

Regio- and Stereoselective Cross-Coupling of *tert*-Propargyl Alcohols with Bis(trimethylsilyl)acetylene and Its Utilization in Constructing a Fluorescent Donor–Acceptor System[†]

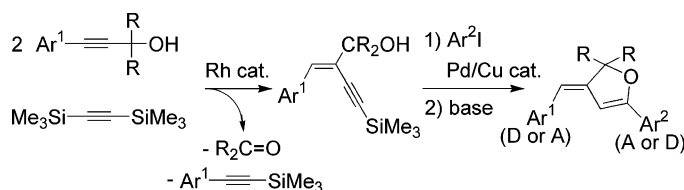
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ABSTRACT



1,1-Disubstituted 3-aryl-2-propyn-1-ols undergo regio- and stereoselective cross-coupling on treatment with bis(trimethylsilyl)acetylene in the presence of a rhodium catalyst via cleavage of C(sp)–C(sp³) and C(sp)–Si bonds to produce the corresponding 2-hydroxymethyl-(*E*)-enynes. The subsequent desilylative Sonogashira coupling followed by base-promoted cyclization affords fluorescent dihydrofuran derivatives.

The catalytic synthesis of π -conjugated enyne compounds has attracted considerable interest, due to the presence of the skeleton in natural products and their utility as versatile building blocks in organic synthesis.¹ While a variety of enynes can now be prepared by the dimerization of alkynes, the selective cross-coupling of two different alkynes is, in general, still difficult owing to the fact that the formation of regio- and stereoisomers as well as homodimers is possible, and thus a major challenge.^{2,3} Among the rare, leading examples of the cross-coupling is the palladium-catalyzed

reaction of terminal alkynes with internal ones having an electron-withdrawing group.² As in this instance, a key alkynylmetal intermediate generated by C(sp)–H bond activation of a terminal alkyne is usually involved in such a reaction.

Meanwhile, various unique and useful catalytic processes involving C–C bond cleavage via β -carbon elimination in metal alcoholate intermediates have recently been developed.⁴ In the course of our work on the transformations,⁵ we found that in the presence of a rhodium catalyst, γ -arylated *tert*-propargyl alcohols, i.e., ketone-masked aryl acetylenes **1**,

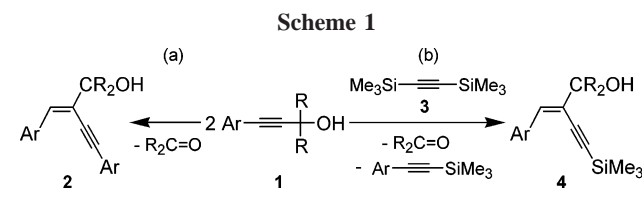
[†] This paper is dedicated to the heartfelt memory of the late Professor Yoshihiko Ito of Kyoto and Doshisha Universities.

(1) (a) Trost, B. M. *Angew. Chem., Int. Ed.* **1995**, *34*, 259. (b) Tsuji, J. *Transition Metal Reagents and Catalysts*; Wiley: Chichester, UK, 2000.

(2) (a) Trost, B. M.; Chan, C.; Ruhter, G. *J. Am. Chem. Soc.* **1987**, *109*, 3486. (b) Trost, B. M.; Harms, A. E. *Tetrahedron Lett.* **1996**, *37*, 3971. (c) Trost, B. M.; Sorum, M. T.; Chan, C.; Harms, A. E.; Ruhter, G. *J. Am. Chem. Soc.* **1997**, *119*, 698.

(3) (a) Akita, M.; Yasuda, H.; Nakamura, A. *Bull. Chem. Soc. Jpn.* **1984**, *57*, 480. (b) Yi, C. S.; Liu, N. *Organometallics* **1998**, *17*, 3158. (c) Lucking, U.; Pfaltz, A. *Synlett* **2000**, 1261. (d) Wang, J.; Kapon, M.; Berthet, J. C.; Ephritikhine, M.; Eisen, M. S. *Inorg. Chim. Acta* **2002**, *334*, 183. (e) Chen, L.; Li, C.-J. *Tetrahedron Lett.* **2004**, *45*, 2771. (f) Hirabayashi, T.; Sakaguchi, S.; Ishii, Y. *Adv. Synth. Catal.* **2005**, *347*, 872. (g) Katayama, H.; Yari, H.; Tanaka, M.; Ozawa, F. *Chem. Commun.* **2005**, 4336.

efficiently undergo homocoupling with liberation of a ketone molecule through C(sp)–C(sp³) bond cleavage to regio- and stereoselectively produce 1,4-diaryl-2-hydroxymethyl-(*E*)-enynes **2** (Scheme 1, a).^{5e} Notably, the products readily



cyclize in the presence of a base to form dihydrofuran derivatives, some of which exhibit strong fluorescence in the solid state. During the examination of cross-coupling using **1**, we have observed that the propargyl alcohols selectively react with bis(trimethylsilyl)acetylene (**3**) via activation of one of the C(sp)–Si bonds to afford the corresponding 1-aryl-4-trimethylsilyl-(*E*)-enynes **4** (Scheme 1, b). The silyl function has also been subjected to further structural elaboration to lead to a donor–acceptor (D–A) π -conjugated system on the dihydrofuran scaffold.

When a mixture of 1,1,3-triphenyl-2-propyn-1-ol (**1ap**) (0.5 mmol) was treated with **3** (3 mmol) in the presence of [(cod)Rh(OH)]₂/dppb (4 mol %) in refluxing toluene for 2 h, 2-[(*E*)-benzylidene]-4-trimethylsilyl-1,1-diphenyl-3-butyn-1-ol (**4ap**) was produced along with the homocoupling product **2ap** in 64% and 16% yields (calculated as 2[product]/[**1a**], see below), respectively (entry 1 in Table 1). Addition of **1ap** in a slow manner through a cannula to keep the concentration of **1ap** low could successfully suppress the formation of **2ap** to allow the almost exclusive formation of **4ap** (entry 2). Analysis of the reaction mixture by GC-MS confirmed the formation of 1-phenyl-2-(trimethylsilyl)-acetylene and benzophenone in quantitative yields (0.5 equiv) as the byproducts, which may provide important mechanistic information. Decreasing the amount of **3** to 1.0 mmol still gave 76% of **4ap** (entry 3). The reaction was found to be sensitive to the variation of ligand. The use of dppp or dppe in place of dppb reduced the yield of **4ap** and induced the formation of its (*Z*)-isomer in a small, but considerable amount (entries 4 and 5). The reaction without any phosphine ligand was sluggish (entries 7 and 8). The use of [(cod)-

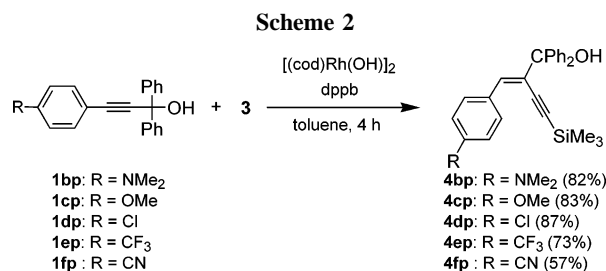
Table 1. Rhodium-Catalyzed Reaction of *tert*-Propargyl Alcohols with Bis(trimethylsilyl)acetylene^a

entry	X	R	ligand	time (h)	4 , % yield ^b	2 , % yield ^b
1	OH	Ph	dppb	2	64 ^e	16
2 ^c	OH	Ph	dppb	4	99 (98) ^e	trace
3 ^{c,f}	OH	Ph	dppb	4	76 ^e	16
4 ^c	OH	Ph	dppp	4	76 (71) (96:4) ^g	trace
5 ^c	OH	Ph	dppe	9	64 (57) (89:11) ^g	3
6 ^c	OH	Ph	PPh ₃ ^d	8	43 (37) ^e	8
7 ^c	OH	Ph		4	32	20
8 ^c	Cl	Ph		8	6	6
9 ^c	Cl	Ph	dppb	2	60	trace
10	OH	Me	dppb	4	62 ^e	31
11 ^c	OH	Me	dppb	4	89 (88) ^e	trace

^a Reaction conditions: [Rh]:[ligand]:[**1**]:[**3**] = 0.02:0.02:0.5:3.0 (in mmol), in refluxing toluene (4 mL) under N₂. ^b GC yield based on the half amount of **1** used. Value in parentheses indicates isolated yield. ^c Toluene solution of **1** (2 mL, 0.25 mM) was added over 2 h. ^d PPh₃ (0.04 mmol) was used. ^e Exclusively (*E*)-isomer. ^f [**3**] = 1.0 mmol. ^g *E/Z* ratio.

RhCl]₂ in place of the hydroxyl complex together with dppb was less effective (entry 9). The reaction of 2-methyl-4-phenyl-3-butyn-2-ol (**1am**) proceeded similarly (entries 10 and 11).

The cross-coupling reactions of various 3-(4-substituted phenyl)-1,1-diphenyl-2-propyn-1-ols **1bp–1fp** with the disilylacetylene **3** in the double scale of entry 2 in Table 1 gave the corresponding products **4bp–4fp** with good isolated yields irrespective of the nature of the 4-substituents (Scheme 2).

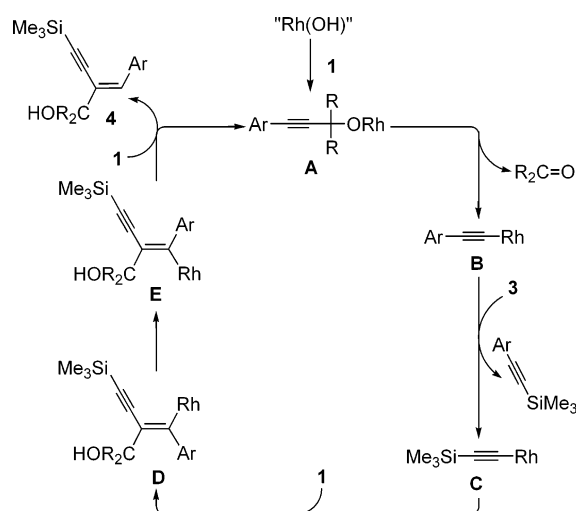


A plausible mechanism for the formation of the enyne **4** is illustrated in Scheme 3, in which neutral ligands are omitted. The first step involves the reaction of **1** with hydroxyrhodium(I) species to form rhodium alcoholate **A** and the successive β -carbon elimination with liberation of benzophenone or acetone gives arylalkynylrhodium **B**. Then, the alkynyl exchange between **B** and **3** takes place to form

(4) (a) Satoh, T.; Miura, M. *Top. Organomet. Chem.* **2005**, *14*, 1. (b) Murakami, M.; Ito, Y. *Top. Organomet. Chem.* **1999**, *3*, 97. (c) Rybtchinski, B.; Milstein, D. *Angew. Chem., Int. Ed.* **1999**, *38*, 870. (d) Mitsudo, T.; Kondo, T. *Synlett* **2001**, 309. (e) Perthuisot, C.; Edelbach, B. L.; Zubris, D. L.; Simhai, N.; Iverson, C. N.; Müller, C.; Satoh, T.; Jones, W. D. *J. Mol. Catal. A: Chem.* **2002**, *189*, 157. (f) Jun, C.-H.; Moon, C. W.; Lee, D.-Y. *Chem. Eur. J.* **2002**, *8*, 2423. (g) Catellani, M. *Synlett* **2003**, 298. (h) Nishimura, T.; Araki, H.; Maeda, Y.; Uemura, S. *Org. Lett.* **2003**, *5*, 2997. (i) Nishimura, T.; Uemura, S. *Synlett* **2004**, 201. (j) Murakami, M.; Makino, M.; Ashida, S.; Matsuda, T. *Bull. Chem. Soc. Jpn.* **2006**, *79*, 1315.

(5) (a) Terao, Y.; Wakui, H.; Satoh, T.; Miura, M.; Nomura, M. *J. Am. Chem. Soc.* **2001**, *123*, 10407. (b) Terao, Y.; Wakui, H.; Nomoto, N.; Satoh, T.; Miura, M.; Nomura, M. *J. Org. Chem.* **2003**, *68*, 5236. (c) Terao, Y.; Nomoto, N.; Satoh, T.; Miura, M.; Nomura, M. *J. Org. Chem.* **2004**, *69*, 6942. (d) Wakui, H.; Kawasaki, S.; Satoh, T.; Miura, M.; Nomura, M. *J. Am. Chem. Soc.* **2004**, *126*, 8658. (e) Funayama, A.; Satoh, T.; Miura, M. *J. Am. Chem. Soc.* **2005**, *127*, 15354.

Scheme 3

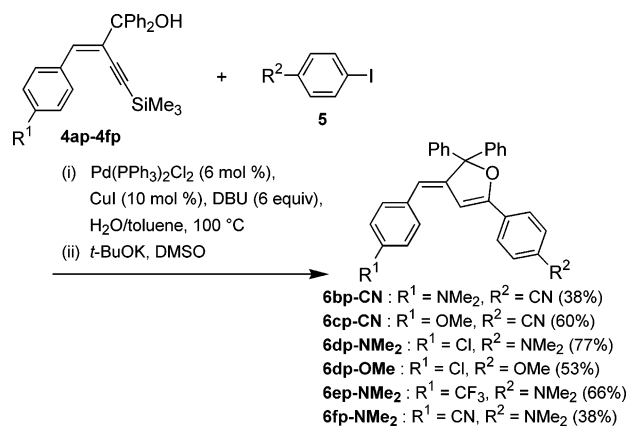


silylalkynylrhodium **C** via activation of the C–Si bond,⁶ presumably by an oxidative addition–reductive elimination sequence.⁷ The successive regioselective insertion of another molecule of **1** to the rhodium–carbon bond affords vinylrhodium **D**. Enyne **4** is formed after the geometrical isomerization of **D** to **E** and protonolysis by **1** with regeneration of **A** as in the case of the homocoupling of **1**. The geometrical isomerization may occur via a zwitterionic form.⁸ The interaction of the oxygen with the metal may intervene to stabilize **E**.

The synthetic utility of the cross-coupling products may be demonstrated by the transformation of the silyl group to various aromatic substituents and the subsequent cyclization to produce dihydrofuran derivatives that show solid-state luminescence.^{5e,9} Since π -conjugated D–A molecules are often useful as flexible systems with respect to emission range,¹⁰ the dihydrofuran scaffold may allow observation of the change of optical properties depending on the different substituents of the two aromatic rings.

As expected, the arylation of **3** with aryl iodides by a reported method for desilylative Sonogashira coupling,¹¹ followed by treatment with *t*-BuOK gave the corresponding dihydrofurans **6** in fair to good yields, although the conditions were not optimized (Scheme 4). They showed solid-state

Scheme 4



fluorescence in a range of 492–589 nm (see Figure S2 in the Supporting Information). Notably, **6cp-CN** showed a relatively strong emission compared to a typical green emitter, tris(8-hydroxyquinolino)aluminum (Alq₃), by a factor of 1.6. Dihydrofurans **6bp-CN** and **6cp-CN**, both of which possess an electron-donating group at R¹ and an electron-withdrawing group at R², showed strong fluorescence in solution (Φ = 0.76 and 0.53 in dioxane, respectively). This contrasts with the fact that each of the other compounds **6dp-NMe₂**, **6dp-OMe**, **6ep-NMe₂**, and **6fp-NMe₂** as well as the homocoupling product **2ap** shows only a weak emission in solution. Furthermore, **6bp-CN** showed a large positive solvatochromism of fluorescence (λ_{max} = 508 nm in hexane; 539 nm in dioxane; 594 nm in MeCN), suggesting a strong intramolecular charge-transfer character in the excited state, while the chromism of **6cp-CN** was very small (see Figure S3 in the Supporting Information). These results indicate that the electronic nature of R¹ and R² strongly affects not only the fluorescent wavelength but also the efficiency in both solid and solution.

In summary, we have developed a new, selective alkyne cross-coupling reaction that proceeds via cleavage of C–C and C–Si bonds. The reaction has enabled the construction of a series of π -conjugated D–A systems involving a dihydrofuran skeleton that show intriguing optical properties.

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Supporting Information Available: Standard experimental procedure and characterization data for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(6) Transmetalation between [RhCl(CO)₂]₂ and acyl- and vinylsilanes: Yamane, M.; Uera, K.; Narasaka, K. *Bull. Chem. Soc. Jpn.* **2005**, *78*, 477.

(7) (a) Fangou, K.; Lautens, M. *Chem. Rev.* **2003**, *103*, 169. (b) Müller, J.; Tschampel, M.; Pickardt, J. *J. Organomet. Chem.* **1988**, *355*, 513.

(8) Hydrosilylation reactions by rhodium and iridium complexes involving *E/Z* isomerization: (a) Jun, C.-H.; Crabtree, R. H. *J. Organomet. Chem.* **1993**, *447*, 177. (b) Ojima, I.; Clos, N.; Donovan, R. J.; Ingallina, P. *Organometallics* **1990**, *9*, 3127.

(9) Filler, R.; Piasek, E. J.; Mark, L. H. *J. Org. Chem.* **1961**, *26*, 2659.

(10) Grabowski, Z. R.; Rotkiewicz, K.; Rettig, W. *Chem. Rev.* **2003**, *103*, 3899.

(11) Matthew, J. M.; Lucas, C. K.; Julia, B. B.; Tendai, L. G.; Kami, L. H.; Ronald, G. B.; Christophor, J. M.; Paul, A. G. *Org. Lett.* **2002**, *4*, 3199.