

Induction and Control of Columnar Mesophase by Charge Transfer Interaction and Side Chain Structures of Tetrasubstituted Pyrenes

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The induction of the columnar phase is described in which intrinsically crystalline 1,4,6,9-tetraacyloxy pyrenes have been mixed with tetranitrofluorenone, TeNF, as an electron acceptor. The pyrene derivative with linear nonyl chains, PC₉, has not shown any mesophases using trinitrofluorenone, suggesting that stronger charge transfer interaction is favored by TeNF. The shape of the substituents works to control the mesophase type: the mixture of PC₉ and TeNF has formed a rectangular columnar phase, and a hexagonal columnar phase is attained by rounder derivatives bearing branched side chains.

Keywords: charge transfer interaction; columnar mesophase; side chain structures; tetranitrofluorenone; tetrasubstituted pyrenes

INTRODUCTION

The organization of molecular materials is attracting a lot of interest. In recent years, self-assembly has been recognized as a powerful method of molecular organization in various fields [1]. Highly organized molecular assembly is important in nanotechnology as well as in biological systems [2]. Various interactions are applied for self-assembly and, among them, π - π interaction is a well known but powerful tool to build up one-dimensional aromatic architecture [3]. In spite of the fact that one-dimensional stacking is a simple architecture, the liquid crystalline (LC) one, an ordered columnar phase, is

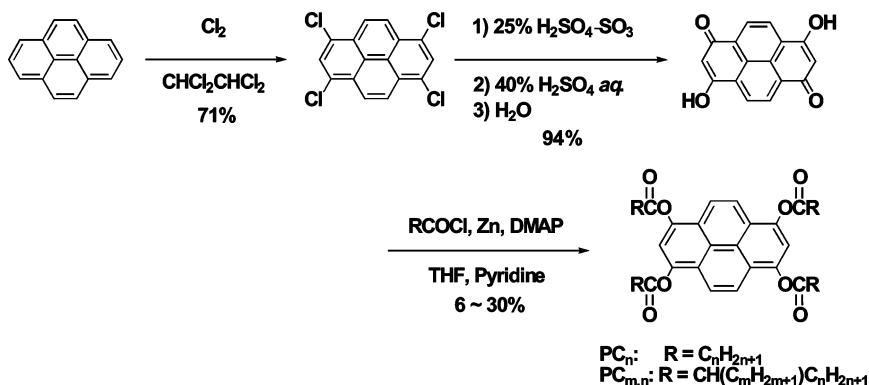
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important for various applications [4] due to its anisotropic and thermotropic properties, especially for electronic applications [3b, 5–9]. However, the methodology to intentionally manifest the LC phase has not been well established. Here we report on the induction and control of the columnar mesophase from crystalline pyrene derivatives by taking advantage of the electronic and structural characteristics of 2,4,5,7-tetranitro-9-fluorenone, TeNF [10].

Most discotic compounds showing a columnar phase have large and highly symmetric aromatic cores resulting from high melting temperatures [4–6] and less solubility in common organic solvents [11]. In this study, therefore, we chose pyrene, which has an aromatic core of C_2 symmetry and designed some derivatives to achieve columnar LC phases of lower melting temperatures. As far as we know, only a few simple pyrene derivatives are known to have low melting temperatures [7,8].

RESULTS AND DISCUSSION

The typical synthetic procedure for pyrene derivatives is shown in Scheme 1 [7,12]. Simple crystalline esters PC_n ($n = 8, 9, 11$) [7] were prepared with long linear alkyl halides while branched side chains were introduced as $PC_{m,n}$ ($m,n = 3,3, 2,4, 5,5$) to reduce the π - π interactions of the central cores. The DSC measurements showed, however, that they were all crystalline and I-Cr transition temperatures were higher for $PC_{m,n}$ than PC_n except $PC_{3,3}$ (Table 1). The transition of $PC_{3,3}$ at 71°C was expected to be a crystal-crystal one from polarizing optical microscopic (POM) observation. Although the effect of



SCHEME 1 Synthesis of tetrasubstituted pyrenes.

TABLE 1 Thermal Properties of PC_n and PC_{m,n}

Pyrene derivatives	I		X		Cr
PC ₈	•	76 (36.8)	—		•
PC ₉	•	67 (49.1)	—		•
PC ₁₁	•	72 (46.5)	—		•
PC _{3,3}	•	172 (48.1)	•	71 (35.2)	•
PC _{2,4}	•	130 (23.5)	—		•
PC _{5,5}	•	151 (35.8)	—		•

PC_n: On cooling at 5°C min⁻¹, PC_{m,n}: On cooling at 10°C min⁻¹.

I: Isotropic, X: Undetermined, Cr: Crystalline.

branched side chains is ambiguous, Table 1 still shows that they reduce the transition enthalpy about 10 J g⁻¹.

It is known that thermotropic LCs can be formed when 2,4,7-trinitro-9-fluorenone, TrNF [13], is mixed with discotic compounds, which do not show mesomorphism themselves [14–16]. Electron-rich discotic compounds are expected to form charge transfer (CT) complexes with electron-deficient TrNF. In this study, PC₉ and TrNF were mixed in 2:1, 3:2, and 1:1 ratios to obtain a red solid. Although the color change suggested the formation of CT complexes, no mesophases were observed for any mixtures. The interaction between PC₉ and TrNF may not be strong enough to form stable cores for LC phases.

Accordingly, TeNF was prepared [10] to increase the interaction with the pyrene derivatives. The mixture of PC₉ and TeNF gave a dark brown solid in various ratios, suggesting stronger CT interaction than that between PC₉ and TrNF. Their thermal properties were studied with DSC and POM, and the results are summarized in Table 2. Table 2 suggests the LC phase for 2:1 ~ 1:1 PC₉–TeNF mixtures, and they were all Col_r phases as demonstrated by the mosaic and sea grass textures for the 1:1 PC₉–TeNF mixture (Fig. 1). The 4:3 mixture was found to have the most stable LC phase between 67–112°C.

To compare the difference between PC₉–TrNF and PC₉–TeNF, the UV-vis spectra of each 1:1 mixture were measured in the solid state, and they are shown in Fig. 2. Both spectra reach to the visible region (above 450 nm), which proves CT, considering that none of PC₉, TrNF, and TeNF have absorption in the same region. The λ_{max} at longer wavelength (555 nm) for PC₉–TeNF indicated a smaller energy gap between the HOMO of PC₉ and the LUMO of TeNF than that for PC₉–TrNF (λ_{max} = 501 nm). Higher electron acceptability of TeNF seems to affect the CT from PC₉, making a stable core and leading to the LC phase emergence. The HOMO-LUMO energy gaps were

TABLE 2 Thermal Properties of the Mixtures of PC₉, PC_{3.3}, PC_{2.4}, and PC_{5.5}-TeNF

Pyrene derivatives	PC _n -TeNF ratio	I	Col _r	Col _h	Cr
PC ₉	4:1	• 61 (45.9)	–	–	•
	5:3	• 83 (9.0)	•	– 66 (41.5)	•
	4:3	• 112 (19.4)	•	– 67 (27.6)	•
	6:5	• 75 (5.3)	•	– 66 (46.3)	•
	5:5	• ^a 75 (6.6)	–	• ^b 67 (35.4)	•
	5:6	• 56 (35.6)	–	–	•
PC _{3.3}	1:1	•	– 138 (5.6)	• 88 (24.9)	•
PC _{2.4}	4:3	•	– 102 (5.2)	• 40 (16.7)	•
PC _{5.5}	1:1	•	– 156 (5.1)	• 84 (19.1)	•

On cooling at 5°C min^{−1}.

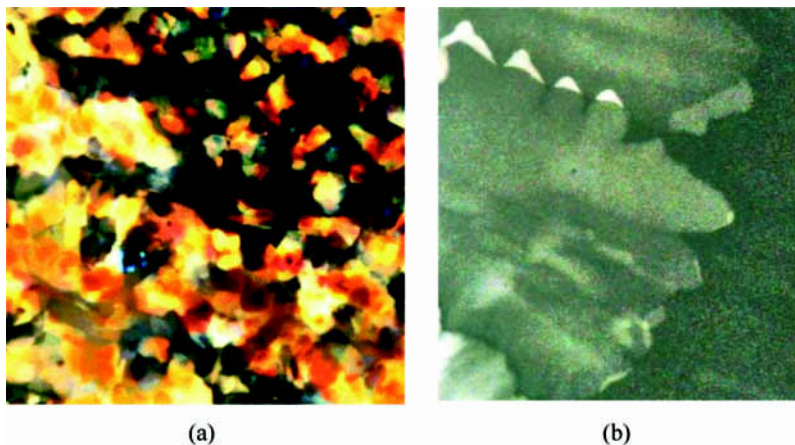
I: Isotropic, Col_r: Discotic rectangular columnar, Col_h: Discotic hexagonal columnar, Cr: Crystalline.

^aPhase separation of I and Col_h was observed.

^bPhase separation of Col_h and Cr was observed.

estimated for tetraacetyloxy pyrene, PC₁, TrNF, and TeNF by PM3 molecular orbital calculations [17]. The computations were performed with MM2 and MOPAC 97 programs. The HOMO energy of PC₁ was −7.98 eV and the LUMO energy levels were −2.94 and −3.34 eV for TrNF and TeNF, respectively.

Although all mixtures were crystalline at rt, no single crystals were obtained for the present PCs in Table 1. To obtain structural

**FIGURE 1** Mosaic (a) and sea grass (b) textures of Col_r of PC₉-TeNF (1:1) at 122°C.

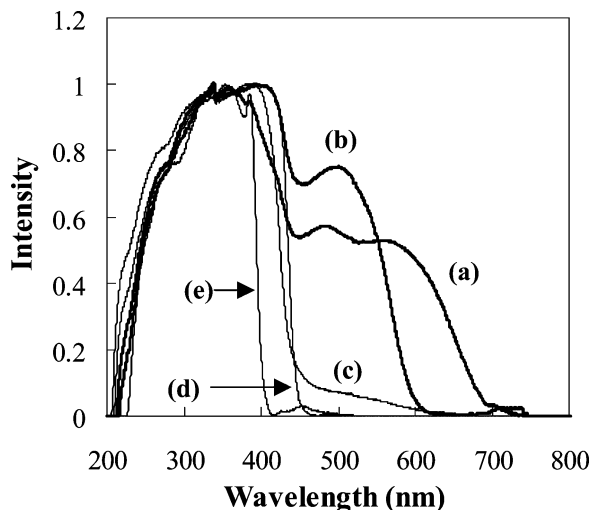


FIGURE 2 UV-vis spectra of (a) PC₉-TeNF (1:1), (b) PC₉-TrNF (1:1), (c) TeNF, (d) TrNF, and (e) PC₉ in their crystalline phases.

information, therefore, PC₁ was prepared, and the mixtures with TrNF and TeNF have colors similar to PC₉-TrNF and PC₉-TeNF. Their single crystals were analyzed by X-ray crystallography and the Ortep diagrams for PC₁-TrNF and PC₁-TeNF are shown in Figs. 3 and 4, respectively.

For PC₁-TrNF, the columnar structure of their rings was formed in a 1:1 ratio with PC₁ and TrNF aromatic rings parallel to each other. The distance was 3.34 Å but TrNF was disordered (Fig. 3). In addition, two characteristics were found: (1) the long axes of PC₁ and TrNF are almost crossed with each other and (2) the center of PC₁ deviated from that of TrNF. These should be due to the asymmetrical charge density of the LUMO. On the other hand, the PC₁-TeNF crystal structure was constructed in a 1:2 ratio, even prepared from the 1:1 mixture. The units were arranged in a zigzag way along the *b* axis direction, and the unit was perpendicular to the neighboring units, as shown in Fig. 4. In addition, three interesting features were observed: (1) the long axes of PC₁ and TeNF are almost eclipsed, but (2) the PC₁ and TeNF rings are not parallel to each other with the center-to-center distance of 3.84 Å, whereas (3) the two TeNF rings are parallel. The symmetrical structure and the four bulky nitro groups of TeNF should be their cause but the reason for the 1:2 composition ratio is unexplained at present.

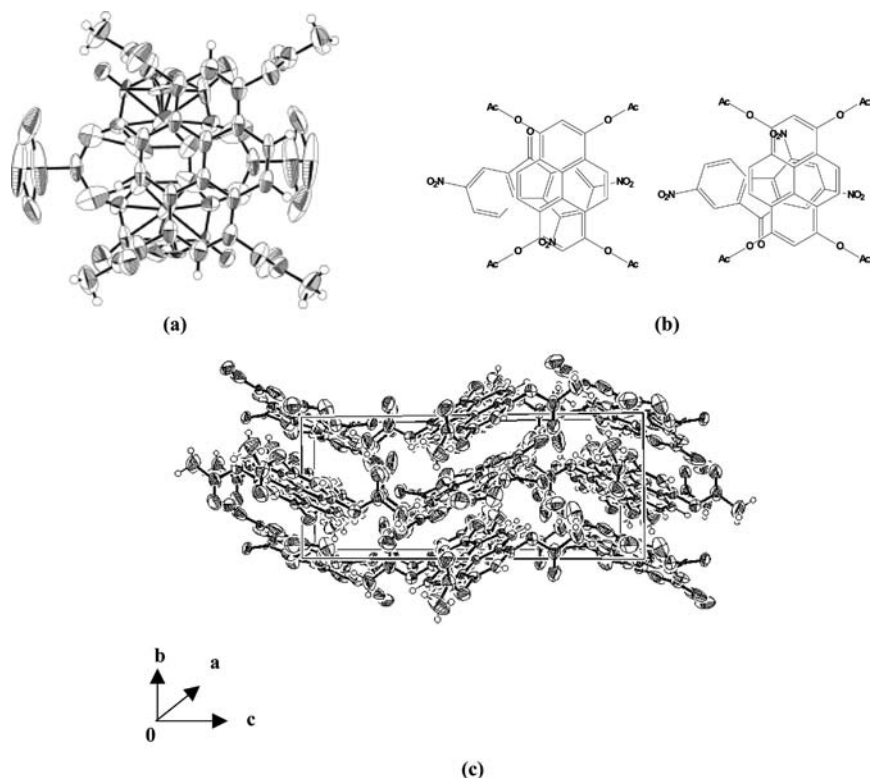


FIGURE 3 Ortep diagram of PC₉-TrNF (a) and view showing the frame format of Ortep (b) and crystal packing diagram (c) of the PC₉-TrNF (1:2) complex.

Based on the result of PC₉, PC_{5.5} with branched side chains was also blended with TeNF. The PC_{5.5}:TeNF mixture in a 1:1 ratio was observed to have a hexagonal columnar phase Col_h by POM (Fig. 5). The phase transition behaviors are summarized in Table 2. In the PC_{m,n}-TeNF composition range studied, that is, between 2:1 and 2:3, all mixtures of PC_{2.4}, PC_{3.3}, and PC_{5.5} formed Col_h phases (Table 2).

Table 2 shows a couple of interesting features of PC-TeNF mixtures: (1) the rectangular PC derivative with linear side chains like PC₉ formed a Col_r phase (Fig. 6a); (2) rounder PC_{3.3} and PC_{5.5} with branched side chains turned to the Col_h phase (Fig. 6b) with higher transition temperatures; and (3) PC_{2.4}, having asymmetrical branches with the same carbon number as PC_{3.3}, showed the lowest Col_h-Cr transition temperature at 40°C. It is noteworthy that the LC-Cr transition temperatures were higher for all PC_{m,n} with smaller

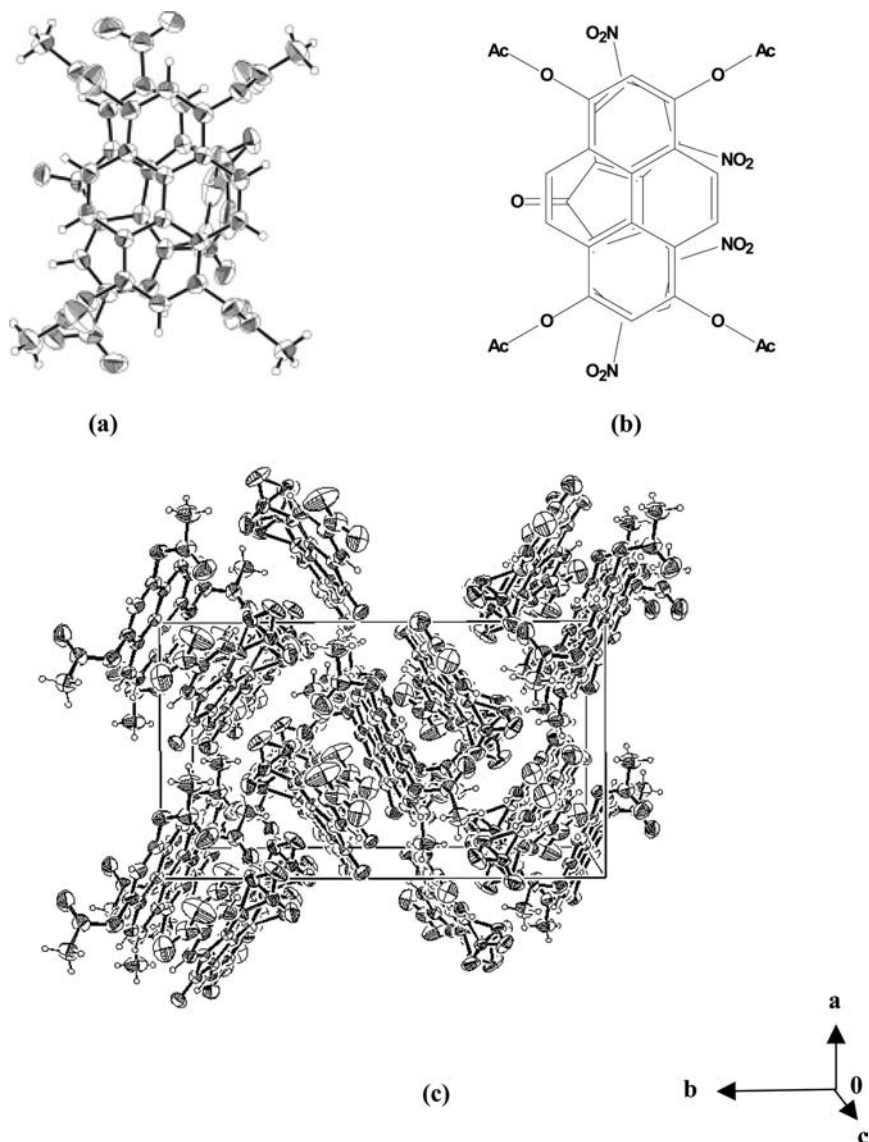


FIGURE 4 Ortep diagram of PC₉-TeNF (a) and view showing the frame format of Ortep (b) and crystal packing diagram (c) of the PC₉-TeNF (1:2) complex.

transition enthalpies than PC_n. It indicates that the side chain structure can control the morphology for the LC phase effectively as presented for porphyrine derivatives by Ohta et al. [18].



FIGURE 5 Snow-like texture of Col_h of PC_{5.5}-TeNF (1:1) at 122°C.

DATA

X-ray crystallographic data of PC₁-TrNF: empirical formula C_{18.50}H_{11.50}N_{1.50}O_{7.50}; formula weight 374.79; black prisms; crystal dimensions 0.20 × 0.20 × 0.30 mm³; monoclinic; space group $P2_1/n$ (#14); $a = 11.855(5)$, $b = 7.770(5)$, $c = 18.721(5)$ Å, $\beta = 102.907(5)^\circ$,

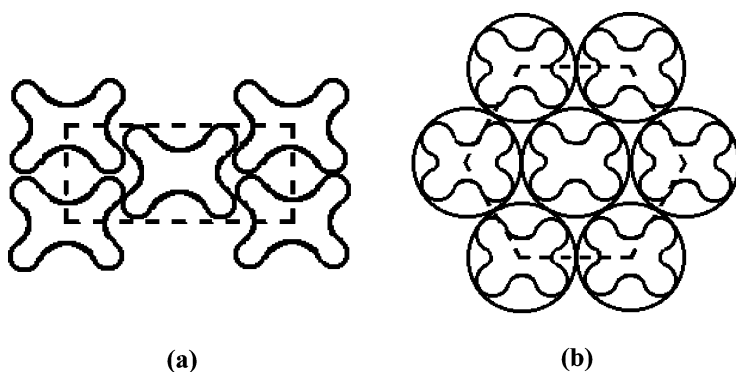


FIGURE 6 Schematic view showing the frame format of rectangular (a) and hexagonal (b) structures of PC derivatives.

$V = 1680.9(14) \text{ \AA}^3$, $Z = 4$; $D_{\text{calcd}} = 1.481 \text{ g/cm}^3$; $F(000) = 772.00$; $\mu(\text{MoK}\alpha) = 0.117 \text{ cm}^{-1}$; $2\theta_{\text{max}} = 55.0^\circ$; no. of reflections measured 3847; no. of observations ($I > 2.00\sigma$ (I), $2\theta < 55.0^\circ$) 3847; no. of variables 338; reflection/parameter ratio was 11.38; $R = 0.0792$; $R_w = 0.1538$; GOF = 0.841.

X-ray crystallographic data of PC₁-TeNF: empirical formula C₂₅H₁₃N₄O₁₃; formula weight 577.39; black prisms; crystal dimensions $0.2 \times 0.3 \times 0.2 \text{ mm}^3$; monoclinic; space group $P2_1/a$ (#14); $a = 10.7720(4)$, $b = 18.7790(9)$, $c = 12.4590(5) \text{ \AA}$, $\beta = 90.958(3)^\circ$, $V = 2519.95(18) \text{ \AA}^3$, $Z = 4$; $D_{\text{calcd}} = 1.522 \text{ g/cm}^3$; $F(000) = 1180.00$; $\mu(\text{MoK}\alpha) = 0.126 \text{ cm}^{-1}$; $2\theta_{\text{max}} = 55.0^\circ$; no. of reflections measured 17314; no. of observations ($I > 2.00\sigma$ (I), $2\theta < 55.0^\circ$) 4922; no. of variables 420; reflection/parameter ratio was 11.72; $R = 0.0690$; $R_w = 0.2167$; GOF = 1.562.

Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publications no. CCDC-244511 (PC₁-TrNF) and 244510 (PC₁-TeNF). Copies of the data can be obtained free of charge upon request to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [Fax: +44(1223)336033; E-mail: deposit@ccdc.cam.ac.uk].

SUMMARY

Some pyrene derivatives with four acyloxy chains were prepared. Although all of them were crystalline, the addition of TeNF allowed for LC phase emergence caused by CT complexation. It was shown that the LC phase type can be controlled by the structure of the side chain: linear side chains formed Col_r whereas branched ones led to Col_h. A more comprehensive study on the LC emergence of PC derivatives is under way, and the relationships with the side chain structure and the LC properties will be reported in the near future.

REFERENCES

- [1] Lehn, J.-M. (1995). *Supramolecular chemistry*, VHC: Weinheim, 139.
- [2] (a) Cornelissen, J. J. L. M., Rowan, A. E., Nolte, R. J. M., & Sommerdijk, N. A. J. M. (2001). *Chem. Rev.*, *101*, 4039.
 (b) Kleppinger, R., Lillya, C. P., & Yang, C. (1997). *J. Am. Chem. Soc.*, *119*, 4097.
 (c) Trzaska, S. T. & Swager, T. (1998). *Chem. Mater.*, *10*, 438.
- [3] (a) Lifshitz, E., Goldfarb, D., Vega, S., Luz, Z., & Zimmermann, H. (1987). *J. Am. Chem. Soc.*, *109*, 7280.
 (b) O'Neill, M. & Kelly, S. M. (2003). *Adv. Mater.*, *15*, 1135.

- [4] Borden, N. & Movaghar, B. (1998). Chapter IX application properties of columnar discotic liquid crystals. In: *Handbook of Liquid Crystals*, Demus, D., Goodby, J., Gray, G. W., Spiess, H.-W., & Vill, V. (Eds.), Wiley-VCH: Weinheim, 781.
- [5] Adam, D., Schuhmacher, P., Simmer, J., Haussling, L., Siemensmeyer, K., Etzbachi, K. H., Ringsdorf, R., & Haarer, D. (1994). *Nature*, 371, 141.
- [6] (a) Warman, J. M. & Schouten, P. G. (1995). *J. Phys. Chem.*, 99, 17181.
(b) Warman, J. M. & van de Craats, A. M. (1999). *Adv. Mater.*, 11, 1469.
- [7] Bock, H. & Helfrich, W. (1995). *Liquid Cryst.*, 18, 707.
- [8] Hassheider, T., Benning, S. A., Kitzerow, H.-S., Achard, M.-F., & Bock, H. (2001). *Angew. Chem. Int. Ed. Engl.*, 40, 2060.
- [9] Liu, C.-Y., Fechtenkötter, A., Watson, M. D., Müllen, K., & Bard, A. J. (2003). *Chem. Mater.*, 15, 124.
- [10] Newman, M. S. & Boden, H. (1962). *Org. Synth.*, 42, 95.
- [11] Mullen, K., Spiess, H. W., Fechtenkötter, A., Saalwächter, K., & Harbison, M. A. (1999). *Angew. Chem. Int. Ed. Engl.*, 38, 3039.
- [12] Vollmann, H., Becker, H., Correll, M., & Streeck, H. (1937). *Liebigs Ann. Chem.*, 531, 1.
- [13] Horning, E. C. & Orchin, M. (1955). *Org. Synth.*, 3, 837.
- [14] (a) Ringsdorf, H., Bengs, H., Karthaus, O., Wüstefeld, R., Ebert, M., Wendroff, J. H., Kohne, B., & Praefcke, K. (1990). *Adv. Mater.*, 2, 141.
(b) Ringsdorf, H., Wüstefeld, R., Zerta, E., Ebert, M., Joachim H., & Wendorff, J. H. (1989). *Angew. Chem. Int. Ed. Engl.*, 7, 914.
- [15] Baehr, C., Evert, M., Frick, G., & Wendorff, J. H. (1990). *Liquid Crystals*, 4, 601.
- [16] Lee, S. & Chang, J. (2003). *Tetrahedron Lett.*, 44, 7493.
- [17] (a) Dewar, M. J. S., Zoebisch, E. G., Healy, E. F., & Stewart, J. J. P. (1985). *J. Am. Chem. Soc.*, 107, 3902.
(b) Stewart, J. J. P. (1998). *WinMOPAC Ver.2*, Fujitsu Ltd: Tokyo, Japan.
- [18] Ohta, K., Yamaguchi, N., & Yamamoto, I. (1998). *J. Mater. Chem.*, 8, 2637.

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