

Enantioselective Total Syntheses of Methanoquinolizidine-Containing Akuammiline Alkaloids and Related Studies

Elias Picazo, Lucas A. Morrill, Robert Susick, Jesus Moreno, Joel Smith, and Neil K. Garg J. Am. Chem. Soc., Just Accepted Manuscript • DOI: 10.1021/jacs.8b03404 • Publication Date (Web): 25 Apr 2018 Downloaded from http://pubs.acs.org on April 25, 2018

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Enantioselective Total Syntheses of Methanoquinolizidine-Containing Akuammiline Alkaloids and Related Studies

Elias Picazo,[†] Lucas A. Morrill,[†] Robert B. Susick, Jesus Moreno, Joel M. Smith, and Neil K. Garg*

Department of Chemistry and Biochemistry, University of California, Los Angeles, California 90095

ABSTRACT: The akuammiline alkaloids are a structurally diverse class of bioactive natural products isolated from plants found in various parts of the world. A particularly challenging subset of akuammiline alkaloids are those that contain a methanoquinolizidine core. We describe a synthetic approach to these compounds that has enabled the first total syntheses of (+)-strictamine, (-)-2(S)-cathafoline, (+)-akuammiline, and (-)- Ψ -akuammigine. Our strategy relies on the development of the reductive interrupted Fischer indolization reaction to construct a common pentacyclic intermediate bearing five contiguous stereocenters, in addition to late-stage formation of the methanoquinolizidine framework using a deprotection–cyclization cascade. The total syntheses of (-)- Ψ -akuammigine and (+)-akuammiline mark the first preparations of akuammiline alkaloids containing both a methanoquinolizidine core and vicinal quaternary centers. Lastly, we describe the bioinspired reductive rearrangements of (+)-strictamine and (+)-akuammiline to ultimately provide (-)-10-demethoxyvincorine and a new analogue thereof.

Introduction

The akuammiline alkaloids are an important class of natural products isolated from plants found in India, Africa, and Southeast Asia. These alkaloids, thought to be the active ingredients in traditional medicines used to treat a multitude of ailments in humans and livestock.¹ have been the subject of intense investigations spanning structural elucidation, biosynthesis, biological evaluation, and chemical synthesis.² More than 75 akuammilines have now been unambiguously characterized,³ many of which display exciting biological profiles, and all of which possess provocative chemical structures. Despite their enticing characteristics and biosynthetic relation to the popular *Strychnos* alkaloids,⁴ the akuammiline family has been comparatively less studied by synthetic chemists relative to the Strychnos alkaloids. It was not until recently that the synthetic community launched efforts toward the akuammiline alkaloids, which have now culminated in total syntheses of several subclasses.⁵ This is particularly noteworthy, as even the simplest akuammilines contain at least five interconnected rings, a multitude of stereocenters, and one or more basic nitrogen atoms.

We became interested in a subset of the akuammiline alkaloids that possess a methanoquinolizidine core. This cagelike scaffold, along with several methanoquinolizidinecontaining akuammilines, are shown in Figure 1 (i.e., 1–4).⁶ (+)-Strictamine (1)⁷ features an indolenine and four stereocenters, whereas (–)-2(*S*)-cathafoline (2)⁸ has an indoline core resulting in five contiguous stereocenters. (+)-Akuammiline (**3**) and (–)- Ψ -akuammigine (**4**)⁹ display an additional layer of complexity with vicinal quaternary centers at C7 and C16. Moreover, (–)- Ψ -akuammigine (**4**) also bears a furoindoline motif, leading to six interconnected rings and five contiguous stereocenters, thereby rendering it the most complex member of this akuammiline subclass. The complexity of **1–4**, coupled with attractive biological activities (vide infra), have prompted numerous synthetic efforts. Following seminal studies by Dolby,¹⁰ Bosch and Bennasar,¹¹ and Sakai¹² beginning more than four decades ago, the laboratories of Cook,¹³ Tokuyama,¹⁴ and Matsuo¹⁵ have more recently described promising synthetic approaches. However, a total synthesis of any methanoquinolizidine-containing akuamiline alkaloid had proven elusive.¹⁶ Moreover, to date, no total syntheses of methanoquinolizidine-containing akuamilines that also bear vicinal quaternary centers (e.g., **3** and **4**) have been reported.

Herein, we describe a full account of our enantioselective approach to several methanoquinolizidine-containing akuammilines and subsequent biomimetic manipulations. Key to the success of these studies is the development of the reductive interrupted Fischer indolization reaction^{17,18,19} to construct a complex pentacyclic intermediate, as well as the late-stage formation of the methanoquinolizidine framework using a deprotection–cyclization cascade. These studies result in the



Figure 1. Representative methanoquinolizidine-containing akuammiline alkaloids.

first total syntheses of (+)-strictamine (1) and (-)-2(*S*)cathafoline (2), which we communicated in 2016.⁵¹ We then exploit a previously undesired epimerization event to access (+)-akuammiline (3) and (-)- Ψ -akuammigine (4), each of which bear vicinal quaternary stereocenters embedded in their methanoquinolizidine cores. Lastly, we describe late-stage reductive rearrangements of (+)-strictamine (1) and (+)akuammiline (3), inspired by the proposed biosynthesis of akuammiline alkaloids²⁰ to give pyrrolidinoindoline scaffolds.

Results and Discussion

(+)-Strictamine and (-)-2(S)-Cathafoline: Isolation, Biological Activity, and Retrosynthetic Analysis

(+)-Strictamine (1) and (-)-2(*S*)-cathafoline (2) were selected as our initial synthetic targets. (+)-Strictamine (1) was isolated in 1966⁷ from the plant *Rhazya stricta*, whereas (-)-2(*S*)-cathafoline (2) was extracted from *Alstonia macrophylla* in 2014.⁸ (+)-Strictamine (1) inhibits the transcription factor NF- κ B,²¹ and may therefore serve as a lead compound for the discovery of new drugs for the treatment of cancer or inflammatory diseases. (-)-2(*S*)-Cathafoline (2) shows moderate activity in overturning drug resistance in vincristine-resistant KB cells.⁸ Despite the aforementioned synthetic pursuits, neither 1 or 2 had succumbed to total synthesis. In the case of (+)-strictamine (1), this is especially noteworthy given that the compound had been known for more than 50 years.

Our retrosynthetic analysis of (+)-strictamine (1) and (-)-2(*S*)-cathafoline (2) is shown in Scheme 1. We envisioned forming the methanoquinolizidine scaffold by late-stage construction of the C5–N4 bond. Thus, 1 and 2 would be accessed from primary alkyl chloride 5 through a tandem deprotection–cyclization reaction, followed by manipulation of the indoline ring system. Chloride 5 would derive from indoline lactone 6 through methanolysis and halide formation. In a key retrosynthetic disconnection, we envisaged indoline lactone 6 arising from ketolactone 9 and phenylhydrazine (8) via a

Scheme 1. Retrosynthetic analysis of (+)-strictamine (1) and (-)-2(*S*)-cathafoline (2).



reductive interrupted Fischer indolization reaction.¹⁷ If successful, this transformation would introduce two new rings. Moreover, two stereogenic centers would be generated, including the C2 stereocenter seen in 2(S)-cathafoline (2) (via 7) and the challenging C7 quaternary center common to all akuammilines. Ketolactone 9 would ultimately be obtained from enone 10, which, in turn, would arise from a gold-catalyzed cyclization of silyl enol ether 11.²² Finally, 11 would be accessed in enantioenriched fashion from sulfonamide 12 and dibenzoate 13 via a Pd-catalyzed desymmetrization and subsequent manipulations.²³

Enantioselective Synthesis of the Fischer Indolization Substrate

To initiate our forward synthetic studies, we sought to prepare enone 16 in enantioenriched form, as this could serve as a precursor for later construction of the necessary [3.3.1]azabicycle (Scheme 2). Although we initially examined several possible routes, including enzymatic desymmetrization²⁴ and the use of Koga bases,²⁵ we found a Pd-catalyzed Trost desymmetrization²³ was ultimately most successful. Upon treatment of dibenzoate 13 and sulfonamide 12 with a suitable Pd precatalyst and (R,R)-DACH-phenyl Trost ligand 14, the desired desymmetrization took place. Direct saponification of the product furnished alcohol 15 in 89% yield. Subsequent PCC-mediated oxidation of the resultant secondary alcohol delivered enone 16 in excellent yield. Enone 16 could be crystallized at the interface of EtOAc and heptane; X-ray diffraction studies confirmed the desired absolute stereochemistry had been obtained in the desymmetrization reaction.²⁶

Scheme 2. Trost desymmetrization and synthesis of enone 16.



Having synthesized enone 16, we sought to introduce the [3.3.1]-azabicycle of the natural product and perform further elaborations (Scheme 3). Upon treatment with TBDPSOTf and 2,6-lutidine, enone 16 was converted to silyl enol ether 11 in high yield. This set the stage for a pivotal gold-mediated cyclization, inspired by precedent from Li and coworkers.²² Using our optimized protocol, silyl enol ether 11 was treated with (PMe₃)AuCl and AgOTf, followed by quenching with tosic acid hydrate in the same pot, to ultimately afford bicycle 10 as the major product. This sequence is thought to proceed by initial formation of oxocarbenium ion intermediate 17, which undergoes nucleophilic attack by *t*-BuOH to give silyl enol ether 18. In the presence of acid and water, 18 is hydrolyzed to give the desired enone 10. Chiral SFC analysis

at this stage revealed that **10** had been prepared in 96% ee. It should also be noted that the key gold-mediated cyclization gave an undesired by product, **19**, in minor quantities. Although **10** and **19** were inseparable, the mixture could be taken forward through an epoxidation and Wittig homologation.^{5h,k} Aqueous workup of the Wittig product led to enol ether hydrolysis with epoxide fragmentation to cleanly deliver enal **20** as a single constitutional isomer in 49% yield from silyl enol ether **11**.

Scheme 3. Gold-catalyzed cyclization to construct the [3.3.1]-azabicycle and elaboration to enal **20**.



Using the sequence shown in Scheme 4, enal 20 was elaborated to ketolactone 9, the substrate for our key reductive interrupted Fischer indolization reaction. One of the synthetic challenges we encountered was the introduction of a hydroxyethyl group on C7. Initial approaches using metal-mediated conjugate addition strategies were met without success, so instead we targeted Ueno–Stork cyclization methodology.²⁷ En route to testing this approach, aldehyde 20 was converted to the corresponding methyl ester, 21, using an NIS-based oxidative esterification protocol.²⁸ Next, 21 was treated with ethyl vinyl ether (22) and NIS to give a mixed iodoacetal

Scheme 4. Synthesis of ketolactone **9**, the substrate for the reductive Fischer indolization reaction.



intermediate. Subsequent reaction with tributyltin hydride and AIBN led to cyclization at C7 to deliver acetal **23** as an inconsequential mixture of diastereomers.²⁷ Acid-mediated hydrolysis of the acetal and reduction of the resultant lactol provided diol **24**. Finally, lactonization, followed by alcohol oxidation, afforded the desired ketolactone **9**.

Reductive Interrupted Fischer Indolization Reaction

With ketolactone 9 in hand, we developed the key reductive interrupted Fischer indolization reaction. Our laboratory had extensive experience in developing Fischer indolizations in the context of methodology¹⁸ and complex molecule synthesis.^{5b,h,k} Thus we began by surveying a range of acid sources that were utilized in the Fischer indolization reaction literature and those we had previously deemed most promising. As shown in Table 1, the reactions of phenylhydrazine (8, 3 equiv) and ketolactone 9 using 5.0 equiv of acid were performed at 60 °C for 6 h to provide meaningful comparison data. Most attempts, including those with Lewis acids and very strong protic acids, were met with disappointment (entries 1–5). Whereas the use of AcOH gave modest yield using water as the solvent (entry 6), an improvement was seen in switching the solvent to acetic acid (entry 7). Lastly, the use of TFA gave the most promising result, delivering Fischer indolization product 26 in 38% yield (entry 8). It should be noted that this transformation proceeds with complete diastereoselectivity and sets the C7 quaternary stereocenter. The diastereoselectivity is thought to be governed by approach of the phenyl ring on the less hindered faced of the [3.3.1]-azabicycle during the charge-accelerated [3,3]-sigmatropic rearrangement (see transition structure 25).

 Table 1. Survey of acids to promote the Fischer indolization reaction of ketolactone 9 and 8.

Page 3 of 11



^{*a*} Yields were determined by ¹H NMR analysis using 1,3,5-trimethoxybenzene as an external standard.

Our next objective was to further optimize the Fischer indolization reaction and develop the desired reductive variant.¹⁷ We were delighted to find that slight changes in our reaction conditions (i.e., 40 °C and 12 h), greatly improved conversion to indolenine lactone 26 (Scheme 5). Moreover, it was found that by simply adding triethylsilane and additional TFA to the Fischer indolization reaction with stirring at 23 °C, we obtained the reductive Fischer indolization product 6 in 83% yield. Overall, the conversion of ketolactone 9 to indoline 6 proceeds with installation of two new rings, three new bonds, and two new stereocenters, all with complete diastereoselectivity. This transformation represents one of the most complex Fischer indolization reactions reported in the literature.²⁹ Interestingly, hydride approach from the more congested face of the [3.3.1]-azabicycle appears to be favorable (see transition structure 7). Although counterintuitive in some respects, the stereochemical outcome may be rationalized by the kinetic preference for the formation of a cis ring juncture between the pyrrolidine and cyclohexyl units. The stereochemical outcome is consistent with reports from the Styrchnos alkaloid literature³⁰ as well as a similar C2 reduction in akuammiline synthesis.5v

Scheme 5. Reductive interrupted Fischer indolization reaction delivers 6.



Undesired C16 Epimerization and Synthesis of Unnatural C16 Epimers of Strictamine and 2(S)-Cathafoline

Having executed the key reductive interrupted Fischer indolization reaction, we sought to prepare chloroester 30 en route to the desired natural products through a seemingly straightforward sequence (Scheme 6). Boc protection of indoline 6 proceeded smoothly to furnish 27. Subsequently, a two step hydrolysis/methylation protocol was used to arrive at hydroxyester 28. Although we found 28 to be somewhat unstable, we found it could be used directly in the subsequent step. Treatment under Appel-like conditions cleanly provided 29. which unfortunately possessed the undesired stereochemistry at C16. Despite a survey of alternative methanoloysis conditions prior to performing the chlorination, the undesired C16 epimer, 29, was routinely formed as the major product. We surmise that C16 epimerization of 27 occurs more favorably than methanolysis or hydrolysis of the lactone. Several workarounds were pursued, such as attempted lactone opening via amidation,³¹ Pd-catalyzed ring-opening,³² or S_N2 displacements,³³ none of which were successful. Additionally, efforts to epimerize C16 of **29** and related compounds to provide the desired C16 epimer also proved fruitless.

Scheme 6. Undesired C16 epimerization.



Faced with the undesired epimerization leading to chloroester 29, we saw an opportunity to develop late-stage transformations and synthesize unnatural epimeric analogs of strictamine and 2(S)-cathafoline. As shown in Scheme 7, the C16 epimer of strictamine 34 could be accessed in three steps from chloroester 29. Deprotection of 29 furnished indoline 31 in quantitative yield. Treatment of 31 with PCC led to indoline oxidation, thus providing the corresponding indolenine motif. Lastly, we pursued the key deprotection-cyclization step, analogous to that outlined in our retrosynthetic analysis (see Figure 1). Exposure of the indolenine intermediate to a solidsupported thiol resin 32^{34} and Cs_2CO_3 in acetonitrile at 65 °C delivered 16-epi-strictamine (34), presumably by way of transition structure 33. Analogous to our earlier study toward the total synthesis of picrinine,^{5h,k} the use of 32 proved instrumental in facilitating purification of the basic tertiary amine product. More importantly, this sequence demonstrated

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18 19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

Page 4 of 11

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48 49

50

51

52 53

54

55 56

57 58 59

60

that our overall approach involving a late-stage deprotectioncyclization could be used to access the challenging methanoquinzolidine framework.

Scheme 7. Synthesis of 16-epi-strictamine (34).



We also explored two complementary routes to the C16 epimer of 2(S)-cathafoline **35**, which are depicted in Figure 2. Beginning from intermediate alkyl chloride 31, reductive amination using formaldehyde led to N-methylation.^{5c} Subsequent deprotection-cyclization provided 35, albeit in only 10% yield. Given the modest synthetic efficiency of this sequence, alternative path we pursued an that involved methanoquinolizidine formation prior to N-methylation. Deprotection-cyclization of sulfonamide 29, followed by acidic deprotection of the indoline, furnished the methanoquinolizidine-containing product 36 in 76% yield over two steps. Finally, a reductive amination involving the indoline nitrogen afforded 16-epi-2(S)-cathafoline (35). The synthetic routes to 34 and 35 not only permitted access to unnatural analogs of strictamine and 2(S)-cathafoline, but also gave us critical experience in performing late-stage operations, particularly for introduction of the methanoquinolizidine unit and purification of polar natural product-like compounds.

1. HCHO, NaBH₃CN MeCN, AcOH, 23 °C 'н Si) 32 CO2We ĆO₂Me Cs₂CO₃, MeCN, 65 °C 16-epi-2(S)-(10% yield, 2 steps) 31 Cathafoline (35) (si) 32 HS Cs₂CO₃, MeCN, 65 °C 2. TFA, CH2Cl2, 23 °C ′н ́со.м4 CO₂Me (76% vield, 2 steps) 29 36 HCHO, NaBH₂CN MeCN, AcOH, 23 °C (48% yield) 16-epi-2(S)-Cathafoline (35)

Figure 2. Synthetic routes to 16-epi-2(S)-cathafoline (35).

Enantioselective Total Syntheses of (+)-Strictamine and (-)-2(S)-Cathafoline

Despite the C16 epimerization setback, we returned to the primary task of completing the total syntheses of (+)strictamine and (-)-2(S)-cathafoline (Scheme 8). We first sought a means to open the lactone ring of 6 under non-basic reaction conditions, to hopefully avoid the undesired epimerization. To this end, indoline lactone 6 was treated with lithium borohydride to give the corresponding diol with the C16 stereochemistry unperturbed. Subsequent protection of the more accessible primary alcohol furnished silyl ether 37. Next, a straightforward three-step sequence was performed to elaborate the C16 hydroxymethyl group of 37 to the corresponding methyl ester, 38. From hydroxyester 38, a twostep chlorination protocol gave alkyl chloride 5^{35} which proceeded in the presence of the free N-H of the indoline unit and set the stage for the final steps of the total syntheses. (+)-Strictamine (1) was accessed in quantitative yield from 5 by indoline oxidation, followed by deprotection-cyclization. Similarly, the total synthesis of (-)-2(S)-cathafoline (2) was achieved by N-methylation of 5, followed by deprotectionpentacyclic cyclization. In both cases, the methanoquinolizidine framework was established in the final step of the total synthesis. NMR spectra of synthetic (+)-1 and (-)-2 correlated with spectra of the natural samples. This synthetic effort marked the first total syntheses of 1 and 2, which was initially communicated concurrently with Zhu's synthesis of 1.¹⁶ Other synthetic routes to 1 have subsequently been reported by the labs of Fujii and Ohno,⁵⁰ Gaich,^{5p,} Snyder, ^{5r} Qin, ^{5u} and Zu.^{5v}

Scheme 8. Total syntheses of (+)-strictamine (1) and (-)-2(S)-cathafoline (2).



(-)-Ψ-Akuammigine: Isolation, Biological Activity, and Retrosynthetic Analysis

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55 56

57 58 59

60

Having validated our strategy toward the total syntheses of natural products containing the methanoquinolizidine core, we directed our attention toward (-)- Ψ -akuammigine (4), the most structurally complex of the methanoquinolizidinecontaining akuammiline alkaloids. $(-)-\Psi$ -Akuammigine (4) was isolated from Picralima klaineana in 1932.9 Biological studies later revealed promising activity as an antiinflammatory agent.³⁶ Moreover, (-)-Ψ-akuammigine (4) belongs to a subset of akuammilines bearing vicinal quaternary centers, a structural motif which continues to be a notable synthetic challenge. In fact, only recently have there been reported total syntheses of akuammiline alkaloids bearing this motif.^{5w-y} Despite these triumphant efforts, akuammilines containing both the methanoquinolizidine and vicinal quaternary centers have yet to succumb to total synthesis.3

Our initial retrosynthetic analysis of (-)-Ψ-akuammigine (4) is shown in Scheme 9. The natural product (4) was proposed to arise from late-stage intermediate 39, possessing a furoindoline motif and an alkyl chloride. In the forward sense, removal of the nosyl group was expected to enable C5-N4 bond formation and forge the methanoquinolizidine core, analogous to the strategy used to synthesize (+)-strictamine (1) and (-)-2(S)-cathafoline (2). Chloride 39 would be synthesized from furoindoline 40 through oxidative esterification and elaboration of the silvl ether. In a key maneuver, furoindoline 40 would ultimately be derived from aldehyde 42 through a sequence involving C16 hydroxymethylation, followed by Nmethylation and cyclization (see transition structure 41). Although not depicted, it was envisioned that the namesake of the natural product family, (+)-akuammiline (4), could also be accessed from 42 or derivatives thereof. Finally, aldehyde 42 was anticipated to be available from indoline lactone 6, which should be readily accessible by the reductive interrupted Fischer indolization reaction¹⁷ described above (see Scheme 5).

Scheme 9. Retrosynthetic analysis of (-)- Ψ -akuammigine (4).



Attempted Introduction of the Furoindoline Motif

To initiate our studies toward (–)- Ψ -akuammigine (4), we focused on the scalable preparation of aldehyde 42 following the route shown in Scheme 10. Knowing the commonly encountered difficulties associated with late-stage endeavors in akuammiline total synthesis, we expected our expedition would require us to perform the key, reductive interrupted Fischer indolization¹⁷ step on a large scale. Thus, we first prepared more than 10 grams of ketolactone 9. In the key reductive Fischer indolization step, we were heartened to find that gram-scale reaction of ketolactone 9 and phenylhydrazine (8) under acidic conditions proceeded smoothly to deliver intermediate indolenine 26. Reduction in the same pot using triethylsilane then furnished the pentacyclic indoline lactone 6 Scheme 10. Scalable preparation of aldehyde 42.



in 93% yield. Elaboration of indoline lactone 6 through a 3step sequence involving lactone reduction, selective alcohol protection, and simultaneous oxidation of the indoline and remaining alcohol afforded indolenine aldehyde 42.

With aldehyde 42 in hand, we sought to tackle three structural challenges presented by Ψ -akuammigine (4): the vicinal quaternary centers, the furoindoline moiety, and the methanoquinolizidine core. As shown in Scheme 11, treatment of aldehyde 42 with paraformaldehyde under basic conditions^{5x,y} provided diol **43** in 58% yield. Of note, this transformation establishes the second of the vicinal quaternary centers and proceeds with Tischenko-type reduction of the aldehyde.³⁸ Next, we pursued introduction of the furoindoline moiety by exposing diol 43 to excess methyl iodide and K_2CO_3 ³⁹ This allowed for differentiation of the C16 hydroxymethyl groups and furnished furoindoline alcohol 44 in 66% yield. Lastly, we sought to forge the methanoquinolizidine core of the natural product. A three-step oxidation/esterification sequence delivered ester 45, which was poised to undergo C5-N4 bond formation upon C5 activation and removal of the nosyl group. Unfortunately, removal of the silvl ether group present in 45 failed to deliver the desired alcohol 47. Instead, we exclusively obtained an undesired isomer, furoindoline 46, in 67% yield using Ph₃PBr₂⁴⁰ and in high efficiencies under other conditions (e.g., acetic acid, TBAF). It is hypothesized that the presumed in situ structural rearrangement of furoindoline 47 to isomer 46 is thermodynamically driven, with 46 being significantly more stable.41



Scheme 11. Attempted introduction of the furoindoline motif and undesired isomerization to afford **46**.



Enantioselective Total Synthesis of (-)- Ψ -Akuammigine

The discovery of the undesired furoindoline isomerization prompted us to reassess our synthetic approach toward (–)- Ψ akuammigine (4) (Scheme 12). Rather than installing the furoindoline prior to the methanoquinolizidine, we sought to reverse the order in which these key structural features were formed. Thus, we targeted epimeric esters **36** or **48** and derivatives thereof that contain the methanoquinolizidine core, but lack the vicinal quaternary centers and furoindoline, as potential precursors to (–)- Ψ -akuammigine (4).⁴² It should be noted that the C16 stereochemistry was considered inconsequential, as this stereocenter would be controlled during a subsequent alkylation step. However, the strategy would likely require the handling and purification of polar compounds due to the earlier installation of the tertiary amine. Given limited options, we pursued this revised strategy.

Scheme 12. Reassessment of synthetic approach.



With the aim of installing the methanoquinolizidine and setting the stage for furoindoline formation, we developed the sequence described in Scheme 13. Beginning with the product of our reductive interrupted Fischer indolization reaction, indoline lactone 6, we performed a one-pot saponification/methylation reaction. This net methanolysis of the lactone proceeded smoothly, with the expected epimerization of the C16 stereocenter. Several attempts were made to activate the resultant primary alcohol (e.g., tosylate, mesylate, halides, etc.), but the desired products of such efforts were highly un-

stable and could not be isolated. We ultimately found that the alcohol intermediate could be activated by conversion to diphenyl phosphate ester **49**.⁴³ This set the stage for methanoquinolizidine formation. Treatment of **49** with solid-supported thiol resin **32** under basic conditions provided ester **36**.^{5h,k,l} Reduction of ester **36** with LiBH₄ afforded the corresponding alcohol with concomitant borylation of the tertiary amine to give indoline alcohol **50** in 69% yield.⁴⁴ Although unintentional, the introduction of the amine-borane functional group proved beneficial, as it masked the basic tertiary amine and dramatically facilitated purification of late-stage intermediates.

Scheme 13. Introduction of the methanoquinolizidine framework and fortuitous amine-borane complexation.



The final steps to complete the total synthesis of $(-)-\Psi$ akuammigine (4) focused on installation of the C16 quaternary stereocenter and the furoindoline motif (Scheme 14). Double oxidation of indoline alcohol 50 with PCC yielded the corresponding indolenine aldehyde, which was a suitable substrate for C16 alkylation. Exposure of the indolenine aldehyde to paraformaldehyde (9 equivalents) and Cs₂CO₃ provided diol 51 in 43% yield over two steps. In a key ploy, we pursued furoindoline formation by exposing 51 to methyl iodide under basic conditions.³⁹ This gave rise to the desired product, furoindoline alcohol 53, in 45% yield, with 33% recovered 51. We presume the reaction proceeds through activated indoleninium 52 to deliver an intricate scaffold bearing both the methanoquinolizidine and the furoindoline motifs. Notably, if the reaction was allowed to proceed with higher conversion, borane removal and N4-methylation occured to give an undesired ammonium salt byproduct. Nonetheless, oxidation to the aldehyde, followed by oxidative esterification²⁸ with borane removal,⁴⁵ furnished (-)-Ψ-akuammigine (4) in 86% yield over two steps. All spectral data for (-)-Y-akuammigine (4) were consistent with reported data and authentic samples.⁴⁰

Scheme 14. Total synthesis of (-)- Ψ -akuammigine (4).



Enantioselective Total Synthesis of (+)-Akuammiline

Lastly with regard to the synthesis of methanoquinolizidine-containing akuammilines, we sought to prepare (+)-akuammiline (**3**, Scheme 15), the namesake of the natural product family. Akuammiline (**3**) was isolated alongside (-)- Ψ -akuammigine (**4**) in 1932 and has not been synthesized previously.⁹ (+)-Akuammiline (**3**) is structurally similar to (-)- Ψ -akuammigine (**4**), but possesses a C16 acetoxymethyl group, rather than a furoindoline unit.

We were delighted to find that late-stage intermediate alcohol **50** served as a suitable precursor to (+)-akuammiline (**3**) (Scheme 15). Double oxidation of indoline alcohol **50** provided borane-protected indolenine aldehyde **54**. C16 alkylation of **54** proceeded smoothly to establish the vicinal quaternary centers and provide aldehyde **55**.⁴⁷ It was optimal to use only 1.2 equivalents of paraformaldehyde to preserve the aldehyde moiety and avoid reduction to the corresponding diol as we had observed earlier (see **51**, Scheme 14). To complete the total synthesis of (+)-akuammiline (**3**), aldehyde **55** was elaborated through a multistep protocol involving acetylation, conversion to the methyl ester, and borane removal.⁴⁸ Synthetic (+)-akuammiline (**3**) was spectroscopically comparable to an authentic sample of the natural product.⁴⁹

Scheme 15. Total synthesis of (+)-akuammiline (3).



Structural Rearrangement of Methanoquinolizidines to Pyrrolidinoindolines

One interesting notion regarding the methanoquinolizidine-containing akuammiline alkaloids is that, biosynthetically, they serve as precursors to the pyrrolidinoindoline-containing akuammilines.²⁰ As such, we wondered if such structural rearrangements could efficiently be performed in the laboratory under mild reaction conditions. Exciting precedent for such conversions exists from the Zn/AcOH-mediated rearrangement of strictamine reported by Banerji and Chakrabarti in 1966, albeit without a reported yield.50 We were particularly enticed to attempt the rearrangement of the methanoquinolizidine-containing natural products (+)-strictamine (1) and (+)-akuammiline (3) using the mild reducing agent samarium diodide⁵¹ (Figure 3). Samarium diiodide has been utilized to reductively cleave C-N bonds of α -amino carbonyls, even in the context of akuammiline chemistry.^{5v} However, there have been no previous examples of the corresponding reductive cleavage of C-N bonds with SmI₂ using α -amino imine substrates. Given that previous reductive rearrangements of the methanoquinolizidine core have been low yielding,⁵⁰ we viewed this as an opportunity to improve their efficiencies.

The results of our rearrangement studies are shown in Figure 3. Upon treatment of (+)-strictamine (1) with SmI_2 in THF and methanol at 23 °C, the desired reductive rearrangement took place to furnish pyrrolidinoindoline compound 56. Subsequent methylation afforded (-)-10-demethoxyvincorine (57), a bioactive natural product isolated in 2014⁸ from *Alstonia macrophylla*, that had yet to be synthesized. Next, we applied the same reductive rearrangement conditions to (+)-akuammiline (3). To our delight, the rearrangement proceeded in quantitative yield to provide the unnatural akuammiline derivative 58. Of note, 58 bears a pyrrolidinoindoline core with vicinal quaternary stereocenters at C7 and C16, and is therefore reminiscent of complex akuammilines bearing those structural features.⁵²



Figure 3. Structural rearrangement to access pyrrolidinoindolines (-)-10-demethoxyvincorine (57) and 58.

60

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43 44

45

46

47

48

49

50

51

52

53

54

55

56

57 58 59

60

Conclusion

We have completed the first enantioselective total syntheses of five akuammiline alkaloids, in addition to several unnatural compounds. Our synthetic approach to the methanoquinolizidine-containing natural products 1-4 features a number of key steps, including: a) a Trost desymmetrization to govern absolute stereochemistry, b) a gold-mediated cyclization to construct the [3.3.1]-azabicycle, c) a reductive interrupted Fischer indolization to arrive at pentacycle 6, and d) late-stage formation of the methanoquinolizidine using a deprotection-cyclization cascade. In the case of (+)akuammiline (3) and what is arguably the most complex methanoquinolizidine-containing alkaloid. (-)-Ψakuammigine (4), late-stage introduction of the vicinal quaternary stereocenters was achieved by employing challenging C16 alkylations. Additionally, in the case of $(-)-\Psi$ akuammigine (4), a means to install the furoindoline was crafted, while avoiding an undesired furoindoline isomerization that plagued our early approaches. Lastly, bioinspired reductive rearrangements of (+)-strictamine (1) and (+)akuammiline (3) provided (-)-10-demethoxyvincorine (57) and a C16 analogue thereof 58, by rearrangement of the methanoquinolizidine core to pyrrolidinoindoline scaffolds.

These efforts mark the first total syntheses of some of the most complex akuammilines known, including those that possess both a methanoquinolizidine core and vicinal quaternary stereocenters. It is expected that our studies will enable biological investigations of akuammilines and unnatural analogs. Moreover, the lessons learned from our synthetic endeavors, including the delicate late-stage manipulations and various undesired setbacks, will inform synthetic forays toward akuammilines and other challenging natural products.

ASSOCIATED CONTENT

Supporting Information

Detailed experimental procedures and compound characterization data. This material is available free of charge via the Internet at <u>http://pubs.acs.org</u>.

AUTHOR INFORMATION

Corresponding Author

neilgarg@chem.ucla.edu

Author Contributions

[†]E.P. and L.A.M. contributed equally.

ACKNOWLEDGMENT

The authors are grateful to the National Science Foundation (CHE-0955864 and CHE-1464898), Bristol–Myers Squibb, the Dreyfus Foundation, the UCLA Gold Shield Alumnae, and the University of California, Los Angeles for financial support. We are grateful to the NIH-NIGMS (F31-GM113642 to J. M., F31-GM117945 to E.P., and F31-GM121016 to L.A.M.), the Foote Family (J.M., E.P., and L.A.M.), the Majeti-Alapati family (E.P.), and the UCLA Cota-Robles Fellowship Program (E.P.). J. M. S. acknowledges the National Science Foundation GRFP (DGE-1144087) and the UCLA Graduate Division for a Dissertation Year Fellowship. We are grateful to Professor T.-S. Kam (University of Malaya) for providing authentic samples of aspidophylline A, strictamine, and 2(*S*)-cathafoline and Professors L. Evanno and E. Poupon (Université Paris-Saclay) for providing authentic samples of (+)-akuammiline and (-)-Ψ-akuammigine. These studies were supported by shared instrumentation grants from the NSF (CHE-1048804) and the National Center for Research Resources (S10RR025631).

REFERENCES

¹ Baliga, M. S. Chin. J. Integr. Med. 2012, DOI: 10.1007/s11655- 011-0947-0.

² For reviews on akuammiline alkaloids, see: (a) Ramírez, A.; García-Rubio, S. *Curr. Med. Chem.* **2003**, *10*, 1891–1915. (b) Eckermann, R.; Gaich, T. *Synthesis* **2013**, *45*, 2813–2823. (c) Smith, J. M.; Moreno, J.; Boal, B. W.; Garg, N. K. *Angew. Chem., Int. Ed.* **2015**, *54*, 400–412. (d) Adams, G. L.; Smith, A. B. The Chemistry of the Akuammiline Alkaloids. In *The Alkaloids*; Knolker, H.-J., Ed.; Elsevier: New York, 2016; Vol. *76*, p 171.

³ www.reaxys.com (accessed March 26, 2018).

⁴ Bonjoch, J.; Solé, D. Chem. Rev. 2000, 100, 3455-3482.

⁵ (a) Zhang, M.; Huang, X.; Shen, L.; Qin, Y. J. Am. Chem. Soc. 2009, 131, 6013-6020. (b) Zu, L.; Boal, B. W.; Garg, N. K. J. Am. Chem. Soc. 2011, 133, 8877-8879. (c) Zi, W.; Xie, W.; Ma, D. J. Am. Chem. Soc. 2012, 134, 9126-9129. (d) Adams, G. L.; Carroll, P. J.; Smith III, A. B. J. Am. Chem. Soc. 2012, 134, 4037-4040. (e) Horning, B. D.; MacMillan, D. W. C. J. Am. Chem. Soc. 2013, 135, 6442-6445. (f) Adams, G. L.; Carroll, P. J.; Smith III, A. B. J. Am. Chem. Soc. 2013, 135, 519-528. (g) Smith, M. W.; Snyder, S. A. J. Am. Chem. Soc. 2013, 135, 12964-12967. (h) Smith, J. M.; Moreno, J.; Boal, B. W.; Garg, N. K. J. Am. Chem. Soc. 2014, 136, 4504-4507. (i) Teng M.; Zi, W.; Ma, D. Angew. Chem., Int. Ed. 2014, 53, 1814-1817. (j) Ren, W.; Wang, Q.; Zhu, J. Angew. Chem., Int. Ed. 2014, 53, 1818-1821. (k) Smith, J. M.; Moreno, J.; Boal, B. W.; Garg, N. K. J. Org. Chem. 2015, 80, 8954-8967. (1) Moreno, J.; Picazo, E.; Morrill, L. A.; Smith, J. M.; Garg, N. K. J. Am. Chem. Soc. 2016, 138, 1162-1165. (m) Ren, W.; Wang, Q.; Zhu, J. Angew. Chem., Int. Ed. 2016, 55, 3500-3503. (n) Jiang, S.-Z.; Zeng, X.-Y.; Liang, X.; Lei, T.; Wei, K.; Yang, Y.-R. Angew. Chem., Int. Ed. 2016, 55, 4044-4048. (o) Nishiyama, D.; Ohara, A.; Chiba, H.; Kumagai, H.; Oishi, S.; Fujii, N.; Ohno, H. Org. Lett. 2016, 18, 1670-1673. (p) Eckermann, R.; Breunig, M.; Gaich, T. Chem. Commun. 2016, 52, 11363-11365. (q) Wang, T.; Duan, X.; Zhao, H.; Zhai, S.; Tao, C.; Wang, H.; Li, Y.; Cheng, B.; Zhai, H. Org. Lett. **2017**, *19*, 1650–1653. (r) Smith, M. W.; Zhou, Z.; Gao, A. X.; Shimbayashi, T.; Snyder, S. A. Org. Lett. **2017**, *19*, 1004–1007. (s) Wang, D.; Hou, M.; Ji, Y.; Gao, S. Org. Lett. 2017, 19, 1922-1925. (t) Eckermann, R.; Breunig, M.; Gaich, T. Chem. Eur. J. 2017, 23, 3938-3949. (u) Xiao, T.; Chen, Z.-T.; Deng, L.-F.; Zhang, D.; Liu, X.-Y.; Song, H.; Qin, Y. Chem. Commun. 2017, 53, 12665-12667. (v) Xie, X.; Wei, B.; Li, G.; Zu, L. Org. Lett. 2017, 19, 5430-5433. (w) Li, Y.; Zhu, S.; Li, J.; Li, A. J. Am. Chem. Soc. 2016, 138, 3982-3985. (x) Li, G.; Xie, X.; Zu, L. Angew. Chem., Int. Ed. 2016, 55, 10483-10486. (y) Zhang, B.; Wang, X, Cheng, C.; Sun, D.; Li, C. Angew. Chem., Int. Ed. 2017, 56, 7484-7487.

⁶ Equilibrium geometry calculations for the methanoquinolizidine scaffold were performed using Spartan '16 (B3LYP/6-31(G)). The 3D representation was prepared from the calculated equilibrium geometry using CYL-view. For CYLview, see: Legault, C. Y. CYLview, 1.0b; Université de Sherbrooke: Quebec, 2009; http://www.cylview.org.

try 2014, 98, 204-215.

1677-1681.

1

15 16 17

13

14

18 19

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55 56

57 58

¹⁴ Komatsu, Y.; Yoshida, K.; Ueda, H.; Tokuyama, H. *Tetrahedron Lett.* **2013**, *54*, 377–380.

(4), see: Henry, T. A. J. Chem. Soc. 1932, 2759-2768.

Bosch, J. J. Org. Chem. 1996, 61, 1239-1251.

Cook, J. M. J. Nat. Prod. 2012, 75, 181-188.

by, L. J.; Nelson, S. J. J. Org. Chem. 1973, 38, 2882-2887.

¹⁵ Kawano, M.; Kiuchi, T.; Negishi, S.; Tanaka, H.; Hoshikawa, T.; Matsuo, J.; Ishibashi, H. Angew. Chem., Int. Ed. 2013, 52, 906–910.

⁷ For the initial isolation of (+)-strictamine (1), see: (a) Schnoes, H. K.;

Biemann, K.; Mokrý, J.; Kompis, I.; Chatterjee, A.; Ganguli, G. J. Org.

Chem. 1966, 31, 1641-1642. For the X-ray structure of strictamine, see:

(b) Ahmad, Y.; Fatima, K.; Occolowitz, J. L.; Solheim, B. A.; Clardy, J.;

Garnick, R. L.; Le Quesne, P. W. J. Am. Chem. Soc. 1977, 99, 1943-1946.

⁸ For the initial isolation of (-)-2(S)-cathafoline (2), see: Lim, S.-H.; Low,

Y.-Y.; Sinniah, S. K.; Yong, K.-T.; Sim, K.-S.; Kam, T.-S. Phytochemis-

⁹ For the initial isolation of (+)- akuammiline (3) and (-)- Ψ -akuammigine

¹⁰ (a) Dolby, L. J.; Esfandiabi, Z. J. Org. Chem. 1972, 37, 43-46. (b) Dol-

¹¹ (a) Bennasar, M.-L.; Zulaica, E.; López, M.; Bosch, J. Tetrahedron Lett.

1988, 29, 2361-2364. (b) Bennasar, M.-L.; Zulaica, E.; Ramírez, A.;

¹² Koike, T.; Takayama, H.; Sakai, S.-I. Chem. Pharm. Bull. 1991, 39,

¹³ Edwankar, R. V.; Edwankar, C. R.; Namjoshi, O. A.; Deschamps, J. R.;

¹⁶ For our laboratory's communication detailing the first total syntheses of (+)-strictamine and (-)-2(*S*)-cathafoline, see ref 51. Simultaneously Zhu's group reported the total synthesis of (\pm) -strictamine, see ref 5m.

¹⁷ For the reductive interrupted Fischer indolization reaction, see: Maligres, P. E.; Houpis, I.; Rossen, K.; Molina, A.; Sager, J.; Upadhyay, V.; Wells, K. M.; Reamer, R. A.; Lynch, J. E.; Askin, D.; Volante, R. P.; Reider, P. J.; Houghton, P. *Tetrahedron* **1997**, *53*, 10983–10992.

¹⁸ For methodology studies of the interrupted Fischer indolization, see: (a) Boal, B. W.; Schammel, A. W.; Garg, N. K. Org. Lett. 2009, 11, 3458–3461. (b) Schammel, A. W.; Boal, B. W.; Zu, L.; Mesganaw, T.; Garg, N. K. Tetrahedron 2010, 66, 4687–4695. (c) Simmons, B. J.; Hoffmann, M.; Champagne, P. A.; Picazo, E.; Yamakawa, K.; Morrill, L. A.; Houk, K. N.; Garg, N. K. J. Am. Chem. Soc. 2017, 139, 14833–14836.

¹⁹ For a pertinent review of our laboratory's Fischer indolization studies, see: Susick, R. B.; Morrill, L. A.; Picazo, E.; Garg, N. K. Synlett **2017**, 28, 1–11.

²⁰ Scott, A. I. Acc. Chem. Res. **1970**, *3*, 151–157.

²¹ Hou, Y.; Cao, X.; Wang, L.; Cheng, B.; Dong, L.; Luo, X.; Bai, G.; Giao, W. J. Chromatogr. B **2012**, *908*, 98–104.

²² (a) Lu, Z.; Li, Y.; Deng, J.; Li, A. Nat. Chem. 2013, 5, 679–683. (b)
 Xiong, X.; Li, Y.; Lu, Z.; Wan, M.; Deng, J.; Wu, S.; Shao, H.; Li, A. Chem. Commun. 2014, 50, 5294–5297.

²³ For a review on catalytic asymmetric allylic alkylation employing heteroatom nucleophiles, see: Trost, B. M.; Zhang, T.; Sieber, J. D. Chem. Sci. 2010, *1*, 427–440.

²⁴ Ohno, M.; Otsuka, M. Org. React. 1989, 37, 1–55.

²⁵ Aoki, K.; Koga, K. Chem. Pharm. Bull. 2000, 48, 571–574.

²⁶ For the crystallographic data of **16**, refer to CCDC number 1440684.

²⁷ Salom-Roig, X. J.; Dénès, F.; Renaud, P. Synthesis 2004, 1903–1928.

²⁸ McDonald, C.; Holcomb, H.; Kennedy, K.; Kirkpatrick, E.; Leathers, T.; Vanemon, P. J. Org. Chem. **1989**, *54*, 1213–1215. ²⁹ For examples of the Fischer indolization being employed in late-stage total synthesis, see references 5 b,h,k,l, and 18c in addition to the following: (a) Ueda, H.; Satoh, H.; Matsumoto, K.; Sugimoto, K.; Fukuyama, T.; Tokuyama, H. *Angew. Chem., Int. Ed.* **2009**, *48*, 7600–7603. (b) Satoh, H.; Ueda, H.; Tokuyama, H. *Tetrahedron* **2013**, *69*, 89–95. (c) Krüger, S.; Gaich, T. *Eur. J. Org. Chem.* **2016**, 4893–4899.

³⁰ For an example of an indolenine reduction to an indoline in *Strychnos* alkaloid synthesis, see: Kim, J.-Y.; Suhl, C.-H.; Lee, J.-H.; Cho, C.-G. *Org. Lett.* **2017**, *19*, 6168–6171; see also references therein.

³¹ (a) Basha, A.; Lipton, M.; Weinreb, S. M. *Tetrahedron Lett.* **1977**, *18*, 4171–4174. (b) Shimizu, T.; Osako, K.; Nakata, T. *Tetrahedron Lett.* **1997**, *38*, 2685–2688.

³² Iwata, A.; Oshita, J.; Tang, H.; Kunai, A. J. Org. Chem. **2002**, 67, 3927–3929.

³³ (a) Liotta, D.; Sunay, U.; Santiesteban, H.; Markiewicz, W. J. Org. Chem. **1981**, 46, 2605–2610. (b) McMurry, J. Org. React. **1976**, 24, 187– 224.

³⁴ For the removal of nosyl groups using a solid-supported thiol reagent, see: Cardullo, F.; Donati, D.; Merlo, G.; Paio, A.; Salaris, M.; Taddei, M. *Synlett* **2005**, 2996–2998.

³⁵ Wang, H.; Reisman, S. E. Angew. Chem., Int. Ed. 2014, 53, 6206–6210.

³⁶ Duwiejua, M.; Woode, E.; Obiri, D. D. J. Ethnopharmacol. **2002**, *81*, 73–79.

³⁷ For a reported effort toward (–)-Ψ-akuammigine (**4**), see: Andreansky, E. S.; Blakey, S. B. *Org. Lett.* **2016**, *18*, 6492–6495.

³⁸ (a) Sung, M. J.; Lee, H. I.; Lee, H. B.; Cha, J. K. J. Org. Chem. 2003, 68, 2205–2008. (b) Shimada, K.; Kaburagi, Y.; Fukuyama, T. J. Am. Chem. Soc. 2003, 125, 4048–4049. (c) Yu, J.; Wearing, X. Z.; Cook, J. M. J. Am. Chem. Soc. 2004, 126, 1358–1359. (d) Yu, J.; Wearing, X. Z.; Cook, J. M. J. Org. Chem. 2005, 70, 3963–3979.

³⁹ García, A.; Castedo, L.; Domínguez, D. *Tetrahedron* **1996**, *52*, 5929–5932.

⁴⁰ Le, D. D.; Cortesi, A. T.; Myers, S. A.; Burlingame, A. L.; Fujimori, D. G. J. Am. Chem. Soc. 2013, 135, 2879–2882.

⁴¹ Equilibrium geometry calculations were performed using Spartan '16 (B3LYP/6-31(G)). The results indicated that furoindoline **46** is nearly 8 kcal/mol lower in energy than desired furoindoline **47**.

 42 Additional strategies were investigated. For example, we attempted to introduce the C16 quaternary stereocenter prior to Fischer indolization on a model system. However, Fischer indolization efforts using **i** were unsuccessful, presumably due to the difficulty of performing the [3,3]-sigmatropic rearrangement when trying to access vicinal quaternary stereocenters. With regard to late-stage efforts, intermediates such as diol **ii** could not be elaborated to substrates suitable for methanoquinolizidine formation.



⁴³ Wilk, A.; Grajkowski, A.; Phillips, L. R.; Beaucage, S. L. J. Org. Chem. 1999, 64, 7515–7522.

⁴⁴ (a) Stotter, P. L.; Friedman, M. D.; Dorsey, G. O.; Shiely, R. W.; Williams, R. F.; Minter, D. E. *Heterocycles* **1987**, *25*, 251–258. (b) Lu, X.; Cook, J. M. Org. Lett. **2001**, *3*, 4023–4026. (c) Lu, X.; Deschamp, J. R.; Cook, J. M. Org. Lett. **2002**, *4*, 3339–3342. (d) Liao, X.; Zhou, H.; Wearing, X. Z.; Ma, J.; Cook, J. M. Org. Lett. **2005**, *7*, 3501–3504.

59 60

⁴⁵ Schwartz, M. A.; Rose, B. F.; Vishnuvajjala, B. J. Am. Chem. Soc. 1973, 95, 612–613.

⁴⁶ For (–)-Ψ-akuammigine (4) characterization data, see: Hu, W.-L.; Zhu, J.-P.; Hesse, M. *Planta Med.* **1989**, *55*, 463–466.

 47 Attempts to install the furoindoline from aldehyde **55** en route to (–)- Ψ -akuammigine (**4**) were unsuccessful.

⁴⁸ Picot, A.; Lusinchi, X. Bull. Soc. Chim. Fr. 1977, 1227–1234.

⁴⁹ For (+)-akuammiline (3) characterization data, see: Benayad, S.; Ahamada, K.; Lewin, G.; Evanno, L.; Poupon, E. *Eur. J. Org. Chem.* 2016, 1494–1499. ⁵⁰ For the Zn / strong acid-promoted rearrangement of the methanoquinolizidine in low yield (or no reported yield), see ref 7a and: (a) Savaskan, S.; Kompis, I.; Hesse, M.; Schmid, H. *Helv. Chim. Acta* **1972**, *55*, 2861– 2867. (b) Banerji, J.; Chakrabarti, R. *Indian J. Chem. Sect. B* **1984**, *23B*, 453–454.

⁵¹ Honda, T.; Ishikawa, F. Chem. Commun. **1999**, 1065–1066.

⁵² An exemplary akuammiline natural product of this variety is echitamine, first isolated in 1875. For the seminal isolation report, see: Group-Besanez, v. *Justus Liebigs Ann. Chem.* **1875**, *176*, 88–89.



