

Enantioselective Synthesis of Guaianolides in the Osmitopsin Family by Domino Metathesis

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Supporting Information

ABSTRACT: Relay metathesis enabled an improved access from (S)-citronellal to the marine trisnorguaiane (-)-clavukerin A. This hydroazulene was applied as an advantageously functionalized building block for the asymmetric synthesis of the sesquiterpene lactone osmitopsin and the proposed structure of 4,5-epoxyosmitopsin using a chemo-, regio-, and diastereoselective diepoxide opening as the key step.



S esquiterpene lactones of the guaianolide type are often associated with interesting biological properties¹ and have recently attracted much interest from the synthetic community.^{2,3} The guaianolides osmitopsin (1) and 4,5-epoxyosmitopsin (2) were isolated from the aerial parts of *Osmitopsis asteriscoides*,⁴ a South African medicinal plant known for its antimicrobial activity (Figure 1).⁵



Figure 1. Naturally occurring guaianolides 1 and 2 (proposed structure) and non-natural diastereomers 3 and 4 of 2.

Assignment of the relative configuration depicted for 1 and 2 was mainly based on ¹H NMR arguments, while the absolute configuration was unknown. In particular, experiments using the shift reagent $Eu(fod)_3$ with epoxy lactone 2 and the isomeric compound 3 obtained by epoxidation of 1 with *meta*-chloroperbenzoic acid led Bohlmann to propose a *cis* relationship between the epoxide oxygen and the γ -lactone attached to the hydroazulene core in 2.^{4a} Whereas the

bioactivity of 1 and 2 has not yet been investigated, the racemic C-10 epimer 4 of 2 synthesized by Posner shows significant activity against schistosomal cercariae,⁶ worm larvae that cause bilharziosis.⁷ Herein we report a concise synthesis of osmitopsin (1) and the proposed structure of 4,5-epoxyosmitopsin (2) from (S)-citronellal (9).

We first focused on the synthesis of 4,5-epoxyosmitopsin (2), and Scheme 1 illustrates our original retrosynthetic analysis for this structure. Since many methods are known for attachment of an *exo* methylene unit to a γ -butyrolactone,⁸ we selected the saturated lactone **5** as a precursor to **2** that should be available by intramolecular Mitsunobu reaction with inversion of configuration⁹ from hydroxy acid **6**. Carboxylic acid **6** in turn might be obtained by selective nucleophilic opening of





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diepoxide 7 that was reported to be formed by epoxidation of the marine natural product (-)-clavukerin A (8).¹⁰ Hydroazulene 8 can be prepared from (S)-citronellal (9) in only four steps using a combined organocatalytic/metal-catalyzed strategy.¹¹

Since the chemoselectivity of the domino metathesis reaction of dienyne 11 to give hydroazulene 8 and hexalin 14 was only moderate (8:14 = 2:1) so far,¹¹ we tried to improve this issue by application of a relay strategy (Scheme 2).¹² As before,





asymmetric organocatalytic Michael addition of 9 to methyl vinyl ketone and subsequent chemoselective dibromoolefination efficiently led to ketone 10,¹¹ which now was subjected to a Wittig reaction with unsaturated ylide 12^{13} followed by alkyne formation with butyllithium in a one-pot procedure.^{14,15} The monosubstituted olefin of the resulting trienyne 13 was envisioned to react preferably with a ruthenium carbene complex to place the metal left to the triple bond after release of cyclopentene. Thus, higher selectivity for production of the desired hydroazulene was anticipated.¹⁶ Indeed, using the Grubbs-II catalyst **G-II** in the presence of *p*-benzoquinone, 1^{7} an efficient relay metathesis succeeded to give a 13:1 mixture of hydroazulene and hexalin, from which pure (-)-clavukerin A (8) was isolated in 92% yield. If the Grubbs catalysts G-I and G-II were used successively, the isomer ratio was even improved to 25:1, and the isolated yield of 8 increased to 96%.

With this more selective access to (-)-clavukerin A (8), we then studied bisepoxidation of the 1,3-diene (Scheme 3). To our surprise, using *meta*-chloroperbenzoic acid we obtained the *anti* diepoxide 7 reported by Kitagawa and secured by X-ray diffraction analysis¹⁰ only in small amounts next to the *syn* diepoxide 15 as the major product, while the total epoxide yield was rather low. Finally, we found that methyl(trifluoromethyl)-dioxirane¹⁸ produces only 15 in nearly quantitative yield, and





we could also confirm the relative configuration of this compound by X-ray diffraction analysis.¹⁹ Due to the undesired configuration of the epoxide at the five-membered ring of 15, we had to modify our synthetic strategy. As originally intended, we first studied opening of the diepoxide with different carbon nucleophiles. Whereas experiments with dilithiated acetic acid,^{20*} diethylethoxyalkynylalane,^{20c,21} or a vinylcuprate²² were not successful, reaction with an allylcopper species from allylmagnesium bromide and catalytic amounts of copper cyanide²³ afforded the desired product 16 cleanly in a completely chemo-, regio-, and diastereoselective fashion. Deoxygenation of the remaining epoxide then proceeded with high efficiency by oxidation of **16** with pyridinium dichromate²⁴ followed by reaction of the resultant epoxy ketone 17 with molybdenum hexacarbonyl²⁵ to give dienone 18. Reduction of this substrate with the bulky hydride lithium selectride provided the desired β -alcohol 19 as a single stereoisomer. Unexpectedly, vanadium-catalyzed hydroxyl-directed epoxidation²⁶ of this alcohol with tert-butylhydroperoxide not only furnished the desired β -epoxide 21 but led to a 1.2:1 mixture of diastereomers 20 and 21. Under catalysis with molybdenum hexacarbonyl, 26b the β -epoxide 21 was formed preferentially, albeit with a lower total epoxide yield. As the unwanted α epoxide 20 can be oxidized quantitatively with PDC in DMF to return epoxy ketone 17, vanadium catalysis is overall more efficient, especially since separation of the diastereomeric epoxides 20 and 21 succeeds without problems.

Before proceeding with β -epoxide 21 to the target molecule 2, we tested a range of conditions for α -methylene γ -

Scheme 4. Completion of the Synthesis of Guaianolide 2 and Its C-4/C-5 Bisepimer 3



butyrolactone construction commencing with α -epoxy diastereomer 20 (Scheme 4). The optimum conditions found for conversion of this substrate to lactone 22 involved catalytic dihydroxylation with osmium tetroxide and subsequent 1,2-diol cleavage with (diacetoxyiodo)benzene²⁷ to give a mixture of cyclic hemiacetals followed by pyridinium dichromate oxidation.²⁸ The relative configuration of epoxy lactone 22 was unambiguously established by X-ray diffraction analysis.¹¹ Condensation of 22 with Bredereck's reagent 23^{29} then delivered enamino lactone 24 in high yield that was readily converted to α -methylene lactone 3 by DIBAL reduction.³ Compound 3 turned out to be identical to the epoxy lactone obtained by Bohlmann through epoxidation of the guaianolide osmitopsin (1) with meta-chloroperbenzoic acid.44,31 Transformation of β -epoxide 21 into lactone 5 was best carried out by oxidative scission of the terminal olefin via dihydroxylation/ 1,2-diol cleavage as for 20 followed by oxidation with TEMPO and (diacetoxyiodo)benzene.³² Epoxy lactone 5 also provided suitable crystals for X-ray diffraction analysis that unequivocally proved its relative configuration.¹⁹ While condensation of 5 with Bredereck's reagent 23 led to the corresponding enamino lactone in high yield (91%), DIBAL reduction of this compound did not proceed as smoothly as for 24, and furnished the target molecule 2 in 44% yield next to a presumed overreduction product, the separation of which was not trivial.³³ After considerable experimentation, we found that exo methylenation of 5 was best achieved by alkylation of the lactone enolate with Eschenmoser's salt $25^{6,34}$ under strict temperature control followed by quaternization of the resultant tertiary amine 26 with methyl iodide and β -elimination with DBU.³⁵ From this reaction sequence we finally isolated compound 2 in good overall yield. Surprisingly, the data obtained for this compound showed significant deviation from the values reported in the literature. Ha, 36 Thus, we have to conclude that the structural assignment for 4,5-epoxyosmitopsin has been disproved by our synthesis of 2. The natural product is most likely a diastereomer of structure 2, and current efforts are devoted to identifying the correct stereochemistry.

For the synthesis of osmitopsin (1) the route depicted in Scheme 5 eventually turned out to give optimal results. Epoxy ketone 17 available in three steps from clavukerin A (8) was directly converted into epoxy alcohol 20 by L-selectride reduction, which saved two steps compared to its preparation from 8 according to Scheme 3. Transformation of 20 to epoxy lactone 22 as before then allowed a chemoselective epoxide deoxygenation with a lower valent tungsten halide to give Scheme 5. Streamlined Access to Epoxy Alcohol 20 and Synthesis of Guaianolide 1



olefinic lactone 27.^{37,38} Finally, *exo* methylenation of 27 was accomplished as for 5 using Eschenmoser's salt 25 to yield α -methylene lactone 1 as a colorless oil. This compound had ¹H NMR data identical to those reported for the guaianolide osmitopsin and furnished epoxy lactone 3 upon reaction with *meta*-chloroperbenzoic acid as described by Bohlmann.^{4a,31,39}

Due to the relay metathesis strategy and the highly selective diepoxide opening reaction as the crucial elements of our synthetic route, only 14 steps were required to arrive at the sesquiterpene lactone osmitopsin (1) and 15 steps to secure the proposed structure of 4,5-epoxyosmitopsin (2) from (S)-citronellal (9). Moreover, with the improved access to (-)-clavukerin A (8) developed in the course of this synthesis, further applications of this compound as an advanced enantiomerically pure building block for structurally more complex hydroazulenes are easily foreseeable.

ASSOCIATED CONTENT

Supporting Information

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The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.6b01619.

Letter

X-ray structure of 15 (CIF) X-ray structure of 5 (CIF) X-ray structure of 22 (CIF)

Experimental procedures, spectroscopic data, and ¹H and ¹³C NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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