water was distilled with toluene under nitrogen, and the glassy residue was washed twice with a small amount of ether and dissolved in 10 ml of water. The slightly pink solution was stirred with Dowex 1-X8 for 1 hr at which time a slight precipitate formed and was filtered. The water was removed with toluene and 1.15 g (64%) of tan powder was obtained: mp 295°. Anal. (C₁₀- $H_{13}NO_5$) C, H, N.

References and Notes

(1) C. A. Stone and C. C. Porter, Adv. Drug Res., 4, 71 (1967).

Notes

- (3) C. S. Rondestvedt, Jr., B. Manning, and S. Tabibian, J. Am. Chem. Soc., 72, 3183 (1950).
- (4) S. Searles and G. J. Cvejanovich, J. Am. Chem. Soc., 72, 3200 (1950).
- (5) H. E. Carter, Org. React., 3, 198 (1946).
- (6) M. Sletzinger, R. A. Firestone, and D. F. Reinhold, U.S. Patent 3395176; Chem. Abstr., 64, 19769f (1966).
- (7) R. Rabei, S. Spector, N. L. Louis, and A. Sjoerdsma, Clin. Pharmacol. Ther., 11, 289 (1970).

Central Nervous System Depressants. 13. s-Triazolo-1,5-benzodiazepin-5-ones^{†,1}

Robert Bruce Moffett,* Bharat V. Kamdar, and P. F. VonVoigtlander

Research Laboratories, The Upjohn Co., Kalamazoo, Michigan 49001. Received April 28, 1975

Several 8-chloro-6-phenyl-4H-s-triazolo[4,3-a][1,5]benzodiazepin-5(6H)-ones, substituted in the 1 position, were prepared. These were tested for CNS activity. The most active, as a depressant, was the 1-methyl compound.

In continuation² of our search for better drugs acting on the central nervous system a series of s-triazolo-1,5benzodiazepin-5-ones was prepared. These were made as outlined in Scheme I. P_2S_5 on 1³ was found to give a mixture of mono- and dithiones, 2 and 3. Since completion of this work Weber and Bauer⁴ reported a similar mixture. We were able to separate the mixture by chromatography; however, this was not necessary since the mixture could be used and the products 4 or 5 purified.

The triazolo compounds could be obtained from the thione mixture by treatment with a hydrazide or via the hydrazine compound 5 and subsequent acylation and cyclization. The chloromethyl group in the 1 position provided access to other triazolo compounds by nucleophilic displacement.

Pharmacology. Compounds 4, 7, and 8 were subjected to a battery of tests designed to detect CNS activity. All three compounds were nontoxic as shown by no loss of righting reflex or loss of traction in standard tests⁵ at 100 mg/kg. Compound 4 was active in the nicotine antagonism test⁵ (ED₅₀ 5 mg/kg), the strychnine antagonism test⁶ (ED₅₀ 28 mg/kg), the pentylenetetrazole antagonism test⁷ (ED₅₀ 7 mg/kg), and the hypoxic stress test (ED₅₀ 60 mg/kg). A standard benzodiazepine, chlordiazepoxide, had the following ED₅₀'s in these antagonism tests: nicotine 1.0, strychnine 13, pentylenetetrazole 2.6, and hypoxic stress 10.5 mg/kg. Compound 8 was weakly active (nicotine ED_{50} 89 and pentylenetetrazole ED_{50} 25 mg/kg). Compound 7 was inactive in these tests at 100 mg/kg. These tests indicate that the 1-methyl compound 4 possesses moderate CNS depressant or tranquilizing activity.

Experimental Section⁸

8-Chloro-1-methyl-6-phenyl-4H-s-triazolo[4,3-a][1,5]benzodiazepin-5(6H)-one (4). A solution of 3 g of a mixture of 2 and 3⁴ and 2 g of acethydrazide in 300 ml of *n*-BuOH was heated under reflux for 24 hr during which time N₂ was passed through the solution. After evaporation in vacuo the residue was washed (H₂O) and chromatographed on silica gel eluting with 3% MeOH in CHCl₃. The product (1.2 g) was recrystallized from EtOH yielding 0.8 g of white crystals, mp 297-298°. The principal spectral bands are ir (Nujol mull) 1695 (C=O), 1600, 1540, 1500 (C=C/C=N), 1320, 1205, 1105, 830, 755, 720, 695 cm⁻¹ (C=C/arom); NMR (CDCl₃) δ 2.66 (s, 3, CH₃), ab centered at 3.58 and 4.23 (2, J = -14 Hz, 4-CH₂), and between 6.95 and 7.55 (m, 8, arom H's); mass spectrum M·+ 424 (1 Cl). Anal. (C₁₇H₁₃ClN₄O) C, H, Cl, N.

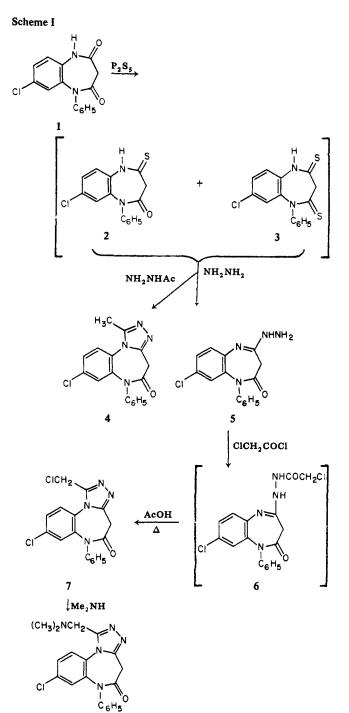
8-Chloro-1,3-dihydro-4-hydrazino-1-phenyl-2H-1,5benzodiazepin-2-one Hydrate (5). To a suspension of 12.9 g of a mixture of 2 and 3 in 350 ml of MeOH was added dropwise 9.6 ml of hydrazine hydrate during which time N₂ was passed through the mixture. After stirring at room temperature overnight the resulting white solid was collected, washed (MeOH), dried, and recrystallized from MeOH yielding 4 g of crystalline solid, mp 102-103°. An additional 2.5 g of less pure material was obtained from the filtrates. Ir and NMR support the structure and NMR indicates that it is a hydrate. Anal. (C₁₅H₁₃ClN₄-O·1.5H₂O) C, H, Cl, N.

8-Chloro-1-(chloromethyl)-6-phenyl-4H-s-triazolo[4,3a][1,5]benzodiazepin-5(6H)-one (7). A mixture of 4 g (0.0133 mol) of 5 and 30 ml of THF, under N₂, was cooled to 0° and 1.5 g (0.0133 mol) of ClCH₂COCl in 5 ml of THF was added dropwise with stirring. After stirring at 0° for 35 min and at room temperature for 1 hr, the solution was poured into ice water, mixed with a little CHCl₃, and neutralized with NaHCO₃. The resulting solid was collected, washed (H₂O and Et₂O), and dried yielding 4 g of white solid, mp 210–215°. Ir and NMR indicate this is the expected 2-(7-chloro-5-phenyl-3H-1,5-benzodiazepin-2-yl)chloroacetyl hydrazide (6).

Without further purification 1.5 g of 6 in 20 ml of AcOH was heated, under N₂, in a bath at 140° for 4 hr. After cooling the resulting solid was collected, washed (H₂O and Et₂O), and dried yielding 1.1 g of 7, mp 282–285° dec. A sample was recrystallized from MeOH-CH₂Cl₂: mp 306–308° dec. The principal spectral bands are ir (Nujol mull) 3060 (=CH), 1695 (C=O), 1600, 1595, 1525, 1505, 1495 (C=C/C=N), 1320, 855, 760, 695 cm⁻¹ (C=C/arom); uv (EtOH) 229 nm (ϵ 34,200), 283 (1950), 291 (sh, 1650); mass spectrum M.⁺ 258 (2 Cl). Anal. (C₁₇H₁₂Cl₂N₄O) C, H, N; Cl: calcd, 19.74; found, 20.30.

8-Chloro-1-[(dimethylamino)methyl]-6-phenyl-4H-s-triazolo[4,3-a][1,5]benzodiazepin-5(6H)-one (8). A mixture of 1.07 g (3.0 mmol) of 7 and 30 ml of THF was cooled to 0° under N₂. A methanolic solution of 5 g of Me₂NH and 0.5 g of KI was added and the mixture was stirred at room temperature for 4 hr. The solution was evaporated in vacuo, mixed with aqueous NaHCO₃, and extracted with CHCl₃. After washing (H₂O) and drying (Na₂SO₄) the solution was evaporated. On trituration with EtOAc the oil crystallized yielding 0.98 g of white solid, mp 232-234°. The principal spectral bands are ir (Nujol mull) 3060,

[†] Dedicated to the memory of Professor Edward E. Smissman.



8

3040 (=CH), 2780 (CH of N alkyl), 1695 (C=O), 1600, 1580, 1535, 1505, 1495 (C=C/C=N), 1320, 835, 765, 710 cm⁻¹ (C=C/arom); uv (EtOH) 228 nm (ϵ 37,650), 282 (sh, 2100), 290 (sh, 1850); NMR (CDCl₃) δ 2.35 (s, 6, CH₃), 370 (s, 2, NCH₂-), ab centered at 3.59 and 4.27 (2, J = -14 Hz, 4-CH₂), and between 6.95 and 8.32 (m, 8, arom H's). Anal. (C₁₉H₁₈ClN₅O) C, H, Cl, N.

Biological Test Methods. Male Carworth Farms (CF-1) mice were used in all studies reported here. The test compounds were dissolved or suspended in 0.25% aqueous carboxymethylcellulose and administered ip.

The tests were repeated at dose intervals of 0.3 log units until activity was no longer noted. ED50's (mg/kg) were calculated by the method of Spearman and Karbes.⁹ Methods for determining acute lethality,⁵ loss of righting reflex,⁵ loss of traction,⁵ nicotine antagonism,⁵ strychnine antagonism,⁶ and pentylenetetrazole antagonism⁷ have been published.

Hypoxic stress antagonism was tested by placing mice in 125-ml flasks and measuring the time which the animal survived after stoppering the flask. Mice were scored as displaying hypoxic stress antagonism if their survived times were 2 SD above the control \bar{x} survival time.

Acknowledgment. The authors wish to thank the following who contributed to this work: our Physical and Analytical Chemistry Unit for analytical and spectral data; Drs. Jacob Szmuszkovicz and Allan Rudzik for helpful consultations; and Mr. Walter Friis for technical assistance.

References and Notes

- Presented in part at the 9th Great Lakes Regional Meeting of the American Chemical Society, St. Paul, Minn., June 4, 1975.
- (2) R. B. Moffett, J. Org. Chem., 39, 568 (1974) (paper 12).
- K.-H. Weber, A. Bauer, and K.-H. Hauptmann, S. African Patent Spec. 680803 (1968) [Chem. Abstr., 70, 106579f (1969)]; Justus Liebigs Ann. Chem., 756, 128 (1972).
- (4) K.-H. Weber and A. Bauer, Justus Liebigs Ann. Chem., 1973, 000 (1974).
- (5) G. H. Youngdale, D. G. Anger, W. C. Anthony, J. P. Da-Vanzo, M. E. Greig, R. V. Heinzelman, H. H. Keasling, and J. Szmuszkovicz, J. Med. Chem., 7, 415 (1964).
- (6) H. H. Keasling, E. L. Schumann, and W. Veldkamp, J. Med. Chem., 8, 548 (1965).
- (7) J. B. Hester, A. D. Rudzik, and B. V. Kamdar, J. Med. Chem., 14, 1078 (1971).
- (8) Melting points were taken in capillary tubes with a partial immersion thermometer. Calibration of the apparatus against standard compounds showed no need for correction. Spectra were taken on the following: ir on PE 421, uv on Cary 15, mass spectra on Varian Mat CH5, and NMR on Varian A-60D spectrometers.
- (9) D. J. Finney, "Statistical Method in Biological Assay", Hafner Publication Co., New York, N.Y., 1952.