ISSN 1070-3632, Russian Journal of General Chemistry, 2018, Vol. 88, No. 2, pp. 338–341. © Pleiades Publishing, Ltd., 2018. Original Russian Text © I.E. Mikhailov, Yu.M. Artyushkina, G.A. Dushenko, O.I. Mikhailova, Yu.V. Revinskii, V.I. Minkin, 2018, published in Zhurnal Obshchei Khimii, 2018, Vol. 88, No. 2, pp. 342–345.

LETTERS TO THE EDITOR

Spectral Luminescent Properties of 2-Aryl-5-(2,6-dimethoxyphenyl)-1*H*-1,3,4-oxadiazoles

I. E. Mikhailov^{*a,b**}, Yu. M. Artyushkina^{*a*}, G. A. Dushenko^{*a*}, O. I. Mikhailova^{*a*}, Yu. V. Revinskii^{*b*}, and V. I. Minkin^{*a*}

^a Research Institute of Physical and Organic Chemistry, Southern Federal University, pr. Stachki 194/2, Rostov-on-Don, 344090 Russia *e-mail: mikhail@ipoc.sfedu.ru

^b Southern Scientific Center, Federal Research Center, Russian Academy of Sciences, Rostov-on-Don, Russia

Received September 18, 2017

Abstract—2-Aryl-5-(2,6-dimethoxyphenyl)-1*H*-1,3,4-oxadiazoles were synthesized by cyclization reaction of benzoylbenzohydrazides in SOCl₂, and their spectral luminescent properties were studied. The oxadiazoles containing phenyl or *o*-methoxyphenyl substituents in the position 2 were shown to luminesce with a high quantum yield in polar and nonpolar solvents. The replacement of these substituents by 2-hydroxyphenyl group led to oxadiazole intensely emitting only in highly polar aprotic DMSO.

Keywords: 2,5-diaryl-1,3,4-oxadiazoles, luminescence, luminescence quantum yield, organic phosphors **DOI:** 10.1134/S107036321802024X

1,3,4-Oxadiazoles and their derivatives belong to one of the most intensively studied classes of fivemembered nitrogen-containing heterocycles characterized by good synthetic accessibility and high structural variability, as well as by wide range of practical applicability. These compounds have diverse biological activity [1] and are extensively used in various fields of medicinal chemistry [2] and pesticide chemistry [3]. Also, they possess interesting spectral luminescent properties making them suitable for the preparation of widely demanded organic [4–7] and metal complex-based [8, 9] phosphors.

For expanding the range of compounds of this class and for studying their spectral luminescent properties we applied here the reaction of hydrazides **1a–1c** with 2,6-dimethoxybenzoic acid chloride in the presence of triethylamine to obtain benzoylbenzohydrazides 2a-2c whose subsequent cyclization with thionyl chloride led to 2-aryl-5-(2,6-dimethoxyphenyl)-1*H*-1,3,4-oxadiazoles **3a**-**3c** (Scheme 1).

In the absorption spectra of oxadiazoles **3a–3c** the maximum of the long-wave band due to $\pi \rightarrow \pi^*$ electron transitions is observed within the 256–316 nm range, consistent with their benzenoid structure. The luminescence spectra of oxadiazole **3b** in isooctane, toluene, and acetonitrile contain two bands: a shortwave band (λ_{max}^{fl} 353–378 nm, ϕ 0.001–0.012) which has a normal Stokes shift (3019–4992 cm⁻¹) and a long-wave band (λ_{max}^{fl} 479–482 nm, ϕ 0.006– 0.013) which has an anomalously large Stokes shift (10471–



10799 cm⁻¹). Based on the fluorescence excitation spectra the short-wave band was assigned to the initial benzenoid structure 3, and the long-wave luminescence was associated with the emission of a shortlived phototautomer resulting from an excited-state intramolecular proton transfer (ESIPT) from the phenolic hydroxy group to the closest nitrogen atom of the heterocycle [10, 11]]. A low total luminescence quantum yield exhibited for oxadiazole **3b** (ϕ 0.013– 0.019) is due to nonradiative deactivation of its excited state by the ESIPT mechanism. However, in highly polar DMSO the spectrum of oxadiazole 3b contains only one short-wave high-intensity band (λ_{max}^{fl} 363 nm, φ 0.15) with a normal Stokes shift (4098 cm⁻¹) assigned to the initial benzenoid structure 3b basing on the fluorescence excitation spectra. This is explained by strong intermolecular hydrogen bonding between the phenolic hydroxy group of oxadiazole 3b and highly polar aprotic DMSO inhibiting the ESIPT process. Oxadiazoles 3a and 3c where an ESIPT process is impossible because of the absence of a mobile proton of the phenolic OH group emit intensely in a closely adjacent spectral region (λ_{max}^{fl} 352–363 nm, φ 0.11–0.33). Similar spectral behavior was previously observed in 1,3,4-oxadiazoles structurally similar to compounds **3a–3c** [11–15].

Thus, the replacement of the *ortho*-phenolic substituent in 2-(2-hydroxyphenyl)-5-(2,6-dimethoxyphenyl)-1H-1,3,4-oxadiazole **3b** by phenyl or *ortho*-methoxyphenyl group blocks the ESIPT process in oxadiazoles **3a** and **3c** formed, leading to strong enhancement of the luminescence quantum yield in nonpolar solvents and in acetonitrile compared to oxadiazole **3b**. This allows these compounds to be included among widely demanded oxadiazole series organic phosphors emitting in the short-wave region of the visible spectrum.

N'-Benzoyl-2,6-dimethoxybenzohydrazide (2a). To a solution of 1.36 g (0.01 mol) of benzoic acid hydrazide 1a in 20 mL of anhydrous acetonitrile, 5 mL of triethylamine and 2.01 g (0.01 mol) of 2,6-dimethoxybenzoic acid chloride in 30 mL of anhydrous acetonitrile were added successively with stirring. The reaction mixture was left for 1 day at room temperature and then refluxed for 3 h. The solvent was distilled off in a vacuum to give an oily residue which was washed with water (2 × 10 mL) and air-dried, whereupon the resulting product was isolated by column chromatography (eluent ethyl acetate-petroleum ether, 1 : 2), with the R_f 0.75–0.78 fraction being collected. After the solvent was distilled off, the

product was recrystallized from 2-propanol (2 × 15 mL). Yield 2.46 g (82%), colorless crystals, mp 196–198°C. IR spectrum, v, cm⁻¹: 765, 821, 1010, 1124, 1170, 1237, 1260, 1278, 1478, 1499, 1542; 1573, 1597, 1621 (C=C); 1655, 1698 (C=O); 3180 (NH). ¹H NMR spectrum (CDCl₃), δ , ppm: 3.83 s (6H, 2,6-OCH₃), 6.57 d (2H, H_{Ar}, *J* = 8.4 Hz), 7.24 t (1H, H_{Ar}, *J* = 8.4 Hz), 7.41–7.55 m (3H, H_{Ar}), 7.84 d.d (2H, H_{Ar}, *J* = 1.2, *J*₂ = 7.5 Hz), 9.12 d (1H, NH, *J* = 3.8 Hz), 9.32 d (1H, NH, *J* = 3.8 Hz). ¹³C NMR spectrum (DMSO-*d*₆), δ_{C} , ppm: 56.32 (2OCH₃), 101.32 (C_{Ar}), 104.83 (2C_{Ar}), 127.92 (2C_{Ar}), 128.03 (C_{Ar}), 128.99 (2C_{Ar}), 132.32 (C_{Ar}), 133.05 (C_{Ar}), 158.13 (2C_{Ar}), 163.42 (C=O), 166.33 (C=O). Found, %: C 64.05; H 5.35; N 9.38. C₁₆H₁₆N₂O₄. Calculated, %: C 63.99; H 5.37; N 9.33.

N'-(2-Hydroxybenzoyl)-2,6-dimethoxybenzohydrazide (2b) was prepared in a similar manner by the reaction of salicylic acid hydrazide 1b with 2,6-dimethoxybenzoic acid chloride. Yield 2.31 g (73%), colorless crystals, mp 223–224°C (mp 222–225°C [16]). IR and ¹H and ¹³C NMR spectra of 2b corresponded to those presented in [16].

N'-(2-Methoxybenzoyl)-2,6-dimethoxybenzohydrazide (2c) was prepared in a similar manner by the reaction of o-methoxybenzoic acid hydrazide 1c with 2,6-dimethoxybenzoic acid chloride. Yield 2.61 g (79%), colorless crystals, mp 150–152°C. IR spectrum, v, cm⁻¹: 732, 757, 804, 1003, 1125, 1168, 1188, 1233, 1264, 1297, 1479; 1562, 1625 (C=C); 1670, 1701 (C=O); 3137 (NH). ¹H NMR spectrum (CDCl₃), δ , ppm: 3.82 s (6H, 2,6-OCH₃), 4.06 s (3H, OCH₃), 6.55 d (2H, H_{Ar}, J = 8.4 Hz), 6.90–7.11 m (2H, H_{Ar}), 7.29 d.d (1H, H_{Ar}, $J_1 = 7.5, J_2 = 7.6$ Hz), 7.47 d.d (1H, H_{Ar}, $J_1 = 7.6, J_2 =$ 8.2 Hz), 8.08 d.d (1H, H_{Ar} , $J_1 = 1.2$, $J_2 = 7.5$ Hz), 9.65 d (1H, NH, *J* = 4.2 Hz), 11.17 d (1H, NH, *J* = 4.2 Hz). ¹³C NMR spectrum (CDCl₃), $\delta_{\rm C}$, ppm: 55.93 (2OCH₃), 56.01 (OCH₃), 102.17 (C_{Ar}), 105.15 (2C_{Ar}), 112.47 (C_{Ar}), 115.18 (C_{Ar}), 121.63 (C_{Ar}), 128.32 (C_{Ar}), 129.17 (C_{Ar}), 132.84 (C_{Ar}), 157.93 (C_{Ar}), 159.12 (2C_{Ar}), 161.33 (C=O), 165.74 (C=O). Found, %: C 61.88; H 5.47; N 8.53. C₁₇H₁₈N₂O₅. Calculated, %: C 61.81; H 5.49; N 8.48.

2-(2,6-Dimethoxyphenyl)-5-phenyl-1,3,4-oxadiazole (3a). A solution of 2.40 g (0.008 mol) of benzohydrazide 2a in 50 mL of thionyl chloride was refluxed for 6 h. The solvent was distilled off in a vacuum, and the reaction mass was cooled to room temperature and poured over 50 g of crushed ice. The resulting precipitate was filtered off, and the crystalline product was purified by column chromatography

(eluent ethyl acetate-petroleum ether, 1: 2), with the $R_{\rm f}$ 0.75-0.80 fraction being collected. After the solvent was distilled off the product was recrystallized from 2-propanol. Yield 1.40 g (62%), colorless crystals, mp 164–166°C. IR spectrum, v, cm⁻¹: 755, 787, 804, 963, 986, 1012, 1037, 1125, 1169, 1240, 1276, 1285, 1438, 1464, 1471; 1480; 1551, 1560, 1593 (C=C); 1610, 1632 (C=N). UV spectrum, λ_{max} , nm [$\epsilon \times 10^{-4}$ L mol⁻¹ cm⁻¹, $\lambda_{exc} 300 \text{ nm}$]: isooctane, 205 [2.92], 290 [0.95], λ_{max}^{fl} (ϕ) 358 (0.11); acetonitrile, 257 [1.36], 318 [0.83], λ_{max}^{fl} (ϕ) 359 (0.12); DMSO, 315 [1.18], λ_{max}^{fl} (ϕ) 357 (0.23). ¹H NMR spectrum (CDCl₃), δ, ppm: 3.79 s (6H, 2,6-OCH₃), 6.62 d (2H, H_{Ar}, J = 8.4 Hz), 7.43 t (1H, H_{Ar}, J = 8.4 Hz), 7.45–7.52 m (3H, H_{Ar}), 8.03– 8.17 m (2H, H_{Ar}). ¹³C NMR spectrum (CDCl₃), δ_{C} , ppm: 56.10 (20CH₃), 102.89 (C_{Ar}), 103.87 (2C_{Ar}), 124.46 (C_{Ar}), 127.00 (2C_{Ar}), 128.93 (3C_{Ar}), 131.40 (C_{Het}), 133.56 (C_{Het}), 159.96 (2C_{Ar}), 165.27 (C_{Ar}). Found, %: C 68.15; H 4.99; N 9.97. C₁₆H₁₄N₂O₃. Calculated, %: C 68.08; N 5.00; N 9.92.

2-[5-(2,6-Dimethoxyphenyl)-1,3,4-oxadiazol-2-yl]phenol (3b) was prepared in a similar manner by the cyclization of benzohydrazide 2b in thionyl chloride. Yield 1.17 g (49%), colorless crystals, mp 93–95°C. IR spectrum, v, cm⁻¹: 758, 831, 974, 1003, 1035, 1056, 1141, 1158, 1240, 1272, 1285, 1300, 1402, 1453, 1471, 1490; 1544, 1559 (C=C); 1598, 1629 (C=N); 3201 (OH). UV spectrum, λ_{max} , nm [$\epsilon \times 10^{-4}$ L mol⁻¹ cm⁻¹ λ_{exc} 300 nm]: isooctane, 257 [1.85], 263 [1.77], 307 [1.25], 318 [1.12], $\lambda_{max}^{fl}(\phi)$ 364 (0.001), 481 (0.012); toluene, 309 [1.18], 317 [1.08], $\lambda_{max}^{fl}(\phi)$ 378 (0.006), 482 (0.013); acetonitrile, 304 [0.74], 319 [0.62], λ_{max}^{fl} (φ) 353 (0.012), 479 (0.006); DMSO, 304 [0.74], 316 $[0.62], \lambda_{max}^{fl}$ (ϕ) 363 (0.15). ¹H NMR spectrum (CDCl₃), δ, ppm: 3.81 s (6H, 2,6-OCH₃), 6.65 d (2H, H_{Ar} , J = 8.4 Hz), 6.97 t (1H, H_{Ar} , J = 8.4 Hz), 7.11 d.d $(1H, H_{Ar}, J_1 = 1.1, J_2 = 7.6 \text{ Hz}), 7.36-7.48 \text{ m} (2H, H_{Ar}),$ 7.78 d.d (1H, H_{Ar} , $J_1 = 1.8$, $J_2 = 7.9$ Hz), 10.31 (1H, OH). ¹³C NMR spectrum (CDCl₃), δ_{C} , ppm: 56.14 (20CH₃), 103.91 (2C_{Ar}), 107.88 (C_{Ar}), 108.62 (C_{Ar}), 117.47 (C_{Ar}), 119.72 (C_{Ar}), 126.79 (C_{Ar}), 133.34 (C_{Ar}), 133.60 (C_{Ar}), 157.60 (C_{Het}), 158.90 (C_{Het}), 160.02 (2C_{Ar}), 164.89 (C_{Ar}). Found, %: C 64.46; H 4.75; N 9.44. C₁₆H₁₄N₂O₄. Calculated, %: C 64.42; H 4.73; N 9.39.

2-(2,6-Dimethoxyphenyl)-5-(2-methoxyphenyl)-1,3,4-oxadiazole (3c) was prepared in a similar manner by the cyclization of benzohydrazide 2c in thionyl chloride. Yield 1.72 g (69%), colorless crystals, mp 138–140°C. IR spectrum, v, cm⁻¹: 753, 771, 789, 807, 965, 982, 1010, 1043, 1128, 1166, 1244,

1266, 1283, 1439, 1456, 1468; 1487; 1554, 1565, 1598 (C=C); 1617, 1641 (C=N). UV spectrum, λ_{max} , nm $[\epsilon \times 10^{-4} \text{ L mol}^{-1} \text{ cm}^{-1}, \lambda_{exc} 300 \text{ nm}]$: isooctane, 257 [1.16], 285 [0.81], 297 [0.75], $\lambda_{m\,ax}^{fl}(\phi)$ 352 (0.16); acetonitrile, 254 [1.37], 287 [0.88], 297 [0.79], λ_{max}^{fl} (ϕ) 357 (0.15); DMSO, 288 [1.16], 300 [1.05], $\lambda_{max}^{fl}(\phi)$ 363 (0.33). ¹H NMR spectrum (CDCl₃), δ, ppm: 3.79 s (6H, 2,6-OCH₃), 3.95 s (3H, OCH₃), 6.61 d (2H, H_{Ar}, J = 8.4 Hz), 6.99–7.11 m (2H, H_{Ar}), 7.36–7.53 m (2H, H_{Ar}), 7.95 d.d (1H, H_{Ar} , $J_1 = 1.8$, $J_2 = 7.9$ Hz). ¹³C NMR spectrum (CDCl₃), δ_{C} , ppm: 56.09 (2OCH₃), 56.10 (OCH₃), 103.17 (C_{Ar}), 103.90 (2C_{Ar}), 111.98 (C_{Ar}), 113.61 (C_{Ar}), 120.65 (C_{Ar}), 130.61 (C_{Ar}), 132.71 (C_{Ar}), 133.24 (C_{Ar}), 157.93 (C_{Het}), 159.16 (C_{Het}), 159.98 (2CAr), 163.86 (CAr). Found, %: C 65.44; H 5.13; N 9.04. C₁₇H₁₆N₂O₄. Calculated, %: C 65.38; H 5.16; N 8.97.

IR spectra were recorded on a Varian Excalibur 3100 FT-IR spectrometer in Nujol. ¹H (250.13 MHz) and ¹³C (62.90 MHz) NMR spectra were obtained on a Bruker DPX-250 instrument. The absorption and fluorescence spectra were measured on a Cary Scan 100 spectrophotometer and a Cary Eclipse fluorescence spectrophotometer, respectively. The fluorescence quantum yields were determined relative to anthracene in acetonitrile [17].

ACKNOWLEDGMENTS

This study was financially supported by the Ministry of Education and Science of the Russian Federation in the framework of the governmental contract (project no. 4.979.2017/PCh).

REFERENCES

- Boström, J., Hogner, A., Llinas, A., Wellner, E., and Plowright, A.T., *J. Med. Chem.*, 2012, vol. 55, no. 5, p. 1817. doi 10.1021/jm2013248
- Farshori, N.N., Banday, M.R., Ahmad, A., Khan, A.U., and Rauf, A., *Bioorg. Med. Chem. Lett.*, 2010, vol. 20, no. 6, p. 1933. doi 10.1016/j.bmcl.2010.01.126
- Jha, K.K., Samad, A., Kumar, Ya., Shaharyar, M., Khosa, R.L., Jain, Ja., Kumar, V., and Singh, P., *Eur. J. Med. Chem.*, 2010, vol. 45, no. 11, p. 4963. doi 10.1016/j.ejmech.2010.08.003
- Yang, H., Mu, J., Chen, X., Feng, L., Jia, J., and Wang, J., Dyes Pigm., 2011, vol. 91, no. 3, p. 446. doi 10.1016/ j.dyepig.2011.03.035
- Beldovskaya, A.D., Dushenko, G.A., Vikrishchuk, N.I., Popov, L.D., Revinskii, Yu.V., Mikhailov, I.E., and

Minkin, V.I., *Russ. J. Org. Chem.*, 2013, vol. 49, no. 12, p. 1861. doi 10.1134/S1070428013120312

- Mikhailov, I.E., Popov, L.D., Vikrishchuk, N.I., Beldovskaya, A.D., Revinskii, Yu.V., Dushenko, G.A., and Minkin, V.I., *Russ. J. Gen. Chem.*, 2015, vol. 85, no. 1, p. 203. doi 10.1134/S1070363215010363
- Mikhailov, I.E., Vikrishchuk, N.I., Popov, L.D., Beldovskaya, A.D., Revinskii, Yu.V., Dushenko, G.A., and Minkin, V.I., RF Patent 2568640, *Byul. Izobret.*, 2015, no. 32.
- Beldovskaya, A.D., Dushenko, G.A., Vikrishchuk, N.I., Popov, L.D., Revinskii, Yu.V., Mikhailov, I.E., and Minkin, V.I., *Russ. J. Gen. Chem.*, 2014, vol. 84, no. 1, p. 171. doi 10.1134/S1070363214010290
- Mikhailov, I.E., Popov, L.D., Vikrishchuk, N.I., Beldovskaya, A.D., Dushenko, G.A., Revinskii, Yu.V., Kurbatov, S.V., and Minkin, V.I., RF Patent 2610527, *Byul. Izobret.*, 2017, no. 5.
- Serdyuk, O.V., Evseenko, I.V., Dushenko, G.A., Revinskii, Yu.V., and Mikhailov, I.E., *Russ. J. Org. Chem.*, 2012, vol. 48, no. 1, p. 78. doi 10.1134/ S1070428012010113
- Doroshenko, A.O., Posokhov, E.A., Verezubova, A.A., and Ptyagina, L.M., *J. Phys. Org. Chem.*, 2000, vol. 13, no. 5, p. 253. doi 10.1002/1099-1395(200005)

13:5<253::AID-POC238>3.0.CO;2-D

- Mikhailov, I.E., Artyushkina, Yu.M., Burov, O.N., Dushenko, G.A., Revinskii, Yu.V., and Minkin, V.I., *Russ. J. Gen. Chem.*, 2016, vol. 86, no. 2, p. 406. doi 10.1134/S1070363216020341
- Mikhailov, I.E., Artyushkina, Yu.M., Dushenko, G.A., Mikhailova, O.I., Revinskii, Yu.V., Burov, O.N., and Minkin, V.I., *Russ. J. Org. Chem.*, 2016, vol. 52, no. 11, p. 1700. doi 10.1134/S1070428016110270
- Artyushkina, Yu.M., Mikhailov, I.E., Burov, O.N., Dushenko, G.A., Mikhailova, O.I., Revinskii, Yu.V., and Minkin, V.I., *Russ. J. Gen. Chem.*, 2016, vol. 86, no. 12, p. 2702. doi 10.1134/S1070363216120239
- Artyushkina, Yu.M., Mikhailov, I.E., Dushenko, G.A., Mikhailova, O.I., Revinskii, Yu.V., Burov, O.N., and Kurbatov, S.V., *Russ. J. Gen. Chem.*, 2017, vol. 53, no. 5, p. 808. doi 10.1134/S1070428017050281
- Liu, X., Liu, W., Lee, K., Park, M., Ri, H.-Ch., Kim, G.H., and Lah, M.S., *Dalton Trans.*, 2008, no. 46, p. 6579. doi 10.1039/b810711d
- Beldovskaya, A.D., Dushenko, G.A., Vikrishchuk, N.I., Popov, L.D., Revinskii, Y.V., and Mikhailov, I.E., *Russ. J. Gen. Chem.*, 2013, vol. 83, no. 11, p. 2075. doi 10.1134/S1070363213110200