## LETTERS TO THE EDITOR

## Asymmetric Synthesis of a Phosphorus Analog of Natural γ-amino-β-hydroxybutyric Acid

V. V. Nesterov and O. I. Kolodyazhnyi

Institute of Bioorganic Chemistry and Petrochemistry, National Academy of Sciences of Ukraine, ul. Murmanskaya 1, Kiev, 02094 Ukraine

Received August 1, 2006

## DOI: 10.1134/S1070363206100318

We have developed a simple synthesis of a phoshorus analog of natural  $\gamma$ -amino- $\beta$ -hydroxybutyric acid (GABOB) which is important in the biological aspect and used for treatment of some deseases of nervous system [1, 2]. In the first step, the pure (*S*)-dimenthyl 3-chloro-2hydroxypropylphosphonate (**I**) we prepared earlier [3, 4] was dehydrochlorinated with potassium carbonate in acetonitrile in the presence of potassium iodide to form epoxide (*R*)-**II** with a high stereospecifity and without loss of optical purity.



Compound II was reacted with dibenzylamine to obtain crystalline hydroxyaminophosphonate (*R*)-III. The latter was further dealkylated by heating with hydrochloric acid in dioxane to afford crystalline phosphonic acid (*R*)-IV which was debenzylated by hydrogenation in methanol with palladium on charcoal. As a result, a crystalline phosphorus analog of natural  $\gamma$ -amino- $\beta$ -hydroxybutyric acid (*R*)-V was prepared with good yield and high optical purity. The structure of compound (*R*)-V was confirmed by elemental analysis and NMR spectroscopy. The enantiomeric purity of diastereomeric compounds I– III was monitored by NMR spectroscopy. The *R* configuration of compound V was established by measuring its optical rotation [5, 6]. Due to the presence of menthyl groups, compounds **I–III** could be isolated and purified by crystallization, i.e. we could exclude chromatographic purification from the synthetic procedure and prepare an optically pure phosphorus analog (R)-V.

**Bis**[(1*R*,2*S*,5*R*)-menthyl] (2*R*)-(oxiran-2-yl)methylphosphonate (II). To a solution of 1.156 g of compound I [4] in 19 ml of acetonitrile–DMF (10:3.5) we added 0.707 g of potassium carbonate and a catalytic amount of potassium iodide. The mixture was refluxed for 8 h, the precipitate was then filtered off, and the solvent removed to obtain compound II as a yellowish oil, yield 98%,  $[\alpha]_D^{20}$  –62.4° (*c* 7.0, CHCl<sub>3</sub>). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm (*J*, Hz): 0.8 d (3H, *J* 7); 0.88 d (3H, *J* 7); 0.89 d (3H, *J* 7); 0.90 d (3H, *J* 7); 0.91 d (3H, *J* 7); 0.8–1.65 m (16H, CH<sub>2</sub>C); 2.15–2.05 m (2H, PCH<sub>2</sub>); 2.4 d.d (1H, *J* 2.1, *J* 5.1); 3.0 d.d (1H, *J* 4, *J* 4.8); 3.0 m (1H, CHOH); 4.24 m (2H, CHOH). <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>):  $\delta_{\rm P}$  25.3 ppm.

Bis[(1R,2S,5R)-menthyl] (2R)-[3-(dibenzylamino)-2-hydroxypropyl]phosphonate (III). To a solution of 1.1 g of compound II in 20 ml of methanol we added 0.52 g of dibenzylamine. The mixture was refluxed for 20 h. The solvent was removed, and the residue was recrystallized from acetonitrile. The product is a colorless crystalline compound, yield 70%, mp 73°C,  $[\alpha]_D^{20}$  –45.92 (*c* 4.57, CHCl<sub>3</sub>). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>,  $\delta$ , ppm (J, Hz):  $\delta$ , ppm (*J*, Hz): 0.73 d (3H,  $J_{\rm HH}$  7.2, CH<sub>3</sub>); 0.80 d (3H,  $J_{\rm HH}$  6.9, CH<sub>3</sub>); 0.87 d (3H,  $J_{\rm HH}$  7.0, CH<sub>3</sub>); 0.90 d (3H,  $J_{\rm HH}$ 7.0, CH<sub>3</sub>); 0.91 d (3H, J<sub>HH</sub> 7.0, CH<sub>3</sub>); 0.92 d (3H, J<sub>HH</sub> 7.0, CH<sub>3</sub>); 2.0 d.d.d ( $J_{\rm HP}$  12.5,  $J_{\rm HH}$  1.2, PCH<sup>a</sup>); 2.13 d.d.d (J 14.1, J 7.8, J 2.7, PCH<sup>b</sup>), 1.2-2.2 m (16H, CH<sub>2</sub>C); 4.1 m (2H, CHO), 3.55 c (4H, CH<sub>2</sub>Ph). 1.829 d.d.d (J 33.9, J 18.6, J 3.6, CH<sub>2</sub>N), 7.1-7.3 m (10H, C<sub>6</sub>H<sub>5</sub>). <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>),  $\delta_{P}$ , ppm: 30.0. Found, %: C 72.64; H 9.51; P 5.08. C<sub>37</sub>H<sub>58</sub>. NO<sub>4</sub>P. Calculated, %: C 72.63; H 9.55; P 5.06.

(2*R*)-[3-(Dibenzylamino)-2-hydroxypropyl]phosphonic acid hydrochloride (IV). To 0.75 g of compound III in 30 ml of dioxane we added 10 ml of conc. HCl, and the mixture was kept for 3 days at 80–85°C. The mixture was evaporated, and the residue was dissolved in 15 ml of water and washed with benzene. After removal of water in a vacuum, the product was obtained as a colorless hygroscopic crystalline compound, yield 85%. <sup>1</sup>H NMR spectrum (CDOD),  $\delta$ , ppm (*J*, Hz): 1.73 d.d.d (*J*<sub>HP</sub> 32.4, *J*<sub>HH</sub> 17.4, *J*<sub>HH</sub> 8.1, CH<sup>a</sup>P), 1.93 d.d.d (*J*<sub>HP</sub> 34.5, *J*<sub>HH</sub> 14.7, *J*<sub>HH</sub> 4.5 CH<sup>b</sup>P), 3.00 d.d (*J*<sub>HH</sub> 13.2, *J*<sub>HH</sub> 9.3, CH<sup>b</sup>N), 4.08–4.21 m [1H, CH(OH)], 4.95 s (3H, OH); 7.52–7.54 m (10H, C<sub>6</sub>H<sub>5</sub>). <sup>31</sup>P NMR spectrum (H<sub>2</sub>O):  $\delta_P$  26.5 ppm.

(2*R*)-(3-amino-2-hydroxypropyl)phosphonic acid (V). To a solution of 0.39 g of compound IV in 45 ml of ethanol we added 0.37 g of palladium on charcoal (10%). Gaseous hydrogen was passed through a stirred mixture for 20 h. Then the mixture was filtered, the filtrate was evaporated, and the product was washed with acetone and dried in a vacuum. A colorless hygroscopic compound was obtained,  $[\alpha]_D^{20}$  +8.3° (*c* 1.2, H<sub>2</sub>O), <sup>31</sup>P NMR spectrum:  $\delta_P$  22.8 ppm. The product was additionally purified on KU-2 cationite, yield 90%,  $[\alpha]_D^{20}$  +10.2° (*c*<sub>1</sub>, H<sub>2</sub>O). <sup>1</sup>H NMR spectrum (D<sub>2</sub>O),  $\delta$ , ppm (*J*, Hz): 1.78 d.d.d (*J*<sub>HH</sub> 31.2, *J*<sub>HH</sub> 15.0, *J*<sub>HH</sub> 6.3, CH<sup>a</sup>P), 1.86 d.d.d (*J*<sub>HH</sub> 32.1, *J*<sub>HH</sub> 14.7, *J*<sub>HH</sub> 6.3, CH<sup>b</sup>P), 2.99 d.d (*J*<sub>HH</sub> 13.5, *J*<sub>HH</sub> 8.5, CH<sup>a</sup>N), 3.27 d.d (*J*<sub>HH</sub> 13.1, *J*<sub>HH</sub> 3.4, CH<sup>b</sup>N), 4.14 m [1H, CH(OH)]. <sup>31</sup>P NMR spectrum (D<sub>2</sub>O):  $\delta_P$ 18.1 ppm. Found N, %: 9.43. Calculated N, %: 9.03.

The NMR spectra were measured on a Varian-300 instrument against internal TMS (<sup>1</sup>H) and external 85%  $H_3PO_4$  in  $D_2O$  (<sup>31</sup>P). The optical rotations were measured on a Polax-2L polarimeter (Japan).

## REFERENCES

- 1. Aminophosphonic and aminophosphinic acids. Chemistry and Biological Activity, Kukhar, V.P. and Hudson, H.R., Eds., Chichester: Wiley, 2000, p. 518.
- 2. Otzuka, M., Obata, K., Miyata, Y. and Yaneka Y., *J. Neurochem.*, 1971, vol. 18, no. 2, p. 287.
- Kolodiazhnyi, O.I., *Tetrahedron: Asymmetry*, 2005, vol. 16, no. 20, p. 3295.
- 4. Nesterov, V.V. and Kolodiazhnyi O.I., *Tetrahedron: Asymmetry*, 2006, vol. 17, no. 17, p. 1023.
- 5. Wroblewski, A.E. and Halajewska-Wosik, A., *Tetrahedron: Asymmetry*, 2003, vol. 14, no. 21, p. 3359.
- Ordbñez, M., González-Morales, A., Ruiz, C., De la Cruz-Cordero, R. and Fernández-Zertuche M., *Tetrahedron: Asymmetry*, 2003, vol. 14, no. 13, p. 1775.