

Constituents of Labiatae.¹ Anagadiol, a New Triterpene from *Salvia Broussonetii* Benth.

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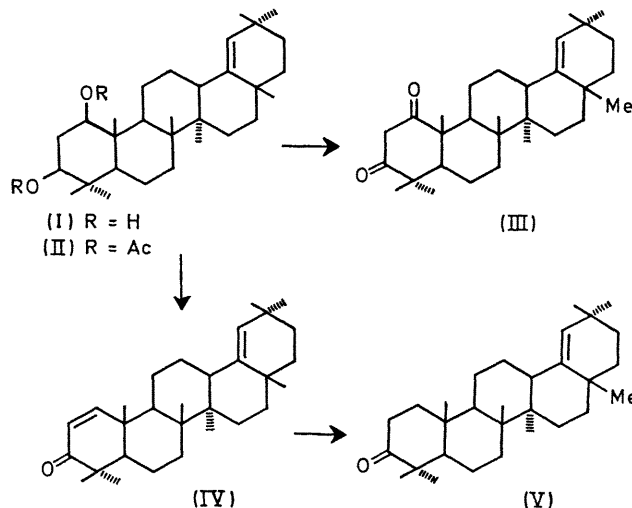
Summary Chemical and spectral evidence for the $1\beta,3\beta$ -dihydroxyolean-18-ene structure of anagadiol (I), a new triterpene isolated from the aerial parts of *Salvia Broussonetii* Benth., are reported.

In connection with our work on the chemical components of Labiatae we have investigated *Salvia Broussonetii* Benth., endemic to the Canary Islands.² The concentrated ethanolic extract of the aerial parts of this plant was percolated through silica gel. The fractions eluted with CHCl_3 showing positive Liebermann-Burchard test were acetylated and chromatographed on a dry column. Elution with benzene gave several triterpenes. One of them has the empirical formula $\text{C}_{34}\text{H}_{54}\text{O}_4$ (M^+ 526), m.p. $247-255^\circ$. Its n.m.r. spectrum shows the presence of eight angular Me groups (τ 8.92–9.25), one vinyl hydrogen (5.18), two Me-CO (8.00, 8.04), and two hydrogen atoms on carbon bearing oxygen (5.31, 5.42, q). Hence, this substance (II) is a diacetate. Its mass spectrum shows the characteristic pattern³ of the triterpenes of the Δ^{18} -oleanene series (peaks at m/e 205, 204 base peak, 190, 189, 177). The corresponding alcohol was named anagadiol (I), $\text{C}_{30}\text{H}_{50}\text{O}_2$, m.p. $212-214^\circ$, $[\alpha]_D +5^\circ$ (CHCl_3).

Jones oxidation of (I) yielded a diketone (III), $\text{C}_{30}\text{H}_{46}\text{O}_2$, m.p. $112-114^\circ$, $[\alpha]_D +61^\circ$ (CHCl_3); i.r. (CHCl_3) 1720, 1703 cm^{-1} ; u.v. (EtOH) 260 nm. By adding alkali the u.v. maximum suffers a bathochromic shift to 293 nm. This behaviour is typical of the enolizable $\text{O}=\text{C}-\text{CH}_2-\text{C}=\text{O}$ group.⁴ The presence of a β -diketone system as shown in (III) is supported by examining its n.m.r. spectrum: it shows a doublet (probably a degenerated quartet, 2H) at τ 6.60, 6.64 which disappears by equilibration with D_2O .⁵

Oppenauer oxidation of (I) gave an $\alpha\beta$ -unsaturated ketone (IV), $\text{C}_{30}\text{H}_{46}\text{O}$, m.p. $163-164^\circ$, $[\alpha]_D +29^\circ$ (CHCl_3); i.r. (CHCl_3) 1670, 1620 cm^{-1} ; u.v. (EtOH) 232 nm; n.m.r. τ 4.14 and 2.77 (2d, J 10 Hz). Catalytic reduction of (IV)

yielded germanicone (V), m.p. $189-190^\circ$, $[\alpha]_D +40^\circ$ (CHCl_3). Hence, anagadiol possesses structure (I).



Its stereochemistry at C-1 and C-3 was deduced from the n.m.r. spectrum of (II) at 100 MHz. The two protons geminal to the acetyl groups give rise to partially superimposed quartets, part X of two independent ABX systems, whose $J_{AX} + J_{BX}$ are 16 and 17 Hz, respectively. This suggests axial-axial and axial-equatorial couplings which are compatible only with a $1\beta\text{-OH}, 3\beta\text{-OH}$ configuration. From all these results we conclude that anagadiol must be $1\beta,3\beta$ -dihydroxyolean-18-ene (I). This is the first time that a natural product of the oleanene series has been found to be oxidized at C-1.

Satisfactory analyses were obtained for all the compounds

mentioned. The n.m.r. spectra were recorded in CDCl_3 with Me_4Si as internal reference.

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¹ For previous paper in the series, see J. M. Arteaga, J. L. Breton, B. M. Fraga, and A. G. González, *Anales de Quím.*, 1970, **66**, 181.

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