

# A NEW DIAMINO-DERIVATIVE OF AN INORGANIC HETEROCYCLE CONTAINING SILICON, NITROGEN, PHOSPHORUS AND OXYGEN

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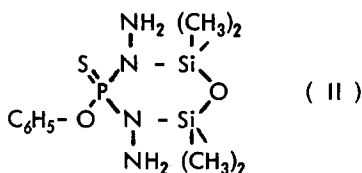
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A few years ago WANNAGAT, GIESEN and RABET (1) reported the synthesis of some inorganic heterocycles containing the elements silicon, nitrogen, phosphorus and oxygen

e.g. type I :



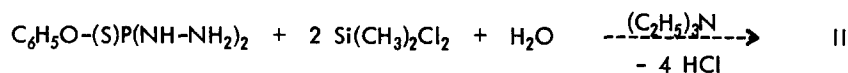
We now report the synthesis of another heterocycle of this type:



3,5-Diamino-2,2,6,6-tetramethyl-4-phenoxy-4-thioxo-1-oxa-3,5-diaza-4-phospha(λ-5)

-2,6-disila-cyclohexane

II is formed by hydrolysis during the reaction of phenoxy-thiophosphoryl-dihydrazide in tetrahydrofuran with dichloro-dimethyl-silane in the presence of triethylamine:



II is obtained in small amounts by distillation of the reaction mixture at 110°C/10<sup>-4</sup> mmHg.

The pure, colourless, wax-like crystals melt at 100 - 101°C.

The constitution of II was confirmed by analysis, <sup>1</sup>H-NMR-, Mass- as well as IR- and RAMAN-spectra.

Analysis (theory in parenthesis) : C : 34.69 (34.46)% , N : 15.55(16.07)% ,

H : 6.38 ( 6.07)% .

The  $^1\text{H}$ -NMR-spectrum (solvens  $\text{CDCl}_3$ ) indicates the phenyl group  $\delta = 7.22$  (multiplett),  $\text{NH}_2$ -groups  $\delta = 3.55$  (singlett) and two different methyl groups (none-equivalent positions below and above flexible ring)  $\delta = 0.26$  and  $\delta = 0.16$  (singletts). The ratio of intensities corresponds very nearly to the theoretical of 5 : 4 : 6 : 6 .

TABLE 1

Some Fragments of II in the Mass-spectrum	
m/e	m/e
348 $\text{M}^+$	149 $\text{O-P(N-Si)}_2\text{OH}_2^+$
255 $\text{SP(N(NH}_2\text{)Si(CH}_3\text{)}_2\text{O)}^+$	132 $(\text{CH}_3)_2\text{Si-OSi(CH}_3\text{)}_2^+$
225 $\text{SP(N(NH}_2\text{)Si(CH}_3\text{)}_2\text{O)}^+$	131 $\text{P(N-Si)}_2\text{O}^+$
209 $\text{O-P(N(NH}_2\text{)Si(CH}_3\text{)}_2\text{O)}^+$	77 $\text{C}_6\text{H}_5^+$

TABLE 2

Main Peaks in the IR- and RAMAN-spectra of II					
IR	RAMAN	Assignement	IR	RAMAN	Assignement
3335 s	3329 w	$\nu_{\text{as}} (\text{NH}_2)$	1597 s	1595 m	$\nu (\text{CC}) \text{ Ar.}$
3265 m	3255 m	$\nu_{\text{s}} (\text{NH}_2)$	1497 s		$\nu (\text{CC}) \text{ Ar.}$
2970 s	2965 m	$\nu_{\text{as}} (\text{CH}_3)$	1198 s	1195 w	$\nu (\text{P-O-Ar})$
	2909 vs	$\nu_{\text{s}} (\text{CH}_3)$	1173 m	1165 m	$\nu (\text{P-O-Ar})$
1630 w sh		$\delta (\text{NH}_2)$	1045 vs		$\nu_{\text{as}} (\text{Si-O-Si})$

#### Reference

1. U. WANNAGAT, K. GIESEN and F. RABET, Z. anorg. allg. Chem. 382(1971) 195