

### Synthesis of 3-Substituted 1-Mesitylenesulfonyloxyureas

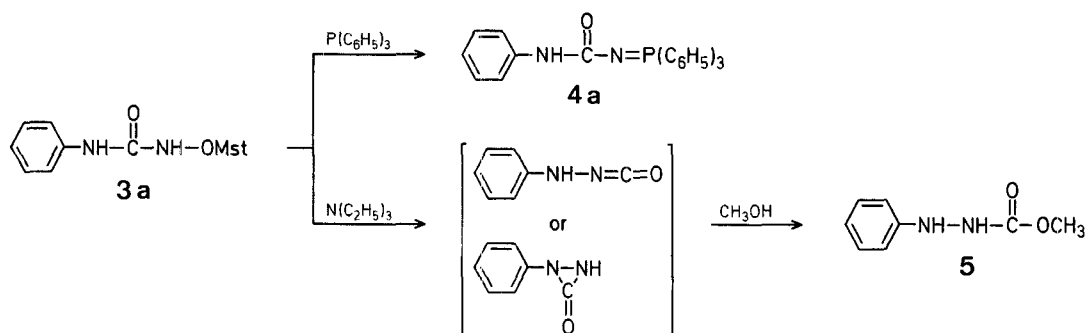
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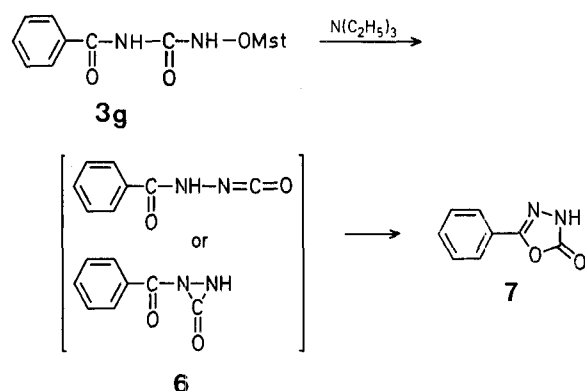
We report here a synthesis of new urea derivatives, 3-substituted 1-mesitylenesulfonyloxyureas (**3**), which are expected not only to react as aminocarbonylaminating agents<sup>1</sup>, but also to be potential precursors of aminocarbonyl nitrenes<sup>2</sup>, aminoisocyanates<sup>3</sup>, or oxodiaziridines<sup>4</sup>.

In general, compounds **3** were prepared by the reaction of equimolar quantities of the isocyanates **2** and *O*-mesitylenesulfonylhydroxylamine<sup>6</sup> (**1**) in dichloromethane at room temperature. The structures of **3** were determined by the elemental and spectral analyses and the chemical evidence: for example, the infrared spectrum of **3a** showed strong absorption bands at 1680 (C=O), 3370, and 3150 cm<sup>-1</sup> (NH). Treatment of **3a** with potassium iodide in acetic acid liberated iodine. Treatment of **3a** with triphenylphosphine in dichloromethane followed by addition of 5% sodium hydroxide gave *N*-phenylaminocarbonyliminotriphenylphosphorane (**4a**) in 61% yield. Similarly **3b** gave *N*-*p*-bromophenylaminocarbonyliminotriphenylphosphorane (**4b**) in 58% yield.





The behavior of **3a** as a precursor of an aminoisocyanate (or an oxidiaziridine) was demonstrated by the reaction with triethylamine in methanol at room temperature to give 1-methoxycarbonyl-2-phenylhydrazine (**5**) in 39% yield. Similar treatment of **3g** in chloroform or methanol, however, gave 2-phenyl-5-oxo-4,5-dihydro-1,3,4-oxadiazole (**7**), the formation of which may involve an initial formation of benzamidoisocyanate (or *N*-benzoyloxodiaziridine) (**6**) followed by an intramolecular cyclization.



at room temperature for 20 min and then the solvent was removed under reduced pressure below 30°. The residual solid was recrystallized from methanol to give white crystals of **3a**; yield: 4.3 g (65%).

#### 1-Methoxycarbonyl-2-phenylhydrazine (**5**):

To an ice-cooled solution of **3a** (334 mg, 1 mmol) in methanol (100 ml) was added dropwise a solution of triethylamine (101 mg, 1 mmol) in methanol (10 ml). After evaporation of methanol, compound **5** was isolated by preparative T.L.C. on alumina (Alumina PF<sub>254</sub>) using chloroform as solvent; yield: 64 mg (39%); m.p. 114–115° (from methanol) (lit.<sup>7</sup> m.p. 115–117°). Its I.R. spectrum (KCl) was identical with that of an authentic sample prepared by a known procedure<sup>7</sup>.

#### 2-Phenyl-5-oxo-4,5-dihydro-1,3,4-oxadiazole (**7**):

To an ice-cooled solution of **3g** (362 mg, 1 mmol) in chloroform (5 ml) was added dropwise triethylamine (101 mg, 1 mmol). After 5 min, ether was added and precipitated triethylammonium mesitylenesulfonate was filtered off. The filtrate was concentrated to dryness and the residual solid was recrystallized from ether/petroleum ether to give **7**; yield: 103 mg (64%); m.p. 137° (lit.<sup>8</sup> m.p. 139°).

Mass spectrum:  $m/e = 162$  ( $M^+$ ), 118, 105, 91, 77.

**Table 1.** 3-Substituted 1-Mesitylenesulfonyloxyureas (**3a–g**)

	R	Yield (%)	m.p. (from CH <sub>3</sub> OH)	Elemental Analyses				
<b>3a</b>		65	161–163	C <sub>16</sub> H <sub>18</sub> N <sub>2</sub> O <sub>4</sub> S (334.39)	calc.	C 57.48	H 5.43	N 8.38
					found	57.74	5.52	8.30
<b>3b</b>		46	188–190	C <sub>16</sub> H <sub>17</sub> BrN <sub>2</sub> O <sub>4</sub> S (413.30)	calc.	C 46.50	H 4.15	N 6.78
					found	46.40	4.17	6.72
<b>3c</b>		34	190–191	C <sub>17</sub> H <sub>20</sub> N <sub>2</sub> O <sub>4</sub> S (348.42)	calc.	C 58.61	H 5.79	N 8.04
					found	58.55	5.84	7.91
<b>3d</b>		67	177–178°	C <sub>16</sub> H <sub>17</sub> ClN <sub>2</sub> O <sub>4</sub> S (368.84)	calc.	C 52.10	H 4.65	N 7.60
					found	52.11	4.64	7.54
<b>3e</b>		36	165–166	C <sub>16</sub> H <sub>17</sub> ClN <sub>2</sub> O <sub>4</sub> S (368.84)	calc.	C 52.10	H 4.65	N 7.60
					found	52.15	4.69	7.55
<b>3f</b>		60	94–95°	C <sub>28</sub> H <sub>56</sub> N <sub>2</sub> O <sub>4</sub> S (510.78)	calc.	C 65.85	H 9.87	N 5.49
					found	65.92	9.97	5.42
<b>3g</b>		55	182–184	C <sub>17</sub> H <sub>18</sub> N <sub>2</sub> O <sub>5</sub> S <sup>a</sup> (362.40)	calc.	C 56.35	H 5.01	N 7.73
					found	56.12	5.14	7.59

<sup>a</sup> I.R. (KCl):  $\nu_{\max} = 1720, 1670$  (C=O)  $\text{cm}^{-1}$ .

Further studies on the chemical properties of compounds **3** are in progress.

#### Reaction of *O*-Mesitylenesulfonylhydroxylamine with Phenyl Isocyanate (**2a**):

To a solution of phenyl isocyanate **2a** (2.4 g, 20 mmol) in dichloromethane (10 ml) was added dropwise a solution of *O*-mesitylenesulfonylhydroxylamine (**1**) (4.3 g, 20 mmol) in dichloromethane (5 ml). The reaction mixture was allowed to stand

#### Reaction of **3a** with Triphenylphosphine:

To an ice-cooled solution of **3a** (167 mg, 0.5 mmol) was added a solution of triphenylphosphine (131 mg, 0.5 mmol) in dichloromethane (5 ml) and the reaction mixture was stirred for 10 min and then treated with 5% sodium hydroxide (5 ml). The organic layer was separated and dried over magnesium sulfate. After evaporation of the solvent, a residual mass was recrystallized from benzene/ether to give **4a**; yield: 120 mg (61%); m.p. 178–179° (lit.<sup>9</sup> m.p. 178–179°).

Similarly **4b** was obtained from **3b** in 58% yield; m.p. 175–176° (lit.<sup>9</sup> m.p. 169–171°).

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- <sup>1</sup> Other known aminocarbonylaminating agents involve *N*-chlorourea<sup>2</sup> and aminocarbonylazide<sup>3</sup>.
- <sup>2</sup> R. A. Wiesboeck, *J. Org. Chem.* **30**, 3161 (1965).  
S. Oae, T. Masuda, K. Tsujihara, N. Furukawa, *Bull. Chem. Soc. Japan* **45**, 3586 (1972).  
Y. Wada, R. Oda, *Bull. Chem. Soc. Japan* **43**, 2167 (1970).
- <sup>3</sup> W. Lwowski, in: S. Patai, *The Chemistry of the Azide Group*, Interscience Publ., New York, 1971, p. 520–527.  
E. Lieber, R. L. Minnis, Jr., C. N. R. Rao, *Chem. Rev.* **65**, 377 (1965).
- <sup>4</sup> R. F. Smith, T. C. Rosenthal, P. T. Hussong, P. G. Bori, *Tetrahedron Lett.* **1970**, 4007.
- <sup>5</sup> W. Lwowski, R. A. de Mauriac, T. W. Mattingly, Jr., E. Scheffele, *Tetrahedron Lett.* **1964**, 3285.  
H. J. Timpe, H. G. O. Becker, *J. Prakt. Chem.* **314**, 325 (1972).
- <sup>6</sup> Y. Tamura, J. Minamikawa, K. Sumoto, M. Ikeda, *J. Org. Chem.* **38**, 1239 (1973).
- <sup>7</sup> G. Heller, *Liebigs Ann. Chem.* **263**, 281 (1891).
- <sup>8</sup> R. Stolle, K. Krauch, *Ber. dtsch. chem. Ges.* **45**, 3311 (1912).
- <sup>9</sup> I. N. Zhmorova, A. P. Martynyuk, *Zh. Obshch. Khim.* **38**, 876 (1968).