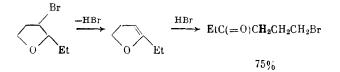
THERMAL ISOMERIZATION OF 2-ETHYL-3-BROMOTETRAHYDROFURAN

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We have shown that 2-ethyl-3-bromotetrahydrofuran (ETHF) undergoes an unusual thermal isomerization. Heating of an equimolar mixture of RR(SS) and RS(SR) diastereomers of 2-ethyl-3-bromotetrahydrofuran leads to their isomerization to 6-bromo-3-hexanone. This isomerization may proceed through consecutive dehydrobromination and opening of the dihydrofuran ring by hydrogen bromide:



Thermolysis of 2-Ethyl-3-bromotetrahydrofuran. Freshly distilled ETHF was heated for 1.5 h in a sealed glass ampul in a finger autoclave under argon at 200°C. The reaction mixture was passed through a layer of alumina and subjected to fractionation.

6-Bromo-3-hexanone. Bp 42°C (60 mm Hg). Found: C, 40.43; H, 6.24%. Calculated for C₆H₁₁BrO: C, 40.25; H, 6.19%. PMR spectrum in CDCl₃ (δ, ppm): 1.1 t (3H, CH₃), 2.45 q (2H, CH₂), 2.6 t (2H, CH₂), 2.1 m (2H, CH₂), 3.4 t (2H, CH₂Br). Mass spectrum (I_{rel}, %): 178/180 (10) $[M]^{+}$, 149/151⁽³²⁾ $[M - Et]^{+}$, 121/123 (20) $[M - EtCO]^{+}$, 107/109 (3) $[CH_2CH_2Br]^{+}$, 93/95 (2) $[CH_2Br]^{+}$, 72 (50) $[EtC(=CH_2)OH]^{+}$, 57 (100 $[EtCO]^{+}$, 41 (50) $[C_3H_5]^{+}$. IR spectrum (ν , cm⁻¹): 1706 (COEt).

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