TRANSFORMATION OF ENT-KAUR-15-ENES BY GIBBERELLA FUJIKUROI

BRAULIO M. FRAGA, MELCHOR G. HERNANDEZ, MANUEL D. RODRIGUEZ, CARMEN E. DIAZ,* PEDRO GONZALEZ* and JAMES R. HANSON[†]

Instituto de Productos Naturales Organicos del CSIC, La Laguna, Tenerife, Canary Islands, Spain; *Departmento de Quimica Organica, Universidad de La Laguna, Tenerife, Canary Islands, Spain; †School of Molecular Sciences, University of Sussex, Brighton, Sussex, BN1 9QJ, U.K.

(Received 5 November 1986)

Key Word Index-Gibberella fujikuroi; microbiological transformation; ent-kaur-15-enes; gibberell-15-enes; diterpenoids.

Abstract—The microbiological transformation of *ent*-kaur-15-ene and of some derivatives by *Gibberella fujikuroi* into the gibberell-15-ene analogues of GA₃, GA₇, GA₁₃ and GA₁₆ and into 7,18-dihydroxykaur-15-enolide and the Δ^{15} -isomer of fujenal is described.

INTRODUCTION

ent-Kaur-16-ene (1) is a key precursor of the gibberellin plant growth hormones [1]. The dwarf d₅ mutant of maize is blocked for gibberellin biosynthesis [2]. Instead of the normal gibberellin precursor ent-kaur-16-ene, it produces ent-kaur-15-ene (5). It has been suggested [3] that the gibberellin biosynthetic system in the dwarf maize mutant is unable to handle the double bond isomer, entkaur-15-ene. In this connection it is worth noting that whereas hydroxylated ent-kaur-15-enes are found quite commonly (e.g. in Sideritis species [4]), the corresponding gibberell-15-enes have not been isolated. Useful information concerning the structural requirements for specific biosynthetic steps in the gibberellin pathway has accrued from feeding false substrates to Gibberella fujikuroi [5]. It was therefore of interest to incubate ent-kaur-15-ene and some relatives with the fungus to see if the inability of the gibberellin pathway to handle ent-kaur-15-ene was a more general phenomenon. Conflicting results have been obtained in the past. It has been reported [6] that ent-kaur-15-ene was not metabolized to gibberellin analogues by strain GF1a but subsequently a series of gibberellin analogues were obtained [7] by the transformation of ent-17-hydroxykaur-15-ene.

RESULTS

Preparation of substrates

ent-Kaur-15-ene (5) was prepared from the more readily available ent-kaur-16-en-19-oic acid (2) [8]. Reduction of the methyl ester with LAH afforded ent-19hydroxykaur-16-ene (3) which on oxidation with PDC gave the 19-aldehyde (4). Wolff-Kischner reduction (Huang-Minlon conditions) brought about some double bond isomerization but the yield of ent-kaur-15-ene (5) was increased by further treatment with toluene-psulphonic acid. The hydrocarbons were then separated by chromatography on silver nitrate-silica gel.

ent- 7α -Hydroxykaur-15-ene (9) was obtained from siderol (ent- 7α -acetoxy-18-hydroxykaur-15-ene, 6), by chlo-



rination with triphenylphosphine–CCl₄ to afford 7 and subsequent reduction with tri-*n*-butyltin hydride. Hydrolysis of the resultant *ent*-7 α -acetoxykaur-15-ene (8) afforded *ent*-7 α -hydroxykaur-15-ene (9). *ent*-18-Hydroxykaur-15-ene (10) [9] was obtained by hydrolysis of isocandol B acetate.

Feeding experiments

The fermentations were carried out in the presence of AMO-1618 to inhibit the formation of *ent*-kaur-16ene [10] and consequently its endogenous metabolites. The transformations were carried out for a period of 6 days. The metabolites were then isolated and the acids were purified as their methyl esters. *ent*-Kaur-15-ene (5) gave a 15-isofujenal derivative (isolated as its dimethyl ester, 13), 15-isogibberellin A₃ (14), 15-isogibberellin A₁₆ (16) and 7 β ,18-dihydroxykaur-15-enolide (18). *ent*-7 α -Hydroxykaur-15-ene (9) gave 15-isogibberellin A₇ (15) and 15-isogibberellin A₁₃ (16) and *ent*-7 α -hydroxy-15 β ,16 β -epoxykaurane (19). *ent*-18-Hydroxykaur-15-ene (10) gave *ent*-7 α ,18-dihydroxykaur-15-en-19-oic acid (20) and 7 β ,18-dihydroxykaur-15-enolide (18).

The metabolites were identified by comparison of their ¹H NMR spectra with the spectra of their Δ^{16} -isomers. Thus 15-isogibberellin A₃ methyl ester (14) showed the typical H-5: H-6 doublets (δ 3.15 and 2.69, J = 10 Hz) and signals characteristic of ring A (δ 6.26, d, J = 9 Hz, H-1; 5.90, dd, J = 4 and 9 Hz, H-2; 4.16, d, J = 4 Hz, H-3). The H-15 resonance appeared as a broad singlet at 5.36 and the H-17 resonance as a doublet (1.68, J = 1.5 Hz). The corresponding carboxylic acid has been described previously [12]. 15-Isogibberellin A₇ methyl ester (15) showed a similar pattern of signals. 15-Isogibberellin A₁₆ methyl ester (16) contained the signal assigned to H-1 as a quartet (J = 6 and 10 Hz) at $\delta 4.00$ (cf. A₁₆ and A₅₇, 4.00, J = 6 and 10 Hz [13, 14]). The multiplicity of this signal and the absence of a downfield shift in the H-5 signal compared to 15-isoGA₇ (15) are in accord with the location of a hydroxyl group at C-1 α . Furthermore, the ¹³C NMR spectrum (see Table 1) closely paralleled that of gibberellin A₁₆. 15-Isogibberellin A₁₃ trimethyl ester (17)

Table 1. ¹³C NMR data for isogibberellin A₁₆ and gibberellin A₁₆ methyl esters* (in CDCl₃ at 50.32 MHz)

Carbon	IsoGA ₁₆	GA16	Carbon	IsoGA16	GA16
1	71.5	71.8	11	19.4	18.7
2	38.9	39.7	12	20.8	32.3
3	70.4	70.4	13	41.0	39.1
4	54.1	54.8	14	41.1	37.0
5	50.2	51.1	15	131.8	45.2
6	52.1	53.0	16	150.2	158.2
7	173.4	174.9	17	15.4	107.4
8	56.3	52.1	18	14.0	14.9
9	48.0	54.0	19	178.4	179.3
10	95.4	96.2			

*From ref. [11].











20	R	=	Н
21	R	=	Ac







17



19

revealed the characteristic H-3 (δ 3.70), H-5 and H-6 (2.46 and 3.53, J = 11 Hz) resonances, three methoxyl signals at 3.57, 3.60 and 3.65 together with the H-15 and H-17 signals at 5.42 and 1.60. 15-Isofujenal dimethyl ester (13), methyl ent-7 α ,18-dihydroxykaur-15-en-19-oate (20) [15] and 7 β ,18-dihydroxykaur-15-enolide (18) were identical to authentic samples. The latter were characterized as their diacetates. The epoxide, ent-7 α -hydroxy-15 β ,16 β epoxykaurane (19) was assigned the ring D stereochemistry as shown since it was identical to material obtained by epoxidation of ent-7 α -hydroxykaur-15-ene (9) with mchloroperbenzoic acid. Furthermore the H-15 and H-17 proton resonances are very close in position to those of the corresponding epoxide from siderol [16].

DISCUSSION

These results, taken in conjunction with earlier experiments [7, 15] show that the gibberellin biosynthetic pathway in Gibberella fujikuroi is capable of accepting the 15-enes and metabolizing them, albeit less efficiently to the 15-ene isomers of the natural fungal metabolites. However, the relative proportions of 15-isogibberellin A₃ (14), A_7 (15) and A_{16} (16) differ from the normal series. In particular, gibberellin A16 is usually a minor metabolite. It is possible that 13-hydroxylation may be less efficient in the 15-ene series. The difference between the 7- and 18hydroxykaur-15-enes in which only the latter forms a kaurenolide, parallels the difference which has been observed [15] in the 16-enes and is a reflection of the intervention of a 6-ene in kaurenolide formation [17]. Finally, the formation of a 15,16-epoxide may be a modification of the hydration at C-16 which is a probable dumping mechanism on this pathway. Alternatively, the 15-ene may be a substrate for the enzyme system which epoxidizes the 6-ene.

EXPERIMENTAL

Preparation of ent-kaur-15-ene (5). The methyl ester of entkaur-16-en-19-oic acid (2) (1.4 g) in dry THF (10 ml) was added dropwise to a suspension of LAH (450 mg) in THF (10 ml). After 2 hr reflux, EtOAc was added and the mixture was washed with dil. HCl and the product recovered in EtOAc. It was chromatographed on silica gel in petrol-EtOAc (95:5) to give ent-kaur-16en-19-ol (3) (1.05 g); ¹H NMR (200 MHz, CDCl₃): 80.95 and 0.99 (each 3H, s), 3.42 and 3.73 (each 1H, d, J = 11 Hz; H-19), 4.72 and 4.78 (each 1H, br s, H-17). EIMS (rel. int.): 288 [M]* (22), 273 (27), 257 (100), 255 (11), 245 (14), 229 (15), 187 (15). The alcohol (3, 600 mg) in CH₂Cl₂ (16 ml) was treated with pyridinium dichromate (760 mg) at room temp. for 3 hr. It was then diluted with Et2O and percolated through silica gel to afford ent-kaur-16-en-19-al (4, 530 mg); ¹H NMR: δ0.89 and 1.00 (3H, s) 4.80 (2H, br s, H-17), 9.73 (1H, s, H-19), EIMS (rel, int.): 286 (31), 271 (15), 257 (40), 243 (78), 225 (28), 199 (29), 187 (40). Hydrazine hydrate (3 ml) was added to a soln of the aldehyde (4, 510 mg) in diethyleneglycol (10 ml) and the mixture was then heated under reflux for 2 hr. KOH pellets (500 mg) were added and the mixture was then heated under reflux for a further 45 min. The condenser was removed and the temp. was allowed to reach 200° over 2 hr. The soln was cooled and poured into H₂O. The products were recovered in EtOAc to afford a three-component mixture of entkaurane (11), ent-kaur-16-ene (1) and ent-kaur-15-ene (5) as revealed by TLC. The mixture was heated with toluene-psulphonic acid (10 mg) in C_6H_6 (5 ml) under reflux for 2 hr and then chromatographed on 10% AgNO3-silica gel in petrol to give ent-kaurane (11) (35 mg); ¹H NMR (90 MHz CDCl₃); $\delta 0.80-1.02$ (12H) EIMS (rel. int.): 274 [M]⁺ (52), 259 (77), 245 (11), 231 (21), 189 (21), 177 (12). Further elution gave ent-kaur-16ene (1, 40 mg); ¹H NMR (200 MHz); $\delta 0.81$, 0.85 and 1.02 (each 3H, s), 4.73 and 4.78 (each 1H, br s); EIMS (rel. int.): 272 [M]⁺ (27), 257 (68), 229 (41), 213 (22), 201 (11), 187 (20). Further elution afforded ent-kaur-15-ene (5, 150 mg); ¹H NMR (80 MHz, CDCl₃); $\delta 0.81$, 0.85 and 1.03 (each 3H, s), 1.69 (3H, d, J = 1.5 Hz, H-17), 5.06 (1H, br s, H-15); EIMS (rel. int.): 272 [M]⁺ (35), 257 (32), 244 (10), 229 (16), 213 (7), 201 (7), 187 (17). Further elution afforded ent-kauranol (12, 80 mg); ¹H NMR (200 MHz, CDCl₃); $\delta 0.79$, 0.83 and 1.01 (each 3H, s), 1.35 (3H, s, H-17). EIMS (rel. int.): 290 [M]⁺ (11), 275 (18), 272 (44), 257 (55), 232 (62), 229 (17), 217 (28), 201 (7), 187 (13).

ent-18-Chloro-7 α -acetoxykaur-15-ene (7). Siderol (6), obtained from Sideritis spinulosa, in dry pyridine (18 ml) and CCl₄ (20 ml) was heated with triphenylphosphine (750 mg) under reflux for 4 hr. The products were recovered with EtOAc and chromatographed on silica gel in petrol-EtOAc (90:10) to afford ent-18chloro-7 α -acetoxykaur-15-ene, mp 109-111°. (Found: [M]⁺ 364.215; C₂₂H₃₃O₂Cl requires 364.217.) ¹H NMR (90 MHz, CDCl₃): δ 0.84 and 1.08 (each 3H, s), 1.69, (3H, d, J = 2 Hz, H-17), 2.06 (3H, s, OAc), 3.05 and 3.41 (each 1H, d, J = 11 Hz, H-18) 4.69 (1H, br s H-7), 5.30 (1H, br s, H-15). EIMS (rel. int.): 364 [M]⁺ (2), 322 (11), 304 (38), 289 (10), 276 (4), 255 (11), 199 (3), 185 (5).

Preparation of ent-7a-acetoxykaur-15-ene (8). The above chloro compound (7, 400 mg) in toluene (10 ml) was treated with tri-*n*-butyltin hydride (0.3 ml) and azobisisobutyronitrile (5 mg) under reflux for 16 hr. The solvent was evapd and the residue dissolved in Et₂O and stirred with H₂O satd with KF for 15 min. The soln was filtered, dried and the solvent evapd to give a gum which was chromatographed on silica gel. Elution with petrol-EtOAc (90:10) gave ent-7a-acetoxykaur-15-ene (350 mg), mp 103-105°. (Found: $[M]^+$ 330.257; C₂₂H₃₄O₂ requires 330.256.) ¹H NMR (90 MHz): $\delta 0.79$ (6H, s), 1.05 (3H, s), 1.69 (3H, d, J = 2 Hz, H-17), 4.75 (1H, t, J = 6 Hz, H-7), 5.27 (1H, br s, H-15). EIMS (rel. int.): 330 (3), 288 (20), 270 (93), 255 (52), 242 (13), 227 (22).

Preparation of ent-7 α -hydroxykaur-15-ene (9). The above acetoxy derivative (8, 340 mg) was treated with 5% methanolic KOH (20 ml) at room temp. for 24 hr. The soln was neutralized and the product recovered in EtOAc to afford ent-7 α hydroxykaur-15-ene (9, 250 mg), mp 134–136°. (Found: [M]⁺ 288.246; C₂₀H₃₂O requires 288.245.) ¹H NMR (80 MHz, CDCl₃): δ 0.81, 0.85 and 1.04 (each 3H, s), 1.73 (3H, d, J = 1.5 Hz, H-17), 3.60 (1H, br s, H-7), 5.50 (1H, br s, H-15). EIMS (rel. int.): 288 [M]⁺ (39), 273 (9), 270 (16), 255 (16), 245 (12), 227 (7), 199 (9), 164 (55), 149 (20).

Incubation experiments. Gibberella fujikuroi (ACC 917) inhibited with 5×10^{-5} M AMO 1618, was grown in shake culture at 25° for 1-2 days in 80-100 conical flasks (250 ml) each containing sterile medium (50 ml) [15]. The substrate (see below) in EtOH (16-20 ml) was distributed equally between the flasks and the fermentation was allowed to continue for a further 6 days. The broth was filtered, adjusted to pH 2 with dil. HCl and extracted with EtOAc. The extract was separated into acidic and neutral fractions with NaHCO₃. The acidic fraction was methylated with CH₂N₂. The fractions were then chromatographed on silica gel in petrol-EtOAc.

ent-Kaur-15-ene (5, 150 mg) gave the methyl ester of isofujenal (13, 5 mg), isogibberellin A₃ (14, 10 mg) and isogibberellin A₁₆ (16, 11 mg) from the acid fraction. 7β , 18-Dihydroxykaur-15-enolide (18, 12 mg) was obtained from the neutral fraction.

ent-7 α -Hydroxykaur-15-ene (9, 240 mg) gave the methyl esters of isogibberellin A₇ (15, 20 mg) and isogibberellin A₁₃ (17, 6 mg) and from the neutral fraction, ent-7 α -hydroxy-15 β ,16 β epoxykaurane (19, 15 mg). ent-18-Hydroxykaur-15-ene (10, 260 mg) gave the methyl ester of ent-7 α ,18-dihydroxykaur-15-en-19-oic acid (20, 33 mg) and from the neutral fraction, 7β ,18-dihydroxykaur-15-enolide (18, 19 mg).

Isogibberellin A_3 methyl ester (14). Mp 190–193°. (Found: [M]⁺ 360.159. $C_{20}H_{24}O_6$ requires 360.157.) ¹H NMR (200 MHz, CDCl₃): δ 1.25 (3H, s, H-18), 1.68 (3H, d, J = 1.5 Hz, H-17), 2.69 and 3.15 (each 1H, d, J = 10 Hz, H-5 and H-6), 3.71 (3H, s, OMe), 4.16 (1H, d, J = 4 Hz, H-3), 5.44 (1H, d, J = 1.5 Hz, H-15), 5.90 (1H, dd, J = 4 and 9 Hz, H-2), 6.26 (1H, d, J = 9 Hz, H-1), EIMS (rel. int.): 360 [M]⁺ (8), 342 (2), 328 (14), 301 (13), 297 (8), 237 (22), 227 (13), 209 (27).

Isogibberellin A_7 methyl ester (15). Mp 160–162°. (Found: [M]⁺ 344.161; $C_{20}H_{24}O_5$ requires 344.162.) ¹H NMR (200 MHz, CDCl₃): δ 1.25 (3H, s, H-18), 1.68 (3H, d, J = 1.5 Hz, H-17), 2.67 and 3.13 (each 1H, d, J = 10 Hz, H-5 and H-6), 3.70 (3H, s, OMe), 4.14 (1H, d, J = 3 Hz, H-3) 5.36 (1H, br s, H-15), 5.90 (1H, dd, J = 3 and 9 Hz, H-2), 6.27 (1H, d, J = 9 Hz, H-1). EIMS (rel. int.): 344 [M]⁺ (55), 326 (1), 313 (2), 312 (6), 300 (8), 298 (4), 284 (17), 282 (53), 267 (24), 253 (16), 239 (22), 222 (45), 214 (50).

Isogibberellin A_{13} methyl ester (17). Gum. (Found: [M]⁺ 420.215; $C_{23}H_{32}O_7$ requires 420.215.) ¹H NMR (200 MHz, CDCl₃): δ 1.09 (3H, s, H-18), 3.60 (3H, d, J = 1.5 Hz, H-17), 2.46 and 3.53 (each 1H, d, J = 11 Hz), 3.57, 3.60 and 3.65 (each 3H, s, OMe), 3.70 (1H, m, H-3), 5.42 (1H, br s, H-15). EIMS (rel. int.): 420 [M]⁺ (4) 388 (1), 328 (8), 300 (5), 285 (7).

Isogibberellin A_{16} methyl ester (16). Gum. (Found: 362.175; $C_{20}H_{26}O_6$ requires 362.173.) ¹H NMR (200 MHz, CDCl₃): δ 1.14 (3H, s, H-18), 1.68 (3H, d, J = 1.5 Hz, H-17), 2.62 and 3.15 (each 1H, d, J = 10 Hz, H-5 and H-6), 3.68 (3H, s, OMe), 3.83 (1H, br s, H-3), 4.00 (1H, dd, J = 6 and 10 Hz, H-1), 5.34 (1H, br s, H-15). EIMS (rel. int.): 362 [M]⁺ (24), 344 (13), 300 (38), 288 (32), 256 (11), 241 (21).

Isofujenal dimethyl ester (13). Gum. (Found: $[M-H]^+$ 375.214; C₂₂H₃₃O₅ requires 375.217.) ¹HNMR (200 MHz, CDCl₃): δ 1.04 and 1.29(each 3H, s), 1.71 (3H, d, J = 1.5 Hz, H-17), 2.63 (1H, s, H-5), (3.61 and 3.69 (each 3H, s, OMe), 5.23 (1H, br s, H-15), 9.80 (1H, s, H-7). EIMS (rel. int.): 376 [M]⁺ (7), 375 (20), 344 (13), 316 (10), 312 (6), 284 (11), 227 (15).

ent-7a, 18-Dihydroxykaur-15-en-19-oic acid methyl ester (20). ¹H NMR (90 MHz, pyridine- d_5): δ 1.06 (3H, s, H-20), 1.74 (3H, d, J = 2 Hz, H-17), 3.70 (3H, s, OMe), 3.78 and 4.41 (each 1H, d, J = 11 Hz, H-18), 3.90 (1H, br s, H-7), 6.00 (br s, H-15). The diacetate, prepared with Ac₂O in pyridine was a gum. (Found: [M] * 432.251; C₂₅H₃₆O₆ requires 432.251.) ¹H NMR (90 MHz, CDCl₃): δ 0.91 (3H, s, H-20), 1.73 (3H, br s, H-17) 2.05 and 2.12 (each 3H, s, OAc), 3.70 (3H, s, OMe), 3.88 and 4.30 (each 1H, d, J = 11 Hz, H-18), 4.84 (1H, t, H-7), 5.22 (1H, br s, H-15), EIMS (rel. int.): 432 [M] * (1), 390 (11), 372 (6), 357 (4), 312 (28), 253 (11).

ent-7 α -Hydroxy-15 β ,16 β -epoxykaurane (19). Mp 193–195°. (Found: [M]⁺ 304.242; C₂₀H₃₂O₂ requires 304.240.) ¹H NMR (200 MHz, CDCl₃): δ 0.78, 0.85 and 0.97 (each 3H, s), 1.42 (3H, s, H-17), 3.06 (1H, br s, H-15), 3.78 (1H, t, J = 2 Hz, H-7). EIMS (rel. int.): 304 [M]⁺ (3), 289 (3), 286 (7), 271 (11), 260 (8), 243 (11), 201 (12), 189 (7), 173 (8), 149 (60). The epoxide was identical to material prepared by the epoxidation of isocandol A (45 mg) with *m*-chloroperbenzoic acid (20 mg) in CHCl₃ (3 ml) at room temp. for 10 hr.

ent-6 β ,7 α ,18-*Trihydroxykaur*-15-*en*-19-*oic acid* 19,6 β -*lactone* (18). ¹H NMR (200 MHz): δ 0.91 (3H, s, H-20), 1.73 (3H, s, H-17), 1.90 (1H, d, J = 7 Hz, H-5), 3.71 (2H, br s, H-18), 4.12 (1H, d, J = 7 Hz, H-7), 4.79 (1H, t, J = 7 Hz, H-6), 5.50 (1H, br s, H-15). EIMS (rel. int.): 332 [M]⁺ (4), 302 (4), 271 (9), 241 (8), 213 (6). The diacetate, prepared with Ac₂O-pyridine was a gum. (Found: [M]⁺ 416.225; C₂₄H₃₂O₆ requires 416.220.) ¹H NMR (90 MHz, CDCl₃): δ 1.02 (3H, s, H-20), 1.71 (3H, d, J = 2 Hz, H-17), 2.07 and 2.10 (each 3H, s, OAc), 4.15 (2H, br s, H-18), 4.27 (1H, t, J = 7 Hz, H-6), 5.50 (1H, br s, H-15), 5.60 (1H, d, J = 7 Hz, H-7). EIMS (rel. int.): 416 [M]⁺ (3), 374 (12), 356 (4), 341 (9), 314 (25), 299 (10).

Acknowledgements—We thank CAICYT (Madrid, Spain) for financial support and Professor Panizo (CSIC, Madrid) for samples of *ent*-kaur-16-en-19-oic acid. C.E.D. thanks the Ministry of Education and Science (Spain) for a research fellowship.

REFERENCES

- Cross, B. E., Galt, R. H. B. and Hanson, J. R. (1964) J. Chem. Soc. 295.
- 2. Hedden, P. and Phinney, B. O. (1979) Phytochemistry 18, 1475.
- Coolbaugh, R. C. (1983) in Biochemistry and Physiology of Gibberellins (Crozier, A., ed.) Vol. 1, p. 75. Praeger, New York.
- Gonzalez, A. G., Fraga, B. M., Hernandez, M. G., Luis, J. G. and Larruga, F. (1979) Biochem. Syst. Ecol. 7, 115.
- 5. for review see: Bearder, J. R. (1983) in Biochemistry and Physiology of Gibberellins, Vol. 1. p. 251.
- Hedden, P., Phinney, B. O., MacMillan, J. and Sponsel, V. M. (1977) Phytochemistry 16, 1913.
- Wada, K., Imai, T. and Shibata, K. (1979) Agric. Biol. Chem. (Japan) 43, 1157.
- Martin-Panizo, F. and Rodriguez, B. (1979) Anales Quim. 75, 431.
- 9. Piozzi, F., Venturella, P., Bellino, A. and Marino, M. L. (1973) J. Chem. Soc. Perkin Trans. 1, 1164.
- 10. Cross, B. E. and Myers, P. L. (1969) Phytochemistry 8, 79.
- 11. Yamaguchi, I., Takahashi, N. and Fujita, K. (1975) J. Chem. Soc. Perkin Trans. 1, 992.
- 12. Lischewski, M., Adam, G. and Serebryakov, E. I. (1980) Tetrahedron Letters 21, 45.
- 13. Galt, R. H. B. (1968) Tetrahedron 24, 1337.
- 14. Bearder, J. R. and MacMillan, J. (1973) J. Chem. Soc. 2824.
- Fraga, B. M., Hanson, J. R., Hernandez, M. G. and Sarah, F. Y. (1980) *Phytochemistry* 19, 1087.
- Piozzi, F., Venturella, P., Bellino, A. and Mondelli, R. (1968) Tetrahedron 24, 4073.
- 17. Graebe, J. and Hedden, P. (1981) Phytochemistry 20, 1011.