

DARUTOSIDE, A DITERPENOID FROM *SIEGESBECKIA PUBESCENS* AND ITS STRUCTURE REVISION

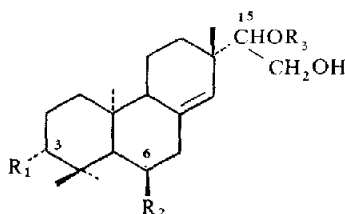
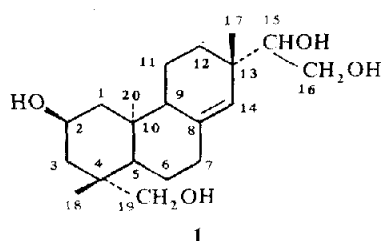
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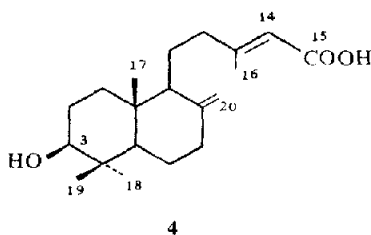
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Key Word Index—*Siegesbeckia pubescens*; *S. orientalis*; Compositae; ^{13}C NMR: *ent*-pimarane-type glucoside; darutoside.

From *Siegesbeckia pubescens* (Compositae), which is an Asian folk medicine 'Hi-cheom', several *ent*-kaurane-type diterpenes and an *ent*-pimarane-type diterpene (1) have been isolated [1-3]. In addition, a diterpene glucoside (2), $\text{C}_{26}\text{H}_{44}\text{O}_8 \cdot \text{H}_2\text{O}$, mp 225-226°, $[\alpha]_D^{25} -48^\circ$ (MeOH; c 1.0) which gave an aglycone (3), $\text{C}_{20}\text{H}_{34}\text{O}_3$, mp 163°, $[\alpha]_D^{25} -10^\circ$ (MeOH; c 1.0) by hydrolysis with β -D-glucosidase, was also isolated and the structures 2a and 3a were tentatively proposed for this glucoside and its aglycone, respectively [3].



	R ₁	R ₂	R ₃
2	OGlc*	H	H
2a	H	OGlc	H
2b	OH	H	Glc
3	OH	H	H
3a	H	OH	H



* Glc = β -D-Glucopyranosyl

Table 1. ^{13}C -Chemical shifts (δ_c) for 1, 2, 3 (in $\text{C}_5\text{D}_5\text{N}$) and 4 (in CDCl_3)

Carbon No.	1	2	3	4*
1	49.2	36.9	37.5	37.1
2	63.9	23.9	28.5	27.8
3	45.5	85.1	78.2	78.9
4	39.7	38.5	39.5	39.2
5	55.5	54.8	54.7	54.7
6	22.6	22.5	22.8	24.0
7	36.7	36.3	36.5	38.2
8	138.0	138.1	138.5	147.7
9	51.3	50.6	51.0	56.0
10	40.9	37.9	38.3	39.5
11	19.0	18.6	18.8	21.7
12	32.7	32.7	32.9	40.0
13	37.9	37.8	37.6	163.7
14	129.7	129.3	129.6	114.9
15	76.5	76.5	76.7	171.4
16	63.9	63.8	64.0	19.2
17	23.2	23.2	23.3	14.6
18	28.2	28.8	29.0	28.3
19	64.8	14.7	14.9	15.4
20	16.9	17.1	16.6	106.9
Glc-1		102.2		
2		74.9		
3		77.9		
4		71.8		
5		78.3		
6		63.0		

* Quoted from ref. [8].

The assignments of ^{13}C NMR spectra of pimarane-type diterpenes have been reported [4-7]. Referring to these results, we assigned the carbon resonances of 1, reconfirming the structure proposed by Murakami *et al.* [2]. On the basis of these assignments, the spectrum of 3 was compared with those of the 3β -(equatorial)-hydroxy-labdane-type diterpene (4) [8] and the 3β -(equatorial)-hydroxy-triterpenes [9]. As shown in Table 1, signals due to the A-ring carbons (C-1, -2, -3, -4 and -5) of 3 were found at almost the same positions as those of 4, and a signal at δ 22.8 (triplet) of 3 must be attributable to the unsubstituted methylene group of C-6. It follows that the former proposal of the allocation of the third hydroxyl group of 3 at C-6 must be revised to be located at C-3 (equatorial).

The stereochemistry of carbon signal displacements of both aglycone and sugar moieties of glycosides has been reported [10]. On going from 3 to 2, a β -D-glucoside of 3, a signal due to C-3 was deshielded by +6.9

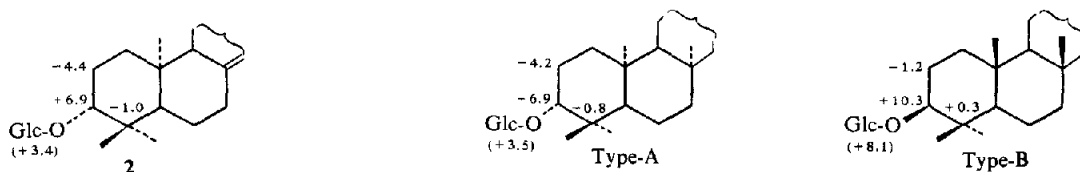


Fig. 1. β -D-Glucosylation NMR chemical shifts. Each value indicates the glucosylation shift ($\Delta\delta = \delta_{\text{glucoside}} - \delta_{\text{aglycone}}$). Values in parentheses indicate the glucosylation shift of anomeric carbon ($\Delta\delta = \delta_{\text{glucoside}} + \delta_{\beta\text{-glucose}}$). Glc = β -D-Glucopyranosyl.

ppm and signals attributable to C-2 and C-4 were displaced upfield by -4.2 and -0.8 ppm, respectively, while other carbon signals including C-15 and C-16 remained almost unshifted. Comparison of the magnitude of these displacements with those of β -D-glucopyranosides, types-A and -B (Fig. 1) [10] led to the conclusion that in the structure of **2**, the β -D-glucosyl linkage must be located at the 3-equatorial hydroxyl group and the chirality of this hydroxyl group should be represented by the *R* configuration. This means that the absolute configuration of **3** should be the *ent*-type, the same as for type-A. With regard to the carbon resonances of the sugar residue, the spectrum of **2** exhibited signals corresponding to a β -D-glucopyranosyl moiety at almost the same positions as those of type-A. It should be noted that the anomeric carbon signal of type-B appears at a remarkably higher field than that of type-A [10]. Consequently, the structure of **2** can be now formulated as illustrated.

Previously, Diara *et al.* [11] isolated a glucoside named darutoside from *Siegesbeckia orientalis*, for which the structure **2b** was proposed. Direct comparison of the ^{13}C NMR spectrum, mp and optical rotation of **2** with those of an authentic sample of darutoside definitely indicated the identity of both glucosides, leading to the revision of the structure of darutoside from **2b** to **2**.

EXPERIMENTAL

^{13}C NMR spectra were recorded on a JEOL PFT-100 spectrometer, equipped with EC-6 computer, operating at 25.15 MHz (pulse flip angle of 90° , at pulse interval of 1–2 sec). Samples were studied as 0.2–0.5 M soln in $\text{C}_5\text{D}_5\text{N}$ with TMS as internal

standard; 500–15000 accumulations with 8192 data points for 4 KHz were used.

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REFERENCES

1. Canonica, L., Rindone, B., Scolastico, C., Han, K. D. and Kim, J. H. (1969) *Tetrahedron Letters* 4801.
2. Murakami, T., Isa, T. and Satake, T. (1973) *Tetrahedron Letters* 4991.
3. Han, K. D., Kim, J. H. and Oh, S. J. (1975) *J. Pharm. Soc. Korea* **19**, 129.
4. Matsuo, A., Uto, S., Nakayama, M., Hayashi, S., Yamasaki, K., Kasai, R. and Tanaka, O. (1976) *Tetrahedron Letters* 2451.
5. Wenkert, E. and Buckwalter, B. L. (1972) *J. Am. Chem. Soc.* **94**, 4367.
6. Polonsky, J., Baskevitch, Z., Bellavita, N. C., Ceccherelli, P., Buckwalter, B. L. and Wenkert, E. (1972) *J. Am. Chem. Soc.* **94**, 4369.
7. Cambie, R. C., Burfitt, I. R., Goodwin, T. E. and Wenkert, E. (1975) *J. Org. Chem.* **40**, 3789.
8. Brawn, S. and Breitenbach, H. (1977) *Tetrahedron* **33**, 145.
9. Asakawa, J., Kasai, R., Yamasaki, K. and Tanaka, O. (1977) *Tetrahedron* **33**, 1935.
10. Kasai, R., Suzuo, M., Asakawa, J. and Tanaka, O. (1977) *Tetrahedron Letters* 175.
11. Diara, A., Asselineau, C., Laszlo, P. and Pudles, J. (1963) *Bull. Soc. Chim. Fr.* 99.