

Tetrakis(2,2,3,3-tetrafluoropropoxy)methylphosphorane (IIIb) was obtained in 72% yield, $\delta P = -48$ ppm. PMR spectrum in CD_3CN : 1.90 d (PCH_3 , 3H, $^2J_{PCH} = 16$; s in the $^1H-(^{31}P)$ DNMR spectrum, $\omega = 40477.621$ Hz), 4.46 m ($POCH_2$, 8H, $^3J_{FH} = 13.5$, $^3J_{PH} = 6$, $^4J_{FH} = 1$; br.t in the $^1H-(^{31}P)$ DNMR spectrum), 6.32 t.t (CHF_2 , 4H, $^2J_{FH} = 52.5$, $^3J_{FH} = 4.7$). Found: C, 26.79; H, 2.57%. Calculated for $C_{13}H_{15}F_{16}O_4P$: C, 26.85; H, 2.65%.

Tetrakis(1,1,5-trihydroperfluoropentyloxy)methylphosphorane (IIIc) was obtained in 68% yield, bp 130-135°C (0.04 mm), $\delta P = -49$ ppm. PMR spectrum in 1:1 $CCl_4 + 50\%$ CD_3CN : 1.80 d ($P-CH_3$, 3H, $^2J_{PCH} = 17$), 4.40 br.t.d ($POCH_2$, 8H, $^3J_{FH} = 13$, $^3J_{PH} = 6$), 6.10 t.t (CHF_2 , 4H, $^2J_{PCH} = 50.5$, $^3J_{FH} = 5$). Found: C, 26.09; H, 1.77%. Calculated for $C_{21}H_{15}F_{32}O_4P$: C, 25.97; H, 1.54%.

Bis(2,2,2-trifluoroethoxy)methylphosphonate (IV) was obtained from 0.011 mole salt (Ia) and 0.012 mole 2-propanol in 89% yield, bp 75-77°C (10 mm), $\delta P = 33$ ppm. PMR spectrum in CCl_4 : 1.52 d ($P-CH_3$, 3H, $^2J_{PCH} = 18.5$; s in the $^1H-(^{31}P)$ DNMR spectrum, $\omega = 40480.908$ Hz), 4.36 two d.q ($POCH_2$, 4H, $^3J_{PH} \approx ^3J_{FH} = 9$; two q in the $^1H-(^{31}P)$ DNMR spectrum). Found: C, 23.12; H, 2.77%. Calculated for $C_5H_7F_6O_3P$: C, 23.07; H, 2.69%.

The 1H and $^1H-(^{31}P)$ NMR spectra were taken on a Varian HA 100D spectrometer at 100 MHz for (Ia), (Ic), (IIIa), (IIIb), and (IV) and on a Tesla BS 467A spectrometer at 60 MHz for (Ib) and (IIIc). The $^{31}P-(^1H)$ NMR spectra were taken on an RYa-2303 spectrometer at 8 MHz by E. I. Gol'dfarb, to whom the authors are sincerely grateful.

LITERATURE CITED

1. R. Hudson, Topics in Phosphorus Chemistry, Vol. 11, Interscience Publ., New York-London-Sydney-Toronto (1983), p. 339.
2. S. E. Fry and N. J. Pienta, J. Org. Chem., 19, No. 25, 4877 (1984).
3. E. S. Lewis and K. S. Colle, J. Org. Chem., 43, No. 5, 571 (1978).
4. E. S. Lewis and D. Hamp, J. Org. Chem., 48, No. 12, 2025 (1983).

REACTION OF PHENYLSULFENYL CHLORIDE WITH C-(N,N-DIMETHYLAMINO)- METHYLENE-P-PHENYLPHOSPHINE

A. S. Ionkin, N. V. Nikolaeva,
and B. A. Arbuzov

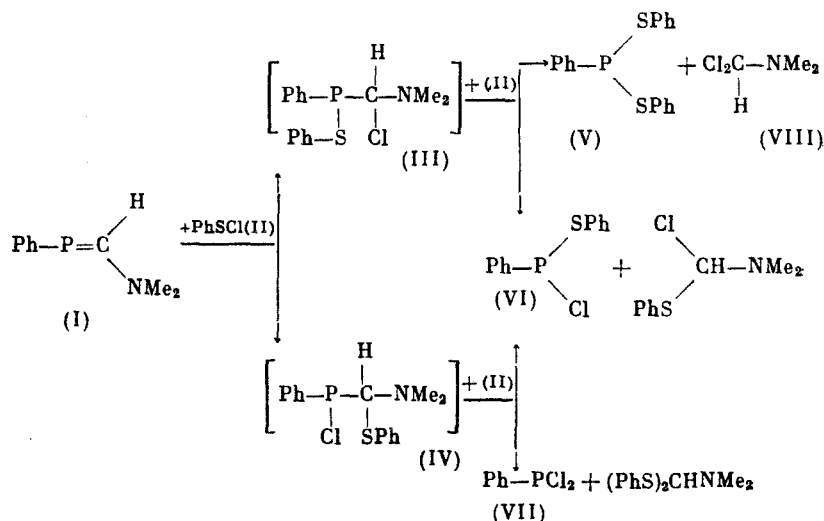
UDC 542.91:547.53.024+546.
223.2'13:547.558.1'233

PhSCl adds regiospecifically to the $P=C$ double bond in C-(N,N-dimethylamino)-methylene-P-phenylphosphine to give $PhP(SPh)_2$ and Cl_2CHNMe_2 .

Regitz [1] and Schmidtpeter [2] have shown that sulfenyl chlorides add at the $P=C$ bond of 1,2,3-diazaphospholes to give unstable addition products with sulfur-carbon and phosphorus-chlorine bonds. The regiospecificity of the addition of sulfenyl chlorides to these dicoordinated phosphorus compounds corresponds to the polarization of the $P=C$ bond arising due to the difference in the electronegativities of the component atoms. C-dialkyl-aminophosphaalkenes have opposite polarization of the $P=C$ double bond [3], which, in turn, may lead to reverse addition in the reaction of sulfenyl chlorides with such phosphoalkenes.

In order to check this hypothesis, we carried out the reaction of C-(N,N-dimethylamino)methylene-P-phenylphosphine (I) with phenylsulfenyl chloride (II). Completion of this reaction requires a (I):(II) mole ratio equal to 1:2.

A. E. Arbuzov Institute of Organic and Physical Chemistry, Kazan Branch, Academy of Sciences of the USSR. Translated from Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya, No. 4, pp. 948-949, April, 1991. Original article submitted June 11, 1990.



We have found that the reaction does not stop at the addition of one mole of (II) and proceeds with complete cleavage of the P-C bond in (I). The major phosphorus-containing product of this reaction is bis(phenylthio)phenylphosphinite (V). Phenylchlorophosphine (VII) is the product of the reverse addition of the sulfonyl chloride but was not detected even in the ^{31}P NMR spectra of the reaction mixtures. (Phenylthio)phenylchlorophosphine (VI) was noted in the spectra of the reaction mixtures but only in small amounts. The ^{31}P NMR signal for (VI) is found at 141.9 ppm [4,5]. The (V):(VI) ratio in the reaction mixtures ranges from 6:1 to 7:2 as indicated by ^{31}P NMR spectroscopy. Product (VI) may be obtained either from intermediate (III) with a P-S fragment upon reverse addition of a second molecule of (II) or from intermediate (IV) with a P-Cl fragment by an addition scheme differing from that in the initial step for the addition of sulfonyl chloride (II). N,N-Dimethylamino(dichloro)methane (VII) was isolated as the only nitrogen-containing compound in this reaction.

EXPERIMENTAL

The ^{31}P NMR spectra were taken on a Bruker M-250 spectrometer. All the operations were carried out in argon.

A sample of 2.2 g (0.012 mole) (I) was added to 6.4 g (0.052 mole) phenylsulfonyl chloride (II) in 15 ml acetonitrile at 0°C. The crystalline precipitate formed upon completion of the exothermal reaction was filtered off and recrystallized from acetonitrile. The yield of N,N-dimethylamino(dichloro)methane (VIII) was 1.25 g (73%), mp 140°C (sealed capillary) [6]. Distillation of the filtrate gave 2.04 g (47%) (V), bp 146-147°C (0.01 mm) [7]. ^{31}P NMR spectrum: $\delta = 90.0$ ppm [7]. Found: P, 8.80%. Calculated for $\text{C}_{18}\text{H}_{15}\text{PS}_2$: P, 9.51%.

LITERATURE CITED

1. W. Rösh and M. Regitz, *Synthesis*, 591 (1984).
2. J. Högel and A. Schmidpeter, *Chem. Ber.*, 118, No. 4 (1985).
3. A. S. Sherman, I. I. Patsanovskii, E. A. Ishmaeva, et al., *Dokl. Akad. Nauk SSSR*, 309, No. 5, 1145 (1989).
4. N. I. Rizpolozhenskii, V. D. Akamsin, and T. M. Dosova, *Izv. Akad. Nauk SSSR, Ser. Khim.* No. 3, 622 (1970).
5. L. N. Shiitov and B. M. Gladshtein, *Zh. Obshch. Khim.*, 39, No. 6, 1251 (1969).
6. H. Eilingsfeld and M. Seefelder, *Chem. Ber.*, 96, No. 10, 2671 (1963).
7. A. I. Razumov, E. A. Krasil'nikova, T. V. Zykova, et al., *Zh. Obshch. Khim.*, 42, No. 6, 1250 (1972).