${\bf 2\text{-}Methylene bicyclo} \ [{\bf 3,2,1}] octa-{\bf 3,6\text{-}diene}, \ a \ Bishomoful vene-type \ Structure$

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Summary From a study of its n.m.r. and u.v. spectra and its dipole moment, it has been shown that the resonance contribution of a bishomofulvene-type structure is important in 2-methylenebicyclo[3,2,1]octa-3,6-diene (2).

WE report the syntheses and properties of 2-methylenebicyclo[3,2,1]octa-3,6-diene (2) and 2-methylenebicyclo-

[3,2,1]oct-3-ene (4). The olefin (2) was obtained in good yield either from the reaction of (1)¹ with triphenylphosphine methylide in THF or from the addition of methylmagnesium iodide to (1) followed by dehydration of the resulting methylcarbinol. The structure of (2) followed from its ¹H n.m.r. $^{+}$ [τ 3·60—4·20 (4H), 4·95 (1H, m), 5·19 (1H, m), 6·71 (1H, m), 7·18 (1H, m), and 7·85 (2H, m)] and mass

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[‡] Detailed analysis of ¹H and ¹⁸C n.m.r. spectra will be reported in a forthcoming full paper.

spectra. The latter spectrum confirmed the molecular weight (118) and the base peak was attributed to $(M-1)^+$,



corresponding to C₉H₉+. Compound (4) was similarly prepared from $(3)^2 \tau 4.20$ (2H, m), 5.20 (2H, m), and 7.5— 8.5 (8H, m)].

The most striking physical properties of (2) are its ¹³C n.m.r. and u.v. spectra and its dipole moment.3

The 13 C n.m.r. chemical shifts of C-2 and C-9 in (2) are δ (Me₄Si) 144.5 and 110.3, respectively. Under identical conditions, the ¹³C n.m.r. chemical shifts of C-2 and C-9 in (4) are δ (Me₄Si) 153.5 and 102.5, respectively, i.e., it could be taken that the C-2 in (2) is more electronegative than that in (4) and the C-9 in (2) is more positive than that in (4).4 This suggests that resonance contribution of (2a) is important in (2).

Measurement of the u.v. spectrum of (2) showed a broad absorption having fine structure at 243 nm (ϵ 12,000) (calculated⁵ λ_{max} 232 nm). The olefin (4) has a spectrum with λ_{\max} at 230 nm, and ϵ_{\max} 5000. This bathochromic effect of 13 nm in (2) can be accounted for in a similar manner $\lceil cf. \ (5) \longleftrightarrow (5a) \rceil.$

The dipole moment was found to be 0.59 D, which suggests a high degree of charge separation in (2).§

These data for (2) are in accord with a delocalized bishomofulvene-type resonance structure (2a).6

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§ The dipole moment was determined by measuring the dielectric constants and densities of solutions of varying molarity at 20° in

benzene and extrapolating.

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