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Keramamides E, G, H, and J, New Cyclic Peptides Containing an Oxazole or a Thiazole Ring from a *Theonella* Sponge

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Abstract. Four new cyclic peptides, keramamides E (1), G (2), H (3), and J (4), containing an oxazole or a thiazole ring have been isolated from the Okinawan marine sponge *Theonella* sp. and the structures elucidated by 2D NMR data and degradation experiments. The sequence of amino acid residues in 1 ~ 4 was determined on the basis of FAB MS/MS data.

Marine sponges of the genus *Theonella* have been shown to be a rich source of unique secondary metabolites with intriguing structures and interesting biological activities.¹ In our continuing studies on bioactive substances from marine organisms,² we previously isolated some cyclic peptides, keramamides A³, B ~ D⁴, and F⁵, from an Okinawan marine sponge *Theonella* sp. Further investigation on extracts of this sponge led to isolation of four new cyclic peptides, keramamides E (1), G (2), H (3), and J (4), containing an oxazole or a thiazole ring and many unusual amino acids. This paper describes the isolation and structure elucidation of 1 ~ 4 on the basis of spectral data, especially FAB MS/MS data, and chemical means.

The methanol/toluene (3:1) extract of the sponge collected off Kerama Islands, Okinawa, was partitioned between toluene and water. The chloroform extract of the aqueous phase was subjected to silica gel and Sephadex LH-20 columns, and HPLC on ODS to yield keramamides E (1, 4.6 x 10⁻⁵ %, wet weight of the sponge), G (2, 5.3 x 10⁻⁵ %), H (3, 2.8 x 10⁻⁵ %), and J (4, 5.0 x 10⁻⁵ %) together with known cyclic peptides, keramamides B (5)⁴, C (6)⁴, D (7)⁴, and F (8)⁵.

The molecular formula of keramide E (1) was determined as C₅₃H₇₅N₁₀O₁₂Br by HRFABMS [*m/z* 1123.4818, (M+H)⁺, Δ +1.0 mmu].⁶ Amino acid analyses of the acid hydrolysate of 1 showed the presence of 1 mol each of proline (Pro), ornithine (Orn), isoleucine (Ile), alanine (Ala), and norvaline (nVal). Extensive analyses of the ¹H (Table 1) and ¹³C NMR data of 1 including ¹H-¹H COSY, HMQC, and NOESY spectra by comparison with those of keramide B (5) suggested the gross structure of 1 containing a 2-hydroxy-3-methylpentanoic acid (Hmp), a 2-bromo-5-hydroxytryptophan (BhTrp), and segments a and b. The presence of an Hmp group at Ile-NH was revealed by the NOESY correlation between Ile-NH and Hmp-α. The ¹H and ¹³C NMR spectral data of 1 [δ_H 8.22 (1H, s); δ_C 136.9 s, 139.5 d, and 165.2 s] indicated the presence of an oxazole ring in segment b. The sequence of amino acid residues in 1 was determined by FAB MS/MS experiments. The MS/MS product ions observed in the FAB MS/MS spectra from the protonated molecular ions (*m/z* 1123 and 1125) are shown in Table 2 and Fig. 1. Chiral HPLC analysis (SUMICHIRAL OA-5000) of the acid hydrolysate of 1 clarified that Ala, Ile, Orn, Pro, and nVal were L-form. The BhTrp residue was converted into Asp by treatment of 1 with ozone followed by

$\text{CH}_3\text{CO}_3\text{H}$,⁴ while the α -keto- β -amino acid (a) was transformed into Leu by treatment of **1** with $\text{H}_2\text{O}_2/\text{NaOH}$.⁵ Both Asp and Leu in the degradation products were determined to be L by the chiral HPLC analysis. The absolute stereochemistry of Hmp generated by alkaline hydrolysis of **1** was determined to be 2*S*,3*S* by comparison of retention times in chiral HPLC with those of (2*S*,3*S*)-, (2*R*,3*R*)-, (2*S*,3*R*)-, and (2*R*,3*S*)-Hmp derived from L-Ile, D-Ile, L-allo-Ile, and D-allo-Ile through deamination with NaNO_2 , respectively.⁷ Thus the structure of keramide E was concluded to be **1**.

Similarly the absolute configurations of Hmp in keramamides B ~ D (**5** ~ **7**)⁴ which had remained to be determined were concluded to be 2*S*,3*S* by the same method as described above.

The HRFABMS [m/z 921.3886 ($\text{M}+\text{H}$)⁺, Δ -3.7 mmu] of keramide G (**2**) established the molecular formula to be $\text{C}_{43}\text{H}_{56}\text{N}_{10}\text{O}_{11}\text{S}$, which was the same as that of keramide F (**8**). Amino acid analyses of the acid hydrolysate of **2** revealed 1 mol each of Ala, isoserine (Ise), 2,3-diaminopropionic acid (Dpr), and Ile. Extensive analyses of the ^1H (Table 1) and ^{13}C NMR data of **2** including ^1H - ^1H COSY, HMQC, and

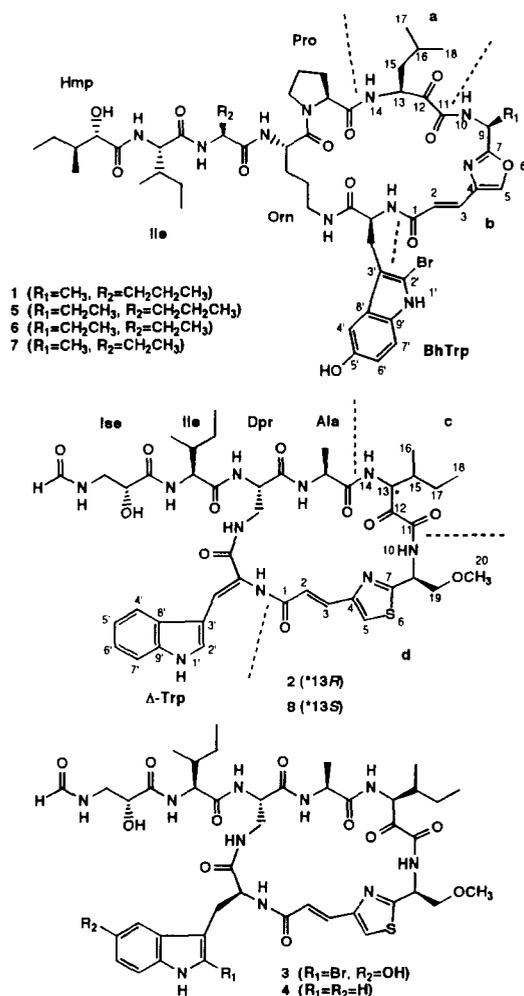


Table 1. ¹H NMR Data of Keramamides E (1), G (2), H (3), and J (4) in DMSO-d₆

position	1	J (Hz)	2	J (Hz)	3	J (Hz)	4	J (Hz)
Hmp	OH	5.50* (brd)	CHO	8.03(s)	8.03 (s)	8.03 (s)		
	α	3.74 (m)						
	β	1.70 (m)	Ise	NH	7.97 (m)	7.97 (m)	7.98 (m)	
	γ-CH ₂	1.34, 1.14 (m)		α	4.04 (m)	4.01 (m)	4.02 (m)	
	γ-CH ₃	0.82 (m)		β	3.14 (m)	3.14 (m)	3.13 (m)	
	δ-CH ₃	0.81 (m)			3.50 (m)	3.48 (m)	3.43 (m)	
				OH		5.96 (d) 5.9	5.98 (d) 5.4	
Ile	NH	7.52 (d) 9.3						
	α	4.26 (m)	Ile	NH	7.67 (d) 8.8	7.67 (d) 9.8	7.68 (d) 8.8	
	β	1.70 (m)		α	4.24 (m)	4.25 (m)	4.25 (dd) 8.8, 6.8	
	γ-CH ₂	1.40, 0.98 (m)		β	1.74 (m)	1.75 (m)	1.74 (m)	
	γ-CH ₃	0.82 (m)		γ-CH ₃	0.85 (m)	0.83 (m)	0.85 (m)	
	δ-CH ₃	0.81 (m)		γ-CH ₂	1.43 (m)	1.40 (m)	1.42 (m)	
					1.05 (m)	1.00 (m)	1.05 (m)	
nVal	NH	8.09 (d) 7.8		δ-CH ₃	0.85 (m)	0.85 (m)	0.84 (m)	
	α	4.23 (m)						
	β	1.43, 1.53 (m)	Dpr	α-NH	8.08 (d) 7.8	8.12 (d) 7.8	8.14 (d) 6.8	
	γ	1.30 (m)		α	4.42 (m)	4.47 (m)	4.46 (m)	
	δ	0.86 (m)		β	2.92 (m)	2.85 (m)	2.83 (m)	
					3.60 (m)	3.57 (m)	3.46 (m)	
Om	α-NH	7.90 (d) 6.8		β-NH	8.04 (m)	7.67 (m)	7.82 (m)	
	α	4.45 (m)						
	β	1.63, 1.36 (m)	Ala	NH	8.02 (m)	8.06 (d) 5.9	8.09 (d) 7.3	
	γ	1.37 (m)		α	4.72 (m)	4.63 (m)	4.67 (m)	
	δ	3.44, 2.64 (m)		β	1.24 (d) 6.8	1.36 (d) 6.8	1.34 (d) 7.3	
	δ-NH	7.49 (m)						
Pro	α	4.34 (m)	d	2	6.91 (d) 15.1	6.69 (d) 15.1	6.61 (d) 14.9	
	β	2.17, 1.85 (m)		3	7.42 (d) 15.1	7.15 (d) 15.1	7.15 (d) 14.9	
	γ	1.86, 1.47 (m)		5	7.90 (s)	7.76 (s)	7.75 (s)	
	δ	3.70, 3.51 (m)		9	4.81 (m)	4.76 (m)	4.75 (dd) 10.0, 6.8	
				10	9.31 (d) 6.8	9.25 (d) 7.3	9.28 (d) 6.8	
b	2	6.58 (d) 15.1	c	13	5.49 (m)	5.40 (dd) 9.5, 3.4	5.46 (dd) 9.8, 3.4	
	3	7.14 (d) 15.1		14	8.53 (d) 9.3	8.45 (d) 9.5	8.61 (d) 9.8	
	5	8.22 (s)		15	2.46 (m)	2.45 (m)	2.46 (m)	
	9	4.84 (m)		16	0.75 (d) 6.8	0.79 (d) 7.3	0.77 (d) 7.3	
	10	9.20 (d) 5.4		17	1.18 (m)	1.22 (m)	1.22 (m)	
	13	4.89 (m)		18	0.84 (m)	0.89 (t) 7.3	0.82 (m)	
a	14	8.18 (d) 5.0		19	3.96 (m)	3.93 (dd) 9.0, 5.6	3.92 (m)	
	15	1.37 (m)		20	4.02 (m)	4.02 (m)	3.99 (m)	
	16	1.46 (m)			3.39 (s)	3.39 (s)	3.38 (s)	
	17	0.83 (m)		α'		4.66 (m)	4.64 (m)	
	18	0.83 (m)		β'	7.86 (s)	2.89 (m)	3.13 (m)	
	19	1.49 (d) 6.8				3.35 (m)	3.35 (m)	
				1'-NH	11.51 (s)	11.20 (s)	10.77 (s)	
BhTrp	NH	8.12 (d) 11.2		2'	7.33 (d) 2.4		6.98 (d) 2.0	
	α	4.58 (m)		4'	7.71 (d) 8.3	6.91 (d) 2.4	7.56 (d) 7.8	
	β	3.06, 2.83 (m)		5'	7.15 (t) 7.3		6.93 (t) 7.8	
	1'	11.23 (s)		6'	7.10 (t) 7.3	6.59 (dd) 8.5, 2.4	7.02 (t) 7.8	
	4'	6.88 (d) 2.0		7'	7.41 (d) 7.8	7.04 (d) 8.5	7.28 (d) 7.8	
	6'	6.58 (dd) 8.8, 2.0		α'-NH	9.07 (s)	8.08 (d) 9.0	8.05 (d) 7.0	
	7'	7.02 (d) 8.8		5'-OH		8.76 (s)		
	OH	8.69 (s)						

a) δ in ppm

NOESY spectra by comparison with those of keramamide F (8) showed the presence of a formyl group, an α,β-dehydrotryptophan (Δ-Trp), and segments c and d. The presence of a thiazole ring in segment d was

Table 2. FAB MS/MS Data of Keramamide E (1)

<i>m/z</i>	<i>m/z</i>	Assignment of MS/MS product ions ^c
1123 ^a	1125 ^b	M + H (parent ion)
1105	1107	M - H ₂ O + H
1067	1069	M - C ₄ H ₉ + 2H
1037	1039	M - C ₅ H ₁₁ O + 2H
1009	1011	Ile-nVal-cyclo(Om-Pro-a-b-BhTrp) + 2H
896	898	nVal-cyclo(Om-Pro-a-b-BhTrp) + 2H
797	799	cyclo(Om-Pro-a-b-BhTrp) + 2H
683	685	Pro-a-b-BhTrp + 2H
655	657	Pro-a-b-BhTrp - CO + 2H
403	403	Pro-a-b + H
327	327	Hmp-Ile-nVal + H
306	306	a-b + H
228	228	Hmp-Ile
200	200	Hmp-Ile - CO
165	165	b + H
122	122	b - C ₂ H ₅ N
115	115	Hmp

a)⁷⁹Br parent ion. b)⁸¹Br parent ion. c)The amide bond cleavages are assumed to occur between NH and CO (the B-type fragmentation).⁸

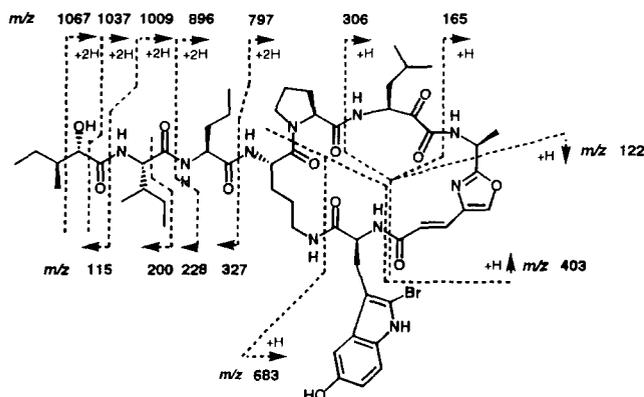


Fig. 1 FAB MS/MS Fragmentations of Keramamide E (1)

indicated by the ¹H and ¹³C NMR spectral data of **2** [δ_{H} 7.90 (1H, s); δ_{C} 149.4 s, 123.1 d, and 165.9 s]. Further substantial evidences for the structure of **2** were obtained from the FAB MS/MS spectrum from the (M + H)⁺ ion (*m/z* 921), which afforded MS/MS product ions corroborating well the amino acid sequence as shown in Fig. 2. Thus the gross structure of keramamide G (**2**) was elucidated to be the same as that of keramamide F (**8**). The differences between **2** and **8** were found only for carbon chemical shifts at C-13 (**2**, δ_{C} 56.7; **8**, δ_{C} 59.7), C-16 (**2**, δ_{C} 14.3; **8**, δ_{C} 16.0), and C-17 (**2**, δ_{C} 26.7; **8**, δ_{C} 23.4), suggesting that the stereochemistry at C-13 of the α -keto- β -amino acid (**c**) in **2** was different from that of **8**. Oxidation of **2** with H₂O₂/NaOH⁵ followed by acid hydrolysis furnished Ile of which the absolute configuration was determined to be D by chiral HPLC analysis. Chiral HPLC analysis of the acid hydrolysate of **2** clarified that Ala, Ile, and Dpr were L-form, while Ise was D-form. Treatment of **1** with ozone for degradation of the (*O*-methylserine)thiazole (**d**) yielded *O*-methylserine, which was L-form from chiral HPLC analysis. Thus the structure of keramamide G was assigned to be **2**.

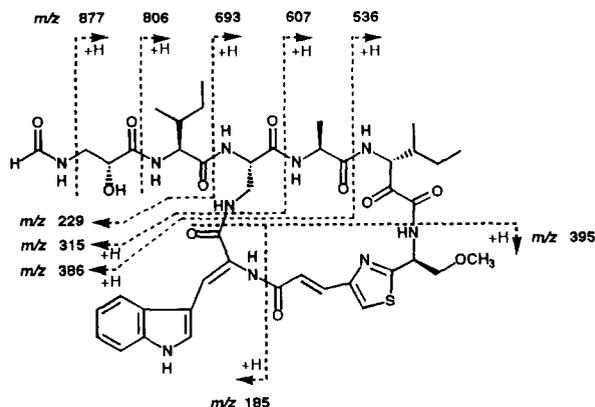


Fig. 2 FAB MS/MS Fragmentations of Keramide G (2)

The HRFABMS [m/z 1018.3190, M^+ , Δ -2.9 mmu] of keramide H (3) established its molecular formula to be $C_{43}H_{57}N_{10}O_{12}BrS$. Amino acid analyses of the acid hydrolysate of 3 revealed 1 mol each of Ala, Ise, Dpr, and Ile. The 1H NMR (Table 1) spectrum of 3 was almost the same as that of 2 except for its tryptophan moiety. Comparison of the 1H - 1H COSY spectrum of 3 with those of 1 and 2 revealed the presence of BhTrp [δ_H 11.20 (s), 6.91 (d, $J = 2.4$ Hz), 6.59 (dd, $J = 8.5$ and 2.4 Hz), 7.04 (d, $J = 8.5$ Hz), 8.08 (d, $J = 9.0$ Hz), and 8.76 (s)] for 3 in place of Δ -Trp for 2. Furthermore, the FAB MS/MS spectra from the protonated molecular ions (m/z 1017 and 1019) of 3 supported the gross structure proposed for keramide H. Ala, Ile, and Dpr in the acid hydrolysate of 3 were L-form by chiral HPLC analysis, while Ise was D-form. The BhTrp residue in 3 was converted into Asp by treatment of 3 with ozone followed by CH_3CO_3H , while the α -keto- β -amino acid (c) was transformed into Ile by treatment of 3 with $H_2O_2/NaOH$. Degradation of the (*O*-methylserine)thiazole (d) with ozone yielded *O*-methylserine. Asp, Ile, and *O*-methylserine in the degradation products were determined to be L by chiral HPLC analysis. Thus the structure of keramide H was assigned to be 3.

The molecular formula of keramide J (4) was determined as $C_{43}H_{58}N_{10}O_{11}S$ by HRFABMS [m/z 923.4099 ($M + H$) $^+$, Δ +1.3 mmu] and the molecular weight was larger than that of keramide G (2) by 2 Daltons. The 1H NMR data (Table 1) of 4 were similar to those of 2 except for resonances of tryptophan moiety. Aromatic proton signals [δ_H 6.98 (d, $J = 2.0$ Hz), 7.56 (d, $J = 7.8$ Hz), 6.93 (t, $J = 7.8$ Hz), 7.02 (t, $J = 7.8$ Hz), 7.28 (d, $J = 7.8$ Hz), and 8.05 (d, $J = 7.0$ Hz)] for 4 were assigned as those of tryptophan by the 1H - 1H COSY data of 4 and comparison with the chemical shifts reported for a tryptophan residue in polydiscamide A.⁹ Hydrolysis of 4 under mild acidic condition followed by amino acid analysis of the products revealed 1 mol each of Ala, Ise, Dpr, Ile, and Trp. The amino acid sequence of 4 was deduced from the FAB MS/MS data. Chiral HPLC analysis of the acid hydrolysate of 4 clarified that Ala, Ile, Dpr, and Trp were L-form, while Ise was D-form. The α -keto- β -amino acid (c) was transformed into Ile by treatment of 4 with $H_2O_2/NaOH$, while the (*O*-methylserine)thiazole (d) was converted into *O*-methylserine by the same method as described above. Both Ile and *O*-methylserine were determined to be L by chiral HPLC analysis. Thus the structure of keramide J was concluded to be 4.

Keramamides E (1), G (2), H (3), and J (4) are new cyclic peptides containing an oxazole or a thiazole ring in addition to various unusual amino acids. Although there are many reports on cyclic peptides having oxazole and/or thiazole ring(s) from tunicates¹⁰ and terrestrial microorganisms,¹¹ very few peptides containing conjugated oxazole or thiazole ring(s) have been isolated from natural origin. Such α -keto- β -amino acids contained in 1–4 have been reported for poststatin, a peptide from *Streptomyces viridochromogenes*¹², while 2-hydroxy-3-methylpentanoic acid moiety has been found in peptides from a black yeast *Aureobasidium pullulans*.¹³ Keramamide E (1) exhibited cytotoxicity against L1210 murine leukemia cells and KB human epidermoid carcinoma cells with IC₅₀ values of 1.60 and 1.55 μ g/mL, respectively, while keramamides G (2), H (3), and J (4) showed weak cytotoxicity (IC₅₀ ~10 μ g/mL).

Experimental Section

General Methods. Optical rotations were measured on a JASCO DIP-370 polarimeter. UV and IR spectra were obtained on JASCO Ubest-35 and JASCO IR report-100 spectrometers, respectively. ¹H and ¹³C NMR spectra were recorded on a JEOL EX-400 spectrometer in DMSO-*d*₆. The 2.50 ppm resonance of residual CD₃SOCHD₂ and 39.5 ppm of (CD₃)₂SO were used for ¹H and ¹³C NMR spectra as internal references, respectively. Mass spectra were obtained on a JEOL JMS-HX/HX 110A tandem mass spectrometer by using glycerol as a matrix.

Collection, Extraction, and Isolation. The sponge *Theonella* sp. was collected off Kerama Island, Okinawa and was kept frozen until used. The toluene/MeOH (1:3, 2 L x 2) extract of the sponge (4 kg, wet weight) was suspended with 1M NaCl (1 L) and was extracted with toluene (600 mL x 2). The aqueous layer was extracted with CHCl₃ (800 mL x 2). The CHCl₃-soluble material (2.1 g) was subjected to a silica gel column with gradient elution of MeOH (2–50%) in CHCl₃. The fraction eluted with 15% MeOH in CHCl₃ was separated by a silica gel column with toluene/MeOH (98:2 to 70:30) to afford crude peptide fractions I and II. The fraction I was further purified by a Sephadex LH-20 column (MeOH) followed by reversed-phase HPLC [YMC-Pack AM-324 ODS, Yamamura Chemical, 10 x 250 mm; flow rate: 2.0 mL/min; eluent: MeOH/H₂O/CF₃CO₂H, 70:30:0.1; UV detection at 254 nm] to give keramamides B (5, 7.4 mg, *t*_R 48.0 min), C (6, 11.8 mg, *t*_R 40.1 min), and D (7, 10.1 mg, *t*_R 31.5 min) and a fraction (12.3 mg, *t*_R 37.1 min), which was purified by the same reversed-phase HPLC column with CH₃CN/H₂O/CF₃CO₂H (42:58:0.1) to afford keramamide E (1, 1.8 mg, *t*_R 63 min). The fraction II was separated by a Sephadex LH-20 column (MeOH) followed by the same HPLC column with CH₃CN/H₂O (40:60) to give keramamide F (8, 5.1 mg, *t*_R 22.0 min) and the fractions III (*t*_R 23–26 min) and IV (*t*_R 26–37 min). Fraction III was subjected to the same HPLC column with MeOH/H₂O/CF₃CO₂H (70:30:0.1) to afford keramamides G (2, 2.1 mg, *t*_R 16 min) and H (3, 1.1 mg, *t*_R 18 min). Fraction IV was purified under the same HPLC condition to yield keramamide J (4, 2.0 mg, *t*_R 36 min).

Keramamide E (1): colorless solid; [α]²²_D -39° (c 0.1, MeOH); IR (KBr) ν_{\max} 3400, 1670, 1640, and 1520 cm⁻¹; UV(MeOH) λ_{\max} 269 (ϵ 30900) and 310 (sh) nm; ¹H NMR (Table 1); ¹³C NMR (DMSO-*d*₆) δ_C 172.6 (Hmp-CO), 74.9 (Hmp- α), 38.1 (Hmp- β), 23.2 (Hmp- γ -CH₂), 15.3 (Hmp- γ -CH₃), 11.7 (Hmp- δ -CH₃), 170.6 (Ile-CO), 55.8 (Ile- α), 37.3 (Ile- β), 24.2 (Ile- γ -CH₂), 15.4 (Ile- γ -CH₃), 10.9 (Ile- δ -CH₃), 171.2 (nVal-CO), 52.0 (nVal- α), 34.0 (nVal- β), 18.5 (nVal- γ), 13.6 (nVal- δ), 169.4 (Orn-CO), 49.7 (Orn- α), 37.8 (Orn- β), 25.1 (Orn- γ), 38.5 (Orn- δ), 171.2 (Pro-CO), 58.4 (Pro- α), 29.5 (Pro- β), 24.4 (Pro- γ), 46.9 (Pro- δ), 163.9 (C-1), 123.5 (C-2), 127.4 (C-3), 136.9 (C-4), 139.5 (C-5), 165.2 (C-7), 43.9 (C-9), 159.6 (C-11), 195.8 (C-12), 52.4 (C-13), 29.1 (C-15), 24.8 (C-16), 20.8 (C-17), 23.1 (C-18), 18.0 (C-19), 170.7 (BhTrp-CO), 53.4 (BhTrp- α), 27.9 (BhTrp- β), 109.5 (BhTrp-2'), 109.0 (BhTrp-3'), 102.3 (BhTrp-4'), 150.7 (BhTrp-5'), 111.5 (BhTrp-6'), 110.9 (BhTrp-7'), 128.2 (BhTrp-8'), and 130.5 (BhTrp-9'); FABMS (positive) *m/z* 1123 and 1125 (1:1) (M+H)⁺; exact mass found *m/z* 1123.4818, calcd for C₅₃H₇₆N₁₀O₁₂Br 1123.4808; FAB MS/MS (Table 2).

Keramamide G (2): colorless solid; [α]²¹_D +10° (c 0.12, MeOH); IR (KBr) ν_{\max} 3400, 1660, and 1520 cm⁻¹; UV(MeOH) λ_{\max} 224 (ϵ 26500), 279 (21200), and 340 (11700) nm; ¹H NMR (Table 1); ¹³C NMR (DMSO-*d*₆) δ_C 161.5 (CHO), 171.4 (Ise-CO), 70.2 (Ise- α), 41.6 (Ise- β), 170.1 (Ile-CO), 56.2 (Ile- α), 37.1 (Ile- β), 15.3 (Ile- γ -CH₃), 24.3 (Ile- γ -CH₂), 11.0 (Ile- δ -CH₃), 169.3 (Dpr-CO), 51.1 (Dpr- α), 41.6 (Dpr- β), 173.8 (Ala-CO), 47.6 (Ala- α), 20.0 (Ala- β), 164.7 (Δ -Trp-CO), 121.9 (Δ -Trp- α), 124.0 (Δ -Trp-

β), 127.1 (Δ -Trp-2'), 109.6 (Δ -Trp-3'), 118.0 (Δ -Trp-4'), 121.9 (Δ -Trp-5'), 120.0 (Δ -Trp-6'), 111.9 (Δ -Trp-7'), 135.4 (Δ -Trp-8'), 127.3 (Δ -Trp-9'), 163.3 (C-1), 124.0 (C-2), 132.1 (C-3), 149.4 (C-4), 123.1 (C-5), 165.9 (C-7), 53.7 (C-9), 161.0 (C-11), 197.2 (C-12), 56.7 (C-13), 37.0 (C-15), 14.3 (C-16), 26.7 (C-17), 11.6 (C-18), 72.6 (C-19), and 58.3 (C-20); FABMS (positive) m/z 921 (M+H)⁺; exact mass found m/z 921.3886, calcd for C₄₃H₅₇N₁₀O₁₁S 921.3923; FAB MS/MS (positive) m/z 921 [M + H]⁺, 903 [M - H₂O + H]⁺, 877 [M - CH₂NO + H]⁺, 833 [M - C₃H₆NO₂ + H]⁺, 806 [Ile-cyclo(Dpr-Ala-c-d- Δ Trp) + 2H]⁺, 790 [Ile-cyclo(Dpr-Ala-c-d- Δ Trp) - NH + H]⁺, 693 [cyclo(Dpr-Ala-c-d- Δ Trp) + 2H]⁺, 677 [cyclo(Dpr-Ala-c-d- Δ Trp) - NH + H]⁺, 607 [Ala-c-d- Δ Trp + H]⁺, 536 [c-d- Δ Trp + H]⁺, 395 [d- Δ Trp + H]⁺, 386 [Ise(CHO)-Ile-Dpr-Ala + H]⁺, 315 [Ise(CHO)-Ile-Dpr + H]⁺, 229 [Ise(CHO)-Ile]⁺, 185 [Δ Trp + H]⁺, and 117 [Ise(CHO) + H]⁺.

Keramamide H (3): colorless solid; $[\alpha]^{20}_D$ -42° (c 0.055, MeOH); IR (KBr) ν_{max} 3400, 1660, and 1520 cm⁻¹; UV (MeOH) λ_{max} 223 (sh) and 277 (ϵ 23100) nm; ¹H NMR (Table 1); FABMS (positive) m/z 1017 and 1019 (1:1, M+H)⁺; exact mass found m/z 1018.3190 (M⁺), calcd for C₄₃H₅₇N₁₀O₁₂⁸¹BrS 1018.3219; FAB MS/MS (positive) m/z (for ⁷⁹Br parent ion) and m/z^* (for ⁸¹Br parent ion) 1017 and 1019* [M + H]⁺, 999 and 1001* [M - H₂O + H]⁺, 973 and 975* [M - CH₂NO + H]⁺, 929 and 931* [M - C₃H₆NO₂ + H]⁺, 902 and 904* [Ile-cyclo(Dpr-Ala-c-d-BhTrp) + 2H]⁺, 886 and 888* [Ile-cyclo(Dpr-Ala-c-d-BhTrp) - NH + H]⁺, 789 and 791* [cyclo(Dpr-Ala-c-d-BhTrp) + 2H]⁺, 774 and 776* [cyclo(Dpr-Ala-c-d-BhTrp) - NH + H]⁺, 703 and 705* [Ala-c-d-BhTrp + H]⁺, 632 and 634* [c-d-BhTrp + H]⁺, 491 and 493* [d-BhTrp + H]⁺, 386 and 386* [Ise(CHO)-Ile-Dpr-Ala + H]⁺, 315 and 315* [Ise(CHO)-Ile-Dpr + H]⁺, 282 and 284* [BhTrp + 2H]⁺, 229 and 229* [Ise(CHO)-Ile]⁺, and 117 and 117* [Ise(CHO) + H]⁺.

Keramamide J (4): colorless solid; $[\alpha]^{18}_D$ +8.4° (c 0.1, MeOH); IR (KBr) ν_{max} 3410, 1680, and 1540 cm⁻¹; UV (MeOH) λ_{max} 222 (ϵ 45200) and 278 (27200) nm; ¹H NMR (Table 1); ¹³C NMR (DMSO-*d*₆) δ_C 161.4 (CHO), 171.4 (Ise-CO), 70.0 (Ise- α), 41.3 (Ise- β), 169.2 (Ile-CO), 58.3 (Ile- α), 37.1 (Ile- β), 15.3 (Ile- γ -CH₃), 24.2 (Ile- γ -CH₂), 11.0 (Ile- δ -CH₃), 169.2 (Dpr-CO), 51.3 (Dpr- α), 41.9 (Dpr- β), 174.1 (Ala-CO), 48.1 (Ala- α), 19.7 (Ala- β), 171.6 (Trp-CO), 58.3 (Trp- α), 28.3 (Trp- β), 127.1 (Trp-2'), 111.3 (Trp-3'), 122.5 (Trp-4'), 118.1 (Trp-5'), 120.6 (Trp-6'), 111.5 (Trp-7'), 137.5 (Trp-8'), 127.3 (Trp-9'), 164.4 (C-1), 124.4 (C-2), 131.2 (C-3), 149.2 (C-4), 122.6 (C-5), 165.5 (C-7), 53.7 (C-9), 161.4 (C-11), 197.0 (C-12), 56.1 (C-13), 37.0 (C-15), 14.2 (C-16), 26.7 (C-17), 11.5 (C-18), 72.5 (C-19), and 58.3 (C-20); FABMS (positive) m/z 923 (M+H)⁺; exact mass found m/z 923.4099 calcd for C₄₃H₅₉N₁₀O₁₁S 923.4086; FAB MS/MS (positive) m/z 923 [M + H]⁺, 905 [M - H₂O + H]⁺, 879 [M - CH₂NO + H]⁺, 835 [M - C₃H₆NO₂ + H]⁺, 808 [Ile-cyclo(Dpr-Ala-c-d- Δ Trp) + 2H]⁺, 792 [Ile-cyclo(Dpr-Ala-c-d- Δ Trp) - NH + H]⁺, 695 [cyclo(Dpr-Ala-c-d- Δ Trp) + 2H]⁺, 679 [cyclo(Dpr-Ala-c-d- Δ Trp) - NH + H]⁺, 609 [Ala-c-d- Δ Trp + H]⁺, 538 [c-d- Δ Trp + H]⁺, 397 [d- Δ Trp + H]⁺, 386 [Ise(CHO)-Ile-Dpr-Ala + H]⁺, 315 [Ise(CHO)-Ile-Dpr + H]⁺, 229 [Ise(CHO)-Ile]⁺, 188 [Trp + 2H]⁺, and 117 [Ise(CHO) + H]⁺.

Amino Acid Analysis by Chiral HPLC. Keramamides E, G, or H (1-3, each 0.1 mg) was hydrolyzed with 6N HCl (1.0 mL) at 110°C for 24 h and keramamide J (4, 0.1 mg) was hydrolyzed with 4N methanesulfonic acid (0.1 mL) at 115°C for 24h. Chiral HPLC analyses were carried out using a SUMICHIRAL OA-5000 [Sumitomo Chemical Industry, 4 x 150 mm, 40 °C, UV detection at 254 nm]. Retention times (min) of standard amino acids were as follows: L-Orn (13.1) and D-Orn (14.2) [eluent: H₂O containing 1.0 mmol/L of CuSO₄, flow rate: 0.2 mL/min]; L-Ala (5.7), D-Ala (8.1), L-(*O*-Me)Ser (9.7), D-(*O*-Me)Ser (14.0), L-Pro (12.7), D-Pro (27.2), L-Dpr (19.7), D-Dpr (22.9), L-nVal (23.8), and D-nVal (44.1) [eluent: H₂O containing 1.0 mmol/L of CuSO₄, flow rate: 1.0 mL/min]; L-Ile (13.3), allo-L-Ile (11.3), D-Ile (21.3), and allo-D-Ile (17.4) [eluent: MeOH/H₂O (15:85) containing 2.0 mmol/L of CuSO₄, flow rate: 1.0 mL/min]; L-Ise (25.7), D-Ise (41.4), L-Trp (34.2), and D-Trp (37.0) [eluent: MeOH/H₂O (30:70) containing 2.0 mmol/L of CuSO₄, flow rate: 1.0 mL/min]. The retention times (min) of hydrolysates of 1 ~ 4 were as follows; L-Orn (13.1), L-Ala (5.7), L-Pro (12.7), L-nVal (23.8), and L-Ile (13.3) in 1; L-Ala (5.7), L-Dpr (19.7), L-Ile (13.3), and D-Ise (41.4) in 2 and 3; L-Dpr (19.7), L-Ile (13.3), L-Trp (34.2), and D-Ise (41.4) in 4.

Preparation of 2-Hydroxy-3-methylpentanoic Acid. A solution of NaNO₂ (50 mg) in water (0.5 mL) was added to a stirred and ice-salt-cooled solution of L-Ile (50 mg) in 1N H₂SO₄ (1 mL). The mixture was stirred for 12 h and then extracted with THF (10 mL). The THF solution was washed with brine, dried (MgSO₄) and concentrated under reduced pressure to give (2*S*,3*S*)-Hmp (30 mg, 60%): ¹H NMR (CDCl₃) δ 0.94 (3H, t, J = 7.5 Hz), 1.03 (3H, d, J = 6.9 Hz), 1.29 (2H, m), 1.90 (1H, m), and 4.18 (1H,

d, $J = 3.3$ Hz). According to essentially the same procedure as described above, DL-Ile, allo-L-Ile, and allo-D-Ile afforded (2*S*,3*S*)- and (2*R*,3*R*)-Hmp, (2*S*,3*R*)-Hmp, and (2*R*,3*S*)-Hmp, respectively. (2*S*,3*R*)-Hmp: ^1H NMR (CDCl_3) δ 0.89 (3H, d, $J = 7.1$ Hz), 0.97 (3H, t, $J = 7.5$ Hz), 1.38 (2H, m), 1.90 (1H, m), and 4.29 (1H, d, $J = 2.7$ Hz).

Determination of the Stereochemistry of 2-Hydroxy-3-methylpentanoic Acid in Keramamides E (1) and B ~ D (5 ~ 7). Keramide E (1, 0.2 mg) was hydrolyzed with 1N NaOH/MeOH (1:4, 0.5 mL) at 50 °C for 8 h. The reaction mixture acidified by 1N HCl (0.3 mL) was then extracted with THF (5 mL x 3) and the residue was subjected to chiral HPLC analysis using a SUMICHIRAL OA-5000 [4 x 150 mm, 30 °C, flow rate: 1.0 mL/min; eluent: $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (20:80) containing 2.0 mmol/L of CuSO_4 ; detection: UV at 254 nm]. Retention times (min) of synthetic 2-hydroxy-3-methylpentanoic acid (Hmp) isomers were as follows; 2*S*,3*R*-Hmp (26.0), 2*S*,3*S*-Hmp (28.9), 2*R*,3*S*-Hmp (38.4), and 2*R*,3*R*-Hmp (43.4). The retention time of Hmp in the hydrolysate of 1 was found to be 28.9 min. Keramamides B, C, or D (5 ~ 7, each 0.2 mg) was hydrolyzed with 1N NaOH/MeOH (1:4, 0.5 mL) at 50 °C for 8 h. The reaction mixture acidified by 1N HCl (0.3 mL) was then extracted with THF (5 mL x 3) and the residue was subjected to chiral HPLC analysis as described above. The retention times of Hmp in the hydrolysates of 5, 6, and 7 were found to be all 28.9 min.

Determination of the Stereochemistry of the (*O*-Methylserine)thiazole in Keramamides G (2), H (3), and J (4). A stream of O_3 was bubbled into a 1 mL MeOH solution of keramamides G, H, or J (2~4, each 0.2 mg) at room temperature for 8 min. The reaction mixture was subjected to hydrolysis and the chiral HPLC analysis using a SUMICHIRAL OA-5000 [4 x 150 mm, 40 °C, flow rate: 1.0 mL/min; eluent: H_2O containing 1.0 mmol/L of CuSO_4]. Retention times of authentic L-(*O*-Me)Ser and D-(*O*-Me)Ser were 9.7 and 14.0 min, respectively. The retention times of (*O*-Me)Ser in the oxidation product of 2, 3, and 4 were found to be all 9.7 min.

Determination of the Stereochemistry of the α -Keto- β -amino Acid in Keramamides E (1), G (2), H (3), and J (4). To a stirred solution of keramide E (1, 0.2 mg) in 5% NaOH (0.5 mL) was added dropwise 30% H_2O_2 (0.1 mL). After stirring at 65 °C for 40 min the reaction mixture was hydrolyzed with 6N HCl (1.0 mL) at 110 °C for 24h and subjected to chiral HPLC analysis using SUMICHIRAL OA-5000 [40°C; flow rate: 1.0 mL/min; detection: UV at 254 nm]. Retention times of authentic L-Leu, D-Leu, L-Ile, D-Ile, allo-L-Ile, and allo-D-Ile [eluent: $\text{MeOH}/\text{H}_2\text{O}$ (15:85) containing 2.0 mmol/L of CuSO_4] were 14.4, 21.9, 13.3, 21.3, 11.3, and 17.4 min, respectively. The retention time of Leu in the degradation product of 1 was found to be 14.4 min. Keramamides G, H, or J (2 ~ 4, each 0.2 mg) was treated with 30% H_2O_2 (0.1 mL) in 5% NaOH (0.5 mL) at 65 °C for 40 min. The reaction mixture was hydrolyzed with 6N HCl (1.0 mL) at 110 °C for 24 h. The hydrolysate was applied to the chiral HPLC analysis as described above. The retention times of Ile in the degradation products of 2, 3, and 4 were found to be 21.3, 13.3, and 13.3 min, respectively.

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