



Corrigendum

Corrigendum to “TMSCl-promoted selective oxidation of sulfides to sulfoxides with hydrogen peroxide” [Tetrahedron Lett. 51 (2010) 6939–6941]

Kiumars Bahrami ^{a,*}, Mohammad M. Khodaei ^{a,*}, Behrooz H. Yousefi ^b, Mehdi Sheikh Arabi ^a

^a Department of Chemistry, Razi University, Kermanshah 67149-67346, Iran

^b Department of Nuclear Medicine, Klinikum rechts der Isar, Technische Universität München, Ismaninger Straße 22, D-81675 Munich, Germany

The authors regret that when this Letter was published the Sulfide and Sulfoxide entries for entry 4 were displayed incorrectly. The correct Table 2 is now reprinted below:

Table 2

Selective oxidation of sulfides to sulfoxides using the H₂O₂-TMSCl system in CH₃CN^a

Entry	Sulfide	Sulfoxide	Yield ^b (%) / time (min)	Mp (°C)
1			98 (2)	120–122 ^{16a}
2			99 (20)	163–164 ^{16b}
3			94 (15)	158 ^{16c}
4			95 (2)	132 ^{16d}
5			96 (3)	68 ^{16d}
6 ^c			94 (10)	199–201 ^{16e}
7			96 (10)	172 ^{15e}

DOI of original article: 10.1016/j.tetlet.2010.10.171

* Corresponding authors. Tel.: +98 831 4274559; fax: +98 831 4274559 (K.B.).

E-mail addresses: Kbahrami2@hotmail.com (K. Bahrami), mmkhoda@razi.ac.ir (M.M. Khodaei).

Table 2 (continued)

Entry	Sulfide	Sulfoxide	Yield ^b (%) / time (min)	Mp (°C)
8			94 (3)	Oil ^{16f}
9			93 (35)	Oil ^{16g}
10			98 (3)	151–153 ^{16h}
11			98 (10)	39–40 ^{16d}
12			92 (3)	29–30 ^{15c}
13			95 (3)	Oil ¹⁶ⁱ

^a The purified products were characterized by mp and ¹H and ¹³C NMR spectroscopy.^b Yield refers to pure isolated product.^c 1,4-Dioxane was used as the solvent.