Polypeptides. XLVII. Effect of the Pyrazole-Imidazole Replacement on the Biological Activity of Thyrotropin-Releasing Hormone^{1,2}

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I-Pyroglutamyl-L- β -(pyrazolyl-3)-alanyl-I-prolinamide, an analog of thyrotropin-releasing hormone (TRH) in which the imidazole residue is replaced by L- β -(pyrazolyl-3)-alanine, has been synthesized and evaluated biologically. The analog exhibited the characteristic biological responses of natural or synthetic TRH; its activity was approximately 5% that of the parent hormone. These results suggest that the acid-base properties of the imidazole ring are not essential for thyrotropin-releasing activity.

In a previous communication³ it was suggested that the replacement of histidine by $L-\beta$ -(pyrazolyl-3)-alanine [Pyr(3)Ala] may provide valuable information regarding the importance for function of the characteristic imidazole pK of histidine in biologically active peptides containing this residue. β -(Pyrazolyl-3)-alanine is isosteric with histidine but the acid-base behavior of the pyrazole ring differs markedly from that of imidazole. The validity of this concept was verified experimentally by the observation that β -(pyrazolyl-3)-alanine¹² S-peptide₁₋₁₄ was a powerful competitive inhibitor of RNase-S.^{4,5} The analog combines with S-protein as firmly as does S-peptide but the pyrazole ring fails to function catalytically because of its inability to accept or donate protons under the conditions of assav.

In order to gain information regarding the functional importance of histidine for hormone activity the Pyr-(3)Ala analogs of Val⁵-angiotensin II⁶⁻⁸ and of Gln⁵- β -corticotropin₁₋₂₀amide^{9,10} were prepared and tested for biological activity. Since both analogs possessed significant levels of biological activity it was concluded that the acid-base characteristics of the imidazole ring cannot be essential for the function of these hormones.

The recent disclosure of the covalent structure of a thyrotropin-releasing hormone TRH (TRF) $(I)^{11-14}$

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(1) See L. Moroder and K. Hofmann, J. Med. Chem., 13, 839 (1970) for paper XLVI in this series.

(2) Supported by grants from the Hoffmann-La Roche Foundation, Nutley, N. J., and U. S. Public Health Service Grant AM-06164-07.

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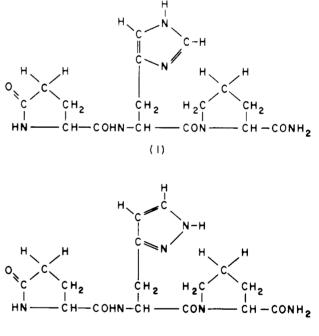
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provided another opportunity to test the importance of histidine dissociation for hormonal activity by the pyrazole-imidazole substitution approach. This communication describes a synthesis of L-pyroglutamyl-L- β -(pyrazolyl-3)-alanyl-L-prolinamide (II) [Pyr(3)Ala TRH] and relates observations pertaining to certain biological properties of this molecule.



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Preparative Aspects.—The synthetic route to the pyrazole analog of TRH takes into account the wellknown tendency of dipeptides containing proline to undergo cyclization with formation of diketopiperazines.¹⁵ N^{α} -Benzyloxycarbonyl-L- β -(pyrazolyl-3)alanine¹⁶ was treated with *tert*-butoxycarbonylhydrazine in the presence of DCI to give an amorphous *tert*butoxycarbonylhydrazide which was subjected to hydrogenolysis over Pd in MeOH. The ensuing crystalline L- β -(pyrazolyl-3)-alanine *tert*-butoxycarbonylhydrazide was then acylated with 2,4,5-trichloro-

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N-benzyloxycarbonyl-L-pyroglutamate¹⁷ phenyl \mathbf{to} N-benzyloxycarbonyl-L-pyroglutamyl-L- β -(pyrgive azolyl-3)-alanine tert-butoxycarbonylhydrazide. The tert-butoxycarbonylhydrazide was converted into the hydrazide TFA salt by treatment with 90% TFA and the latter via the azide was coupled with L-prolinamide.18 The reaction product was distributed between *n*-BuOH and 2% AcOH and crude *N*-benzyloxycarbonyl-L-pyroglutamyl-L-β-(pyrazolyl-3)-alanyl-Lprolinamide was isolated from the *n*-BuOH phases. This material was subjected to hydrogenolysis over Pd in MeOH-AcOH to give crude II which was purified on a Sephadex G 50 partition column with n-BuOH-Ac- $OH-H_2O$ (4:1:5) lower phase as the stationary and the upper phase as the moving component. The process of separation was monitored by tlc using the Cl test for identification of products. Lyophilization from H_2O afforded the monhydrate of II in the form of a highly water-soluble, colorless, amorphous powder.

Experimental Section

A. General Procedures.—Melting points are uncorrected. Rotations were determined with a Zeiss precision polarimeter. Measurements were carried out with a Hg lamp at 546 and 578 $m\mu$ and extrapolated to the 589- $m\mu$ Na line. Elemental analyses were performed by Schwarzkopf Microanalytical Laboratory, Woodside, N. Y.; the O value was actually determined and not computed by difference. The amino acid composition of acid hydrolysates was determined with a Beckmann-Spinco Model 120 amino acid analyzer.¹⁹ Acid hydrolyses were performed in constant boiling HCl at 110° for 24 hr in evacuated tubes; values are not corrected for amino acid destruction. See Hofmann, et al.,²⁰ for designation of solvent systems for tlc.

B. Preparation of Peptides. L- β -(Pyrazolyl-3)-alanine tert-Butoxycarbonylhydrazide.—DCI (2.06 g) was added to an icecold soln of N α -benzyloxycarbonyl-L- β -(pyrazolyl-3)-alanine¹⁶ (2.89 g) and tert-butoxycarbonylhydrazine (1.32 g) in EtOAc (90 ml) and DMF (30 ml). The suspension was stirred for 5 days at room temperature, the DCU was removed by filtration, and the filtrate was evapd. The residue was dissolved in EtOAc and the soln was washed with ice-cold 1 N citric acid, satd aq NaHCO₃, and H₂O, and dried. Evaporation of the solvent gave an oily residue which was hydrogenated for 12 hr over Pd in 60 ml of MeOH. The filtrate from the catalyst was evaporated to give a syrup which soon deposited rosettes of prisms; these were washed with MeOH-Et₂O (1:10) and recrystd from the same solvent: yield 1.47 g (55%); mp 156-157°; [α]²⁷D +6.4° (c 1.15, MeOH). Anal. (C₁₁H₁₉N₅O₃).

N-Benzyloxycarbonyl-L-pyroglutamyl-L-β-(pyrazolyl-3)-alanine tert-Butoxycarbonylhydrazide.—TEA (0.278 ml) was added to an ice-cold soln (20 ml) containing 2,4,5-trichlorophenyl Nbenzyloxycarbonyl-L-pyroglutamate¹⁷ (886 mg) and L-β-(pyrazolyl-3)-alanine tert-butoxycarbonylhydrazide (538 mg). The mixture was stirred at 0° for 30 min and at room temp for 20 hr. The solvent was then removed, the residue was dissolved in EtOAc, and the soln was washed in the usual manner. Evaporation afforded an oil which was triturated with petroleum ether. The ensuing precipitate was collected and purified by 2 pptn from a small vol of EtOH with Et₂O: yield 618 mg (60%); sinters at 120°; mp 140–142° dec; [α]²⁴D –49.3° (c 1.09, MeOH); R_f^{VI} 0.6; R_f^{VII} 0.8, single CI and hydrazide-positive ninhydrin negative spot; amino acid ratios in acid hydrolysate, Glu_{1.1}Pyr-(3)Ala_{0.5}; recovery 95%. Anal. (C₂₄H₃₀N₆O₇).

L-Pyroglutamyl-L- β -(pyrazolyl-3)-alanyl-L-prolinamide (II). N-Benzyloxycarbonyl-L-pyroglutamyl-L- β -(pyrazolyl-3)-alanine tert-butoxycarbonylhydrazide (320 mg) was converted into the free hydrazide trifluoroacetate salt by treatment with 90% TFA in the usual manner; yield 310 mg. This hydrazide salt

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was dissolved in DMF (3 ml), and 6.91 N HCl in dioxane (0.42 ml) followed by t-BuONO (0.074 ml) was added with stirring at -25 to -30° . After 15 min at this temperature the hydrazide test was negative. The soln was cooled at -60° and TEA (0.488 ml) was added followed by a soln containing prolinamide HCl¹⁸ (88 mg), TEA (0.081 ml) in H₂O (0.2 ml), and DMF (2 ml). This mixture was stirred at 5° for 20 hr then the solvent was evapd. This material was distributed between n-BuOH and 2% AcOH, the *n*-BuOH phases were evaporated and the residue was triturated with Et₂O. The ensuing solid (308 mg) was hydrogenated over Pd in MeOH (3 ml) and 20% AcOH (1.5 ml) and the product was lyophilized from H_2O ; yield 190 mg. This product (100 mg) was purified on a Sephadex G 50 partition column (0.5 \times 140 cm) with *n*-BuOH-AcOH-H₂O (4:1:5) lower phase as the stationary and the upper phase as the mobile component. Fractions (3 ml each) were collected at a flow rate of 3 ml/hr and the progress of the separation was monitored by the using the Cl test as the indicator. Fractions containing the desired compd $(R_{\rm f}^{\rm I}|0.4)$ were pooled, the solvent evapl, and the residue lyophilized from H₂O: colorless, fluffy, electrostatic, highly water-soluble powder; yield 80 mg; $[\alpha]^{2_{2_{0}}} - 62.3^{\circ} + c$ 1.41, H₂O) $R_{t}^{v_{1}}$ 0.4; $R_{t}^{v_{11}}$ 0.5; amino acid ratios in acid hydrolysate, Glu_{1.0}Pyr(3)Ala_{1.0}Pro_{0.9}; recovery 99.4%. Anal. $(C_{16}H_{22}N_6O_4,H_5O)$ calcd, C, 50.5; H, 6.4; N, 22.1; O, 21.03, Found, C, 51.0; H, 6.2; N, 21.8; O, 21.13.

Biological Results and Discussion

Linear and parallel dose–response curves (Table I) were obtained with 20, 60, and 90 mµg of Pyr(3)Ala

TABLE 1					
Dose-Responses to TRH and					
Pyr(3)Ala TRH in T3-TRH Assay Mice ^a					
-TRH, mµg/mousePyr(3)Ala TRH, mµg/mouse					
	10 - 20	60 180		20 10,000	
epm ¹²⁵ I, ∆epm	125], Дерт				
$118 591 \ 2319 \ 3961$	$125 \ 403$	2133 427	$4 \ 4497 \ 5$	5956 5890	
^a Each result is mean obtained from 5 mice.					

TRH and 1,3, and 9 mµg of synthetic TRH in triiodothyronine T₃-TRH assay mice. Larger doses of the analog only slightly increase the response. In this assay, the ¹²⁵I blood levels are measured before and 2 hr after iv injection of TRH or the analog. The results, recorded as blood ¹²⁵I Δ cpm, represent the difference in these values and are proportional to the amount of TSH released from the pituitary.²¹

When the T_{3} -TRH assay mice were pretreated with 0.2 μ g of T_{3} 2 hr before the injection of Pyr(3)Ala TRH or TRH, the responses were inhibited (Table II). Thus, as has been shown for TRH, the magnitude of the response of the Pyr(3)Ala TRH depends on the amount of T_{3} present. Studies on the interaction of T_{3} and TRH in regulation of TSH secretion have been described.^{22,23}

TABLE II

INHIBITION OF TRH AND PYR(3)ALA TRH				
Responses by Triiodothyronine in T ₃ -TRH Assay Mice				
	¹²⁵ I Aepm	¹²⁵ I Aepm ^a	p value	
TRH, 9 mµg	4100	20	0.001	
Pyr(3)Ala TRH, 60 mµg	2670	-100	0.001	
^a T_3 (0.2 μ g) was administered sc 2 hr before iv TRH or analog;				
each result is mean obtained from 5 mice.				

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To demonstrate that the site of action of Pyr(3)Ala TRH is the anterior pituitary gland the *in vitro* method was used.²⁴ In this assay corresponding halves of rat anterior pituitary glands are incubated in two beakers (control and experimental) containing Krebs-Ringer bicarbonate solution. The test substance is added to the experimental beaker and after a 1-hr incubation period the amount of TSH in the medium of the control and experimental beakers is measured in T₃-TSH assay mice. The results are recorded as the change in the blood level of ¹²⁵I. The observation (Table III), that more TSH is present in the experimental (E) than in the control (C) medium shows that Pyr(3)Ala TRH, like TRH, stimulates release of TSH into the medium.

TABLE III				
Comparison of TRH and Pyr(3)Ala TRH				
ACTIVITY IN THE in Vitro RAT ANTERIOR PITUITARY				
INCUBATION ASSAY ^a				
	С	\mathbf{E}		
	125I ∆epm	125I ∆epm	p value	
TRH, 9 mµg	327	1506	0.001	
Pyr(3)Ala TRH, 180 mµg	203	3887	0.001	
^a Values recorded represent	t amount of	TSH presen	it in mediu	

^a Values recorded represent amount of TSH present in medium without (C) and with added hormones (E). TSH measured in T_3 -TSH assay mice; each result is the mean obtained from 5 mice.

Incubation of the analog in human serum for 30 min at 37° destroyed biological activity (Table IV). For comparison the analog was added to the serum and assayed without incubation. Parallel experiments were performed using synthetic TRH.

TABLE IV INACTIVATION OF TRH AND Pyr(3)ALA TRH

BY INCUBATION IN HUMAN SERUM^a

	С	\mathbf{E}	
	¹²⁵ I ∆epm	¹²⁵ I ∆cpm	p value
TRH, 9 mµg	3740	592	0.001
Pyr(3)Ala TRH, 180 mµg	3480	-53	0.001

^a Each 0.2 ml of human serum obtained from a normal adult male contained TRH or analog as indicated. Serum without (C) or with (E) incubation (30 min at 37°) was assayed in T_{3} -TRH assay mice; each result is mean obtained from 5 mice.

The ability of Pyr(3)Ala TRH to inhibit the response of TRH in T₃-TRH assay mice was also investigated. Since previous studies indicated that the onset of the TRH response, TRH disappearance rate from plasma, and TRH inactivation by serum were rapid, the *in vivo* inhibitory effect of the analog was assessed

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by administering it at the same time or shortly before the TRH injection. When mice received subactive doses (10–15 mµg) of the analog in amounts 3–5 times the administered dose of TRH (3 mµg), there was no inhibiton of the TRH response (Table V, expt 1 and 2). Inhibitory studies were also performed (expt 3) using active doses of Pyr(3)Ala TRH. When TRH and Pyr(3)Ala TRH were both given to the same mouse, their effects were additive.

TABLE V			
INHIBITION OF THE TRH RESPONSE IN			
T ₃ -TRH MICE BY PYR(3)ALA TRH ^a			

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<u> </u>		p val	Expt 1 - Expt 2- ¹²⁶ Ι Δcpm ¹²⁶ Ι Δcpm p value p value S vs. P S vs. P		Expt 3 ¹²⁵ Ι Δepm p value S vs. P			
\mathbf{S}		\mathbf{TRH}	4015		1221			
Р		TRH	3438	ns	1191	\mathbf{ns}		
	\mathbf{S}	\mathbf{TRH}	2571		1000		2425	
	Ρ	$\mathbf{T}\mathbf{R}\mathbf{H}$	3048	\mathbf{ns}	1230	ns	3637	0.01
		S + TRH			1643		1900	
		P + TRH			1334	ns	4100	0.001
		S	36	8	3	7	-184	

^a Albumin (1%)-0.9% NaCl soln (S) or Pyr(3)Ala TRH (P) was injected iv before (at times recorded) or with synthetic TRH. All mice given TRH received 3 mµg while 10 and 15 mµg of analog were administered in expt 1 and 2, respectively. Mice in expt 3 received 3 mµg of TRH and 10 µg of Pyr(3)Ala TRH according to the above design. The difference in the magnitude of the TRH response is due to variations in the sensitivity of the mice; each result is mean obtained from 5 mice.

From the results presented in this communication it can be concluded that under the conditions of assay Pyr(3)Ala TRH stimulates the release of TSH from the pituitary of the mouse and that it possesses approximately 5% the biological activity of natural or synthetic TRH. The analog fails to compete with TRH and when it is administered in conjunction with the natural hormone exhibits an additive effect. Just as in the case of the natural hormone, the analog is inactivated when incubated with human serum and its effect is inhibited by T_3 . Thus its spectrum of activities as far as it has been investigated parallels that of TRH, albeit at a lower level.

These experimental results eliminate the characteristic acid-base behavior of the imidazole ring of histidine as an essential feature for TSH release. The situation is similar to that which has been observed with Val⁵angiotensin II and Gln⁵- β -corticotropin₁₋₂₀ amide where the replacement of the histidine residue by Pyr(3)Ala resulted in analogs which retained biological activity but at a lower level.