## A Novel Antihypertensive Agent: 1,2,3,5-Tetrahydroimidazo [2,1-b] quinazoline

In the course of a broad investigation of 'amidines' as potential antihypertensive agents, the title compound (II) was synthesized by the lithium aluminum hydride reduction of lactam I². The hydrochloride of II has mp 251–253°C; analysis for  $C_{10}H_{11}N_3\cdot HCl\colon$  Calcd: C, 57.28; H, 5.77; found: C, 56.89; H, 6.00;  $^{\rm EtOH\ (KOH)}_{\lambda_{max}}$  278 nm (log  $\varepsilon=4.09$ ).

This compound is a potent antihypertensive agent which is orally effective in rats, dogs, cats, and rabbits. It lowered the blood pressure significantly in the metacorticoid hypertensive rat³ (5 mg/kg p.o.), in the unanesthetized neurogenic hypertensive dog⁴ (2.5 mg/kg p.o.), and in the normotensive dog (2.5 mg/kg p.o.) without precipitating reflex tachycardia at these doses. The duration of action was sustained up to 48 h. No tolerance developed in a chronic study of 3 weeks duration. Preliminary studies of its mode of hypotensive action seem to exclude a central mechanism. Its peripheral action appears

complex, although  $\alpha$ -adrenergic blockade has been established as one of the components. Catecholamine depletion was demonstrated by a moderate reduction of norepine-phrine level in the rat's heart. Mechanisms involving direct vascular dilation or ganglion blockade were excluded. The possible occurrence of orthostatic hypotension associated with the  $\alpha$ -blockade activity of this agent was not observed in the rabbit tilt test<sup>5</sup>; this is in contrast to the result of a parallel study with phenoxybenzamine. The LD<sub>50</sub> (p.o.) of II (HCl) for rats was found to be 900 mg/kg.

Full reports of the chemistry and pharmacology of this new agent and analogs will be presented in the near future.

Zusammenfassung. 1,2,3,5-Tetrahydroimidazo(2,1-b)-chinazolin wurde als neuer Typ eines bei Tieren gut wirksamen Hypotensivums erkannt. Bei  $\alpha$ -blockierender Komponente wurde keine orthostatische Hypotonie festgestellt.

B. Loev, T. Jen and R. A. McLean

Research Division, Smith Kline and French Laboratories, 1500 Spring Garden Street, Philadelphia (Pennsylvania 19101, USA), 2 March 1971.

- <sup>1</sup> Manuscripts in preparation.
- <sup>2</sup> E. ZIEGLER, W. STEIGER and Th. KAPPE, Mh. Chem. 99, 1499 (1968). R. J. GROUT and M. W. PARTRIDGE, J. chem. Soc. 1960, 3551.
- <sup>3</sup> D. M. GREEN, F. J. SAUNDERS, N. WAHLGREN and R. L. CRAIG, Am. J. Physiol. 170, 94 (1952).
- <sup>4</sup> K. S. GRIMSON, Archs Surg., Lond. 43, 284 (1941).
- J. PARRA and H. VIDRIO, Archs int. Pharmacodyn. Thér. 181, 353 (1969).

## The Synthesis of Toxicarol Isoflavone

Toxicarol isoflavone,  $C_{23}H_{22}O_7$ , was isolated from the root of *Derris malaccensis* along with rotenoids (rotenone, toxicarol (I) etc.) by Harper <sup>1</sup>. Its structure was identified as 2", 2"-dimethylpyrano(6", 5":7,8)-5-hydroxy-2', 4', 5'-trimethoxyisoflavone (II) on the basis of NMR spectral evidence <sup>2</sup>. II is very likely to be related biogenetically to I. The present paper will describe the synthesis of II confirming the proposed structure.

By a modified Hoesch reaction, 2,2-dimethyl-5-hydroxy-7-methoxychroman³ was condensed with 2,4,5-trimethoxy-phenylacetonitrile⁴ in the presence of anhydrous aluminum chloride to give the corresponding desoxybenzoins (III, mp 177–178°, NMR⁵ 14.17 $_8$  (Phenol [intramolecular hydrogen bonding]) and IV, mp 152–154°, NMR 6.8 $_{b8}$  (Phenol[other phenol]). Treatment of III with ethyl formate in the presence of sodium gave the dihydropyranoisoflavone (V, mp 188–189.5°, IR 1663, 1641 cm<sup>-1</sup> (Nujol), UV $\lambda_{max}^{\rm EtOH}$  nm (log  $\varepsilon$ ); 259(4.50), 295(4.17), NMR 1.87 $_t$ (J = 7 Hz), 2.80 $_t$  (J = 7 Hz) (-CH $_2$ -CH $_2$ -), 7.86 $_t$  (C $_2$ -H). Found: C, 67.51; H, 5.86. C $_{24}$ H $_{26}$ O $_7$  requires: C, 67.59; H, 6.15%). V was dehydrogenated with DDQ to give toxicarol isoflavone methyl ether (VI, mp 179–181°, IR 1665, 1641 cm<sup>-1</sup> (Nujol); UV  $\lambda_{max}^{\rm EtOH}$  nm (log  $\varepsilon$ ); 264 (4.58), 293 $_{58}$ (4.08), NMR 1.48 $_t$  (6H) ((CH $_8$ ) $_2$ C < ), 3.72 $_t$ ,

3.83<sub>s</sub>, 3.90<sub>s</sub>(6H) (CH<sub>3</sub>O), 5.57<sub>d</sub> (J = 10 Hz) (C<sub>2</sub><sup>r</sup>-H), 6.31<sub>s</sub> (C<sub>6</sub>-H), 6.59<sub>s</sub> (C<sub>2</sub><sup>r</sup>-H), 6.74<sub>d</sub> (J = 10 Hz) (C<sub>4</sub><sup>r</sup>-H), 6.97<sub>s</sub> (C<sub>6</sub><sup>r</sup>-H), 7.80<sub>s</sub> (C<sub>2</sub>-H). Found: C, 66.71; H, 5.89. C<sub>24</sub>H<sub>24</sub>O<sub>7</sub>·1/2H<sub>2</sub>O requires: C, 66.50; H, 5.81%) (lit.¹ mp 178°). The partial demethylation of VI with anhydrous aluminum chloride in acetonitrile gave the desired isoflavone (II, mp 220–220.5°, IR 1656 cm<sup>-1</sup>, UV  $\lambda_{max}^{\rm ECH}$  nm (log  $\varepsilon$ ); 269(4.62), NMR 1.47<sub>s</sub>(6H) ((CH<sub>3</sub>)<sub>2</sub>C <), 3.79<sub>s</sub>, 3.86<sub>s</sub>, 3.92<sub>s</sub> (CH<sub>3</sub>O), 5.58<sub>d</sub> (J = 10 Hz) (C<sub>3</sub><sup>r</sup>-H), 6.29<sub>s</sub> (C<sub>6</sub><sup>r</sup>-H), 6.65<sub>s</sub> (C<sub>3</sub><sup>r</sup>-H), 6.72<sub>d</sub> (J = 10) (C<sub>4</sub><sup>r</sup>-H), 6.92<sub>s</sub> (C<sub>6</sub><sup>r</sup>-H), 7.92<sub>s</sub> (C<sub>2</sub>-H), 12.92<sub>s</sub> (OH). Found: C, 67.50; H, 5.42. C<sub>23</sub>H<sub>22</sub>O<sub>7</sub> requires: C, 67.31; H, 5.41%) (lit.¹ mp 219°), which was easily converted into

- <sup>1</sup> S. H. HARPER, J. chem. Soc. 1940, 1178.
- <sup>2</sup> S. H. HARPER and W. G. E. UNDERWOOD, J. chem. Soc. 1965, 4203.
- <sup>3</sup> S. N. George and A. Robertson, J. chem. Soc. 1937, 1535. A. P. Johnson and A. Pelter, J. chem. Soc. 1966, 606.
- <sup>4</sup> J. Harley-Mason and A. H. Jackson, J. chem. Soc. 1954, 1165.
- 5 The NMR-spectra were measured with a Hitachi R-20 (60 MHz) spectrometer, using tetramethylsilane as the internal standard (ô-value in CDCl<sub>3</sub>; s, singlet; bs, broad singlet; d, doublet; t, triplet).