 159 (1969).
3. Y. Sato, Y. Sato, H. Kaneko, E. Bianchi, and H. Kataoka, J. Org. Chem., 34, 1577 (1969).
4. G. Kusano, N. Aimi, and Y. Sato, J. Org. Chem., in press.
5. K. Schreiber and H. Ripperger, Ann. Chem., 655, 114 (1962).
6. K. Schreiber and H. Ripperger, Z. Naturforsch., 17b, 217 (1962).
7. K. Schreiber and G. Adams, Ann. Chem., 666, 176 (1963).
8. G. Adams and K. Schreiber, Chem. Ber., 99, 3173 (1966).