THERMAL REARRANGEMENT OF DIALKYL (3-KETOBUTYL) PHOSPHITES

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We have reported the synthesis and certain properties of dialkyl (3-ketobutyl)phosphites obtained by reaction of dialkyl chlorophosphites with β -ketoalcohols in the presence of base [1]. Continuing the study of the properties of these phosphites, we found that upon heating to $160-200^{\circ}$ C for 5-10 h, they rearrange to esters of γ -ketophosphonic acids (by an Arbuzov-type rearrangement).

$$(RO)_2 \ POCH_2CH_2CCH_3 \stackrel{t^{\circ}}{\longrightarrow} (RO)_2 PCH_2CH_2CCH_3$$

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Absorption bands of P=O at 1250 cm⁻¹ and of C=O at 1720 cm⁻¹ are observed in the IR spectrum of the rearrangement products. We heated 7.6 g of diethyl (3-ketobutyl)phosphite for 9 h in a sealed ampule in an atmosphere of nitrogen at 160°C. No pressure was observed upon opening the ampule. Fractionation of the contents of the ampule yielded 5.2 g (68%) of diethyl (γ -ketobutyl)phosphonate having bp 75-78° (0.02 mm); nD²⁰ 1.4388; d₄²⁰ 1.0901. Found %: C 46.30; H 8.15; P 14.70. MR 50.21. C₈H₁₇O₄P. Calculated %: C 46.15; H 8.23; P 14.87. MR 50.02. Literature data [2]: 105-107° (0.3 mm); nD²⁵ 1.4353.

Analogously, 5 g of dibutyl (3-ketobutyl)phosphite was heated for 7.5 h at 190°. We isolated 3.1 g (62%) of dibutyl (γ -ketobutyl)phosphonate having bp 110-112° (0.05 mm); n_D²⁰ 1.4420; d₄²⁰ 1.0168. Found %: C 54.42; H 9.61; P 11.56. MR 68.75. C₁₂H₂₅O₄P. Calculated %: C 54.54; H 9.46; P 11.74. MR 68.49.

LITERATURE CITED

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