

# Unusual Stereospecificity in the Hydrogenation of an Isopropenyl Function with Wilkinson's Catalyst; A Route to Chiral Methyl Valine

By DAVID H. G. CROUT,\* MAX LUTSTORF, and PHILIP J. MORGAN

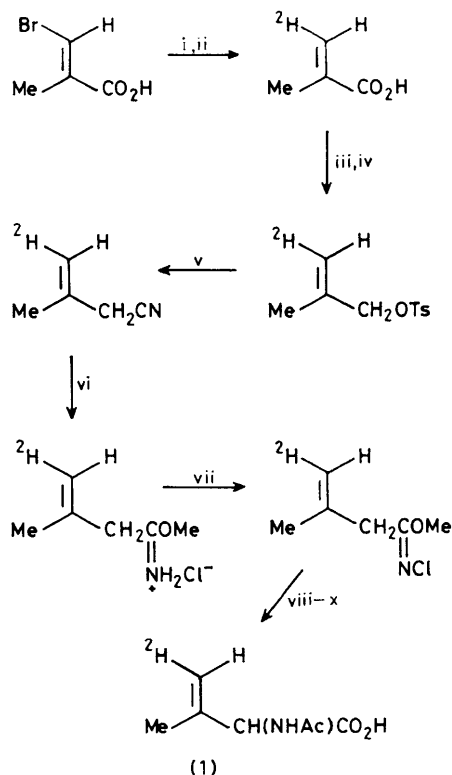
(Department of Chemistry, University of Exeter, Exeter, EX4 4QD)

and ROBERT M. ADLINGTON, JACK E. BALDWIN,\* and MICHAEL J. CRIMMIN

(Dyson Perrins Laboratory, University of Oxford, South Parks Road, Oxford, OX1 3QY)

**Summary** Catalytic hydrogenation with  $\text{H}^3\text{H}$  in the presence of Wilkinson's catalyst of  $(2RS)-(E)-[4-^2\text{H}]-2$ -acetyl-amino-3-methylbut-3-enoic acid gave a mixture of  $(2SR, 3SR, 4SR)-[4-^3\text{H}^2\text{H}]-N$ -acetylvaline [(8) and (9)] and  $2SR, 3RS, 4SR-[4-^3\text{H}^2\text{H}]-N$ -acetylvaline [(6) and (7)] in the ratio 19:1 by  $^3\text{H}$  n.m.r. spectroscopy of the derived valines; a similar reduction with  $^2\text{H}_2$  of 1,5-(*Z*)-4,7-diaza-7-[1-4-(nitrobenzyloxycarbonyl)-2-methylprop-2-enyl]-3-phenoxymethyl-2-thiabicyclo[3.2.0]hept-3-en-6-one gave a mixture of dideuterio-isomers (11) and (12), in the ratio 3:7 by  $^1\text{H}$  n.m.r. spectroscopy of the derived valines.

DURING investigations on valine biosynthesis a sample of valine with stereospecifically labelled methyl groups ('chiral methyl valine') was required.<sup>†</sup> The route chosen (Scheme 2) was realised by catalytic reduction of *N*-acetylisodehydrovaline<sup>1</sup> (1) (Scheme 1) by an equilibrated mixture of hydro-



Ts = toluene-4-sulphonyl

SCHEME 1. *Reagents*. i, NaH; ii, Na-Hg/ $^2\text{H}_2\text{O}$ ; iii,  $\text{LiAlH}_4$ ; iv,  $\text{TsCl}$ ,  $\text{Et}_3\text{N}$ ; v, KCN, 18-crown-6; vi,  $\text{MeOH-HCl}$ ; vii,  $\text{NaOCl}$ ; viii, NaOMe, MeOH; ix, dil.  $\text{HCl}$ ; x,  $\text{Ac}_2\text{O}$ ,  $\text{K}_2\text{CO}_3$ .

<sup>†</sup> The experiments on tritiation of dehydrovaline were carried out at Exeter and those on the penicillin-derived thiazoline at Oxford.

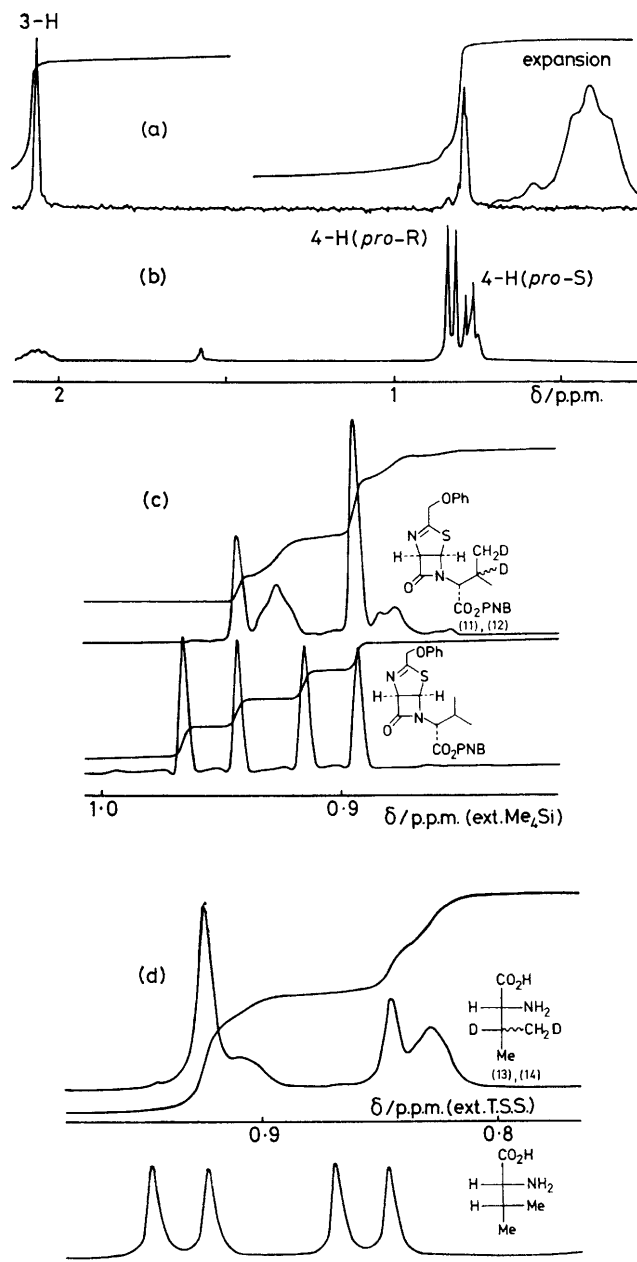
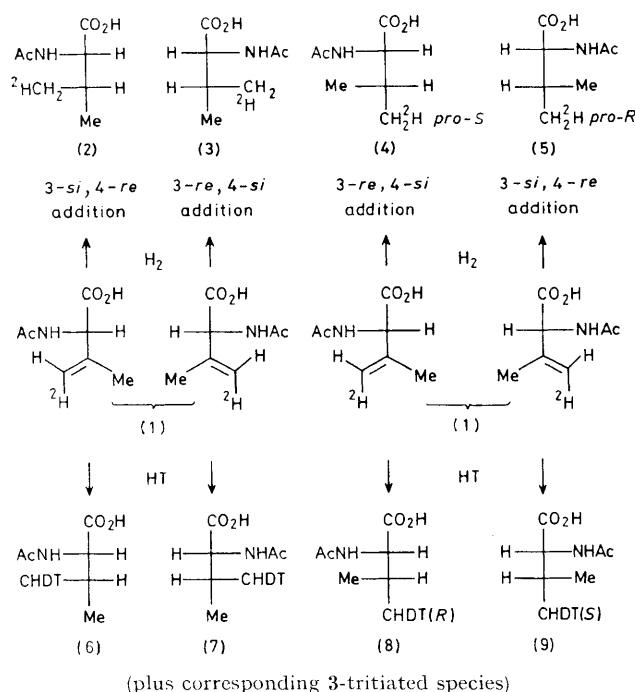


FIGURE. (a)  $^3\text{H}$  N.m.r. spectrum; (b)  $^1\text{H}$  n.m.r. spectrum of the valine produced by hydrogenation of  $(2RS)-(E)-[4-^2\text{H}]-2$ -acetyl-amino-3-methylbut-3-enoic acid [(1)] with an equilibrated mixture of  $\text{H}_2$  and  $^3\text{H}_2$  (7:1) in the presence of Wilkinson's catalyst. TSS = sodium 3-[2,2,3,3- $^3\text{H}_4$ ]trimethylsilylpropionate.

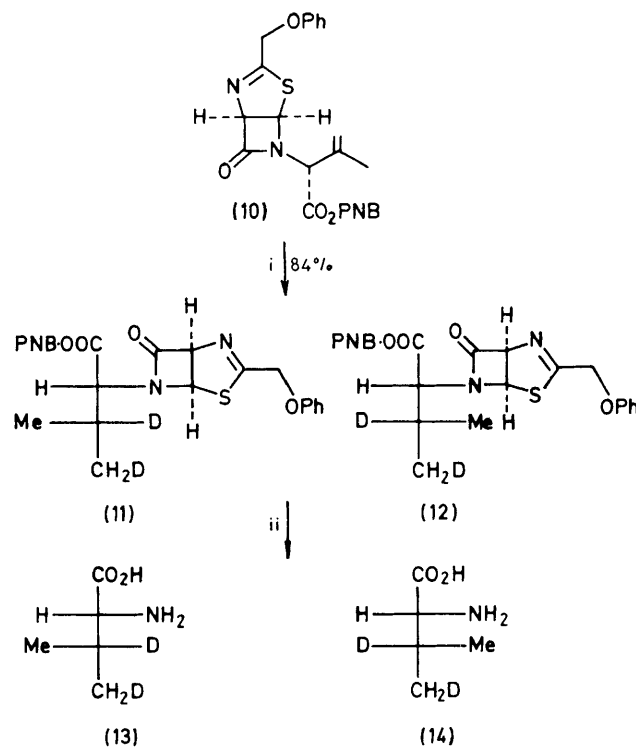
gen and tritium (7:1) in the presence of Wilkinson's catalyst, followed by hydrolysis, giving a mixture of labelled isomers of valine. The  $^1\text{H}$  n.m.r. (300 MHz) spectrum (Figure, b) showed signals of unequal intensity for the diastereotopic methyl groups, from which it was concluded that the diastereomeric pairs [(2) and (3)] and [(4) and (5)] were formed in unequal amounts. Since the higher field methyl doublet in the  $^1\text{H}$  n.m.r. spectrum of *S*-valine was assigned to the *pro-S* methyl group<sup>2</sup> [cf. (4)], then the observed intensity ratio (3:2) of low to high field methyl doublets indicated almost exclusive formation of the pair of diastereoisomers [(4) and (5)] (Scheme 2). This conclusion was



SCHEME 2

supported by the noise-decoupled  $^3\text{H}$  n.m.r. spectrum of the racemic valine so obtained which showed signals due to *pro-S* and *pro-R* methyl groups of the *S*-component [(6) and (8)] [which correspond respectively to the *pro-R* and *pro-S* methyl groups of the *R*-component [as (7) and (9)]] with relative intensity 19:1, and the expected  $^3\text{H}^2\text{H}$  coupling (Figure, a).

Another example of this selectivity was found on catalytic deuteration (Wilkinson's catalyst, 25 °C, 48 h) of the thiazoline azetidinone (10)<sup>3</sup> to the two dideuterio-isomers [(11) and (12), 30:70 respectively, 84%], (Figure, c). Their stereochemistry was determined by hydrolysis to the valines



PNB = 4-nitrobenzyl

SCHEME 3. Reagents. i,  $\text{H}_2$ ,  $(\text{PPh}_3)_3\text{RhCl}$ , PhH, 48 h; ii, 6N-HCl, reflux, 24 h.

(13) and (14), (Scheme 3), the major component of which, (14), corresponds in its  $^1\text{H}$  spectra (Figure, d) to the 2*R*, 3*R*-isomer.<sup>2</sup>

These results show that catalytic reduction by Wilkinson's catalyst occurs with a preference for 3-*re*, 4-*si* attack on the *S*-component and 3-*si*, 4-*re* attack on the *R*-component, for the dehydrovaline (1) and with 3-*re* attack for the 2*R*-thiazoline azetidinone (10). Whether this is a general result for  $\beta\gamma$ -unsaturated amino-acids and other  $\beta\gamma$ -unsaturated systems remains to be seen. However, regardless of the stereoselectivity and also regardless of the C-2 configuration of compound (1), the isotopically stereoisomeric chiral methyl valines produced from the *N*-acetyldehydrovaline (1) have methyl groups with the *R*-configuration in the *pro-S* position and the *S*-configuration in the *pro-R* position.<sup>†</sup>

We thank Dr. L. Field and Lady Richards for assistance with the n.m.r. experiments, and the S.R.C. for financial support (to M. L., P. J. M., M. J. C., R. M. A.).

(Received, 12th June 1981; Com. 690.)

<sup>†</sup> The isomeric valines with opposite configurations were synthesised from (*E*)-3-bromo-2-methylprop-2-enoic acid by a sequence initially involving metallation (BuLi) and deuteration ( $\text{D}_2\text{O}$ ) to (*E*)-3- $[\text{D}]$ -3-bromo-2-methylprop-2-enoic acid, and subsequent processes as before, Scheme 1.

<sup>1</sup> D. H. G. Crout and J. A. Corkill, *Tetrahedron Lett.*, 1977, 4355. The Neber rearrangement employed in this route has precedent, cf., H. E. Baumgarten, J. E. Dirks, J. M. Petersen, and R. L. Zey, *J. Org. Chem.*, 1966, 31, 3708; W. H. Graham, *Tetrahedron Lett.*, 1969, 2223; Y. Nogami, Y. Kawazoe, and T. Taguchi, *J. Pharm. Soc. Jpn.*, 1973, 93, 1058.

<sup>2</sup> R. K. Hill, S. Yan, and S. M. Arfin, *J. Am. Chem. Soc.*, 1973, 95, 7857; D. J. Aberhart and L. J. Lin, *J. Chem. Soc., Perkin Trans.*, 1974, 2320.

<sup>3</sup> R. D. G. Cooper and F. L. José, *J. Am. Chem. Soc.*, 1970, 92, 2575.