Preliminary communication

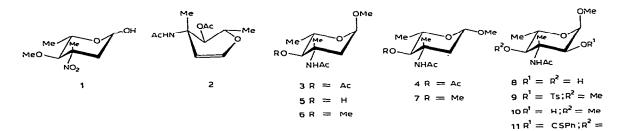
Some approaches to the synthesis of evernitrose and its enantiomer

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Evernitrose (1) is an unusual, branched-chain nitro sugar that occurs¹ as a component of the oligosaccharide antibiotics everninomicins B, C, and D. Although originally assigned² the *L-ribo* configuration, evernitrose is now known³ to be 2,3,6-trideoxy-3-C,4-Odimethyl-3-nitro-L-arabino-hexopyranose (1). This revision of the structure of evernitrose revived our interest in the synthesis of branched-chain amino sugars from the glycal 2^4 , which possesses the same stereochemistry as evernitrose at the asymmetric centres. The glycal 2 is readily prepared⁴ from methyl α -L-rhamnopyranoside.

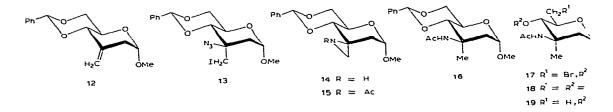
Addition of methanol to 2 in the presence of boron trifluoride gave, after chromatography on silica gel, methyl 3-acetamido-4-O-acetyl-2,3,6-trideoxy-3-C-methyl- α -L-arabino-hexopyranoside (3), m.p. 123–125°, $[\alpha]_D -133°$ (c 1.7, chloroform), and the corresponding β -glycoside 4, m.p. 167–169°, $[\alpha]_D + 8°$ (c 0.95, chloroform), in the ratio of ~3:1. Catalytic O-deacetylation of 3 gave the alcohol 5 (89%), $[\alpha]_D -104°$ (c 0.7, chloroform), which on methylation (methyl iodide and sodium hydride in tetrahydrofuran⁵) gave methyl 3-acetamido-2,3,6-trideoxy-3-C,4-O-dimethyl- α -L-arabino-hexopyranoside (6, 95%), m.p. 136–138°, $[\alpha]_D -71°$ (c 0.85, chloroform). Since the D enantiomer {m.p. 136–138°, $[\alpha]_D +73°$ (c 1, chloroform)} of 6 has recently been transformed into D-evernitrose by Yoshimura *et al.*^{6,7}, the foregoing route formally constitutes a synthesis of L-evernitrose (1), which has also been synthesized^{6,7} by way of the cyanohydrin route originally developed by Bourgeois⁸. A sequence of reactions analogous to that described above was used to transform the β -glycoside 4 into methyl 3-acetamido-2,3,6-trideoxy-3-C,4-Odimethyl- β -L-arabino-hexopyranoside (7), m.p. 151–152°, $[\alpha]_D +57°$ (c 1, chloroform); lit.⁷ (D enantiomer), m.p. 159–160°, $[\alpha]_D -57°$ (c 1, chloroform).

The evernitrose precursor 6 was also obtained by another, though less efficient, route that relied on deoxygenation of the readily available methyl 3-acetamido-3,6-



dideoxy-3-C-methyl- α -L-glucopyranoside⁴ (8) at C-2. Unimolar tosylation⁹ of 8 and methylation (methyl iodide—silver oxide) of the resulting tosylate gave methyl 3-acetamido-3,6-dideoxy-3-C,4-O-dimethyl-2-O-toluene-p-sulphonyl- α -L-glucopyranoside (9), m.p. 173–174°, $[\alpha]_D$ –33° (c 0.9, chloroform), whose structure was readily established by ¹H-n.m.r. spectroscopy. Detosylation of 9 with sodium amalgam gave the alcohol 10 (63%), m.p. 186–187°, $[\alpha]_D$ –118° (c 0.8, chloroform), which afforded a crude 2-thiobenzoate 11 (27%) when treated, in sequence, with N,N-dimethyl- α -chlorobenzylideneammonium chloride and hydrogen sulphide¹⁰. Radical-induced cleavage of the 2-thiobenzoate 11 with tributyltin hydride¹⁰ in refluxing toluene gave 6 (37%), which was identical (m.p., $[\alpha]_D$, and ¹H-n.m.r. spectrum) to the material obtained by way of the glycal 2, as well as a substantial amount (~50%) of an unidentified product. Although the deoxygenation procedure of Barton and McCombie¹⁰ gave disappointingly low yields at both stages, other new deoxygenation procedures¹¹ may possibly be more efficient in converting 10 into 6.

The synthesis of D-evernitrose (D-1) has been approached by a different route, which is based on addition of the pseudohalogen iodine azide¹² (generated *in situ* from sodium azide and iodine monochloride in acetonitrile) to the 3-C-methylene sugar derivative 12*. This reaction gave, *inter alia*, an adduct, m.p. 88–91°, $[\alpha]_D$ +45° (c 1, chloroform), that can reasonably be assigned the structure 13 by analogy** and its reduction (lithium aluminium hydride in ether at 0°) to the spiro-aziridine 14, which was characterized as the N-acetyl derivative 15, m.p. 114–115°, $[\alpha]_D$ +48° (c 1, chloroform).



The configuration at the branch-point in 15 was established by single-crystal X-ray analysis¹⁴. High-pressure hydrogenolysis of 14 in methanol over Raney nickel at ambient temperature, followed by *N*-acetylation of the resulting amine, gave methyl 3-acetamido-4,6-*O*-benzylidene-2,3-dideoxy-3-*C*-methyl- α -D-arabino-hexopyranoside (16, 40%), $[\alpha]_D + 28^{\circ}$ (*c* 1, chloroform). Cleavage of the 4,6-*O*-benzylidene ring of 16 with *N*-bromosuccinimide in refluxing carbon tetrachloride¹⁵ furnished the 6-bromo 4-benzoate 17 (86%), $[\alpha]_D + 78^{\circ}$ (*c* 1, chloroform), which was transformed, with concomitant debenzoylation, into methyl 3-acetamido-2,3,6-trideoxy-3-*C*-methyl- α -D-arabino-hexopyranoside (18), $[\alpha]_D + 96 \pm 3^{\circ}$ (*c* 1, chloroform), on high-pressure hydrogenation in methanol over Raney nickel. The ¹H-n.m.r. spectrum of 18 was indistinguishable from that of its L enantiomer 5. Methylation (as before) of 18 gave methyl 3-acetamido-2,3,6-trideoxy-3-*C*,4-*O*-dimethyl- α -D-arabino-

^{*}Prepared essentially as described¹³ for the L enantiomer.

^{**}Methylenecyclohexane, for example, adds iodine azide regioselectively, to yield 1-azido-1-iodomethylcyclohexane¹².

hexopyranoside (19), m.p. 136–138°, $[\alpha]_D$ +72° (c 1, chloroform), which, as noted earlier, has been converted into D-evernitrose (D-1) by Yoshimura *et al.*⁷.

New compounds had spectroscopic properties in agreement with the structures assigned and, when crystalline, gave satisfactory elemental analyses.

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REFERENCES

- A. K. Ganguly and A. K. Saksena, J. Antibiot., 28 (1975) 707-709; A. K. Ganguly and S. Szmulewicz ibid., 28 (1978) 710-712; A. K. Ganguly, O. Z. Sarre, D. Greeves, and J. Morton, J. Am. Chem. Soc., 97 (1975) 1982-1985.
- 2 A. K. Ganguly, O. Z. Sarre, and H. Reimann, J. Am. Chem. Soc., 90 (1968) 7129-7130.
- 3 A. K. Ganguly, O. Z. Sarre, A. T. McPhail, and K. D. Onan, Chem. Commun., (1977) 313-314.
- 4 J. S. Brimacombe and L. W. Doner, J. Chem. Soc., Perkin Trans. 1, (1974) 62-65.
- 5 J. S. Brimacombe, B. D. Jones, M. Stacey, and J. J. Willard, Carbohydr. Res., 2 (1966) 167-169.
- 6 J. Yoshimura, M. Matsuzawa, K. Sato, and M. Funabashi, Chem. Lett., (1977) 1403-1406.
- 7 J. Yoshimura, M. Matsuzawa, and M. Funabashi, Bull. Chem. Soc. Jpn., 51 (1978) 2064-2067.
- 8 J.-M. Bourgeois, Helv. Chim. Acta, 59 (1976) 2114-2124.
- 9 P. J. Garegg, T. Iversen, and S. Oscarson, Carbohydr. Res., 53 (1977) C5-C7.
- 10 D. H. R. Barton and S. W. McCombie, J. Chem. Soc., Perkin Trans. 1, (1975) 1574-1585.
- 11 See P. M. Collins and V. R. Z. Munasinghe, Chem. Commun., (1977) 927-928, and references cited therein.
- 12 A. Hassner and L. A. Levy, J. Am. Chem. Soc., 87 (1965) 4203-4204; F. W. Fowler, A. Hassner, and L. A. Levy, *ibid.*, 89 (1967) 2077-2082; A. Hassner, Acc. Chem. Res., 4 (1971) 9-16.
- 13 E. H. Williams, W. A. Szarek, and J. K. N. Jones, Can. J. Chem., 47 (1969) 4467-4471.
- 14 J. S. Brimacombe and T. J. R. Weakley, unpublished results.
- 15 S. Hanessian, Carbohydr. Res., 2 (1966) 86-88.