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Article

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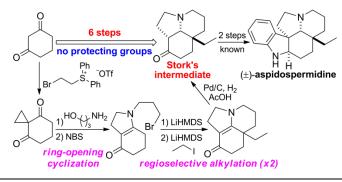
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Protecting-Group-Free Formal Synthesis of Aspidospermidine: Ring-Opening Cyclization of Spirocyclopropane with Amine Followed by Regioselective Alkylations

Hisanori Nambu,* Takafumi Tamura, and Takayuki Yakura

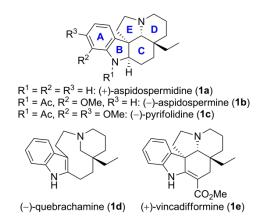
Graduate School of Medicine and Pharmaceutical Sciences, University of Toyama, Sugitani, Toyama 930-0194, Japan. *Supporting Information Placeholder*

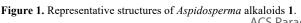


ABSTRACT: A concise formal synthesis of (\pm) -aspidospermidine via Stork's intermediate, which could be used as a divergent synthesis of *Aspidosperma* alkaloids, was achieved by employing a ring-opening cyclization of spirocyclopropane with amine followed by a regioselective intramolecular/intermolecular alkylation sequence. Stork's intermediate was synthesized in only six steps from a simple starting material, 1,3-cyclohexanedione, and was converted into (\pm) -aspidospermidine. To the best of our knowledge, this synthesis of Stork's intermediate involves the least number of steps to date. Furthermore, no protecting groups were used during this synthesis.

INTRODUCTION

Over the past half-century, *Aspidosperma* alkaloids 1 (Figure 1) have elicited considerable attention from synthetic chemists due to their structural diversity and interesting biological activities.¹ Aspidospermidine (1a) and aspidospermine (1b), the prototypical members of the group, contain a pentacyclic [6.5.6.6.5]-ABCDE ring system and have been attractive targets for demonstrating new synthetic approaches to *Aspidosperma* alkaloids.²





As a pioneering Aspidosperma alkaloid synthesis, Stork and Dolfini reported total synthesis of aspidospermine (1b) and quebrachamine (1d) in 1963, in which tricyclic [6.6.5]-CDE ring compound 2 was used as a key synthetic intermediate (Scheme 1A).³ CDE ring 2 was constructed from a starting material possessing the C ring by forming the D ring followed by the E ring.⁴ Conversion of 2 into natural products was achieved by using a Fischer indolization and reduction sequence, although the stereochemistry of 2 was not determined at this point.⁵ In 2000, Aubé and co-workers reported the total synthesis of (+)-aspidospermidine (1a) via intermediate 2 and established the stereochemistry of 2.6 Recently, Jiang and coworkers demonstrated a divergent asymmetric synthesis of Aspidosperma alkaloids including (-)-pyrifolidine (1c) and (+)-vincadifformine (1e) from $2.^7$ Therefore, intermediate 2 has become an attractive and well-known target as a Stork's intermediate, and various synthetic approaches to 2 have been reported to date.8,9

We previously reported a regioselective ring-opening cyclization of spirocyclopropanes **3** with primary amines **4** to generate 2-substituted tetrahydroindol-4-ones **5** (Scheme 1B).^{10,11} Furthermore, regioselective alkylation of **5** at the C-7 position to form alkylated product **6** was achieved by using lithium hexamethyldisilazide (LiHMDS) as a base.¹² Based on these results, we envisioned that these reactions could be applied to synthesize tricyclic CDE ring compound 2. Herein, we report a concise route to *Aspidosperma* alkaloid (\pm) aspidospermidine (1a) via Stork's intermediate 2, employing a ring-opening cyclization of spirocyclopropane 3a with amine 4 to construct CE ring system 5 followed by a regioselective intramolecular/intermolecular alkylation sequence to form the D ring (Scheme 1C).

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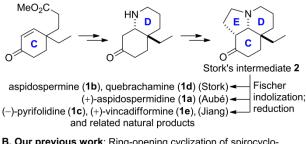
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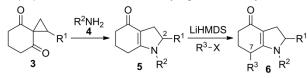
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Scheme 1. Stork's-Intermediate-based Approaches to Aspidosperma Alkaloids 1

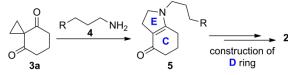
A. Pioneer work: Stork's synthesis of Aspidosperma alkaloids 1



B. Our previous work: Ring-opening cyclization of spirocyclopropane 3 with amine 4 followed by regioselective alkylation



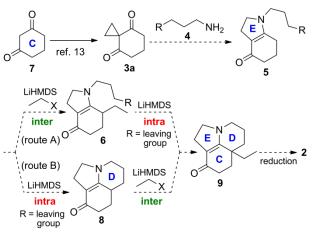
C. This work: A concise synthesis of Stork's intermediate 2 from spirocyclopropane 3a



RESULTS AND DISCUSSION

Our synthetic strategy for Stork's intermediate 2 is illustrated in Scheme 2. We planned to use 1,3-cyclohexanedione (7) as the C ring. After conversion of 7 into spirocyclopropane 3a by our synthetic method,¹³ construction of the E ring could be achieved through a ring-opening cyclization of spirocyclopropane 3a with propylamine 4 to produce bicyclic product 5 possessing a C3 unit at the nitrogen atom for constructing the D ring.¹⁰ We then devised two synthetic routes as follows. One was an intermolecular alkylation of 5 for introducing the ethyl group followed by intramolecular alkylation of resulting product 6 to construct the D ring (route A). The other was to use the two alkylations in reverse order, such as an intramolecular/intermolecular alkylation sequence via tricyclic intermediate 8 (route B). Finally, reduction of the obtained tricyclic CDE ring system 9 would produce key intermediate 2. We predicted that the second alkylation step at the tertiary carbon could be more difficult due to the steric hindrance. Generally, an intramolecular reaction proceeds more easily than an intermolecular one. Therefore, we initially investigated an intermolecular/intramolecular alkylation sequence (route A).

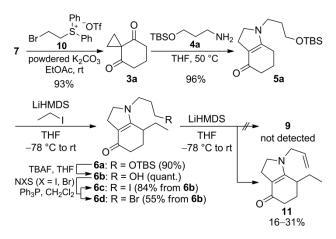
Scheme 2. Synthetic Strategy for Stork's Intermediate 2



route A: intermolecular/intramolecular alkylation sequence route B: intramolecular/intermolecular alkylation sequence

At the outset of this work, synthetic approach by route A starting from 1,3-cyclohexanedione (7) is shown in Scheme 3. The reaction of 7 with sulfonium salt 10 and powdered K_2CO_3 in EtOAc provided spirocyclopropane 3a in 93% yield.¹³ Ring-opening cyclization of **3a** with *tert*-butyldimethylsilyl (TBS)-protected 3-amino-1-propanol 4a proceeded smoothly in THF at 50 °C to produce tetrahydroindol-4-one 5a in 96% yield. An intermolecular regioselective alkylation of 5a with ethyl iodide using LiHMDS as a base afforded ethylated product 6a in 90% yield. After deprotection of TBS ether 6a with tetrabutylammonium fluoride (TBAF), halogenation of resulting alcohol **6b** using *N*-halosuccinimide (NXS, X = I, Br) and triphenylphosphine gave the corresponding iodide 6c and bromide 6d in 84% and 55% yields, respectively. We then examined an intramolecular alkylation using halogenated alkanes 6c and 6d. When these were treated with LiHMDS as a base, cyclization did not proceed. A small amount of corresponding alkene 11 (16%-31% yields) rather than tricyclic compound 9 was obtained by E2 elimination.

Scheme 3. Synthetic Approach Using Regioselective Intermolecular/Intramolecular Alkylations (Route A)



Since the intermolecular/intramolecular alkylation sequence could not be used to synthesize tricyclic compound 9, we next

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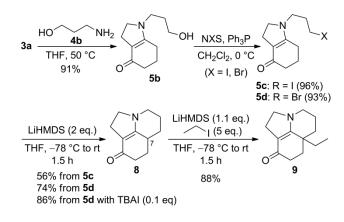
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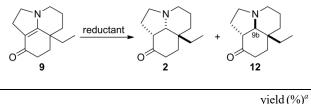
investigated an intramolecular/intermolecular alkylation sequence (route B, Scheme 4). To omit the deprotection step, we used 3-amino-1-propanol 4b for the ring-opening cyclization of 3a. The reaction proceeded uneventfully to provide tetrahydroindol-4-one 5b in 91% yield. In this key reaction, amino group, which is more nucleophilic than hydroxy group, in 4b would attack the cyclopropyl carbon in 3a to cleave the cyclopropane and the subsequent cyclization would form **5b**.¹⁰ Halogenation of alcohol 5b using NXS and Ph₃P gave corresponding iodide 5c and bromide 5d. We then examined an intramolecular alkylation of 5c and 5d using LiHMDS as a base. The use of iodide 5c afforded tricyclic product 8 in 56% yield, accompanied by a small amount of elimination product. When bromide 5d was used in this reaction, the yield of 8 increased to 74%. Gratifyingly, the reaction of 5d in the presence of 0.1 equiv of tetrabutylammonium iodide (TBAI) increased the product vield to 86%. Intermolecular alkylation at the C-7 tertiary carbon of 8 was achieved by using excess amount of alkylating reagent. The reaction of 8 with LiHMDS (1.1 equiv) and ethyl iodide (5 equiv) gave ethylated product $9^{8a,j}$ in 88% yield.

Scheme 4. Synthetic Approach Using Regioselective Intramolecular/Intermolecular Alkylations (Route B)



With tricyclic intermediate 9 in hand,^{8a,j} we then investigated the reduction of 9 (Table 1). The use of lithium aluminum hydride (LiAlH₄) as a reductant afforded 9b-epi-product 12 in 81% yield (entry 1), as well as the previous reports.¹⁴ Additionally, the reaction under Birch reduction conditions also provided 12 in 54% yield (entry 2). We then examined the hydrogenation of 9.15 To our delight, the reaction of 9 with a catalytic amount of Pt/C in acetic acid under hydrogen at 50 °C gave Stork's intermediate 2 in 18% yield (entry 3). Synthetic material 2 showed spectroscopic data (¹H and ¹³C NMR, and IR) consistent with those reported for product 2.8a-c Because this Pt/C-catalyzed hydrogenation required a long reaction time to reach completion (28 h), the product 2 was partially decomposed under the reaction conditions to cause the low vield of 2. After considerable screening of the reaction conditions, we found that hydrogenation with a catalytic amount of Pd/C in acetic acid at 50 °C proceeded faster to completion within 12 h, affording 2 in 34% yield along with decomposition products (entry 6).¹⁶ Consequently, Stork's intermediate 2 was synthesized in only six steps from a simple starting material, 1,3-cyclohexanedione (7), and no protecting groups were used during this route.

Table 1. Stereoselective Reduction of Enaminone 9

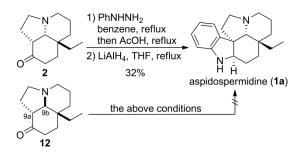


					yield (%)"	
entry	reductant	solvent	temp.	time (h)	2	12
1	LiAlH ₄	THF	reflux	0.5	_	81
2	Na, liq. NH3	THF	-78 °C to rt	2	_	54
3	Pt/C, H ₂	AcOH	50 °C	28	18	-
4	PtO_2, H_2	AcOH	50 °C	24	8	_
5	$Pd(OH)_2, H_2$	EtOAc	50 °C	13	14	_
6	$Pd/C, H_2$	AcOH	50 °C	12	34	_
7	cat. $Rh(I)$, $^{b}H_{2}$	toluene	50 °C	12		NR^d
8	cat. $Ir(I)$, $^{c}H_{2}$	CH_2Cl_2	rt	12		NR^d

^{*a*}Isolated yield. ^{*b*}Wilkinson's catalyst. ^{*c*}Crabtree's catalyst. ^{*d*}No reaction.

The synthesis of aspidospermidine (1a) from Stork's intermediate 2 by employing Fischer indolization with phenylhydrazine in acetic acid and subsequent reduction with LiAlH₄ or NaBH₄ was achieved by several research groups.¹⁷ We also demonstrated this conversion under the above conditions and obtained natural product 1a in 32% yield (Scheme 5).¹⁸ With the expectation that isomerization of the stereochemistry at C-9b of 12 could proceed in the indolization step,¹⁹ conversion of 12 into 1a was also explored. Unfortunately, this approach was unsuccessful, as in previous studies.²⁰

Scheme 5. Synthesis of Aspidospermidine (1a)



CONCLUSION

A concise formal synthesis of (±)-aspidospermidine via Stork's intermediate was achieved by employing a ringopening cyclization of spirocyclopropane followed by a regioselective intramolecular/intermolecular alkylation sequence. Stork's intermediate, which could be converted into various Aspidosperma alkaloids, was synthesized in only six steps from a simple starting material, 1,3-cyclohexanedione, and was converted into (\pm) -aspidospermidine in two steps. To the best of our knowledge, this synthesis of Stork's intermediate involves the least number of steps to date.²¹ Additionally, no protecting groups were used during this synthesis. Further toward the asymmetric synthesis of (+)efforts aspidospermidine and related alkaloids employing an enantioselective intermolecular alkylation are currently in progress.

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EXPERIMENTAL SECTION

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General. Melting points are uncorrected. IR spectra were recorded on a JASCO FT/IR-460 Plus spectrophotometer and absorbance bands are reported in wavenumber (cm⁻¹). ¹H NMR spectra were recorded on JEOL JNM-ECX400P (400 MHz) spectrometer. Chemical shifts are reported relative to internal standard (tetramethylsilane at $\delta_H 0.00$ or CDCl₃ at $\delta_H 7.26$). Data are presented as follows: chemical shift (δ , ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, m = multiplet, br = broad), coupling constant and integration. $^{13}\mathrm{C}\{^{1}\mathrm{H}\}$ NMR spectra were recorded on JEOL JNM-ECX400P (100 MHz) spectrometer. The following internal reference was used (CDCl₃ at δ 77.0). High-resolution mass spectra (HRMS) were recorded on JEOL JMS-GCmate II (EI) and JEOL JMS-AX505HAD (FAB) double-focusing magnetic-sector mass spectrometers. Column chromatography was performed on Silica Gel 60 PF₂₅₄ (Nacalai Tesque) and Kanto silica gel 60 N (63-210 mesh) under pressure. Analytical thin layer chromatography (TLC) was carried out on Merck Kieselgel 60 F254 plates. Visualization was accomplished with UV light and phosphomolybdic acid stain solution followed by heating. An oil bath was used for conventional heating.

All reagents such as 1,3-cyclohexanedione (7), powdered potassium carbonate, 3-amino-1-propanol (4b), 1,1,1,3,3,3-hexamethyldisilazane (HMDS), ethyl iodide, N-iodosuccinimide (NIS), N_{-} bromosuccinimide (NBS), triphenylphosphine, tetrabutylammonium iodide (TBAI), phenylhydrazine, acetic acid, and lithium aluminum hydride are commercially available and were purchased from suppliers such as Sigma-Aldrich Co.; Wako Pure Chemical Industries, Ltd.; Tokyo Chemical Industry Co., Ltd.; Nacalai Tesque, INC. Dehydrated CH₂Cl₂, THF, EtOAc, toluene, and benzene were purchased from Wako Pure Chemical Industries, Ltd. Sulfonium salt 1013 and 3-(tertbutyldimethylsilyloxy)propylamine (4a)²² were prepared according to literature procedures.

Preparation of LiHMDS (0.83 M in THF/hexane): BuLi (1.64 M in hexane, 6.1 mL, 10.0 mmol) was added to a solution of HMDS (2.3 mL, 11.0 mmol) in THF (3.6 mL) at 0 °C, and then the mixture was stirred at 0 °C for 0.5 h.

Intermolecular/intramolecular alkylation sequence (route A). Spiro[2.5]octane-4,8-dione (3a).13 Powdered K₂CO₃ (2.51 g, 18.2 mmol) and 1,3-cyclohexanedione (7) (0.68 g, 6.06 mmol) were added to a suspension of sulfonium salt 10 (2.95 g, 6.67 mmol) in EtOAc (60 mL). After stirring at rt for 3 h, the reaction mixture was filtered through a pad of Celite. The filter cake was rinsed with EtOAc (30 mL) and the filtrate was quenched with water (20 mL) and the whole mixture was extracted with EtOAc (2×10 mL). The combined organic layer was washed with water (20 mL), brine (20 mL) and dried over anhydrous MgSO₄. The filtrate was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, 40% EtOAc in hexane) to provide 3a (778 mg, 93%) as a pale yellow oil: IR (film, cm⁻¹) v 2956, 1682, 1330, 1162, 1026, 956; ¹H NMR (400 MHz, CDCl₃) δ 2.67 (t, J = 6.4 Hz, 4H), 2.15 (quint, J = 6.4 Hz, 2H), 1.76 (s, 4H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 206.8, 40.6, 39.4, 27.4, 17.9; HRMS (EI) m/z calcd for C₈H₁₀O₂ (M⁺) 138.0681, found 138.0668.

1-(3-tert-Butyldimethylsilyloxy)propyl-2,3,6,7-tetrahydro-1H-indol-4(5H)-one (5a). 3-(*tert*-Butyldimethylsilyloxy)propylamine (4a) (232 mg, 1.23 mmol) was added to a solution of spirocyclopropane **3a** (113 mg, 0.82 mmol) in THF (1.6 mL). After stirring at 50 °C for 4.5 h, the reaction mixture was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, 5% MeOH in CH₂Cl₂) to provide **5a** (243 mg, 96%) as a yellow oil: IR (film, cm⁻¹) v 2918, 2849, 1509, 1440, 1290, 1254, 1190, 1094; ¹H NMR (400 MHz, CDCl₃) δ 3.59 (t, *J* = 5.6 Hz, 2H), 3.50 (t, *J* = 9.8 Hz, 2H), 3.27 (t, *J* = 7.0 Hz, 2H), 2.73 (t, *J* = 9.8 Hz, 2H), 2.32 (t, *J* = 6.2 Hz, 2H), 1.94 (tt, *J* = 5.6, 7.0 Hz, 2H), 1.69 (quint, *J* = 6.2 Hz, 2H), 0.02 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 190.6, 168.8, 108.9, 59.3, 51.3, 42.8, 35.6, 30.7, 25.8, 23.9, 22.5, 22.3, 18.1, -5.5; HRMS (EI) *m/z* calcd for C₁₇H₃₁NO₂Si (M⁺) 309.2124, found 309.2127.

> 1-(3-tert-Butyldimethylsilyloxy)propyl-7-ethyl-2,3,6,7-tetrahydro-1H-indol-4(5H)-one (**6a**). A solution of LiHMDS (1.05 mL, 0.83 M

in THF/hexane, 0.88 mmol) was added to a solution of 5a (246 mg, 0.80 mmol) in THF (8 mL) at -78 °C. After stirring at -78 °C for 1 h, ethyl iodide (0.096 mL, 1.2 mmol) was added to the mixture and the whole was allowed to warm to room temperature over 1.5 h. The reaction mixture was quenched with saturated aqueous NH₄Cl (10 mL), and the whole mixture was extracted with CH₂Cl₂ (2 x 10 mL). The combined organic layers were washed with water (10 mL) and brine (10 mL), and dried over anhydrous MgSO₄. Filtration and evaporation in vacuo furnished the crude product, which was purified by column chromatography (silica gel, 5% MeOH in CH₂Cl₂) to provide **6a** (268 mg, 90%) as a yellowish brown oil: IR (film, cm^{-1}) v 2929, 2857, 2360, 2341, 1556, 1504, 1253, 1178, 1092, 836; ¹H NMR (400 MHz, CDCl₃) δ 3.66–3.54 (m, 2H), 3.47 (td, J = 9.8, 2.4 Hz, 2H), 3.33 (dt, J = 15.2, 7.6 Hz, 1H), 3.16 (dt, J = 15.2, 7.6 Hz, 1H), 2.72 (t, J = 9.8 Hz, 2H), 2.41–2.33 (m, 2H), 2.16 (dd, J = 17.6, 2.0 Hz, 1H), 1.98 (dt, J = 13.6, 2.0 Hz, 1H), 1.87 (m, 1H), 1.73–1.71 (m, 2H), 1.59–1.46 (m, 2H), 1.01 (t, *J* = 7.2 Hz, 3H), 0.87 (s, 9H), 0.32 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 190.2, 172.3, 107.8, 59.4, 50.8, 42.9, 33.1, 31.4, 30.7, 25.8, 25.0, 23.9, 23.1, 18.2, 12.5, -5.5; HRMS (FAB) *m/z* calcd for C₁₉H₃₆NO₂Si (M+H)⁺ 338.2515, found 338.2508.

7-Ethyl-1-(3-hydroxy)propyl-2,3,6,7-tetrahydro-1H-indol-4(5H)one (6b). TBAF (1.16 mL, 1 M in THF, 1.16 mmol) was added to a solution of 6a (390 mg, 1.16 mmol) in THF (5.8 mL) at 0 °C. After stirring at room temperature for 0.5 h, the reaction mixture was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, 11% MeOH in CH₂Cl₂) to provide **6b** (258 mg, quant.) as a yellowish brown oil: IR (film, cm⁻¹) v 3278, 2935, 2871, 1579, 1534, 1507, 1446, 1299, 1198, 1181; ¹H NMR (400 MHz, CDCl₃) δ 3.73–3.63 (m, 2H), 3.57–3.40 (m, 4H), 3.28 (dt, J = 13.2, 6.6 Hz, 1H), 2.73 (t, J = 9.6 Hz, 2H), 2.47 (dd, J = 4.8, 4.0 Hz, 1H), 2.39 (ddd, J = 17.8, 14.0, 4.0 Hz, 2H), 2.16 (dd, J = 17.8, 3.2 Hz, 2H), 2.02 (dt, J = 13.2, 2.4 Hz, 1H), 1.93 (dt, J = 14.0, 4.8 Hz, 1H), 1.84 (td, *J* = 12.8, 6.6 Hz, 2H), 1.59–1.49 (m, 2H), 1.04 (t, *J* = 7.2 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 190.0, 173.3, 107.7, 58.9, 50.9, 43.1, 33.2, 31.1, 30.4, 24.9, 23.6, 23.1, 12.5; HRMS (EI) m/z calcd for C₁₃H₂₁NO₂ (M⁺) 223.1572, found 223.1573.

Typical procedure for halogenation of alcohols. 7-Ethyl-1-(3iodo)propyl-2,3,6,7-tetrahydro-1H-indol-4(5H)-one (6c). NIS (170 mg, 0.65 mmol) and triphenylphosphine (146 mg, 0.65 mmol) were added to a solution of 6b (97 mg, 0.43 mmol) in CH₂Cl₂ (2.2 mL). After stirring at room temperature for 0.5 h, the reaction mixture was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, 5% MeOH in CH2Cl2) to provide 6c (122 mg, 84%) as a yellowish brown oil: IR (film, cm⁻¹) v 2934, 2870, 1581, 1542, 1503, 1445, 1299, 1248, 1195, 1181; ¹H NMR (400 MHz, CDCl₃) δ 3.48 (td, J = 9.6, 2.0 Hz, 2H), 3.39–3.14 (m, 4H), 2.75 (t, J= 9.6 Hz, 2H), 2.47-2.36 (m, 2H), 2.21 (ddd, J = 17.2, 5.0, 2.0 Hz, 1H), 2.11–2.03 (m, 3H), 1.96 (tt, J = 13.6, 5.0 Hz, 1H), 1.66–1.45 (m, 2H), 1.08 (t, J = 7.6 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 190.8, 171.9, 108.7, 51.1, 46.8, 33.5, 31.5, 31.1, 25.0, 24.0, 23.3, 12.6, 2.1; HRMS (FAB) m/z calcd for C₁₃H₂₁INO (M+H)⁺ 334.0668, found 334.0676.

l-(*3*-*Bromo*)*propyl*-7-*ethyl*-2,3,6,7-*tetrahydro*-1*H*-*indol*-4(5*H*)-*one* (*6d*). According to the typical procedure for halogenation of alcohols, bromide **6d** was prepared from **6b** (138 mg, 0.62 mmol), NBS (100 mg, 0.93 mmol), and triphenylphosphine (243 mg, 0.93 mmol). The crude product was purified by column chromatography (silica gel, 5% MeOH in CH₂Cl₂) to provide **6d** (98 mg, 55%) as a yellowish brown oil: IR (film, cm⁻¹) v 2936, 2871, 1581, 1542, 1504, 1446, 1300, 1261, 1197, 1181; ¹H NMR (400 MHz, CDCl₃) δ 3.52–3.29 (m, 6H), 2.74 (t, *J* = 9.6 Hz, 2H), 2.47–2.36 (m, 2H), 2.23–2.03 (m, 4H), 1.96 (m, 1H), 1.64–1.50 (m, 2H), 1.07 (t, *J* = 7.6 Hz, 3H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 190.7, 172.1, 108.5, 50.9, 44.5, 33.3, 31.3, 30.3, 29.9, 24.9, 23.9, 23.2, 12.5; HRMS (EI) *m/z* calcd for C₁₃H₂₀BrNO (M⁺) 285.0728, found 285.0714.

l-Allyl-7-ethyl-2,3,6,7-tetrahydro-1H-indol-4(5H)-one (11). A solution of LiHMDS (0.21 mL, 0.83 M in THF/hexane, 0.17 mmol) was added to a solution of **6c** (52 mg, 0.16 mmol) in THF (1.6 mL) at -78 °C. After stirring at -78 °C for 1 h, the reaction mixture was allowed to warm to room temperature over 1.5 h. The reaction mixture was

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quenched with saturated aqueous NH₄Cl (5 mL), and the whole mixture was extracted with CH₂Cl₂ (2 x 5 mL). The combined organic layers were washed with water (5 mL) and brine (5 mL), and dried over anhydrous MgSO₄. Filtration and evaporation in vacuo furnished the crude product, which was purified by column chromatography (silica gel, 5% MeOH in CH₂Cl₂) to provide **11** (5.7 mg, 16%) as a brown oil: IR (film, cm⁻¹) v 2937, 2866, 1601, 1554, 1496, 1444, 1352, 1298, 1244, 1173, 946, 908; ¹H NMR (400 MHz, CDCl₃) δ 5.78 (m, 1H), 5.25 (dd, *J* = 10.4, 2.4 Hz, 1H), 5.21 (dd, *J* = 17.2, 2.4 Hz, 1H), 3.80 (d, *J* = 5.0 Hz, 2H), 3.58–3.43 (m, 2H), 2.78–2.72 (m, 2H), 2.43–2.35 (m, 2H), 2.20 (m, 1H), 2.05 (dt, *J* = 13.6, 3.2 Hz, 1H), 1.95 (tt, *J* = 13.6, 5.0 Hz, 1H), 1.69–1.49 (m, 2H), 1.05 (t, *J* = 7.6 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 190.5, 171.7, 132.4, 117.7, 108.4, 51.0, 48.7, 33.3, 31.3, 25.0, 23.8, 23.2, 12.4; HRMS (FAB) *m*/z calcd for C₁₃H₂₀NO (M+H)⁺ 206.1545, found 206.1541.

Intramolecular/intermolecular alkylation sequence (route B). *1*-(*3-Hydroxy*)*propyl-2,3,6,7-tetrahydro-1H-indol-4(5H)-one* (*5b*). 3-Amino-1-propanol (**4b**) (535 mg, 7.12 mmol) was added to a solution of spirocyclopropane **3a** (492 mg, 3.56 mmol) in THF (7 mL). After stirring at 50 °C for 2.5 h, the reaction mixture was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, 11% MeOH in CH₂Cl₂) to provide **5b** (630 mg, 91%) as a pale yellow solid: mp 80–81 °C; IR (KBr, cm⁻¹) v 3398, 2939, 2869, 1585, 1535, 1511, 1439, 1290, 1190; ¹H NMR (400 MHz, CDCl₃) δ 3.96 (s, 1H), 3.65 (t, *J* = 6.4 Hz, 2H), 3.57 (t, *J* = 9.6 Hz, 2H), 3.37 (t, *J* = 6.4 Hz, 2H), 2.75 (t, *J* = 9.6, 2H), 2.39 (t, *J* = 6.2 Hz, 2H), 2.27 (t, *J* = 6.2 Hz, 2H), 1.97 (quint, *J* = 6.4 Hz, 2H), 1.80 (quint, *J* = 6.2 Hz, 2H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 190.5, 169.6, 108.6, 58.7, 51.4, 43.1, 35.4, 30.4, 23.8, 22.5, 22.1; HRMS (FAB) *m/z* calcd for C₁₁H₁₈NO₂ (M+H)⁺ 196.1338, found 196.1337.

I-(*3*-*IdO*)*propyI-2,3,6,7-tetrahydro-1H-indoI-4(5H)-one* (*5c*). According to the typical procedure for halogenation of alcohols, iodide **5c** was prepared from **5b** (100 mg, 0.51 mmol), NIS (201 mg, 0.77 mmol), and triphenylphosphine (173 mg, 0.77 mmol) at 0 °C for 0.5 h. The crude product was purified by column chromatography (silica gel, 5% MeOH in CH₂Cl₂) to provide **5c** (150 mg, 96%) as a brown oil: IR (film, cm⁻¹) v 2936, 2862, 1594, 1556, 1506, 1187; ¹H NMR (400 MHz, CDCl₃) δ 3.51 (t, *J* = 9.6 Hz, 2H), 3.31 (t, *J* = 6.6 Hz, 2H), 3.19 (t, *J* = 6.6 Hz, 2H), 2.78 (t, *J* = 9.6 Hz, 2H), 2.40 (t, *J* = 6.2 Hz, 2H), 2.32 (t, *J* = 6.2 Hz, 2H), 2.10–1.99 (m, 4H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 191.2, 168.4, 109.7, 51.5, 46.6, 35.7, 31.0, 24.1, 22.9, 22.2, 2.3; HRMS (EI) *m/z* calcd for C₁₁H₁₆INO (M⁺) 305.0277, found 305.0284.

35 1-(3-Bromo)propyl-2,3,6,7-tetrahydro-1H-indol-4(5H)-one (5d)36 According to the typical procedure for halogenation of alcohols, bro-37 mide 5d was prepared from 5b (1.05 g, 5.35 mmol), NBS (1.43 g, 38 8.03 mmol), and triphenylphosphine (2.11 g, 8.03 mmol) at 0 °C for 0.5 h. The crude product was purified by column chromatography 39 (silica gel, 5% MeOH in CH₂Cl₂) to provide 5d (1.27 g, 93%) as a 40 yellowish brown oil: IR (film, cm⁻¹) v 2941, 2868, 1581, 1542, 1508, 41 1438, 1190; ¹H NMR (400 MHz, CDCl₃) δ 3.52 (t, *J* = 9.6 Hz, 2H), 42 3.44 (t, J = 6.4 Hz, 2H), 3.38 (t, J = 6.4 Hz, 2H), 2.77 (t, J = 9.6 Hz, 43 2H), 2.39 (t, J = 6.4 Hz, 2H), 2.30 (t, J = 6.4 Hz, 2H), 2.11 (quint, J = 6.4 Hz, 2H), 2.01 (quint, J = 6.4 Hz, 2H); ¹³C{¹H} NMR (100 MHz, 44 CDCl3) & 191.2, 168.4, 109.6, 51.4, 44.4, 35.7, 30.3, 30.0, 24.0, 22.6, 45 22.2; HRMS (FAB) m/z calcd for C₁₁H₁₇BrNO (M+H)⁺ 258.0494, 46 found 258.0496.

47 1,2,4,5,6,6a,7,8-Octahydro-9H-pyrrolo[3,2,1-ij]quinolin-9-one (8). 48 A solution of LiHMDS (11.9 mL, 0.83 M in THF/hexane, 9.87 mmol) was added to a solution of 5d (1.27 g, 4.93 mmol) and TBAI (182 mg, 49 0.49 mmol) in THF (49 mL) at -78 °C. After stirring at -78 °C for 1 50 h, the reaction mixture was allowed to warm to room temperature 51 over 1.5 h. The reaction mixture was quenched with saturated aque-52 ous NH₄Cl (10 mL), and the whole mixture was extracted with 53 CH₂Cl₂ (2 x 10 mL). The combined organic layers were washed with water (10 mL) and brine (10 mL), and dried over anhydrous MgSO₄. 54 Filtration and evaporation in vacuo furnished the crude product, 55 which was purified by column chromatography (silica gel, 5% MeOH 56 in CH₂Cl₂) to provide 8 (753 mg, 86%) as a pale yellow solid: mp 62-57 63 °C; IR (KBr, cm⁻¹) v 2936, 2860, 1550, 1524, 1436, 1361, 1294,

1196; ¹H NMR (400 MHz, CDCl₃) δ 3.62 (td, J = 10.8, 4.0 Hz, 1H), 3.32–3.24 (m, 2H), 2.86 (td, J = 12.0, 4.0 Hz, 1H), 2.80 (td, J = 12.0, 4.0 Hz, 1H), 2.68 (dtd, J = 14.8, 11.2, 2.0 Hz, 1H), 2.41–2.34 (m, 3H), 2.01–1.94 (m, 3H), 1.85 (m, 1H), 1.61 (tt, J = 12.0, 9.4 Hz, 1H), 1.17 (m, 1H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 190.8, 169.5, 107.8, 53.4, 46.3, 37.0, 33.4, 29.9, 26.7, 24.0, 22.8; HRMS (EI) *m/z* calcd for C₁₁H₁₅NO (M⁺) 177.1154, found 177.1147.

6a-Ethyl-1,2,4,5,6,6a,7,8-octahydro-9H-pyrrolo[3,2,1-ij]quinolin-9-one (9).8a, A solution of LiHMDS (0.88 mL, 0.83 M in THF/hexane, 0.73 mmol) was added to a solution of 8 (120 mg, 0.67 mmol) in THF (6.7 mL) at -78 °C. After stirring at -78 °C for 1 h, ethyl iodide (0.27 mL, 3.35 mmol) was added to the mixture and the whole was allowed to warm to room temperature over 1.5 h. The reaction mixture was quenched with saturated aqueous NH₄Cl (5 mL), and the whole mixture was extracted with CH₂Cl₂ (2 x 5 mL). The combined organic layers were washed with water (5 mL) and brine (5 mL), and dried over anhydrous MgSO₄. Filtration and evaporation in vacuo furnished the crude product, which was purified by column chromatography (silica gel, 5% MeOH in CH₂Cl₂) to provide 9 (123 mg, 88%) as a pale yellow solid: mp 63-64 °C; IR (KBr, cm⁻¹) v 2940, 2860, 1551, 1522, 1463, 1439, 1355, 1302, 1267, 1192; ¹H NMR (400 MHz, CDCl₃) δ 3.60 (ddd, J = 11.2, 10.4, 3.6 Hz, 1H), 3.30 (dd, J = 11.2, 5.2 Hz, 1H), 3.17 (q, J = 10.4 Hz, 1H), 2.85–2.74 (m, 2H), 2.62 (dt, J = 14.8, 11.2 Hz, 1H), 2.44 (ddd, J = 17.6, 13.2, 2.8 Hz, 1H), 2.27 (ddd, J = 17.6, 5.2, 2.4 Hz, 1H), 1.99 (dd, J = 4.4, 2.0 Hz, 1H), 1.96 (dd, J = 5.2, 2.0 Hz, 1H), 1.90 (dt, J = 13.4, 3.2 Hz, 1H), 1.79 (m, 1H), 1.71–1.63 (m, 2H), 1.55 (tdd, *J* = 13.4, 5.2, 1.2 Hz, 1H), 1.13 (td, J = 13.2, 3.6 Hz, 1H), 0.89 (t, J = 7.6 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) & 190.1, 174.8, 107.7, 53.9, 47.2, 35.0, 33.6, 32.5, 28.7, 25.2, 24.1, 18.7, 7.8; HRMS (FAB) m/z calcd for C₁₃H₂₀NO (M+H)⁺ 206.1545, found 206.1542.

Synthesis of (±)-aspidospermidine (1a). rac-(6aR,9aR,9bS)-6a-Ethyldecahydro-4H-pyrrolo[3,2,1-ij]quinolin-9-one (2).6-8 10% Pd/C (82 mg, 200 wt% of 9) was added to a solution of 9 (41 mg, 0.2 mmol) in AcOH (4 mL). The reaction mixture was vigorously stirred under H₂ atmosphere at 50 °C for 12 h. The mixture was filtered through a pad of Celite and the filter cake was rinsed with CH₂Cl₂ (20 mL). The filtrate was concentrated in vacuo, and saturated aqueous NaHCO₃ (20 mL) was added to the residue. The whole mixture was extracted with CH_2Cl_2 (2 × 10 mL), and the combined organic layer was washed with brine (10 mL) and dried over anhydrous MgSO₄. The filtrate was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, 1% MeOH in CHCl₃) to provide 2 (14 mg, 34%) as a colorless oil: IR (film, cm⁻¹) v 2929, 1711, 1457; ¹H NMR (400 MHz, CDCl₃) δ 3.03-2.99 (m, 2H), 2.67 (ddd, J = 9.0, 4.8, 1.6 Hz, 1H), 2.45–2.17 (m, 3H), 1.97–1.60 (m, 7H), 1.51– 1.47 (m, 2H), 1.37–1.28 (m, 2H), 1.10 (td, J = 13.4, 4.8 Hz, 1H), 0.94 (t, J = 7.6 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 211.5, 73.6, 53.2, 52.9, 48.2, 36.8, 34.8, 32.9, 30.1, 29.7, 26.1, 21.3, 7.1; HRMS (EI) *m/z* calcd for C₁₃H₂₁NO (M⁺) 207.1623, found 207.1616.

(±)-Aspidospermidine (1a).6b According to the procedure of Aubé et al., we demonstrated conversion of 2 into aspidospermidine (1a). Phenylhydrazine (3.8 mg, 0.035 mmol) was added to a solution of 2 (6.0 mg, 0.029 mmol) in benzene (0.83 mL). After stirring at reflux for 3 h, the reaction mixture was concentrated in vacuo, and the residue was dissolved in AcOH (0.83 mL). After stirring at reflux for 3.5 h, the reaction mixture was concentrated in vacuo, and the residue was dissolved in THF (0.83 mL). LiAlH₄ (11 mg, 0.29 mmol) was added to a solution of the crude product at 0 °C, and the reaction mixture was stirred at reflux for 11 h. After cooling to 0 °C, the reaction mixture was quenched by addition of one drop of aqueous 10% NaOH. The mixture was filtered through a pad of Celite and the filter cake was rinsed with CH2Cl2 (10 mL). The filtrate was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, 11% MeOH in EtOAc) to provide 1a (2.6 mg, 32%) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 7.07 (d, J = 8.0 Hz, 1H), 7.02 (t, J = 7.2 Hz, 1H), 6.73 (t, J = 7.2 Hz, 1H), 6.64 (d, J = 8.0 Hz, 1H), 3.52 (m, 1H), 3.12-3.06 (m, 2H), 2.35-2.20 (m, 3H), 1.98-1.91 (m, 2H), 1.79-1.66 (m, 3H), 1.53-1.34 (m, 4H), 1.13 (m, 1H), 1.07

58

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.xxxx.

¹H and ¹³C{¹H} NMR spectra for new compounds (PDF)

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Notes

The authors declare no competing financial interest.

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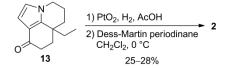
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(14) The Banwell^{8a} and Yang^{8j} groups have independently reported reduction of **9** with LiAlH₄ afforded 9b-*epi*-product **12** in 65% and 82% yields, respectively.

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(16) The Banwell^{8a} and Carreira^{9e} groups independently reported that a stereoselective hydrogenation of pyrrole derivative **13** in the presence of PtO_2 under hydrogen in acetic acid followed by Dess-Martin oxidation provided **2** in 28% and 25% yields, respectively.



(17) The synthesis of **1a** from **2** has been reported by the research groups of Aubé (51% yield),⁶ Gnecco (55% yield),^{9b} Zard (55% yield),^{8b,c} Coldham (42% yield),^{8e,f} Ishikawa and Saito (51% yield),^{9c} Canesi (43% yield),^{8g,h} Cho (45% yield),⁸ⁱ Pandey (50% yield),^{9d} Jiang (53% yield),⁷ and Carreira (42 % yield).^{9e}

(18) Regioisomer during Fischer indolization was also obtained in 11% yield. See the Supporting Information for details.

(19) Stork mentioned that the stereochemistries at C-9a and C-9b of tricyclic intermediates such as 2 and 12 were not operationally significant in the indolization step for synthesizing aspidopermine. See ref. 3.

(20) The Banwell^{8a} and Yang^{8j} groups examined the conversion of **12** into aspidospermidine (**1a**) using Fischer indolization and subsequent reduction but were unsuccessful.

(21) As the shortest-step synthesis of Stork's intermediate **2** to date, Coldham and co-workers reported the synthesis of racemic **2** in six steps from the known 1,5-dibromopentan-3-one, which was prepared in two steps from ethyl 3-bromopropionate. See ref. 8e,f and Denmark, S. E.; Marcin, L. R. Asymmetric Construction of a Quaternary Carbon Center by Tandem [4 + 2]/[3 + 2] Cycloaddition of a Nitroalkene. The Total Synthesis of (–)-Mesembrine. *J. Org. Chem.* **1997**, *62*, 1675– 1686. (22) (a) Prabhakaran, P. C.; Gould, S. J.; Orr, G. R.; Coward, J. K. Synthesis of Chirally Deuteriated Phthalimidopropanols and Evaluation of Their Absolute Stereochemistry. *J. Am. Chem. Soc.* **1988**, *110*, 5779–5784. (b) Ravnsbæk, J. B.; Jacobsen, M. F.; Rosen, C. B.; Voigt, R. N. V.; Gothelf, K. V. DNA-Programmed Glaser–Eglinton Reactions for the Synthesis of Conjugated Molecular Wires. *Angew. Chem. Int. Ed.* **2011**, *50*, 10851–10854.