

STEREOCHEMICAL INTEGRITY OF THE TERMINUS OF THE MIGRATING  
CARBON IN PHOTO-BECKMANN REARRANGEMENTS <sup>1</sup>

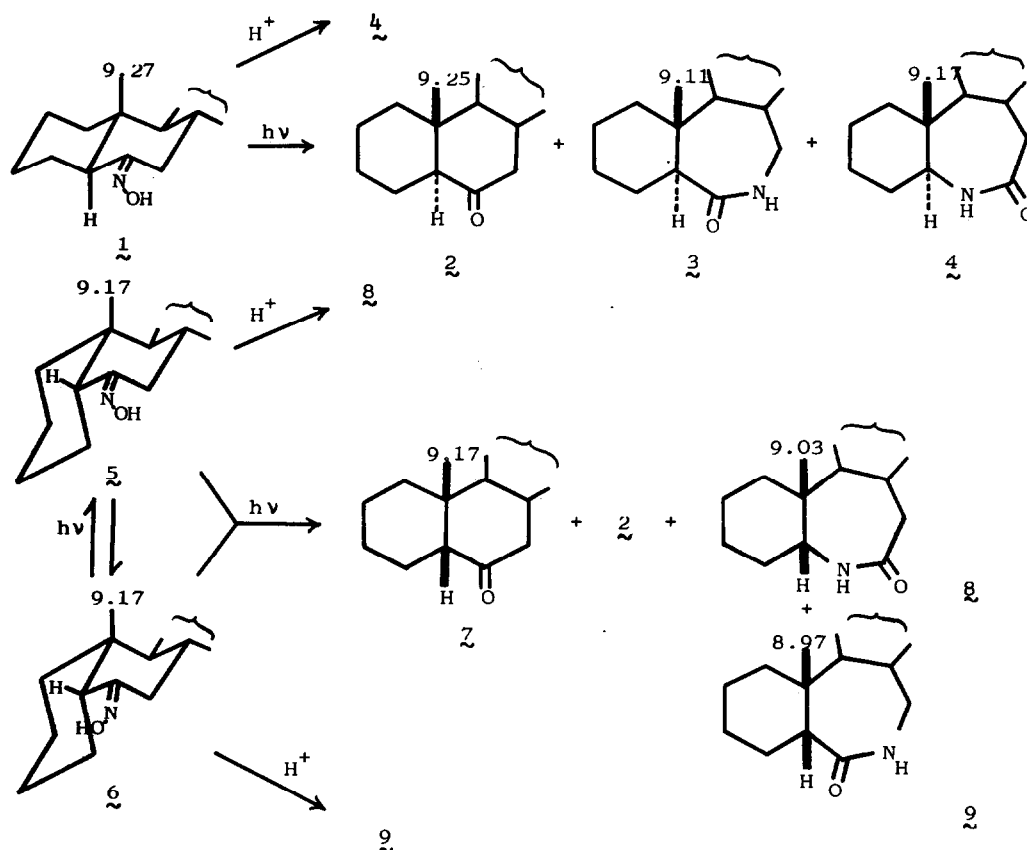
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Since the first report <sup>2</sup> of photo-Beckmann rearrangement a number of additional examples has recently been recorded. <sup>3,4</sup> Oxaziridines were suggested as the intermediate species of the reaction <sup>2,3</sup> and this was demonstrated only recently. <sup>4,5</sup>

We wish to report photo-Beckmann rearrangements of 5  $\alpha$ -cholestan-6-one oxime (1) <sup>6</sup> and the isomeric 5 $\beta$ -cholestan-6-one oximes (5 and 6) which indicate that the photochemical bond migration <sup>7</sup> from intermediary oxaziridines to amides is a stereospecific process, the migration of C-5 occurring with retention of configuration.

Thus, photolysis of 1 (500 mg) in dry methanol (350 ml) by a 150 W low pressure Hg arc lamp for 27 hours afforded three products as confirmed by tlc. By preparative tlc beside the starting oxime (6%), a trace of 5 $\alpha$ -cholestan-6-one (2) and two new compounds both having the same molecular formula C<sub>27</sub>H<sub>47</sub>ON <sup>8</sup> 3, m.p. 110-114°C (11%) and 4, m.p. 174-179°C (15%), were obtained. Both 3 and 4 demonstrated carbonyl absorptions of lactams at 1656 cm<sup>-1</sup> and 1669 cm<sup>-1</sup>, respectively. The structures of these lactams, 3 and 4 were established as 7-aza-B-homo-5 $\alpha$ -cholestan-6-one (3) and 6-aza-B-homo-5 $\alpha$ -cholestan-7-one (4) respectively by the nmr spectra (3, (100 Mz), CONH,  $\tau$  3.96 (1H), N-CH<sub>2</sub>, m,  $\tau$  6.90-7.03 (2H), C-5 H, q,  $\tau$  7.45 (1H), J=5.0 and 11.5 Hz; after deuterium exchange, (60 Mz), N-CH<sub>2</sub>, d,  $\tau$  7.05, J=4.3 Hz; 4, (100 Mz), CONH, d,  $\tau$  4.48 (1H), J=5.5 Hz, C-5 H, q,  $\tau$  6.76 (1H), J=4.0



Arabic numerals denote chemical shifts ( $\tau$ ) of 19 methyl protons.

and 11.0 Hz) as well as by the results of ground state Beckmann rearrangement.

Beckmann rearrangement of **1** with polyphosphoric acid <sup>9</sup> afforded **4** as the sole product (65%), which confirmed that the configuration of the oximino-function in **1** is anti C<sub>5</sub>-C<sub>6</sub> bond. <sup>10</sup>

In contrast to 5 $\alpha$ -cholestan-6-one, 5 $\beta$ -cholestan-6-one <sup>11</sup> gave two isomeric oximes, amorphous **5** and crystalline **6**, m.p. 144-146°, approximately in a ratio of 3:1.

Ground state Beckmann rearrangement of **5** afforded a single crystalline lactam

8, m.p. 95-100°, (80%) and that of 6 yielded amorphous lactam 9 (90%), respectively. These lactams were isomeric and had the molecular formula  $C_{27}H_{47}ON$ . The spectral data (8, i.r.,  $(CHCl_3)$  lactam  $C=O$ ,  $1645\text{ cm}^{-1}$ , n.m.r., (100 Mz), after deuterium exchange, C-5 H, broad d,  $\tau 7.70$ , (1H),  $J=8.0\text{ Hz}$ ; 9, i.r.,  $(CHCl_3)$  lactam  $C=O$ ,  $1643\text{ cm}^{-1}$ , n.m.r., (100 Mz),  $NHCH_2$ , broad s,  $\tau 7.02$  (2H),  $W_H$  10 Hz) are in accord with the structures, 6-aza-B-homo-5 $\beta$ -cholestan-6-one (8) and 7-aza-B-homo-5 $\beta$ -cholestan-6-one (9). On this basis, the configurations of the oximino-groups of 5 and 6 are assigned as anti- $C_5-C_6$  bond and syn- $C_5-C_6$  bond, respectively.<sup>10</sup>

Irradiation of 5 or 6 (500 mg) in dry methanol (250 ml) led to a rapid syn-anti equilibration<sup>12</sup> and it was found that the rate of disappearance of oxime 5 or 6 was considerably faster than that in the case of oxime 1. Careful preparative tlc gave only a pair of lactams 8 and 9 in each case; 5 afforded 8 (28%), 9 (11%), 5 $\beta$ -cholestan-6-one (20%) and 5 $\alpha$ -cholestan-6-one (6%). 6 afforded 8 (30%), 9 (9%), 5 $\beta$ -cholestan-6-one (5%) and 5 $\alpha$ -cholestan-6-one (3%).

The foregoing observations prove the stereochemical integrity of the terminus of the migrating carbon in the photo-Beckmann rearrangement and this may be general in oximes of aliphatic ketones and aldehydes. Moreover, these results suggest that the  $C_5-C_6$  bond of oximes described is not cleaved to give either a radical pair or an ion pair.

It is very probable that the rearrangement of oxaziridines to amides occurs either by heterolytic N-O bond cleavage followed by the formation of  $C=O$  and the migration of C-5 or C-7 to nitrogen concertedly, or through the cleavage of N-O bond and the migration of C-5 or C-7 to nitrogen in a fully concerted manner.

Faster rates observed in cis oximes 5 and 6, when compared with trans oxime 1, must be mainly attributed to the relief of steric compression due to non-bonded interactions of 2 $\alpha$ - and 4 $\alpha$ -axial hydrogens with 7 $\alpha$ - and 9 $\alpha$ -axial hydrogens in going from the oxime 5 or 6 to the lactams 8 and 9.

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- 10) It was pointed out that care must be exercised in relating the stereochemistry of some steroidal oximes with the structure of lactams derived by Beckmann rearrangement. R. H. Mazur, J. Org. Chem., 28, 248 (1963). With regard to the configurations of the oximino-groups of 1, 5 and 6, we confirmed the conclusion from the Beckmann rearrangements by nmr spectroscopy. Hydrogens attached to  $\alpha$ -carbon to oximino-group are known to appear downfield. (W. D. Phillips, Ann. N. Y. Acad. Sci., 70, 817 (1958); H. Saito, K. Nukada, and M. Ohno, Tetrahedron Letters, 2124 (1964); W. F. Trager and A. C. Huitric, Tetrahedron Letters, 825 (1966)). Nmr spectra of 1 and 5 revealed a doublet at  $\tau$ 6.67 ( $J=10.0$  Hz) and at  $\tau$ 6.84 ( $J=10.0$  Hz), respectively. These signals are assignable to the  $7\beta$ -equatorial hydrogens of 1 and 5. Nmr spectrum of 6 demonstrated a diffused single proton quartet centered at  $\tau$ 6.92 ( $J=4$  Hz and 11 Hz), assignable to the angular  $5\beta$ -equatorial hydrogen. These assignments are based on their observed coupling constants coupled with the nmr spectrum and results of Beckmann rearrangement of  $3\alpha,5$ -cyclo- $5\alpha$ -cholestan-6-one oxime. These assignments lead to unambiguous conclusion that the  $\alpha$ -protons syn to oxime hydroxyl of fused cyclohexanones appear downfield from their anti counterpart.
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