Concise total synthesis of aplysinellamides A and B

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Concise and efficient total syntheses of bromotyrosine-derived metabolites aplysinellamides A and B, isolated from Australian marine sponge *Aplysinella* sp., have been accomplished in seven steps. A condensation between cinnamic acid and Boc-D-lysine methyl ester was applied to form the amide skeleton as a key step.

Keywords: bromotyrosine, marine sponge, aplysinella species

Among more than 20,000 marine natural products, 2% of them were discovered from deep-water marine organisms.^{1, 2} Deep-water organisms are expected to produce metabolites having unique structures with interesting bioactivities, such as cytotoxic,³ anti-inflammatory,⁴ anti-bacterial,⁵ anti-malarial,⁶ anti-viral,⁷ anti-thrombotic,⁸ anti-trypanosomal,⁹ anti-neurodegenerative¹⁰ and estrogen.¹¹ For example, tetracyclic compound halenaquinone exhibits extremely potent activity towards farnesyltransferases (IC₅₀ 0.017–0.031 µM) and malaria (IC₅₀ 0.53–0.62 µM) and meroterpene sulfate fascioquinols A and B displayed promising Gram-positive selective antibacterial activity towards Staphylococcus aureus (IC₅₀ 0.9–2.5 µM).⁵

In 2014, Quinn *et al* isolated four new bromotyrosine derivatives aplysinellamides A–C (1–3) and aplysamine-1-N-oxide (4) with ApoE modulation activity from the Australian marine sponge *Aplysinella* sp. and assigned their structures by extensive NMR experiments (Fig. 1).¹² The unique structure of them intrigues many chemists. We initiated the total synthesis of aplysinellamides in order to investigate the potential bioactivity of the derivatives. We now report the first efficient total syntheses of aplysinellamides A (1) and B (2) from the accessible starting material D-lysine (5) in seven steps.

Results and discussion

Our initial retrosynthetic analysis of 1 and 2 is outlined in Scheme 1. We envisaged that the cinnamamide skeleton of 1 and 4 could be constructed through a coupling between amine 9 and the respective acids 11 and 16 (see Scheme 2). Amine 9, in turn, could be obtained from a cheap starting material D-lysine (5). Intermediates including methyl N^{α}-Boc-D-lysinate 9 and the cinnamic acids 11 and 16 were synthesised according to reported literature procedures.¹³⁻¹⁸ Accordingly, D-lysine (5) was reacted with ClCOOBn in the presence of CuSO₄ to form D-lys(Z)-OH 6 in 91% yield, followed by N-protection with (Boc)₂O to afford acid 7 in 96% yield.¹³ Acid 7 was alkylated with CH₃I smoothly and gave methyl ester 8 in 85% yield.¹⁴ Removal of the Cbz group from $\mathbf{8}$ by hydrogenation with 10% Pd/C in MeOH smoothly provided methyl Na-Boc-D-lysinate **9** in 90% yield.¹³ Cinnamic acid **11** was prepared through a two-step sequence. Veratraldehyde was mono-brominated with Br₂/AcOH to give 3-bromo-4-methoxybenzaldehyde 10 in 70% yield, which was subjected to Doebner-Knoevenagel condensation with malonic acid in the presence of pyridine and piperidine to afford cinnamic acid 11 in 92% yield.^{15,16} Similarly, 4-hydroxybenzaldehyde was dibrominated to afford



Fig. 1 Chemical structures of aplysinellamides A-C (1-3) and aplysamine-1-N-oxide (4).



Scheme 1 Retrosynthetic analysis of aplysinellamides A (1) and B (2).

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3,5-dibromo-4-hydroxybenzaldehyde 14 in 96% yield,¹⁷ which was methylated with CH₃I to give 3,5-dibromo-4methoxybenzaldehyde 15 in 97% yield.¹⁸ Then, aldehyde 15 was subjected to Doebner-Knoevenagel condensation to provide cinnamic acid 16 in 79% yield.¹⁶ Coupling of 9 with cinnamic acid 11 was carried out under similar conditions to give cinnamamide 12, which was saponified with 1M LiOH/ MeOH to give amino acid **13** in 61% yield in two steps.¹⁹ Finally, removal of the Boc group from 13 with TFA in DCM smoothly provided aplysinellamide A (1) in 91% yield as a white solid $([\alpha]_{D}^{23} - 3.2^{\circ}, c 0.007 \text{ in MeOH})$. Likewise, coupling of **9** with cinnamic acid 16 was carried out under similar conditions to give cinnamamide 17 in a yield of 70%, which was saponified to give amino acid 18. Subsequently, removal of the Boc group from 18 with TFA in DCM afforded aplysinellamide B (2) in 61% yield in two steps as a white solid $([\alpha]_D^{23} + 0.30^\circ, c \ 0.017$ in MeOH) (Scheme 2). Spectral data of 1 and 2 were consistent with those described for the natural products.¹²

Conclusions

In summary, we have developed a very concise route for the first total syntheses of aplysinellamides A and B starting from D-lysine in seven steps (overall yields: 39% for aplysinellamide A; 31% for aplysinellamide B). Syntheses of these natural products and their derivatives on a large scale could be realised by this route, which will facilitate further biological studies. Studies towards the structure modifications of these natural products for further pharmacological investigation are ongoing and will be reported elsewhere.

Experimental

Melting points were measured on a microscopic melting point apparatus. The IR spectra were recorded on a Bruker Tensor 27 FTIR spectrometer with a KBr disk or film. ¹H NMR and ¹³C NMR spectra were taken on a Bruker AV 300 or AV 400 MHz and 75 or 100 MHz spectrometer in DMSO- d_6 or CD₃OD, chemical shifts are given in parts per million (ppm) relative to TMS as an internal standard. Mass spectra and high resolution mass spectra were performed on Agilent Q TOF 6520 mass spectrometer with electron spray ionisation (ESI) as the ion mode. Optical rotations were recorded using a sodium lamp with a Rudolph Autopol I Automatic Polarimeter with 1 dm tube.

(R)-Methyl 6-amino-2-(tert-butoxycarbonylamino)hexanoate (9): Removal of Cbz group from 8 by hydrogenation with 10% Pd/C in MeOH for 14 h at room temperature smoothly provided methyl N^{α}-Boc-D-lysinate 9 as an oil;¹³ yield 90%. ¹H NMR (400 MHz, DMSO-d₆): δ 7.23 (br d, J = 7.7 Hz, 1H, C2NH), 3.88–3.94 (m, 1H, C2H), 3.61 (s, 3H, 1-COOCH₃), 3.07–3.15 (m, 2H, C6-NH₂), 1.53–1.58 (m, 2H, C1H), 1.37 (s, 9H, C(CH₃)₂), 1.23–1.32 (m, 6H, C3,4,5).

(E)-3-(3-Bromo-4-methoxyphenyl)acrylic acid (11): 3-Bromo-4-methoxybenzaldehyde (10) was subjected to Doebner–Knoevenagel condensation with malonic acid in the presence of pyridine and piperidine to afford cinnamic acid 11 as a yellow solid; yield 92%;¹⁶ m.p. 240–242 °C; ¹H NMR (400 MHz, DMSO- d_6): δ 12.31 (br s, 1H, COOH), 7.96 (s, 1H, C2'H), 7.70 (d, J = 8.6 Hz, 1H, C6'H), 7.51 (d, J = 16.0 Hz, 1H, C3H), 7.14 (d, J = 8.6 Hz, 1H, C5'H), 6.46 (d, J = 16.0 Hz, 1H, C2H), 3.89 (s, 3H, 4'-OCH₃).

(R,E)-6-[3-(3-bromo-4-methoxyphenyl)acrylamido]-2-[(tertbutoxycarbonyl)amino]hexanoic acid (13): Compound 11 (0.08 g, 0.3 mmol) was dissolved in anhydrous DMF (2 mL) and anhydrous



Scheme 2 Synthesis of aplysinellamide A and B. Reagents and conditions: (a) CuSO₄, aq. NaOH, r.t., 5h; CICOOBn, EDTA, r.t., 3h, 91%; (b) (Boc)₂O, 4M NaOH, Dioxane, H₂O, r.t., 14h; MeOH, r.t., 12h, 96%; (c) NaHCO₃, CH₃I, DMF, r.t., 36h, 85%; (d) 10% Pd/C (15 w/w %, MeOH, r.t., 14h, 95%; (e) Br₂, AcOH, r.t., 14h, 70%; (f) Br₂, AcOH, NaOAc, r.t., 1h, 96%; (g) CH₃I, K₂CO₃, DMF, 55 °C, 3h, 97%; (h) Malonic acid, pyridine, piperidine, reflux 12h, 92% for **11** and 79% for **16**; (i) EDC.HCI, HOBt, TEA, dry DMF, dry DCM, r.t., 16h, 70% for **17**; (j) LiOH.H₂O, MeOH, r.t., 16h, 61% for **13** in two steps; (k) TFA, dry DCM, r.t., 40min, 91% for **1**; 62% for **2** in two steps.

DCM (2 mL) and cooled to 0 °C. 1-Hydroxy-benzotriazole hydrate (0.05 g, 0.37 mmol), N-(3-dimethylaminopropyl)-N'ethylcarbodiimide hydrochloride (0.07 g, 0.37 mmol) and TEA (0.11 mL, 0.74 mmol) were added to the solution and the mixture was stirred at 0 °C for 1 h. Then compound 9 (0.07 g, 0.27 mmol) was dissolved in anhydrous DCM (2 mL) and added to the mixture. The mixture was stirred while warming to room temperature for 16 h. The reaction mixture was diluted with DCM (2mL) and washed with saturated NaHCO₃ (2×4 mL). The DCM layer was washed with water (4 mL), brine (4 mL), dried (Na₂SO₄), filtered, and evaporated under reduced pressure, yielding 0.1 g crude product (12) which was used for next step without further purification. The residue was dissolve in methanol (3 mL), water (1 mL) and cooled to 0 °C. LiOH.H₂O (0.04 g, 0.98 mmol) was added to the solution and stirred at room temperature for 16 h. The methanol was evaporated under reduced pressure, and the pH of the remaining water was adjusted to 3 with 3 M HCl. The aqueous solution was extracted with DCM (2×10 mL). The DCM layers were dried with Na2SO4 and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give 13 (0.084 g, 61% yield in two steps) as a white solid; m.p. 119–120 °C; ¹H NMR (400 MHz, DMSO-d_z): δ 8.00 (t, J = 5.6 Hz, 1H, N1H), 7.78 (d, J = 2.0 Hz, 1H, C2"H), 7.55 (dd, J = 8.8, 2.0 Hz, 1H, C6"H), 7.32 (d, J=15.8 Hz, 1H, C3'H), 7.15 (d, J=8.8 Hz, 1H, C5"H), 6.97–7.03 (m, 1H, N2H), 6.52 (d, *J* = 15.8 Hz, 1H, C2'H), 3.88 (s, 3H, OCH₂), 3.80-3.84 (m, 1H, C2H), 3.12-3.17 (m, 2H, C6H), 1.55-1.65 (m, 2H, C3H), 1.41-1.47 (m, 2H, C5H), 1.37 (s, 9H, C(CH₃)₃), 1.35–1.32 (m, 2H, C4H); ¹³C NMR (100 MHz, DMSO-d₆): δ 174.3 (C1), 164.8 (C1'), 156.1 (C4"), 155.6 (C=O of Boc), 136.6 (C3'), 131.6 (C2"), 129.1 (C1"), 128.4 (C6"), 121.4 (C2'), 112.9 (C3"), 111.1 (C5"), 77.9 (C(CH₃)₃), 56.4 (C2), 53.5 (C4"-OCH₃), 38.4 (C6), 30.5 (C3), 28.8 (C(CH₂)₂), 28.2 (C5), 23.1 (C4); HRMS (ESI) calcd for C₂₁H₂₀BrN₂O₆Na (M+Na) *m/z* 507.1101; found: 507.1123.

(R,E)-Methyl 2-[(tert-butoxycarbonyl)amino]-6-[3-(3,5-dibromo-4-methoxyphenyl)acrylamido] hexanoate (17): Compound 16 (0.20 g, 0.6 mmol) was dissolved in anhydrous DMF (3 mL) and anhydrous DCM (3 mL) and cooled to 0 °C. 1-Hydroxy-benzotriazole hydrate (0.10 g, 0.73 mmol), N-(3-dimethylaminopropyl)-N'ethylcarbodiimide hydrochloride (0.14 g, 0.73 mmol) and TEA (0.20 mL, 1.4 mmol) were added to the solution and the mixture was stirred at 0 °C for 1 h. Then compound 9 (0.19 g, 0.72 mmol) was dissolved in anhydrous DCM (2 mL) and added to the mixture. The mixture was stirred while warming to room temperature for 16 h. The reaction mixture was diluted with DCM (5 mL) and washed with saturated NaHCO₂ (2×5 mL). The DCM layer was washed with water (10 mL) and brine (10 mL), dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give 17 (0.24 g, 70% yield) as a white solid; m.p. 100–102 °C; ¹H NMR (400 MHz, DMSO-d₆): δ 8.04 (t, J = 5.8 Hz, 1H, N1H), 7.87 (s, 2H, C2",6"H), 7.32 (d, J = 15.8 Hz, 1H, C3'H), 7.23 (d, J = 8.0 Hz, 1H, N2H), 6.63 (d, J = 15.8 Hz, 1H, C2'H), 3.87–3.96 (m, 1H, C2H), 3.82 (s, 3H, C4"OCH₃), 3.61 (s, 3H, C1OCH₃), 3.12–3.17 (m, 2H, C6H), 1.53–1.66 (m, 2H, C3H), 1.41–1.48 (m, 2H, C5H), 1.37 (s, 9H, C(CH₃)₃), 1.30–1.33 (m, 2H, C4H); ¹³C NMR (100 MHz, DMSO- d_{ϵ}): δ^{-1} 73.2 (C1), 164.2 (C1'), 155.5 (C=O of Boc), 153.8 (C4"), 134.9 (C3'), 134.4 (C1"), 131.5 (C2",6"), 124.6 (C2'), 117.9 (C3",5"), 78.1 (<u>C</u>(CH₂)₂), 60.5 (C4"-O<u>C</u>H₂), 53.5 (C2), 51.7 (C1-OCH₃), 38.4 (C6), 30.3 (C3), 28.6 (C(CH₃)₃), 28.1 (C5), 23.0 (C4); HRMS (ESI) calcd for $C_{22}H_{30}Br_2N_2O_5Na$ (M+Na) m/z599.0363; found: 599.0398.

Aplysinellamide A (1): Compound 13 (0.15 g, 0.31 mmol) was dissolved in DCM: TFA (1 mL: 1 mL) and the reaction was stirred at r.t. for 40 min. The solvents were subsequently evaporated and the product crystallised with diethyl ether to give aplysinellamide A (1) (0.13 g, 91% yield) as a pale powder; m.p.>250 °C; $[\alpha J_D^{23} -3.2 \text{ (c } 0.007 \text{ MeOH)})$ (lit. ¹² $[\alpha J_D^{23} -3.1 \text{ (c } 0.007, \text{ MeOH)})$; IR (KBr): 2923, 2854, 1650, 1595, 1499, 1260, 1050 cm⁻¹; ¹H NMR (400 MHz, CD₃OD and one drop of TFA): δ 7.77 (d, J = 2.0 Hz, 1H, C2"H), 7.50 (dd, $J_q = 8.6 \text{ Hz}$, $J_q = 2.0 \text{ Hz}$, 1H, C6"H), 7.43 (d, J = 15.6 Hz, 1H, C3'H), 7.04 (d,

J = 8.6 Hz, 1H, C5"H), 6.47 (d, J = 15.6 Hz, 1H, C2'H), 3.95–3.98 (m, 1H, C2H), 3.90 (s, 3H, OCH₃), 3.33–3.35 (m, 2H, C6H), 1.90–2.01 (m, 2H, C3H), 1.57–1.65 (m, 2H, C5H), 1.43–1.54 (m, 2H, C4H); ¹³C NMR (100 MHz, CD₃OD and one drop of TFA): δ 172.0 (C1), 169.0 (C1'), 158.8 (C4"), 140.4 (C3'), 133.4 (C2"), 130.5 (C1"), 130.1 (C6"), 120.8 (C2'), 113.5 (C5"), 113.3 (C3"), 57.1 (4"-OCH₃), 54.0 (C2), 40.2 (C6), 31.4 (C3), 30.3 (C5), 23.6 (C4); HRMS (ESI) calcd for C₁₆H₂₇BrN₂O₄ (M+H) *m/z* 385.0765; found: 385.0775.

Aplysinellamide B (2): Compound 17 (0.10 g, 0.17 mmol) was dissolved in methanol (3 mL) and water (1 mL) and cooled to 0 °C. LiOH.H₂O (0.04 g, 0.98 mmol) was added to the solution and the mixture was stirred at room temperature for 16 h. The methanol was evaporated under reduced pressure and the pH of the remaining water was adjusted to 3 with 3 M HCl. The aqueous solution was extracted with DCM (2×10 mL). The DCM layers were dried with Na₂SO₄ and evaporated under reduced pressure, yielding crude product (0.08 g). The residue was dissolved in DCM: TFA (1 mL:1 mL) and the reaction was stirred at r.t. for 40 min. The solvents were subsequently evaporated and the product crystallised with diethyl ether to give aplysinellamide B (2) (0.056 g, 62% yield in two steps) as a pale powder; m.p.>250 °C; $[\alpha]_{D}^{23}$ +0.30 (c 0.017, MeOH) (lit.¹² $[\alpha]_{D}^{23}$ +0.29 (c 0.017, MeOH); IR (KBr): 2927, 2857, 1634, 1419, 1260, 1205cm⁻¹; ¹H NMR (400 MHz, CD₂OD and one drop of TFA): δ 7.79 (s, 2H, C2",6"H), 7.39 (d, J = 15.8 Hz, 1H, C3'H), 6.56 (d, J = 15.8 Hz, 1H, C2'H), 3.97 (t, J = 6.4 Hz, 1H, C2H), 3.88 (s, 3H, C4"-OCH₂), 3.31-3.33 (m, 2H, C6H), 1.92-2.00 (m, 2H, C3H), 1.62-1.66 (m, 2H, C5H), 1.46-1.58 (m, 2H, C4H); 13C NMR (100 MHz, CD3OD and one drop of TFA): δ 172.0 (C1), 168.2 (C1'), 156.6 (C4"), 138.4 (C3'), 135.6 (C1"), 133.2 (C2",6"), 124.1 (C2'), 119.8 (C3",5"), 61.4 (4"-OCH₂), 54.0 (C2), 40.2 (C6), 31.3 (C3), 30.2 (C5), 23.6 (C4); HRMS (ESI) calcd for $C_{16}H_{20}Br_2N_2O_4Na$ (M+Na) m/z = 484.9679; found: 484.9672.

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Electronic Supplementary Information

Supplementary information including ¹H NMR and ¹³C NMR data and spectra of aplysinellamide A (1), aplysinellamide B (2) and compounds 13 and 17, has been deposited online, at stl.publisher.ingentaconnect.com/content/stl/jcr/supp-data

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References

- 1 D. Skropeta, Nat. Prod. Rep., 2008, 25, 1131.
- 2 K. Machida, T. Abe, D. Arai, M. Okamoto, I. Shimizu, N.J. De Voogd, N. Fusetani and Y. Nakao, Org. Lett., 2014, 16, 1539.
- 3 H. Zhang, M.M. Conte and R.J. Capon, Angew. Chem. Int. Ed., 2010, 49, 9904.
- 4 P.L. Katavic, K.W.L. Yong, J.N. Herring, M.A. Deseo, J.T. Blanchfield, V. Ferro and M.J.P. Garson, *Tetrahedron*, 2013, **69**, 8074.
- 5 J. Wang, M.-L. Bourguet-Kondracki, A. Longeon, J. Dubois, A. Valentin and B.R. Copp, *Bioorg. Med. Chem. Lett.*, 2011, 21, 1261.
- 6 E. Gros, A. Al-Mourabit, M.-T. Martin, J. Sorres, J. Vacelet, M. Frederich and M. Aknin, J. Nat. Prod., 2014, 77, 818.
- 7 A. Furuta, K.A. Salam, I. Hermawan, N. Akimitsu, J. Tanaka, H. Tani, A. Yamashita, K. Moriishi, M. Nakakoshi, M. Tsubuki, P. Peng, Y. Suzuki, N. Yamamoto, Y. Sekiguchi, S. Tsuneda and N. Noda, *Mar. Drugs*, 2014, 12, 462.

- 8 E. Sakai, H. Kato, H. Rotinsulu, F. Losung, R.E.P. Mangindaan, N.J. De Voogd, H. Yokosawa and S. Tsukamoto, J. Nat. Med., 2014, 68, 215.
- 9 Y. Feng, R.A. Davis, M.L. Sykes, V.M. Avery and R.J. Quinn, *Bioorg. Med. Chem. Lett.*, 2012, 22, 4873.
- 10 U.R. Abdelmohsen, C. Cheng, C. Viegelmann, T. Zhang, T. Grkovic, S. Ahmed, R.J. Quinn, U. Hentschel and R. Edrada-Ebel, *Mar. Drugs*, 2014, 12, 1220.
- 11 K. Machida, T. Abe, D. Arai, M. Okamoto, I. Shimizu, N.J. De Voogd, N. Fusetani and Y. Nakao, Org. Lett., 2014, 16, 1539.
- 12 L.-W. Tian, Y. Feng, Y. Shimizu, T. Pfeifer, C. Wellington, J.N.A. Hooper and R.J. Quinn, J. Nat. Prod., 2014, 77, 1210.
- 13 H. Shao, J. Seifert, N.C. Romano, M. Gao, J.J. Helmus, C.P. Jaroniec, D.A. Modarelli and J.R. Parquette, *Angew. Chem. Int. Ed.*, 2010, 49, 7688.

- 14 J.J Nestor, M.J. Ernest and T.H. Ho, Eur. Pat. Appl. EP 472220 A1, 19910823, (1991).
- 15 N.U. Baratov, E.G. Mil'grom, V.I. Vinogradov, Y.V. Rashkes and M.S. Yunusov, *Chem. Nat. Compd.*, 1993, 29, 748
- 16 N. Ullah and K.M. Arafeh, Tetrahedron Lett., 2009, 50, 158
- 17 G. Zhu, F. Yang, R. Balachandran, P. Höök, R.B. Vallee, D.P. Curran and B.W. Day, J. Med. Chem., 2006, 49, 2063.
- 18 M. Takahashi, A. Yamamoto, T. Inuzuka, T. Sengoku and H. Yoda, *Tetrahedron*, 2011, 67, 9484.
- 19 S. Gamsey and R. Wessling, World Pat. Appl. WO 2009009756 A2, 20080711, (2008).