THREO- AND ERYTHRO-β-HYDROXY-L-ASPARAGINES A. Singerman and Y. Liwschitz

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The recent findings of Campbell et al. (1) regarding the action of L-asparaginase on various asparagine-requiring tumours have prompted the syntheses and investigation of analogues of this amino acid (2).

We wish to report the preparation of the β -amides of threo- and erythro- β -hydroxy-L-aspartic acids by ammonolysis of β -benzyl threo- β -hydroxy-L-aspartate (3) and β -methyl erythro- β -hydroxy-L-aspartate respectively. The latter ester was prepared by direct esterification of erythro- β -hydroxy-L-aspartic acid obtained by resolution of its racemic N-benzyl derivative by means of L-histidine and subsequent hydrogenolysis (4).

The amides differ in their stability in acidic medium. The one belonging to the threo- series when left for several hours in N HCl is partly hydrolysed to threo- β -hydroxy-L-aspartic acid, detected on chromatograms, which caused a sharp drop in optical rotation. Erythro- β -hydroxy-L-asparagine, however, was stable under the same conditions.

The " β -hydroxy asparagine" isolated from normal human urine (5) could not be identified with either of the isomers synthesised by us, having different properties (R_p values and IR spectrum).

Three- β -hydroxy-L-asparagine - β -Benzyl three- β -hydroxy-L-aspartate (3) (1 g) was disselved in 25% aqueous ammonia (10 ml). After 7 days at room temperature the solution was evaporated to dryness in vacue. The solid residue was disselved in hot water and the pH adjusted to 5 by addition of 6k HCl. The substance precipitated and was recrystallised from water (0.6 g, yield quantitative); m.p. 278° (decomp.); $[\alpha]_{D}^{22}$ -30.6° (c 2.45 in N HCl) (Found: C, 32.2; H, 5.9; N, 18.9; N (Van Slyke), 9.6. $C_4H_8N_2O_4$ requires: C, 32.4; H, 5.4; N, 18.9; N (Van Slyke), 9.4%).

 β -Methyl erythro- β -hydroxy-L-aspartate - To a suspension of erythro- β -hydroxy-L-aspartic acid (4) (5 g) in anhydrous methanol (50 ml) was added concentrated HCl (6 ml). The mixture was heated under reflux for 3 hrs. The

solvents were removed in vacuo, the residue was dissolved in ethanol and the pH adjusted to 8 by addition of pyridine. The resulting precipitate was recrystallised from 50% aqueous ethanol. (3.5 g, 65%); m.p. 229° (decomp.); $[\alpha]_D^{22}$ +65.5° (c, 1.8 in N HCl) (Found: C, 36.8; H, 5.8; N, 8.5; N (Van Slyke), 8.7. $C_5H_9NO_5$ requires: C, 36.8; H, 5.5; N, 8.6; N (Van Slyke), 8.6%).

Erythro- β -hydroxy-L-asparagine was prepared in an analogous manner to the three isomer by ammonolysis of the above ester; m.p. 260° (decomp.); $[\alpha]_D^{22}$ +63.4° (c 1.9 in N HC1) (Found: C, 32.4; H, 5.9; N, 18.9; N (Van Slyke), 9.2. $C_4H_8N_2O_4$ requires: C, 32.4; H, 5.4; N, 18.9; N (Van Slyke), 9.4%).

Paper chromatography in 80/2 phenol-water (light-brown colour with ninhydrin) gave R_f values 0.18 and 0.13 for threo- and erythro- β -hydroxy-L-asparagine respectively. On TLC (silica gel) in n-butanol-acetic acid-pyridine-water (2:1:1:4) they were 0.39 and 0.23 respectively.

REFERENCES

- 1) H.A. Campbell, L.T. Mashburn, E.A. Boyse and L.J. Old, <u>Biochemistry</u>, <u>6</u>, 721 (1967).
- 2) E. Falco and G.B. Brown, <u>J. Medicinal Chem</u>., <u>11</u>, 142 (1968).
-) Y. Liwschitz, A. Singerman and S. Sokoloff, <u>J. Chem. Soc. (C)</u>, 1968 in press.
- 4) Y. Liwschitz, A. Singerman and Y. Wiesel, Israel J. Chem., 1968 in press.
- 5) F. Tominaga, C. Hiwaki, T. Maekawa and H. Yoshida, <u>J. Biochem. (Japan)</u> <u>53</u>, 227 (1963).