Synthesis of the Chrysene Bay Region anti-Diolepoxide from Chrysene

By Peter P. Fu and Ronald G. Harvey*
(The Ben May Laboratory for Cancer Research, University of Chicago, Illinois 60637)

Summary The bay region anti-diolepoxide of chrysene (8) is conveniently synthesized from 3,4,5,6-tetrahydrochrysene obtained from chrysene in two steps.

RECENT research has implicated diolepoxide metabolites as the active forms of benzo[a]pyrene¹ and other carcinogenic hydrocarbons.² Although synthetic approaches to several of the isomeric syn and anti arene diolepoxides have been described,¹ these methods are dependent upon the availability of the appropriate dihydroarenes as synthetic intermediates. In the case of chrysene (1), the requisite 1,2- and 3,4-dihydro-(1) are available only via complex multistep synthesis† from phenanthrene and naphthalene, respectively. We now report the synthesis of the bay region anti-diolepoxide of chrysene (8) directly from the parent hydrocarbon.

In preliminary experiments, hydrogenation of (1) over 10% Pd-C catalyst³ at low pressure (45 lb in⁻²) was shown to afford 5,6-dihydro-(1), while similar reaction over PtO₂ gave 1,2,3,4-tetrahydro-(1) along with several minor hydroaromatic products. Hydrogenation of (1) over a mixed Pd-C-PtO₂‡ catalyst under similar conditions cleanly furnished 1,2,3,4,5,6-hexahydrochrysene (2), m.p. 112·5—113·5 °C. Partial dehydrogenation of 1,2,3,4-tetrahydro-(1) through bromination-dehydrobromination⁴ with N-bromosuccinimide (NBS) and 1,5-diazabicyclo[4.3.0]non-5-ene (DBN) furnished 1,2-dihydro-(1) in low yield. In contrast, partial dehydrogenation of (2) with dichlorodicyanobenzoquinone (DDQ) in refluxing benzene⁴,⁵ took place regiospecifically in the alternative molecular region to provide 3,4,5,6-tetrahydrochrysene (3).

Transformation of (3) into (8) was accomplished through the reaction sequence depicted in the Scheme. The experimental procedures employed were patterned after those utilized for related syntheses described in previous papers⁶ and in a review article.¹ The ¹H n.m.r. spectra of all compounds were in complete agreement with the structural assignments (Table). In particular, the H-4 protons of (6)—(8) exhibited the downfield shift characteristic of bay region protons as a consequence of interaction with the adjacent H-5 proton. In further confirmation of the isomeric structural assignments, the u.v. spectrum of (7) matched that reported⁷ for the 1,2-dihydrodiol which differed distinctively from that of the 3,4-dihydrodiol.

SCHEME. i, Pt-Pd, H₂; ii, DDQ; iii, PhCO₂Ag, I₂; iv, NBS,

DBN; v, NaOMe; vi, m-chloroperbenzoic acid.

† Conventional synthesis of 1,2-dihydrochrysene (W. E. Bachmann and W. S. Struve, *J. Org. Chem.*, 1939, 4, 456) involves Friedel-Crafts succinoylation of phenanthrene, separation of the resulting isomeric keto-acids, Clemmensen or Wolff-Kishner reduction, acid-catalysed cyclization (two isomers possible), reduction of the desired isomeric ketone to the alcohol, and dehydration.

[‡] All new compounds gave satisfactory microanalyses for C and H within $\pm 0.3\%$. ¹H N.m.r. spectra of all compounds were consistent with the assigned structures.

J.C.S. CHEM. COMM., 1978

TABLE. ¹H N.m.r. spectra of compounds (4)—(8) (δ relative to Me₄Si).⁸

Compound	H-1	H-2	H-3	H-4
(4) ^b	6·57 (d)	5·37—5·78 (m)	2·16—2·53 (d)	2·63—3·12 (m)
(5)b	$J_{1,2} = 6.6$ 6.71 (d)	5·45—5·83 (m)	2·37—2·70 (m)	3·29—3·66 (m)
(6) _p	$J_{1,2} = 5.$ 6.85 (d)	6.17 (dd)	6·38 (dd)	c
(7) ^d	4·92 (d)	$\begin{array}{c} 5; J_{2,3} = 3.5; J_{3,4} = \\ 4.41 \text{ (ddd)} \end{array}$	6.20 (dd)	7·30 (dd)
(8)d	4.63 (dd)	0.5 ; $J_{2,3} = 2$; $J_{3,4} = 3.90 \text{ (dd)}$: $J_{2,2} = ca.2$; $J_{3,4} = 6.3$	3.76 (dd)	4·99 (d)

^a Aromatic protons in the appropriate ratio were detected in the aromatic region. The H-5 and H-6 benzylic protons of (4) had $\delta 2.77(s)$. Coupling constants are in Hz. ^b Solvent CDCl₃. ^c In the aromatic region. ^d Solvent (CD₃)₂CO + (CD₃)₂CO + 1 drop of D₂O.

Preliminary experiments indicate (8) to be an effective inhibitor of ϕX 174 DNA viral replication.8 The 1,2-dihydrodiol (7) is reported9 to be highly mutagenic towards

S. typhimurium bacterial cells in the presence of hepatic microsomes.

(Received, 7th March 1978; Com. 246.)

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