## CYCLOASCAULOSIDE A FROM Astragalus caucasicus LEAVES

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The new cycloartane glycoside cycloascauloside A with the structure 20S,24R-epoxycycloartan-3 $\beta$ ,6 $\alpha$ ,16 $\beta$ ,25-tetraol 3-O-[ $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 6)]- $\beta$ -D-(2'-O-acetyl)-glucopyranoside was isolated from leaves of Astragalus caucasicus Pall. The structure was established based on IR, PMR, and <sup>13</sup>C NMR spectra and physicochemical properties of the compound itself and the products of its chemical transformations.

Key words: Astragalus caucasicus, cycloascauloside, cyclogalegigenin, cycloartane.

In continuation of research on isoprenoids from plants of the genus *Astragalus* [1], we isolated from leaves of *A. caucasicus* Pall. (Leguminosae L.) compounds called cycloascaulosides A, B, and C. Herein the structure of cycloascauloside A is reported.

The PMR spectrum of cycloascauloside A (1) contained two 1H doublets coupled in an AB-type system at 0.39-0.54 ppm with SSCC  ${}^{2}J$  = 4.5 Hz in addition to signals for seven methyls at 0.80-1.78 ppm. This enabled 1 to be considered as a cycloartane triterpenoid [1-3]. This conclusion was confirmed by absorption bands in the IR spectrum at 3045 and 1452 cm<sup>-1</sup> and the preparation of cyclogalegigenin (2) upon acid hydrolysis of 1 and its Smith decomposition [2-4]. The carbohydrate part of the hydrolysate contained D-glucose and L-rhamnose in a 1:1 ratio according to paper (PC) and gas chromatography [5].



The IR spectrum of **1** exhibited absorption bands at 1755 and 1240 cm<sup>-1</sup> (ester); the PMR spectrum, a 3H singlet at 1.98 ppm belonging to an acetyl. As expected, the  ${}^{13}$ C NMR spectrum of **1** contained signals for C atoms of one acetyl at 19.70 and 170.02 ppm.

Glycoside 1 treated with dilute base solutions gave progenin 3, which underwent partial hydrolysis. The resulting products included cyclogalegigenin and glycoside 4, acid hydrolysis of which gave D-glucose and genin 2. Therefore, D-glucose was directly bound to cyclogalegigenin.

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C atom	1	2	C atom	1	2
1	32.70	32.50	24	85.00	85.00
2	29.60	31.50	25	70.10	70.10
3	88.10	78.20	26	27.50*	27.20*
4	42.70	42.40	27	28.10*	28.00*
5	53.80	53.30	28	20.30	20.17
6	68.40	68.20	29	29.18	29.40
7	37.80	38.00	30	16.00	16.20
8	46.30	47.00	COCH <sub>3</sub>	170.02	
9	20.30	20.80	CO <u>C</u> H <sub>3</sub>	19.70	
10	29.60	29.90		-D-Glcp	
11	26.20	26.30	1′	104.60	
12	33.70	33.40	2'	79.00	
13	45.50	45.30	3'	78.50	
14	46.30	46.60	4'	$72.00^{a}$	
15	47.40	47.00	5'	77.80	
16	73.30	73.10	6'	67.60	
17	58.80	58.10		-L-Rhap	
18	21.66	21.63	1″	101.00	
19	30.25	30.90	2″	$72.00^{a}$	
20	86.90	86.70	3″	71.70	
21	28.80	28.40	4‴	74.83	
22	34.70	34.80	5″	68.90	
23	26.17	26.10	6″	17.80	

TABLE 1. <sup>13</sup>C NMR Spectrum of 1 and 2 (0 = TMS,  $\delta$ , ppm, C<sub>5</sub>D<sub>5</sub>N)

Signals marked with the same letters overlap; with asterisks, are assigned arbitrarily. Signals were assigned according to the literature [8].

The production of a less polar compound from 1 by absolute acetone and anhydrous copper sulfate suggested that the OH groups on C-2' and C-3' were free in the rhamnosyl units [6].

Enzymatic hydrolysis of 3 produced genin 2 and a biose, which was identical to rutinose [7].

The <sup>13</sup>C NMR spectra of **1** and **2** (Table 1) indicated that only one carbinol C atom of the genin of **1** experienced a glycosylation effect and resonated at 88.1 ppm (C-3). This means that the carbohydrate substituent was a biose and was bonded to C-3 of the genin.

The similarity of the chemical shifts for C-20 and C-24 in the <sup>13</sup>C NMR spectra of **1** and **2** and the production of a genin with constants identical to cyclogalegigenin [1] indicated that the side chain of the glycoside had the 20*S*- and 24*R*- configuration.

Chemical shifts (CS) of C atoms in the carbohydrate units (Table 1) indicated that the CS of C-2 and C-6 of the D-glucose changed most. This was consistent with C-2 and C-6 of the D-glucose being substituted. The production of rutinose from **3** indicated that rhamnose was  $1\rightarrow$ 6-bonded to D-glucose whereas the acyl substituent was located on C-2 of the D-glucose. This was confirmed by the CS of this C atom (Table 1).

The CS of the L-rhamnose and D-glucose C atoms were consistent with the pyranose form of their rings, the  $\alpha$ -configuration of the terminal monosaccharide, and the  $\beta$ -configuration of the D-glucose [9-11].

Thus, the structure of **1** can be represented as 20S,24R-epoxycycloartan- $3\beta,6\alpha,16\beta,25$ -tetraol 3-*O*-[ $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 6)]- $\beta$ -D-(2'-*O*-acetyl)-glucopyranoside.

## EXPERIMENTAL

**General Comments.** TLC used Silufol plates. Column chromatography was carried out over KSK and L silica gel (Czech Rep.) (40-100 µm). Compounds on TLC were developed by spraying with methanolic (25%) phosphotungstic acid with

subsequent heating at 100-110°C for 2-3 min. The solvent systems were CHCl<sub>3</sub>:CH<sub>3</sub>OH (10:1, 1), CHCl<sub>3</sub>:CH<sub>3</sub>OH:H<sub>2</sub>O (70:23.5:2, 2), and C<sub>5</sub>H<sub>5</sub>N:C<sub>6</sub>H<sub>6</sub>:C<sub>4</sub>H<sub>9</sub>OH:H<sub>2</sub>O (3:5:1:3, 3). PMR and <sup>13</sup>C NMR spectra were recorded on Bruker AM-400 and BS-567 (Tesla) spectrometers in C<sub>5</sub>D<sub>5</sub>N. IR spectra were recorded on a UR-20 instrument in KBr. GC was carried out in a Chrom-5 instrument (Czech Rep.) in a glass column (1.5 m × 4 mm) packed with Chromaton-super impregnated with 5% silicone XE-60, column temperature 210°C, vaporizer 230°C, flame-ionization detector at 250°C, He carrier gas, flow rate 50 mL/min. Rotation angles were determined on a SU-2 instrument; melting points, on a Kofler block.

**Isolation of Isoprenoids.** Ground leaves of *A. caucasicus* (1 kg) were extracted three times with ethanol (80%) with heating. The combined extracts were evaporated to an aqueous residue that was worked up three times with  $CHCl_3$  (0.2 L each). The  $CHCl_3$  was evaporated. The aqueous residue was worked up with hot water, filtered, condensed, and chromatographed over a silica-gel column with elution by  $CHCl_3$  and system 1 to afford total isoprenoids, which were separated over a silica-gel column. Compounds were eluted by system 2. Compounds A, B, and C were isolated. The yield of A was 120 mg.

**Cycloascauloside A** (1),  $C_{44}H_{72}O_{16}$ , mp 210-215°C (CH<sub>3</sub>OH);  $[\alpha]_D^{20}$ -4° (*c* 0.1, EtOH), IR spectrum (KBr,  $v_{max}$ , cm<sup>-1</sup>): 3400-3000 (OH), 3045 (CH<sub>2</sub>-cyclopropane ring).

PMR spectrum (C<sub>5</sub>D<sub>5</sub>N,  $\delta$ , ppm, J/Hz): 0.23, 0.50 (2H-19, d, <sup>2</sup>J = 4.2), 0.89, 1.17, 1.23, 1.24, 1.40, 1.57, 1.78 (7 × CH<sub>3</sub>, s), 1.70 (d, J = 6, L-rhamnose CH<sub>3</sub>), 1.98 (s, Ac CH<sub>3</sub>), 3.55 (dd, <sup>3</sup>J<sub>1</sub> = 12, <sup>3</sup>J<sub>2</sub> = 5, H-3), 3.69 (sq, <sup>3</sup>J = 3.6, 9.6, 9.6, H-6), 3.83 (t, <sup>3</sup>J = 7, H-24), 4.70 (q, <sup>3</sup>J<sub>1</sub> = <sup>3</sup>J<sub>2</sub> = <sup>3</sup>J<sub>3</sub> = 8, H-16) [12], 4.90 (d, J = 7.6, D-glucose anomeric proton), 5.89 (d, J = 0.9, L-rhamnose anomeric proton).

**Cyclogalegigenin (2).** Compound **1** (50 mg) was hydrolyzed by methanolic  $H_2SO_4$  (10 mL, 25%) at 60°C for 2 h. The reaction mixture was diluted with water. The methanol was evaporated. The resulting precipitate was filtered off and dried. The resulting genin was purified over a column of silica gel with elution by system 2 to afford genin (28 mg), mp 196-197°C (CH<sub>3</sub>OH), that was identified also by direct TLC comparison with an authentic sample [1].

The carbohydrate part of the hydrolysate (after evaporation) contained D-glucose and L-rhamnose according to PC (system 3). The remaining hydrolysate was reduced with sodium borohydride and acetylated using  $Ac_2/Py$ . The resulting polyol acetates were identified by comparing GC traces with those of authentic samples and detected rhamnite and dulcite acetates in a 1:1 ratio [5].

Smith decomposition [3] of glycoside (10 mg) also gave cyclogalegigenin [1].

**Progenin 3 from 1.** Glycoside (40 mg) was saponified with methanolic base (5 mL, 0.5%) at room temperature for 3 h. After the usual work up of the reaction products, chromatography of the glycoside part over a silica-gel column with elution by system 2 afforded progenin **3** (32 mg), mp 219-221°C (CH<sub>3</sub>OH),  $[\alpha]_D^{23}$ -121.5 ± 0.5° (*c* 0.5, CH<sub>3</sub>OH).

**Enzymatic Hydrolysis of 3.** Progenin **3** (15 mg) was dissolved in  $CH_3OH$ , treated with an aqueous solution of rhamnodiastase, and held at 37°C for 12 h. After work up of the reaction products, the genin part contained the aglycon cyclogalegigenin (**2**). Rutinose was found in the carbohydrate part using PC and system 3 [7].

**Partial Hydrolysis of 3.** Progenin **3** (60 mg) was partially hydrolyzed in methanolic H<sub>2</sub>SO<sub>4</sub> (10 mL, 0.25%) at 50°C for 1 h. After the usual work up of the reaction products, chromatography over a silica-gel column with elution by system 1 afforded **2** (5 mg), mp 195-196°C (CH<sub>3</sub>OH). Continued elution of the column by system 2 afforded glycoside **4** (18 mg), mp 238-239°C (system 1),  $[\alpha]_D^{23}$  +44 ± 0.20° (*c* 0.6, CH<sub>3</sub>OH:CHCl<sub>3</sub>, 1:1) and unreacted starting glycoside (25 mg), mp 210-221°C (system 1),  $[\alpha]_D^{23}$  -121.5° (*c* 0.26, CH<sub>3</sub>OH:CHCl<sub>3</sub>, 1:1).

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